# OU\_158131

UNIVERSAL LIBRARY

OSMANIA UNIVERS	IFY LIBRA	rj <sup>e</sup>
547.2/E 31 P Egloff. G.	Accession No	16847
Egloff. G.	<i>(</i> )	
ysical Constants should be returned on or bef	fore the date last	marked below.

# PHYSICAL CONSTANTS OF HYDROCARBONS

# Volume IV POLYNUCLEAR AROMATIC HYDROCARBONS

### **GUSTAV EGLOFF**

DIRECTOR OF RESEARCH UNIVERSAL OIL PRODUCTS COMPANY RESEARCH LABORATORIES CHICAGO, ILLINOIS



REINHOLD PUBLISHING CORPORATION 330 WEST FORTY-SECOND STREET, NEW YORK, U. S. A.

# COPYRIGHT 1947, BY REINHOLD PUBLISHING CORPORATION

All rights reserved

# Dedicated to JOSEPH G. ALTHER

### GENERAL INTRODUCTION

### American Chemical Society Series of Chemical Monographs

By arrangement with the Interallied Conference of Pure and Applied Chemistry, which met in London and Brussels in July, 1919, the American Chemical Society was to undertake the production and publication of Scientific and Technologic monographs on chemical subjects. same time it was agreed that the National Research Council, in cooperation with the American Chemical Society and the American Physical Society, should undertake the production and publication of Critical Tables of Chemical and Physical Constants. The American Chemical Society and the National Research Council mutually agreed to care for these two fields of chemical development. The American Chemical Society named as Trustees, to make the necessary arrangements for the publication of the monographs, Charles L. Parsons, secretary of the Society, Washington, D. C.; the late John E. Teeple, then treasurer of the Society, New York; and the late Professor Gellert Alleman of Swarthmore College. tees arranged for the publication of the A.C.S. series of (a) Scientific and (b) Technologic Monographs by the Chemical Catalog Company, Inc. (Reinhold Publishing Corporation, successors) of New York.

The Council of the American Chemical Society, acting through its Committee on National Policy, appointed editors (the present list of whom appears at the close of this introduction) to select authors of competent authority in their respective fields and to consider critically the manuscripts submitted.

The first monograph of the series appeared in 1921. After twenty-three years of experience certain modifications of general policy are indicated. In the beginning there still remained from the preceding five decades a distinct though arbitrary differentiation between so-called "pure science" publications and technologic or applied science literature. This differentiation is fast becoming nebulous. Research in private enterprise has grown apace and not a little of it is pursued on the frontiers of knowledge. Furthermore, most workers in the sciences are coming to see the artificiality of the separation. The methods of both groups of workers are the same. They employ the same instrumentalities, and now frankly recognize that their objectives are common, namely the search for new knowledge for the service of man. The officers of the Society therefore have combined the two editorial Boards in a single Board of twelve representative members.

Also in the beginning of the series, it seemed expedient to construe rather broadly the definition of a monograph. Needs of workers had to be recognized. Consequently among the first one hundred monographs appeared works in the form of treatises covering in some instances rather broad areas. Because such necessary works do not now want for publishers, it is considered advisable to hew more strictly to the line of the monograph character which means more complete and critical treatment of relatively restricted areas, and where a broader field needs coverage, to subdivide it into logical sub-areas. The prodigious expansion of new knowledge makes such a change desirable.

These monographs are intended to serve two principal purposes: first, to make available to chemists a thorough treatment of a selected area in form usable by persons working in more or less unrelated fields to the end that they may correlate their own work with a larger area of physical science discipline; second, to stimulate further research in the specific field treated. To implement this purpose the authors of monographs are expected to give extended references to the literature. Where the literature is of such volume that a complete bibliography is impracticable, the authors are expected to append a list of references critically selected on the basis of their relative importance and significance.

### AMERICAN CHEMICAL SOCIETY

### BOARD OF EDITORS

F. W. WILLARD, Editor of Monographs

T. H. CHILTON LINUS C. PAULING

W. M. CLARK W. T. READ L. F. FIESER WALTER A. S

L. F. Fieser Walter A. Schmidt J. Bennett Hill C. A. Thomas

S. C. LIND E. R. WEIDLEIN

C. H. MATHEWSON W. G. WHITMAN

### Preface

The principal source of polynuclear aromatic hydrocarbons is coal-tar. Some of these compounds are also found in petroleum and other natural sources, and many are produced synthetically. Commercial utilization of these hydrocarbons has been confined essentially to the dyestuff industry. Among the polynuclears are compounds of physiological interest such as the carcinogenic or cancer producing hydrocarbons.

Volume IV is a collation of all melting point, boiling point, density, and refractive index data available on polynuclear aromatic hydrocarbons. This survey shows that the opportunities for research on polynuclear aromatics are far from being exhausted. The thousands of known compounds represent only a small fraction of the theoretical possibilities. The physical data on the known compounds are so scanty that there is a great need for investigation of the physical constants of polynuclears.

We are indeed grateful to the staff of John Crerar Library for valuable guidance in the location of material and for privileges granted in connection with use of the outstanding facilities of this institution.

Deep appreciation is hereby expressed for the assistance of the Universal Oil Products Company and the author's colleagues, Mary Alexander, Nancy Corbin, M. S. Beyt, and Madge Spiegler, in this collation and critical study of the physical constants of aromatic hydrocarbons.

GUSTAV EGLOFF

Chicago, Ill., March 1, 1946

# **Contents**

	Volume IV	age
	General Introduction	V
	Preface	vii
	Introduction	1
	Foreword	1
	Nomenclature Order of Tabulating Compounds	13
Τ.	Benzocyclanes $C_nH_{2n-8}$	19
	1 Indane and Its Alkyl Derivatives	20
	<ol> <li>Tetrahydronaphthalene and Its Alkyl Derivatives</li> <li>Higher Benzocyclanes</li> </ol>	28 44
Π.	Polynuclear Aromatics of Empirical Formula $C_nH_{2n-10}$	45
	1 Indone and Its Alkyl Derivatives	46
	2 Indane with One Alkenyl or One Alkylidene Substitution	51 52
	3. Dihydronaphthalene and Its Alkyl Derivatives 4. Tetrahydronaphthalene Derivatives of Empirical Formula  C <sub>n</sub> H <sub>2n-10</sub>	58
	5. Cyclopentanotetrahydronaphthalenes and Their Alkyl Deriva-	co
	tives 6. Octahydroanthracenes and Their Alkyl Derivatives	60 62
	7 Octobardrophononthropograph Their Alkvi Derivatives	64
	8. Miscellaneous Polynuclear Aromatics of Empirical Formula $C_nH_{2n-10}$	68
III.	Polynuclear Aromatics of Empirical Formula C <sub>n</sub> H <sub>2n-12</sub>	74
	1. Indene with One Alkenyl or One Alkylidene Substitution 2. Naphthalene and Its Alkyl Derivatives	75 77
	3. Miscellaneous Polynuclear Aromatics of Empirical Formula $C_nH_{2n-12}$	116
137	Polynuclear Aromatics of Empirical Formula C <sub>n</sub> H <sub>2n-14</sub>	125
1 V .	1. Naphthalene Derivatives of Empirical Formula $C_nH_{2n-14}$	126
	2 Cyclenonenhthelenes and Their Alkyl Derivatives	131
	3. Tetrahydroanthracenes, Tetrahydropnenanthrenes, and Their	137
	Alkyl Derivatives 4. Miscellaneous Polynuclear Aromatics of Empirical Formula $C_nH_{2n-14}$	140
V.	Polynuclear Aromatics of Empirical Formula C <sub>n</sub> H <sub>2n-16</sub>	142
•	1 Independent One Phenyl Substitution	143
	2 Nanh-halana Darivatives of Empirical Formula Cartinate	145 149
	3. Tetrahydronaphthalene with One Phenyl Substitution  A Richardona Fluorene and Their Alkyl Derivatives	151

	P	age
	5. Dihydroanthracenes, Dihydrophenanthrenes, and Their Alkyl Derivatives	158
	6. Miscellaneous Polynuclear Aromatics of Empirical Formula $C_nH_{2n-16}$	166
VI.	Polynuclear Aromatics of Empirical Formula C <sub>n</sub> H <sub>2n-18</sub>	175
	1. Indene with One Phenyl Substitution	176
	2. Anthracenc and Its Alkyl Derivatives 3. Phenanthrene and Its Alkyl Derivatives	$\frac{180}{192}$
	4. Miscellaneous Polynuclear Aromatics of Empirical Formula $C_nH_{2n-18}$	210
VII.	Polynuclear Aromatics of Empirical Formula C <sub>n</sub> H <sub>2n-20</sub>	223
	1. Indene Derivatives of Empirical Formula $C_nH_{2n-20}$	224
	2. Naphthalene Derivatives of Empirical Formula $C_nH_{2n-20}$ 3. Cyclanoanthracenes and Their Alkyl Derivatives	$\frac{226}{231}$
	4. Cyclanophenanthrenes and Their Alkyl Derivatives	233
	5. Miscellaneous Polynuclear Aromatics of Empirical Formula $C_nH_{2n-20}$	239
VIII.	Polynuclear Aromatics of Empirical Formula $C_nH_{2n-22}$	248
	1. Naphthalene with Phenylalkenyl Substitutions	249
	2. Anthracene and Phenanthrene Derivatives of Empirical Formula $C_nH_{2n-22}$	251
	3. Cyclenoanthracenes, Cyclenophenanthrenes and Their Alkyl Derivatives	253
	4. Benzofluorenes and Their Alkyl Derivatives	256
	5. Pyrene and Its Alkyl Derivatives 6. Miscellaneous Polynuclear Aromatics of Empirical Formula $C_nH_{2n-22}$	<ul><li>258</li><li>260</li></ul>
IX.	Polynuclear Aromatics of Empirical Formula C <sub>n</sub> H <sub>2n-24</sub>	269
	1. Fluorene Derivatives of Empirical Formula C <sub>n</sub> H <sub>2n-24</sub>	270
	2. Dihydroanthracenes and Dihydrophenanthrenes with One Phenyl Substitution	274
	3. 1.2-Benzoanthracene and Its Alkyl Derivatives	277
	4. 3,4-Benzophenanthrene and Its Alkyl Derivatives 5. Naphthacene and Its Alkyl Derivatives	288 290
	6. Chrysene and Its Alkyl Derivatives	292
	7. Triphenylene and Its Alkyl Derivatives 8. Miscellaneous Polynuclear Aromatics of Empirical Formula	297
	$C_nH_{2n-24}$	299
X.	Polynuclear Aromatics of Empirical Formula $C_nH_{2n-26}$	312
	1. Indene with Two Phenyl Substitutions	313
	2. Binaphthyls, Their Alkyl Derivatives, and Two Naphthyl Substitutions on Alkanes	317
	3. Anthracene and Phenanthrene with One Phenyl Substitution	325
	4. Cholanthrene and Its Alkyl Derivatives 5. Other Benzocyclanoanthracenes, Benzocyclanophenanthrenes,	333
	and Their Alkyl Derivatives	336
	6. Miscellaneous Polynuclear Aromatics of Empirical Formula $C_nH_{2n-26}$	340
XI.	Polynuclear Aromatics of Empirical Formula $C_nH_{2n-28}$	349
	<ol> <li>Indene Derivatives of Empirical Formula C<sub>n</sub>H<sub>2n-28</sub></li> <li>Naphthalene with Two Phenyl or One Biphenyl Substitutions</li> </ol>	350 353
	3. Benzopyrenes and Their Alkyl Derivatives	358
	4. Miscellaneous Polynuclear Aromatics of Empirical Formula CnH2n-28	361

		age
XII.	Polynuclear Aromatics of Empirical Formula $C_nH_{2n-29}$	<b>37</b> 0
XIII.	<ul> <li>Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-30</sub></li> <li>1. Picene and Its Alkyl Derivatives</li> <li>2. Dibenzoanthracenes, Dibenzophenanthrenes, and Their Alkyl Derivatives</li> <li>3. Miscellaneous Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-30</sub></li> </ul>	372 373 375 379
XIV.	<ul> <li>Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-32</sub></li> <li>1. Indane with Three Phenyl Substitutions</li> <li>2. Fluorene with Two Phenyl Substitutions</li> <li>3. Dihydroanthracenes and Dihydrophenanthrenes with Two Phenyl Substitutions</li> <li>4. Miscellaneous Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-32</sub></li> </ul>	388 389 391 393 396
XV.	Polynuclear Aromatics of Empirical Formula $C_nH_{2n-34}$ 1. Indene with Three Phenyl Substitutions 2. Anthracene or Phenanthrene with Two Phenyl Substitutions 3. Miscellaneous Polynuclear Aromatics of Empirical Formula, $C_nH_{2n-34}$	403 404 407 415
XVI.	Polynuclear Aromatics of Empirical Formula $C_nH_{2n-36}$	429
XVII.	Polynuclear Aromatics of Empirical Formula $C_nH_{2n-38}$	445
XVIII.	Polynuclear Aromatics of Empirical Formula $C_nH_{2n-40}$	457
XIX.	Polynuclear Aromatics of Empirical Formula $C_nH_{2n-42}$	468
XX.	Polynuclear Aromatics of Empirical Formula C <sub>n</sub> H <sub>2n-44</sub>	476
XXI.	Polynuclear Aromatics of Empirical Formula $C_nH_{2n-46}$	482
XXII.	Polynuclear Aromatics of Empirical Formula C <sub>n</sub> H <sub>2n-48</sub>	489
XXIII.	Polynuclear Aromatics of Empirical Formula C <sub>n</sub> H <sub>2n-50</sub>	497
XXIV.	Polynuclear Aromatics of Empirical Formula C <sub>n</sub> H <sub>2n-52</sub>	508
XXV.	Polynuclear Aromatics of Empirical Formula C <sub>n</sub> H <sub>2n-54</sub>	511
XXVI.	Polynuclear Aromatics of Empirical Formula C <sub>n</sub> H <sub>2n-56</sub>	518
XXVII.	Polynuclear Aromatics of Empirical Formula $C_nH_{2n-58}$	524
XXVIII.	Polynuclear Aromatics of Empirical Formula C <sub>n</sub> H <sub>2n-60 to 88</sub>	<b>52</b> 8

### INTRODUCTION

### Foreword

The fourth volume of Physical Constants of Hydrocarbons is a collation and evaluation of the physical constants of polynuclear aromatic hydrocarbons. The compounds included under this classification are fused ring hydrocarbons having at least one aromatic ring.

The study conducted for this volume is a continuation of that made for Volume III.¹ References through April, 1945 were added to those collected concurrently with material for the previous volume. Details concerning the extent of literature coverage are explained in the Introduction to Volume III.

Although thousands of polynuclear aromatic hydrocarbons have been prepared, these compounds represent only a minor fraction of the theoretical possibilities. A great number of the larger polynuclears have been synthesized in recent years particularly in connection with cancer research and other physiological problems.

The present study also shows that the published data for many of the known compounds are insufficient for the identification of these compounds on the basis of physical properties alone. Identification of the carcinogenic and other higher polynuclear hydrocarbons has been markedly facilitated by the study of their ultraviolet, infrared, and Raman spectra.

Critical evaluation of the data in Volume IV was limited by the paucity of data available from the literature. Fewer constants have been determined for the polynuclear aromatics than for the mononuclear aromatics and consequently, the tabulations in this volume are not made with the discrimination possible in Volume III. For example, in Volume IV, melting ranges of over 2° and boiling ranges greater than 5° have been recorded for many compounds because these figures were the only ones available. Likewise, refractive index and density values for which temperature conditions were not stated in the literature appear in Volume IV. Many of these values have apparently been determined on the compound in the solid state, but a notation of the state of aggregation is included in the tabulations only when verified by the reference.

The methods used in the calculation of the most probable values of constants are described in the Introduction of Volume III. The lack of data for the polynuclear aromatics makes it impossible to calculate "best values" for many of the compounds. Only melting point data are available on the greater portion of the polynuclears. Boiling point, density, and refractive index values have been determined for only a very few polynuclears and in most cases the number of values determined is too small to permit the calculation of constants of equations representing them.

<sup>1</sup> Egloff, G., "Physical Constants of Hydrocarbons," Volume III, Reinhold Publishing Corporation, New York, 1946.

1

### Nomenclature

Inasmuch as the nomenclature used for the polynuclear aromatic hydrocarbons in Volume IV does not rigidly follow any previously described system. it will be explained in the following discussion. "The Ring Index" provides the only system sufficiently developed to permit the naming of the nuclei of all polynuclear hydrocarbons. Primarily an attempt is made to conform to these rules, but complications arising in the nomenclature of many derivative compounds necessitate departures from this system. Although the present study gives much evidence for the desirability of an entirely new system, the terminology herein is confined to that already in use. The nomenclature of this volume is also chosen with the intention of offering pictures of the structure which may be readily visualized.

A nomenclature system of basically different character has been proposed by Van Arsdell and Egloff,2 and another was partially elucidated by Clar.3 Both of these systems provide methods for simplifying problems of nomenclature. but neither is broad enough in scope to be applied to all compounds. viously mentioned systems are compared in Table I.

The nomenclature proposed by either Van Arsdell and Egloff or Clar is more scientifically systematic than that of Patterson or that used in this volume. the Van Arsdell and Egloff system, the general term -zene prefixed by the appropriate Greek numerical term designates a straight row of fused rings and, when further prefixed by iso-, a fused structure of the phenanthrene type. Clar makes use of the general term -acene, with the appropriate prefix, to designate straight row fusions and the term -phene for phenanthrene type structures. By using Roman numerals to designate the number of rings in each row, Clar's system can be extended to include compounds having one or more rings fused at an angle. According to Van Arsdell and Egloff, peri-fusions, as shown in example six, are defined by prefixing peri- to the -zene name. Despite the advantages of these systems, classical names such as phenanthrene, anthracene, and naphthacene are used in this volume to facilitate its use by scientists who are accustomed to thinking in terms of these names.

The use of letters to denote ring fusions, a point at which the present volume diverges from Patterson and Capell, will be treated in detail under a discussion on numbering and lettering. In a few cases, the basic names preferred by these authors are objectionable. Essentially different ones, therefore, are used. For example, the name "benzonaphthene," given to compound No. 6 in Table I, is scientifically ambiguous because the term "naphthene" is used indiscriminately throughout the literature to refer to several compounds and to a class of compounds.

In order to classify the compounds and name them with any degree of con-

<sup>&</sup>lt;sup>1</sup> Patterson, A. M., and L. T. Capell, "The Ring Index," Reinhold Publishing Corpora-

tion, New York, 1940.

'Van Arsdell, P. M., and G. Egloff, "Nomenclature of Cyclic Hydrocarbons," paper presented before the American Chemical Society, Atlantic City, September, 1941.

'Clar, E., "Aromatische Kohlenwasserstoffe," Springer-Verlag, Berlin, 1941.

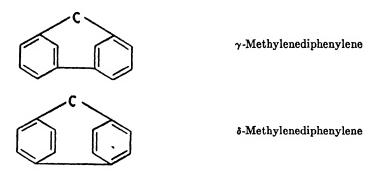
sistency, the name given in the literature is disregarded in many instances. Frequently the literature name bears no relation to the structure, is misleading,

TABLE I

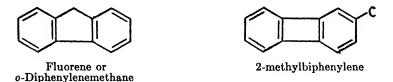
•	I ADDE I			
	Name used in Volume IV	Patterson	Egloff & Van Arsdell	Clar
1.	Phenan- threne	Phenan- threne	Isotri- zene	Triphene
2.	1,2-Benzo- anthra- cene	Benzo [a]- anthra- cene	Isotetra- zene	Tetra- phene- (I, II)
3.	Pentacene	Pentacene	Penta- zene	Pentacene
4.	1,2-Benzo- naph- thacene	Benzo[a]- naphtha- cene	Isopen- tazene	Penta- phene- (I, III)
5.	2,3,6,7-Di- benzo- phenan- threne	Dibenzo- [bh]- phenan- threne		Penta- phene- (II, II)
6.	Phenalene	Benzo- naph- thene	Peritri- zene	

or actually defines the compound incorrectly. The names "phthalacene" and "isophthalacene," for example, give no clue-to structure.

The many names assigned throughout the literature to fluorene and its derivatives provide illustrative examples of faulty nomenclature. Fluorene is often called "diphenylenemethane," a term applicable to more than one compound. The following structures, with which fluorene might be confused if named "diphenylenemethane", are also reported. When the name "di-



phenylenemethane" is qualified by o-, m- or p- it has structural significance, but even then may be confused with "methylbiphenylenes" which have also been called "diphenylenes."

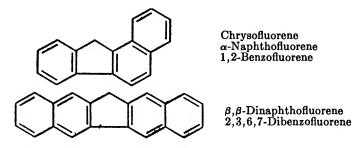


The diphenylene terminology is particularly misleading when used in reference to fluorene derivatives. The following compound, which is named <sup>1</sup>Carnelley, T., J. Chem. Soc. 37, 701 1880.

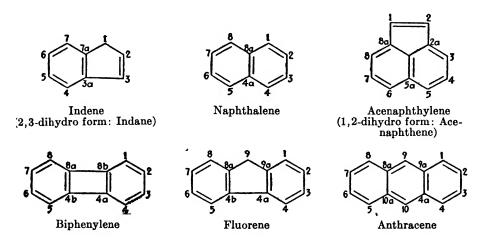
σ-phenethylfluorene in this volume, can be named as an ethane or propane according to the various methods of notation found in the literature:

1-Diphenylene-3-phenylpropane 
$$\alpha$$
-Diphenylene- $\gamma$ -phenylpropane or 1-(9'-Fluoryl)-2-phenylethane  $\alpha$ -9-Fluoryl- $\beta$ -phenylethane

Among the incorrectly used names are chrysofluorene and dinaphthofluorenes. These terms imply a chryso- and two naphtho-fusions, respectively, on fluorene but are used almost exclusively in reference to benzo- and dibenzofluorenes.



The explanation of the nomenclature used in the present study begins with consideration of nuclear structures. The following compounds with the indicated numbering are used as standard nuclei from which more complicated structures are built:



The least hydrogenated forms are considered parent except for the specified dihydro forms of indene, acenaphthylene, phenalene, and cholanthrylene. The largest of these nuclei contained in a given compound is considered as the nuclear structure of that compound. For example, the following structure is named 6-methylchrysene in preference to 1,2-benzo-3-methylphenanthrene.

6-Methylchrysene

Because anthracene is a larger nucleus then fluorene, 2,3-(3',2'-indo)-anthracene rather than 2,3-(3',2'-naphtho)-fluorene is assigned to the following structure.

2,3-(3',2'-Indo)-anthracene

Exception to this ruling is made, however, when the exception results in a decided simplification of the name. Prime numbering, for example, can be eliminated by naming the following structure 1,2,3,4-dibenzofluorene instead of 9,10-(2',3'-indo)-phenanthrene.

1,2,3,4-Dibenzofluorene

Likewise, 2,3-cyclopentano-5,6-benzo-7-methylphenalene includes only one set of numbers while cyclopentano-[a]-(6'-methylbenzo)-[de]-3-hydroanthracene necessitates the use of letters and two sets of numbers.

2,3-Cyclopentano-5,6-benzo-7-methylphenalene

When two parent nuclei of the same size are possible, the one having the greatest number of rings in a horizontal row is chosen as the parent nucleus. For example, 1,2-benzoanthracene is preferred to 2,3-benzophenanthrene.

1,2-Benzoanthracene

Many authors number compounds such as 1,2-benzoanthracene with a single set of numbers.

1,2-Benzoanthracene1

Although this system is advantageous in many respects, it brings in complications when derivative compounds are named. According to this plan, a methyl radical substituted on the benzo group has the number 2-, 3-, or 4- which is ambiguous because 1- and 2- are also used to describe the position of the benzo group on anthracene.

The positions of fused substituents cannot be indicated by numerical notations with simplicity when more than one side of the substituent is common to the parent nucleus. In these cases, letters designate the sides of fusion. The letter "a" represents the side between the 1- and 2-positions, "b" the next side

<sup>&</sup>lt;sup>1</sup> Patterson, A. M., and L. T. Capell, "The Ring Index," Reinhold Publishing Corporation, New York, 1940.

of the formula and so on in alphabetical order. The positions described by letters are illustrated in the following:

When a compound has two or more fused substituents, only one of which can be designated by numbers, letters are used for all.

Benzo-[a]-cyclopentano-[cd]-pyrene not 1,2-Benzocyclopentano-[cd]-pyrene

When all fused substituents are fused to only one side of the nucleus, the positions are indicated by numbers rather than letters.

1,2-Benzo-4,5-cyclopentanopyrene not Benzo-[a]-cyclopentano-[e]-pyrene

All fused substituents other than benzene which require the use of letters, must have a second series of letters or a set of numbers to indicate the positions of carbons relative to the substituent itself. These positions are stated in numbers and are inscribed within brackets preceding the letters used to designate the position of fusion on the nucleus as illustrated in the following examples.

Indo-[3,2,1-de]-anthracene

Naphtho-[3,2,1-bc]-cholanthrene

All numbering and lettering is clockwise except for a few instances which will be explained later. The initial point of numbering on the substituent nucleus is situated so that the carbon atoms common to both the parent and substituent nuclei have the lowest possible numbers consistent with standard numbering. Thus the nomenclature for the following structures is:

1,2-(3'-Methylbenzo)-anthracene not 6,7-(3'-Methylbenzo)-anthracene

1,2-(3',2'-Naphtho)-pyrene not 6,7-(3',2'-Naphtho)-pyrene

For spiro compounds, essentially the same rules apply and the spiro atom is assigned the lowest possible number.

Spiro [3-methylindane-1,1'-cyclohexane] not Spiro [1-methylindane-3,1'-cyclohexane]

Counterclockwise numbering is used for substituent nuclei when clockwise numbering involves a change in the standard numbering. For example, the use of clockwise numbers for the indo-substituent of 2,3-(2',3'-indo)-fluorene would involve numbering the saturated carbon as 3'-rather than 1'-, the standard position.

<sup>1</sup> See Patterson and Capell, "The Ring Index" p. 24. The formula for 7-benz-[de]-anthracene is numbéred clockwise but lettered counter clockwise. Many examples similar to this are found in the names of preference given in "The Ring Index."

The principle of assigning the lowest possible numbers or letters to substituents is followed in this volume as it is for the mononuclear aromatics. For detailed explanation of alkyl and alicyclic derivatives, see the Introduction to Volume III. The problems encountered in numbering the various nuclei of this volume complicate somewhat the general application of the system used for the mononuclears. The nuclear systems must be numbered in accordance with the previous discussion before the rules for using lowest possible numbers are applied. In this volume, there is an exception to this rule. When a partially hydrogenated nuclear structure can be numbered from more than one starting point, the hydrogenated positions are always assigned the lowest numbers. In the naming of the compound, these hydrogen atoms are designated immediately preceding the name of the nuclear structure even though the numbers referring to other substituents are higher.

7,8,9,10-Tetramethyl - 1,2,3,4-tetrahydronaphthacene

not 1,2,3,4-Tetramethyl - 7,8,9,10-tetrahydronaphthacene

1,2,3,4-Tetrahydro - 7,8,9,10-tetramethylnaphthacene

This rule is not followed, however, when assignment of lowest numbers to the hydrogen atoms involves changing of the standard numbering of a ring system.



4,5-Dihydroacenaphthylene

1,2-Dihydroacenaphthylene

Frequently in the literature, a cyclane fused to an aromatic nucleus was names as a cycleno-compound. Although the side common to the nucleus and the substituent has, at least in some of the resonance structures, a double bond, the substituent is called cyclano- in this volume. For example, the following structure is named 1,2-cyclopentanophenanthrene even though it is frequently referred to as a 1,2-cyclopentenophenanthrene in the literature.

1,2-Cyclopentanophenanthrene

Assuming other resonance structures of phenanthrene, a double bond will not always exist on the common side.

Two other resonance structures for 1,2-Cyclopentanophenanthrene

By naming the five carbon ring cyclopentano-, confusion is avoided with such structures as 1,2-(cyclopenten-3'-o)-phenanthrene.

If cyclopenteno- were used for the former structure, cyclopentadieno- would necessarily be used for the latter.

The notation of all possible resonance structures is not practical for this publication. Inclusion of double bonds in the formulae, however, is necessary to indicate the extent of saturation. Double bonds are, therefore, assigned arbitrarily in order to distinguish partially hydrogenated compounds from the others.

*Peri*-fused benzene ring structures, of which phenalene is the simplest, always have one saturated carbon atom.



In this study, the name given to such compounds, except for those included in the list of nuclear structures, specifies the saturated carbon atom by prefixing the appropriate number to -hydro-. For example, the following compound is named benzo-[cd]-5-hydropyrene.

### Order of Tabulating Compounds

The order used for the tabulation in this volume will be explained to facilitate the location of compounds. Constants have been determined on a wide variety of nuclear structures and derivatives thereof. For only a few nuclear structures, however, have sufficient derivatives of one type (i.e. alkyl, alkenyl, phenyl, diphenyl, etc.) been prepared to warrant the classification of these into a separate section. Consequently, a highly arbitrary system is devised for the classification and the order of listing compounds.

All compounds having a common empirical formula are classified in one group. These groups are numbered with Roman numerals and listed in order of decreasing hydrogen content. All compounds of empirical formula  $C_nH_{2n-8}$ , for example, are tabulated under I; compounds of formula  $C_nH_{2n-10}$  are listed under II, etc.

Groups I through XV are divided into sections denoted by Arabic numerals. These subdivisions are made when a lengthy series of compounds having a single nuclear structure or very similar nuclear structures is found. The section including the smallest nuclear structure is tabulated under 1., the next under 2., etc. Thus under Group VII are listed:

- 1. Indene derivatives of empirical formula  $C_nH_{2n-20}$ .
- 2. Naphthalene derivatives of empirical formula  $C_n H_{2n-20}$ .
- 3. Cyclanoanthracenes and their alkyl derivatives.
- 4. Cyclanophanenthrenes and their alkyl derivatives.
- 5. Miscellaneous polynuclears of empirical formula  $C_n II_{2n-20}$ .

When nuclei of two or more sections are of the same size, the section composed of derivatives of the nucleus having the longest straight row of rings precedes the others. Under Group IX, for example, section 5 is naphthacene and its alkyl derivatives, section 6 is chrysene and its alkyl derivatives, etc.

Except for sections entitled miscellaneous and a few sections in which general classification is simplified by listing together such similar compounds as anthracene and phenanthrene derivatives, the compounds of one section have a common nucleus. The compounds are listed in these sections:

- 1. In increasing order of carbon content, i.e. C<sub>12</sub> compounds precede C<sub>13</sub> compounds etc.
- 2. In order of increasing number of substituents, *i.e.* propyl precedes methyl plus ethyl which precedes trimethyl.
- 3. In order of increasing complexity of substituents, *i.e.* n-butyl precedes sec-butyl which precedes tert-butyl. For more detailed explanation see Introduction to Volume III.
- 4. In order of increasing numerical sum of the numbers describing substituent positions, *i.e.* 2,3,5-trimethylnaphthalene precedes 1,2,8-trimethylnaphthalene. If the sum is equal, the compound bearing the lowest number is placed first, *i.e.* 1,4-dimethylnaphthalene precedes 2,3-dimethylnaphthalene.

The order of tabulating compounds in the miscellaneous groups or sections

is more complicated. These hydrocarbons are listed primarily in order of increasing carbon content. After this qualification has been fulfilled, the general order follows:

1. In order of increasing number of rings, *i.e.*, anthracene derivatives precede hexahydronaphthacene derivatives which precede decahydropentacene etc.

(1) 1,2,3,4-Tetraethylanthracene

(2) 1,4-Diethyl-1,2,3,4,5,12-hexahydronaphthacene

(3) 1,2,3,4,4a,5,7,12,14,14a-Decahydropentacene

2. In order of increasing size of rings, *i.e.* a compound having one five- and two six-membered rings precedes one having one five-, one six-, and one seven-membered rings which precedes one having three six-membered rings.

(1) 1,2,3,4-Tetrahydrofluorene (1-5C and 2-6C rings)

(2) Cycloheptano-[cd]-indene (1-5C, 1-6C and 1-7C rings)

(3) 3a,4-Dihydrophenalan (3-6C rings)

3. In order of increasing number of rings fused at angles, i.e., naphthacene precedes chrysene which precedes triphenylene, etc.

(1) Naphthacene

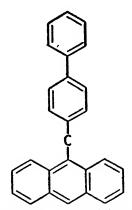
(2) Chrysene

(3) Triphenylene

The foregoing general principles, however, are not sufficient for determining order in every case. Further discussion follows to clarify special problems which arise after the general principles have been followed.

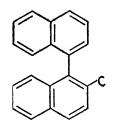
Benzhydrylpolynuclears precede polynuclear biphenylylmethane. For example, 9-benzhydrylanthracene precedes 9-anthryl-p-biphenylylmethane.

(1) 9-Benzhydrylanthracene



(2) 9-Anthryl-p-biphenylylmethane

Biaryls are listed preceding diarylaliphatics, i.e., 2-methyl-1,1'-binaphthyl precedes di-1-naphthylmethane.



(1) 2-Methyl-1,1'-binaphthyl

(2) Di-1-naphthylmethane

Diarylaliphatics, however, precede dialkyldiarylaliphatics. Thus 1,4-di-(1'-indenyl)-butane is listed before 1,2-di-[1'-(2'-methylindenyl)]-ethane.



(1) 1,4-Di-(1'-indenyl)-butane

(2) 1,2-Di-[1'-(2'-methylindenyl)]-ethane

When two polynuclear nuclei have the same carbon structure, the least hydrogenated nucleus is tabulated first. As an illustration, 1-ethylindene precedes 1-ethenylindane.

Spiro compounds follow others having rings of equivalent number and size. Both 7-methyl-5,6-cyclohexanoindene and spiro [indene-1,1'-cyclohexane] have fourteen carbon atoms and are named with reference to indene as the largest nucleus. They also have an equivalent number of rings of the same size. According to our arbitrary decision, the spiro compound is listed last.

(1) 7-Methyl-5,6-cyclohexanoindene (1-5C and 2-6C rings) (2) Spiro [indene-1,1'-cyclohexane] (1-5C and 2-6C rings)

and 2-6C) while the former has only three rings (1-8C and 2-6C).

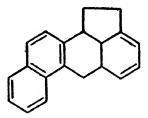
In the assignment of positions in the present tabulation, endo- and bicyclostructures are considered to be two rings. A dibenzoeyelooctene, therefore, would precede a dibenzobicyclooctane because the latter has four rings (2–5C

(1) 1,2,3,4-Dibenzocyclooctene-5

(2) 2,3,6,7-Dibenzobicyclo-[3,3,0]-octane

When two compounds have the same number of rings of the same size, but are not named as derivatives of the same nucleus, the compound having the name with the smallest parent nucleus is tabulated first. For example, benzo-[a]-cyclopentano-[de]-4a,9,9a,10-tetrahydroanthracene precedes 5a,6,12b,12c-tetrahydrocholanthrene.

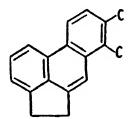
(1) Benzo-[a]-cyclopentano-[de]-4a,9,9a,10tetrahydroanthracene



 $(2)\ 5a, 6, 12b, 12c- Tetra hydrocholan threne$ 

When two compounds have nominally identical ring structures, but one compound has a larger ring structure because of the difference in position of a fused substituent, the compound having the least total substituents is recorded first. For example, 1,2-cyclopentano-3-methylphenanthrene precedes cyclopentano-[jk]-1,2-dimethylphenanthrene.

(1) 1,2 - Cyclopentano - 3 - methyl - phenanthrene



(2) Cyclopentano-[jk]-1,2-dimethylphen-

A review of the Introduction to Volume III is suggested to clarify the order of tabulation of compounds when aliphatic and alicyclic substituents are involved.

# I. BENZOCYCLANES, $C_n H_{2n-8}$

- 1. Indane and Its Alkyl Derivatives
- 2. Tetrahydronaphthalene and Its Alkyl Derivatives
- 3. Higher Benzocyclanes

# 1. INDANE AND ITS ALKYL DERIVATIVES, $C_nH_{2n-8}$

$\mathbf{C}_{9}\mathbf{H}_{10}$		1	0.9559	$D_{25}^{25}$ $^{24}$
Indane			0.9582	$D_{25}^{25\ 88}$
(Hydrindene)			0.9536	$D_{20}^{20}$ 33, 84
		1	0.9601	$D_{20}^{20}$ 88
	)		0.9611	$D_{20}^{20-84}$
		1	0.9547	19.5° 53
		}	0.96250	16.4° 33
M. P., °C		1	0.957	15° 22, 52
$-51.4^{12}$		1	0.9570	$D_{15}^{15}$ 33, 34
B. P., °C @ 760mm		1	0.960	$D_{15}^{15\ 19}$
177		1	0.9645	$D_{15}^{15-33,34}$
176-18044		1	0.95980	10.6° 33
$177.5 - 178.5^{13}$		1	0.9606	$D_{10}^{10-84}$
177-17830			0.9681	$D_{10}^{10}$ 33, 84
$175.5 - 177.5^{53}$			0.9646	$D_5^{5-33}$
$177.0^{33}$ $177^{14}$ , $^{27}$ , $^{29}$			0.9723	$D_5^{5-33}$
176-17734			0.9656	$D_4^{4-33}$
176-176.552			0.9030 $0.9732$	$D_4^4$ $D_4^{4-33}$
$176.25^{33}$			0.9732 $0.9813$	0° 13
$176^{2.47}$		20	0.9813	0. 1.
174-17619, 23		$n_{{ ilde { m D}}}^{20}$		
177–181	$771^{28}$		1.53812	
177–178	76425		1.538313	
176–176.5	76222		1.5352	25° 40
174-174.5	751.526		1.5355	25° 40
176-176.5	$742^{8}$ $29^{12}$	1	1.53512	20.8° 9
$\begin{array}{c} 79.0 \\ 51 \end{array}$	14 <sup>38</sup>		1.5382	20.7° 12
66.5	13 <sup>21</sup>		1.53877	19.25° 34
55	1140		1.5381	17.9° 21
$D_4^{20}$	••		1.53877	16.4° 33
$0.9639^{13}$			1.53896	15.95° <sup>84</sup> 15° <sup>13</sup>
$0.9645^{2}$		- 1	1.5407	20 8 9
0.8942	100° 18	1	1.53038	$n_{\mathrm{H}\alpha}^{-19.25}$ 84
0.9378	50° 13		1.53394	$n_{ m Hlpha}$
0.9560	25° 40		1.53394	$n_{\mathrm{H}\alpha}^{16.4}$ 88
0.9599	25° 40		1.53431	$n_{\mathrm{H}\alpha}^{15.95}$ 34
0.9507	$D_{25}^{25}$ 33, 34	1	1.53431	$n_{ m Hlpha}^{10.6}$ 33

$1.5471_{3}$	$n_{{ m H}m{eta}}^{{ m 20.8}}$
1.55114	$n_{{ m II}m{eta}}^{19\ 25\ 34}$
1.55114	$n_{{ m II}oldsymbol{eta}}^{16}$ 4 33
1.55105	$n_{{ m H}m{eta}}^{15~95~34}$
1.55105	$n_{{ m H}_{m{eta}}}^{10~6~33}$
1.56136	$n_{ m H}^{19}{}^{25}{}^{34}$
1.56136	$n_{{ m II}\gamma}^{16.4-33}$
1.56154	$n_{{ m II}_{\Upsilon}}^{15.95-34}$
1.56154	$n_{\rm II}^{10.6}$ 33
	(a)

(a) Refractive indices at other lines are found in references 33, 34.

# $C_{10}H_{12}$

### 1-Methylindane

B. P., °C @ 760mm  $186 - 187^{31}$ 182-18346 18247 60 - 641240 60  $10^{40}$ 

 $D_{4}^{20}$ 

 $n_{\rm p}^{20}$ 

- $0.940^{37}$
- 0.940240
- $0.9407^{31}$
- 0.9383
- 0.947
  - 21° 40

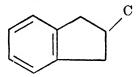
0.9661

- 1.522240
- 1.526037
- 1.5274231
- 1.5204
- 25° 40
- 1.53938
- 16° 46

25° 40

16° 46

### 2-Methylindane



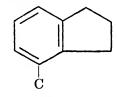
B. P., °C @ 760mm 183-185 74720  $10^{40}$ 70 69 1037, 40  $D_{4}^{20}$ 

> 0.9323725° 40 0.931723° 40 0.931822° 40 0.9321 $D_0^{17}$  20 0.9034

 $n_{\rm p}^{20}$ 

- 1.523537
- 25° 40 1.5189 23° 40 1.5200
- 22° 40 1.5224

### 4-Methylindane

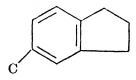


B. P., °C @ 760mm  $203^{24}$ 

 $D_4^{20}$ 

 $0.9350^{24}$ 

### 5-Methylindane



B. P., °C @ 760mm 74 1135

$$D_4^{20}$$

0.949485

 $n_{\scriptscriptstyle D}^{20}$ 

1.533235

# $C_{11}H_{14}$

# 1-Ethylindane

21239

84

 $12^{40}$ 

 $D_4^{20}$ 

 $0.9348 \\ 0.9346$ 

25° 39 23° 40

 $n_{\,\mathrm{D}}^{\,20}$ 

1.5202

23° 40

1.52286

 $n_{
m He}^{25}$  89

### 2-Ethylindane

B. P., °C @ 760mm

1240

 $D_{4}^{20}$ 

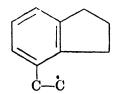
0.9266

 $n_{\rm p}^{20}$ 

1.5160

24° 10

# 4-Ethylindane



B. P., °C @ 760mm

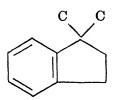
125-130

435

71 - 72

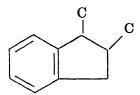
2.55

# 1,1-Dimethylindane



B. P., °C @ 760mm 1911

# 1,2-Dimethylindane



B. P., °C @ 760mm

79.80

 $10^{37}$ 

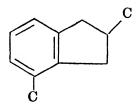
 $D_4^{20}$ 

 $0.927^{37}$ 

 $n_{\rm p}^{20}$ 

 $1.5186^{37}$ 

# 2,4-Dimethylindane



B. P., °C @ 760mm

105-106

254

100-105

233

# 4,7-Dimethylindane

B. P., °C @ 760mm 94-97 10<sup>37</sup> '94 10<sup>38</sup>

 $D_{4}^{20}$ 

0.94937

 $n_{\,{\scriptscriptstyle \mathrm{D}}}^{\,20}$ 

1.534237

1.534636

# $C_{12}H_{16}$

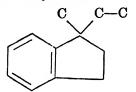
# 1-Isopropylindane

B. P., °C @ 760mm 98-100 17<sup>10</sup>

# 4-Isopropylindane

B. P., °C @ 760mm 88-90 13.6

# 1-Methyl-1-ethylindane



B. P., °C @ 760mm 218<sup>39</sup>

 $D_{4}^{20}$ 

0.9232

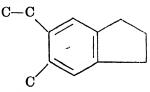
25° 39

 $n_{\rm p}^{20}$ 

1.51563

 $n_{
m He}^{25}$  39

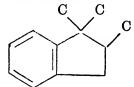
# 5-Methyl-6-ethylindane



B. P., °C @ 760mm 112-116

 $11^{\mathbf{A}}$ 

# 1,1,2-Trimethylindane



B. P., °C @ 760mm 208<sup>1</sup>, 11

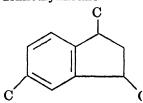
 $D_{4}^{20}$ 

 $0.919^{11}$ 

 $n_{\rm p}^{20}$ 

1.513711

# 1,3,5-Trimethylindane

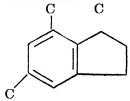


1451

 $n_{\rm p}^{20}$ 

1.520651

# 1,5,7-Trimethylindane



B. P., °C @ 760mm 104-105

 $n_{\,{\scriptscriptstyle D}}^{\,20}$ 

1.5231

25° 50

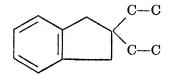
1450

# $C_{13}H_{18}$

## x-Butylindane (a)

- B. P., °C @ 760mm 238<sup>32</sup>
- (a) The structure of this compound was not clearly defined in the literature.

### 2,2-Diethylindane



B. P., °C @ 760mm 118

 $16^{17}$ 

 $D_4^{20}$ 

 $0.9162^{17} \\ 0.9295$ 

13.4° 45

 $n_{\scriptscriptstyle \mathrm{D}}^{20}$ 

1.5135

25° 17

1.51406

 $n_{\mathrm{H}\,\alpha}^{13\,4}$ 

1.52792

n 13.4 4

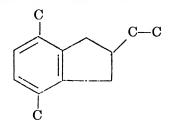
1.53643

213.4 4

1.51820

18.4 4 He

### 2-Ethyl-4,7-dimethylindane



B. P., °C @ 760mm

84-85

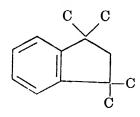
0.149

 $n_{\,\mathrm{D}}^{\,20}$ 

1.5207

22° 49

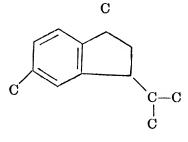
# 1,1,3,3-Tetramethylindane



B. P., °C @ 760mm 206-209<sup>1</sup>

### C14H20

# 1,5-Dimethyl-3-isopropylindane



B. P., °C @ 760mm

115-16

1051

 $n_{\rm p}^{20}$ 

1.5179

21° 51

### 1-Isopropyl-4,6-dimethylindane

$$C \longrightarrow C \longrightarrow C$$

B. P., °C @ 760mm 118-120

 $0.3^{48}$ 

 $n_{\,\mathrm{D}}^{\,20}$ 

1.5215

 $22.5^{\circ}$  48

### 2-Isopropyl-4,7-dimethylindane

 $\mathbf{C}$ 

M. P., °C 23-24<sup>41</sup>

R. P., °C @ 760mm 108-110 3<sup>41</sup>

 $n_{\rm p}^{20}$ 

 $1.518^{41}$ 

### 2,2-Diethyl-5-methylindane

B. P., °C @ 760mm 123-127  $12^{15}$   $D_4^{20}$ 0.9197  $21.5^{\circ 15}$  $n_2^{20}$ 

1.5105515

 $\begin{array}{c} C_{15}H_{22} \\ \\ 2,2,5\text{-Triethylindane} \end{array}$ 

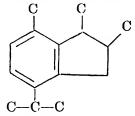
$$c-c$$

 $D_4^{20}$ 

0.9178 16.2° 45 0.9250 14.05° 45

 $n_{\rm p}^{20}$ 

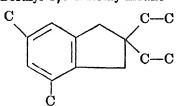
# 1,2,7-Trimethyl-4-isopropylindane



B. P., °C @ 760mm

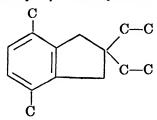
 $D_{4}^{20}$  297  $D_{4}^{20}$  0.9250  $D_{25}^{25}$  7  $D_{5}^{20}$  1.5112 25° 7

# 2,2-Diethyl-4,6-dimethylindane



B. P., °C @ 760mm 265-270<sup>18</sup> 260-265<sup>18</sup>

# 2,2-Diethyl-4,7-dimethylindane



B. P., °C @ 760mm 140-141 13<sup>16</sup>

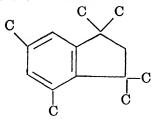
 $D_4^{20}$ 

 $0.923^{16}$ 

 $n_{\rm D}^{20}$ 

1.5159216

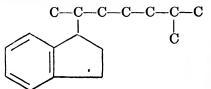
# 1,1,3,3,4,6-Hexamethylindane



B. P., °C @ 760mm 114-117  $13^{43}$   $D_{4}^{20}$ 0.901  $31^{\circ}$  430.905  $D_{31}^{31}$  43  $n_{D}^{20}$ 1.5100  $28.5^{\circ}$  43

#### C17H26

# 2-Methyl-6-indanylheptane



B. P., °C @ 760mm

100-102

0.240

D<sub>4</sub><sup>20</sup>

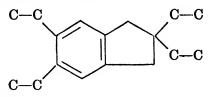
0.9083

19° 40

1.5065

19° 40

# 2,2,5,6-Tetraethylindane



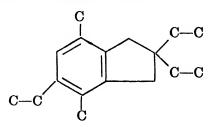
 $D_4^{20}$ 

0.9246 14.35° 48

 $n_{\rm p}^{20}$ 

 $egin{array}{lll} 1.51582 & n_{{
m H}lpha}^{14.35} & 45 \ 1.52935 & n_{{
m H}eta}^{14.35} & 45 \ 1.53776 & n_{{
m H}\gamma}^{14.35} & 45 \ 1.51998 & n_{{
m He}}^{14.35} & 45 \ \end{array}$ 

# 2,2,5-Triethyl-4,7-dimethylindane



B. P., °C @ 760mm 163–164 14<sup>17</sup> D<sub>4</sub><sup>20</sup> 0.916<sup>17</sup>

0.910

 $n_{\,\mathrm{D}}^{\,20}$ 

1.5159217

#### C19H30

### 2,2,4,5,6-Pentaethylindane

$$\begin{array}{c} c-c \\ c-c \\ \end{array}$$

$D_4^{20}$		
	0.9234	$14.1^{\circ}$ 45
$n_{{}^{{}_{\!\!D}}}^{{}_{\!\!\!\!20}}$		
	1.51702	$n_{\mathrm{H}\alpha}^{14.1}$ 45
	1.53043	$n_{\mathrm{H}eta}^{14.1}$
	1.53878	$n_{{ m II}\gamma}^{14.1}$ 45
	1.52115	$n_{ m He}^{14.1}$ 45

#### C21H34

### 2,2,4,5,6,7-Hexaethylindane

$$\begin{array}{c} C-C \\ C-C \\$$

$D_4^{20}$		
	0.9263	14.3° 45
$n_{\rm p}^{20}$		
	1.51856	$n_{\mathrm{H}lpha}^{14.3}$
	1.53198	$n_{\mathrm{H}oldsymbol{eta}}^{14~3~45}$
	1.54022	$n_{ m H_{m \gamma}}^{14.3}$
	1.52265	$n_{ m He}^{14.3}$ 45

#### C25H42

# 1-n-Hexadecylindane

M. P., °C 33.142

B. P., °C @ 760mm 206.5  $1.00^{42}$  $0.50^{42}$ 191.5 $D_4^{20}$ 98.9° 42 0.833360° 42 0.858737.8° 42 0.8738 $n_{\mathrm{p}}^{20}$ 1.494142 40.0° 42 1.4863 30.0° 42 1.4902

#### References on Indane and Its Alkyl Derivatives

- A. Arnold, R. T., and R. A. Barnes, J. Am. Chem. Soc. 66, 960 1944.
- Bogert, M. T., and D. Davidson, J. Am. Chem. Soc. 56, 185 1934.
- 2. Borsche, W., and M. Pommer, Ber. 54, 102 1921.
- Bruce, W. F., J. Am. Chem. Soc. 60, 2277 1938.
- Bruce, W. F., and L. F. Fieser, J. Am. Chem. Soc. 59, 479 1937.
- Bruce, W. F., and S. J. Kahn, J. Am. Chem. Soc. 60, 1017 1938.
- Bruce, W. F., and F. Todd, J. Am. Chem. Soc. 61, 157 1939.
- Carter, C. L., and S. N. Slater, J. Chem. Soc. 1938, 546.
- 8. Clemmensen, E., Ber. 47, 681 1914.
- Cotton, A., and H. Mouton, Ann. chim. phys. [8] 28, 209 1913.
- 10. Courtot, C., Ann. chim. [9] 5, 52 1916.
- Davidson, D., and J. Feldman, J. Am. Chem. Soc. 66, 488 1944.
- Dolliver, M. A., T. L. Gresham, G. B. Kistiakowsky, and W. E. Vaughan, J. Am. Chem. Soc. 59, 831 1937.
- Evans, E. B., J. Inst. Petroleum Tech.
   24, 537 1938.
- Fichter, F., and H. Stenzl, Helv. Chim. Acta 22, 425 1939.
- Fleischer, K., and W. W. Melber, Ann. 422, 242 1921.
- Freund, M., and K. Fleischer, Ann. 411, 14 1916.

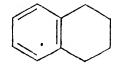
- Freund, M., K. Fleischer, and E. Gofferjé, Ann. 414, 1 1918.
- Freund, M., K. Fleischer, and M. Practorius, Ann. 411, 20 1916.
- Hall, C. C., J. Soc. Chem. Ind. 54, 208
   1935-1936.
- Kizhner, N., J. Russ. Phys. Chem. Soc. 46, 1411 1914; C.A. 9, 2066 1915.
- Kohlrausch, K. W. F., and R. Seka, Ber.
   1551 1938.
- Krämer, G., and A. Spilker, Ber. 23, 3276 1890.
- Krämer, G., and A. Spilker, Ber. 33, 2257 1900.
- 24. Kruber, O., Ber. 57, 1008 1924.
- Lebeau, P., and M. Picon, Compt. rend. 173, 84 1921.
- Lozovol, A. V., M. K. Diakova, and T. G. Stepanzeva, J. Gen. Chem. (U S.-S.R.) 9, 540 1939.
- 27. Mameli, E., and A. Mossini, Giorn. chim. ind. applicata 15, 161 1933.
- MeVieker, W. H., J. K. Marsh, and A. W. Stewart, J. Chem. Soc. 125, 1743
   1924.
- Meyer, R., and W. Meyer, Ber. 51, 1571
   1918.
- 30. Moschner, J., Ber. 33, 737 1900.
- Nenitzescu, C. D., and E. Cioranescu, Ber. 69, 1040 1936.
- 32. Noelting, E., Chimic & industrie 6, 719 1921.
- 33. Perkin, W. H., J. Chem. Soc. 69, 1025 1896.
- Perkin, W. H., Jr., and G. Révay, J. Chem. Soc. 65, 228 1894.
- Plattner, P. A., and H. Roniger, Helv. Chim. Acta 25, 590 1942.
- Plattner, P. A., and H. Roniger, Helv. Chim. Acta 25, 1077 1942.

- Plattner, P. A., and J. Wyss, Helv. Chim. Acta 24, 483 1941.
- Ramart-Lucas, and J. Hoeh, Bull. soc. chim. [5] 2, 327 1935.
- Roblin, R., Jr., D. Davidson, and M. T. Bogert, J. Am. Chem. Soc. 57, 151 1935.
- Ruzicka, L., and E. Peyer, Helv. Chim. Acta 18, 676 1935.
- St. Pfau, A., and P. Plattner, Helv. Chim. Acta 22, 202 1939.
- Schiessler, R. W., D. G. Clarke, C. S. Rowland, W. S. Sloatman, and C. H. Herr, preprint of paper presented at Am. Petroleum Inst., Nov. 10, 1943.
- Smith, L. I., and L. J. Spillane, J. Am. Chem. Soc. 65, 202 1943.
- 44. Stoermer, R., and J. Boes, Ber. 33, 3013 1900.
- 45. von Auwers, K., Ber. 60, 2122 1927.
- von Braun, J., E. Danziger, and Z. Kochler, Ber. 50, 56 1917.
- von Braun, J., and A. Stuckenschmidt, Ber. 56, 1724 1923.
- 48 Wagner-Jauregg, T., H. Arnold, and F. Huter, Ber. 75, 1293 1942.
- Wagner-Jauregg, T., H. Arnold, F. Huter, and J. Schmidt, Ber. 74, 1522 1941.
- Wagner-Jauregg, T., E. Friess, H. Hippehen, and F. Prier, Ber. 76, 1157 1943.
- 51. Wagner-Jauregg, T., and H. Hippchen, Ber. 76, 694 1943.
- 52. Weger, M., Z. angew. Chem. 22, 338 1909.
- 53. Zelinskii, N. D., and I. N. Tits, Ber. 62, 2869 1929.

# 2. TETRAHYDRONAPHTHALENE AND ITS ALKYL DERIVATIVES, C<sub>n</sub>H<sub>2n-8</sub>

#### C10H12

# 1,2,3,4-Tetrahydronaphthalene (Tetralin)



M. P., °C

-35.8

 $-30 - 27^{60}$ 

 $-30^{51, 52}$ 

 $-31^{33}$ 

 $-35.0 \pm 0.5^{23}$ 

29 C<sub>10</sub>H<sub>12</sub>

$-36.22(a)^{38}$		103.0	3028
$-36.3^{12,13}$		100-101	$25^{60}$
B. P., °C @ 760mm		101.1	$22.8^{19}$
207.4		96	2164
$208^{25}$ , 33, 51, 73		93.8	$20^{23}$
206-20860		90.8-91.2	$17^{65, 70}$
205-2086, 52, 57		89	1226
$207.66^{38}$		75.5	$7.2^{19}$
$207.6^{54}$		75.1	$7.1^{19}$
$207.3^{23}$		74.4	$7.0^{19}$
$207.3$ $207.1^{23}$		71-72	641
$207.1^{20}$		54.0	$2.2^{19}$
$207.0$ $207^{14}$ , 79		52-55	246
$206.6 - 207.0^{18}$		49.4	$1.9^{19}$
$206.5 - 207.0^{82}$		48.9	$1.9^{19}$
206.5-20737		46.6	$1.7^{19}$
20620747		65.0	$0.83^{35}$
		41.8	$0.7^{19}$
205–207 (b)	99	40.8	$0.7^{19}$
20610, 21, 34, 48, 61	, 00	40.1	$0.6^{19}$
201-20630		39.3	$0.5^{19}$
$205.4 - 205.8^{32}$		25.0	$0.40^{35}$
$205.7^{62}$		20	$0.263^{50}$
205 (c)		20	$0.26^{50}$
204-2059, 44, 49,	81	-1.2	$0.0235^{35}$
207.6	$768.66^{23}$	-2.4	$0.015^{35}$
206–207	$764^{70}$	$D_4^{20}$	
207	$763^{77}$	0.9702	
$206.8 \pm 0.2$	$759^{7}$	$0.966^{34,83}$	
206.5	$755^{60}$	$0.9663^{77}$	
206.6	$751.28^{23}$	0.968145	
206.2	$740^{23}$	$0.9705^{38}$	
204-205	$722^{4}$	$0.9706^{65}$	
204.0 - 204.5	$716.3^{82}$	$0.9707^{18}$	
204.5-205	$716^{3}$	0.97166, 67, 71	
205-207	$708^{2}$	$0.9712^{82}$	
167.5	27323	$0.9714^{77}$	
150.0	$162^{23}$	$0.972^{47.63}$	
148.6	$147.2^{19}$	$0.973^{22, 31}$	
140.0	11823	0.8496	179.6° 23
126.9	$76.0^{23}$	0.851	170° 22
126.6	$67.8^{19}$	0.8718	150.0° 28
116.8	$52.5^{23}$	0.892	• 120° 22
105.2	3675	0.9072	100° 18

 $C_{10}H_{12}$  30

0.9210	$84.0^{\circ}$ 23	$  n_{D}^{20}  $	
0.924	80° 22	1.539777	
0.9257	$78.45^{\circ 23}$	1.540234, 40, 83	
0.9272	75° 23	1.5416138	
0.9396	60° 23	1.542776	
0.943	50° 15	$1.54282^{82}$	
0.9464	50° 23	$1.5429^{65}$	
0.9469	50° 18	1.543047	
0.949	50° 22	1.543471	
0.9567	$36.7^{\circ}$ 23	1.543767	
0.9638	27° 27	$1.5438^{18}$	
0.9652	25° 23	$1.5439^{19}$	
0.9659	25° 5	1.544177	
0.9665	25° 38	1.544271	
0.9666	25° 41	1.5449316	
0.9675	$25^{\circ}$ 28, 30	1.54631	
0.9678	25° 28	1.5463816	
0.9658	$24.7^{\circ}$ 23	1.5467816	
0.965	$21.5^{\circ}$ 15	1.487	170° 22
0.9694	21.4° 70	1.507	120° 22
0.9729	$20.2^{\circ 31}$	1.524	80° 22
0.9735	$D_{20}^{20}$	1.537	50° 22
0.975	$D_{20}^{20\ 36}$	1.5392	25° 28
0.9732	18.0° 70	1.53952	25° 38
0.9734	17.8° 70	1.5396	25° 28
0.9738	17.6° 70	1.5408	25° 30
0.9718	17° 74	1.5410	25° 41
0.9737	17° 8	1.54222	$21.4^{\circ}$ 70
0.9731	15.1° 23	1.54605	$20.6^{\circ}$ 31
0.9698	15° 15	1.54614	$20.2^{\circ}$ 31
0.972	15° 57	1.5443	$19.9^{\circ}$ 24
0.974	15° 78	1.54511	17.8° 70
0.975	15° 43, 58, 59, 69	1.54529	17.6° 70
0.977	15° 53	1.5468	17° 8
0.9774	15° 68	1.5465	15° 18
0.9743	$13.5^{\circ}$ 23	1.5481	15° 68
0.97572	13.35° 49	1.55200	13.35° 49
0.9745	13.3° 23	1.55312	11.9° 49
0.97634 .	11.9° 49	1.51140	78.1 17
0.9842	2° 23		η <sub>Ηα</sub> 25 88
0.9825	Oo Re	1.53496	$n_{{ m H}_{lpha}}$
0.984	0° 84	1.53765	$n_{{ m H}{m lpha}}$
0.9866	0° 18	1.54612	$n_{\mathrm{H}lpha}^{20.6~81}$
0.984	$D_{f 0}^{0}$ 83	1.54181	$n_{\mathrm{H}\alpha}^{20.2~81}$
	-		

 $C_{10}H_{12}$ 

1.53703	$n_{\mathrm{H}lpha}^{20}$ 88
1.54021	$n_{\mathrm{H}lpha}^{17.8\ 70}$
1.54057	$n_{\mathrm{H}lpha}^{17.6~70}$
1.54703	$n_{\mathrm{H}\alpha}^{13.35}$ 49
1.54819	$n_{\mathrm{H}lpha}^{11.9}$ 49
1.52700	$n_{\mathrm{H}oldsymbol{eta}}^{78.1}$
1.55097	$n_{\mathrm{H}oldsymbol{eta}}^{25}$ 38
1.55405	91 4 70
1.55853	$n_{\mathrm{H}oldsymbol{eta}}^{20.6}$
	$n_{{ m H}oldsymbol{eta}}^{20.2}$ 31
1.55869	$n_{{ m H}m{eta}}$
1.55326	$n_{{ m H}oldsymbol{eta}}^{20~38}$
1.55709	$n_{{ m H}oldsymbol{eta}}^{17.8~70}$
1.55734	$n_{\mathrm{H}oldsymbol{eta}}^{17.670}$
1.56533	$n_{\mathrm{H}oldsymbol{eta}}^{13\ 35\ 49}$
1.56656	$n_{\mathrm{H}oldsymbol{eta}}^{11.9}$
1.53685	$n_{\mathrm{H}\gamma}^{78.1}$
1.56424	$n_{\mathrm{H}\gamma}^{21.4~70}$
1.56907	$n_{\mathrm{H}\gamma}^{20.6~31}$
1.56914	$n_{\mathrm{H}\gamma}^{20}$ 2 31
1.56765	$n_{ m H\gamma}^{17.8\ 70}$
1.56782	17 6 70
	$n_{\mathrm{II}\gamma}$
1.57838	$n_{ m H\gamma}^{11.9}$

#### Additional Data

$$\frac{dD}{dt} = -0.0007761$$
  
(0 to 50°C)

- (a) This constant was given as a freezing point in the literature.
- (b) The boiling point 205-207 is found in references 1, 29, 43, 50, 58, 59, 66, 68, 69, 76, 80.
- (c) The boiling point 205 is found in references 11, 20, 39, 40, 42, 53, 56, 72.

### References on 1,2,3,4-Tetrahydronaphthalene

- Bag, A., T. Egupov, and D. Volokitin, Ind. org. Chem. (U S.S.R.) 2, 141 1936; Chem. Zentr. 1937, I, 1273.
- Bamberger, E., and F. Bordt, Ber. 22, 625 1889.

- 3. Bamberger, E., and M. Kitschelt, Ber. 23, 1561 1890.
- Bamberger, E., and W. Lodter, Ann. 288, 74 1895.
- Berger, G., Z. physik. Chem. 28B, 95 1935.
- Berl, E., and H. Schildwächter, Brennstoff-Chem. 9, 105 1928.
- Bird, L. H., and E. F. Daly, Trans. Faraday Soc. 35, 588 1939.
- 8. Boedtker, E., and O. Rambech, Bull. soc. chim. [4] 35, 631 1924.
- Boes, J., Ber. deut. pharm. Ges. 12, 222 1902.
- Bogert, M. T., and V. G. Fourman, J. Am. Chem. Soc. 55, 4670 1933.
- Booth, J. H. W., Klepzigs Text-Z. 41, 314 1938; Chem. Zentr. 1938, II, 2045.
- de Carli, F., Atti accad. Lincei [6] 4, 460 1926.
- de Carli, F., Atti accad. Lincei [6] 4, 523
   1926.
- Deutsch, P., Gas-u. Wasserfach 74, 245
   1931.
- Doldi, S., Ann. chim. applicata 28, 454
   1938.
- 16. Eckart, II., Brennstoff-Chem. 4, 24 1923.
- 17. Eisenlohr, F., Ber. 54, 2857 1923.
- Evans, E. B., J. Inst. Petroleum Tech. 24, 537 1938.
- Gardner, G. S., and J. E. Brewer, Ind. Eng. Chem. 29, 179 1937.
- 20. Graebe, C., Ber. 5, 677 1872.
- 21. Graebe, C., and P. A. Guye, Ber. 16, 3028
- 22. Grodde, K.-H., Physik. Z. 39, 772 1938.
- Herz, W., and P. Schuftan, Z. physik. Chem. 101, 269 1922.
- Hofmann, F., K. Berlin, and A. W. Schmidt, Brennstoff-Chem. 14, 326 1933.
- Howard, L. B., and G. E. Hilbert, J. Am. Chem. Soc. 60, 1918 1938.
- Hückel, W., and H. Bretschneider, Ann. 540, 157 1939.
- 27. Joglekar, M. S., Z. Physik 101, 398 1936.
- Kagehira, I., Bull. Chem. Soc. Japan 6, 241 1931.
- Kantorowicz, H., Seifenfabr. 40, 129
   1920.
- Kimura, S., Mem. Coll. Sci., Kyoto Imp. Univ. 14, 303 1931.

- 31. Krollpfeiffer, F., Ann. 430, 161 1923.
- Larsen, R. G., R. E. Thorpe, and F. A. Armfield, Ind. Eng. Chem., 34, 183 1942.
- Lebeau, P., and Picon, Compt. rend.
   158, 1514 1914.
- 34. Leroux, H., Compt. rend. 139, 672 1904.
- 35. Linder, E. G., J. Phys. Chem. 35, 531 1931.
- Lovell, W. G., J. M. Campbell, F. K. Signaigo, and T. A. Boyd, Ind. Eng. Chem. 26, 475 1934.
- 37. Maillard, A., Ann. combustibles liquides 9, 1013 1934.
- Mair, B. J., and A. J. Streiff, J. Research Natl. Bur. Standards 27, 343 1941.
- 39. Mameli, E., and A. Mossini, Giorn. chim. ind. applicata 15, 161 1933.
- Martin, E., Ann. combustibles liquides 12, 97 1937.
- Matsuno, K., and K. Han, Bull. Chem. Soc. Japan 11, 321 1936.
- McVicker, W. H., J. K. Marsh, and A. W. Stewart, J. Chem. Soc. 125, 1743 1924.
- 43. Meier, E., Farben-Ztg. 26, 1765 1921.
- 44. Meyer, R., and H. Fricke, Ber. 47, 2765 1914.
- 45. Müller, F. H., Physik. Z. 38, 283 1937.
- 46. Musser, D. M., and H. Adkins, J. Am. Chem. Soc. 60, 664 1938.
- Nametkin, S. S., and L. M. Rosenberg, Bull. acad. sci. U.R.S.S. 1943, No. 11– 12, 3; Survey For. Petrol. Liter. T460, Aug. 18, 1944.
- 48. Ogawa, T., and T. Yokota, Bull. Chem. Soc. Japan 5, 266 1930.
- 49. Pellini, G., Gazz. chim. ital. 31, I, 1 1901.
- 50. Piatti, L., Erdol u. Teer 5, 421 1929.
- 51. Pohl, J., and M. Rawicz, Z. physiol. Chem. 104, 95 1919.
- Röckemann, W., Arch. exptl. Path. Pharmakol. 92, 52 1922.
- Ross, R., and J. P. Leather, Analyst 31, 284 1906.
- Rossini, F. D., Petroleum Engineer 14,
   No. 5, 41 1943.
- 55. Roth, W. A., and K. von Λuwers, Ann. 407, 145 1914.
- Sabatier, P., and J. B. Senderns, Compt. rend. 132, 1254 1901.

- 57. Sachanen, A. N., [Sakhanin] and M. D. Tilicheev, Ber. 62, 658 1929.
- 58. Schön, T., Oesterr. Chem.-Ztg. 23, 69 1920.
- Schrauth, W., and R. Heuter, Farben-Ztg. 25, 535 1919-1920.
- 60: Schroeter, G., Ann. 426, 1 1922.
- 61. Schroeter, G., Umschau 25, 213 1921.
- Schuster, F., Gas-u. Wasserfach 72, 650
   1929.
- Spilker, A., and K. Zerbe, Z. angew. Chem. 39, 1138 1926.
- Straus, F., and L. Lemmel, Ber. 54, 25
   1921.
- Straus through W. A. Roth and K. von Auwers, Ann. 407, 145 1914.
- Treibs, W., and H. Schmidt, Ber. 61, 459 1928.
- Turova-Pollak, M. B., and N. B. Lysibmova, J. Gen. Chem. (U.S.S.R.) 8, 538 1938; C.A. 32, 7909 1938.
- Utz, Gummi-Ztg. 34, 779 1920; C.A. 14, 3002 1920; Chem. Zentr. 1920, IV, 182.
- 69. Vollmann, Farben-Ztg. 24, 1689 1919; C.A. 14, 357 1920.
- 70. von Auwers, K., Ber. 46, 2988 1913.
- 71. von Auwers through G. Schroeter, Ann. 426, 1 1922.
- von Braun, J., and H. Deutsch, Ber. 45, 1267 1912.
- von Braun, J., and A. Stuckenschmidt, Ber. 56, 1724 1923.
- Vorländer, D., and J. Fischer, Ber. 65, 1756 1932.
- Ward, J. J., W. R. Kirner, and H. C. Howard, J. Am. Chem. Soc. 67, 246
- Watermann, H. I., J. J. Leendertse, and J. B. Nieman, Rec. trav. chim. 56, 59 1937.
- Willstätter through W. A. Roth and K. von Auwers, Ann. 407, 145 1914.
- 78. Weger, M., Ber. 36, 309 1903.
- Weissenberger, G., Gas-u. Wasserfach
   73, 819 1930.
- 80. Weissenberger, G., Gas-u. Wasserfach 74, 154 1931.
- Willstätter, R., and V. L. King, Ber. 46, 527 1913.
- 82. Willstätter, R., and F. Seitz, Ber. 56, 1388 1923.
- 83. Wischin, R., Chem. Ztg. 29, 1126 1905.

C11H14

#### 1-Methyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 218-21944 153-155  $55^{1}$  $D_{A}^{20}$ 25° 44 0.9536 25° 44 0.9546 25° 44 0.9547  $n_{\rm p}^{20}$  $n_{
m He}^{25}$  44 1.53316  $n_{\rm He}^{25~44}$ 1.53332 25 44 1.53366

## 2-Methyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 220-222<sup>47</sup> 221<sup>3</sup> 220<sup>32</sup> 218-220<sup>32</sup> 99-101 13<sup>3</sup>

### 5-Methyl-1,2,3,4-tetrahydronaphthalene

M. P., °C  $-23.02 (a)^{34}$ B. P., °C @ 760mm 234.445 234.3534 217.284 184.98 143.76 57.384  $D_4^{20}$  $0.9720^{34}$ 25° 84 0.9682 $n_{\rm p}^{20}$ 1.5439534 25° 34 1.54190  $n_{\mathrm{H}\alpha}^{25~34}$ 1.53738 20 34 1.53937 1.55325 1.55536

(a) This constant was given as a freezing point in the literature

# 6-Methyl-1,2,3,4-tetrahydronaphthalene

M. P., °C -39.92 (a)<sup>34</sup>
B. P., °C @ 760mm  $229.03^{34}$   $229.0^{45}$   $224-226^{27}$  179.91  $217.2^{34}$  138.75  $57.3^{34}$   $D_4^{20}$   $0.950^{27}$ 

0.950<sup>27</sup>
0.9537<sup>34</sup>
0.9500
25° <sup>84</sup>
0.9541
15.1° <sup>27</sup>
1.5328 <sup>21</sup>

- $1.535^{27}$  $1.53572^{34}$ 1.53365 25° 34 15.1° 27 1.53719 1.52920  $n_{\mathrm{H}\alpha}^{20~34}$ 1.53115  $n_{\mathrm{H}\alpha}^{15.1}$  27 1.53316  $n_{\mathrm{H}B}^{25~34}$ 1.54498  $n_{\mathrm{H}eta}^{20~34}$ 1.54706 15.1 27 1.54907  $n_{\rm H_{\Upsilon}}^{15.1}$  27 1.55897
- (a) This constant was given as a freezing point in the literature.

# x-Methyl-1,2,3,4-tetrahydronaphthalene (a)

B. P., °C @ 760mm 115-117

 $15^{2}$ 

 $D_4^{20}$ 

0.936843

(a) The structure of this compound was not clearly defined in the literature.

# $\boldsymbol{C_{12}H_{16}}$

# 1-Ethyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 237-238<sup>44</sup>

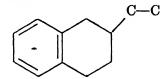
241.5-243.5 73939

 $D_4^{20}$ 

0.952939

0.9498 25° 44 0.9511 25° 44  $n_{D}^{20}$ 1.5388<sup>89</sup>
1.52992  $n_{He}^{25}$ 1.52999  $n_{He}^{25}$ 

2-Ethyl-1,2,3,4-tetrahydronaphthalene

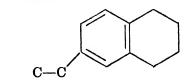


B. P., °C @ 760mm

237<sup>31</sup> 235–235.5 731<sup>30</sup> 63–65 0.5<sup>38</sup>

 $D_4^{20}$  0.9401 15.5° 30 0.9447 15° 31 0.9515  $D_0^{0.30}$  0.9542  $D_0^{0.81}$   $n_D^{20}$  1.5250 15.5° 30 1.5287 15° 31

### 6-Ethyl-1,2,3,4-tetrahydronaphthalene



B. P., °C @ 760mm 245-246<sup>27</sup> 239-239.5

239-239.5  $736^{30}$  127  $21^{22}$ 

 $\begin{array}{c} D_4^{20} \\ & 0.948^{27} \\ & 0.9499 \\ & 0.9608 \\ & 0.9733 \end{array} \qquad \begin{array}{c} 17.6^{\circ \ 27} \\ 15.5^{\circ \ 80} \\ D_0^{0 \ 30} \end{array}$ 

1.53427

1.53474 17.6° 27

1.5350	15.5° 30
1.53072	$n_{\mathbf{H}oldsymbol{lpha}}^{17.6}$
1.54627	$n_{\mathbf{H}oldsymbol{eta}}^{17.6}$
1.55594	$n_{ m H_{oldsymbol{\gamma}}}^{17~6~27}$

# x-Ethyl-1,2,3,4-tetrahydronaphthalene (a)

B. 
$$\Gamma$$
., °C @ 760mm  
 $244^{5}$   
 $119-120$   $12^{5}$   
 $121-122$   $10^{3}$   
 $D_{4}^{20}$   
 $0.9632$   $17^{\circ 5}$   
 $n_{p}^{20}$   
 $1.5414$   $16^{\circ 5}$ 

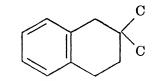
(a) The ethyl group may be in either the 2- or 6-position.

# 1,1-Dimethyl-1,2,3,4-tetrahydronaphthalene

## 1,2-Dimethyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 23544 123.5-124.5 1150  $D_{\scriptscriptstyle A}^{20}$ 0.988500.9433 25° 44 0.9841 23.8° 50 0.9847 23.5° 50  $n_{\scriptscriptstyle \mathrm{D}}^{20}$ 1.559350 23.8° 50 1.55762  $n_{\mathrm{H}lpha}^{23~8~50}$ 1.55298  $n_{\mathrm{H}\beta}^{23~8~50}$ 1.57082 23.8 50 1.58195  $n_{\mathrm{II}_{m{\gamma}}}$ 1.52652

# 2,2-Dimethyl-1,2,3,4-tetrahydronaphthalene



B. P., °C @ 760mm  $123 34^{48}$   $104 12^{15}$   $D_{4}^{20} 0.92483 24.1^{\circ} {}^{48}$   $1.51850 24.1^{\circ} {}^{48}$ 

# 2,3-Dimethyl-1,2,3,4-tetrahydronaphthalene

36

B. P., °C @ 760mm 229-231 (a)<sup>17</sup> 227<sup>32</sup>, <sup>33</sup>

222-224 (a)<sup>17</sup>

(a) These constants were determined on isomeric forms.

# 2,5-Dimethyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm

110-111 1035

 $D_4^{20}$ 

0.9487

16° 35

 $14^{26}$ 

# 2,6-Dimethyl-1,2,3,4-tetrahydronaphthalene

M. P., °C

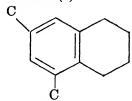
B. P., °C @ 760mm 237-239<sup>16</sup>

 $14 - 17^{16}$ 

### 2,7-Dimethyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 237-238<sup>16</sup>

# 5,7-Dimethyl-1,2,3,4-tetrahydronaphthalene (a)



B. P., °C @ 760mm

 $250-252^{27}$ 

 $D_4^{20}$ 

 $0.960^{27}$ 

0.9589

21.0° 27

 $n_{\scriptscriptstyle 
m D}^{20}$ 

 $1.541^{27} \\ 1.54094$ 

21.0° 27

1.53683

 $n_{\mathrm{H}lpha}^{21.0~27}$ 

1.55287

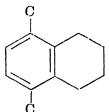
 $n_{\mathrm{H}\beta}^{21.0}$  27

1.56297

21.0 27

(a) The structure of this compound was not clearly defined in the literature.

# 5,8-Dimethyl-1,2,3,4-tetrahydronaphthalene



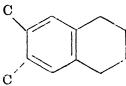
B. P., °C @ 760mm

 $254^{4}$ 

120

14

# 6,7-Dimethyl-1,2,3,4-tetrahydronaphthalene



## $C_{13}H_{18}$

# 1-n-Propyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 253<sup>44</sup> 256-258 759<sup>39</sup>

 $D_4^{20}$ 

0 9395<sup>39</sup>
0.9415 25° <sup>44</sup>

 $n_{\,{\scriptscriptstyle \mathrm{D}}}^{\,20}$ 

1.5308<sup>39</sup>
1.52496  $n_{\text{Ho}}^{25 \text{ 44}}$ 

## 1-Isopropyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 247<sup>44</sup>

 $D_4^{20}$ 

0.9450

25° 44

 $n_{\,{\scriptscriptstyle \mathrm{D}}}^{\,20}$ 

1.52705

 $n_{\rm He}^{25}$ 

# x-Isopropyl-1,2,3,4-tetrahydronaphthalene (a)

B. P., °C @ 760mm

124–126 13<sup>3</sup> 127 7<sup>5</sup>

 $D_4^{20}$ 

0.9518 16° 5

 $n_{\,{\scriptscriptstyle {
m D}}}^{\,{\scriptscriptstyle 20}}$ 

1.5352 16° 5

- (a) The isopropyl group may be in either the 2- or 6-position.
- 2-Methyl-2-ethyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm

 $20^{15}$ 

1-Ethyl-6-methyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 126 10<sup>11</sup>

2-Ethyl-5-methyl-1,2,3,4-tetrahydronaphthalene

$$\bigcup_{C}^{C-C}$$

1628

B. P., °C @ 760mm • 130

2-Methyl-6-ethyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 140-145

3-Ethyl-5-methyl-1,2,3,4-tetrahydronaphthalene

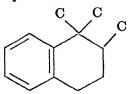
$$\bigcap_{C-C}$$

B. P., °C @ 760mm 129-131

 $14^{23}$ 

1012

1,1,2-Trimethyl-1,2,3,4-tetrahydronaphthalene



B. P., °C @ 760mm 241-242

 $760.7^{42}$ 

 $n_{\,{\scriptscriptstyle D}}^{\,20}$ 

1.5270

 $25^{\circ_{42}}$ 

1,1,6-Trimethyl-1,2,3,4-tetrahydronaphthalene (Ionene)

B. P., °C @ 760mm

240-242 752<sup>9</sup>
238-239 730<sup>37</sup>
130 30<sup>14</sup>
114 14<sup>9</sup>
107 10<sup>9</sup>
88-91 4<sup>8</sup>

 $D_4^{20}$ 

 $0.9303^{\,9}$ 

 $0.9331^{9} 0.9356^{37}$ 

0.9320 25° 8

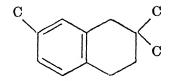
 $n_{\rm p}^{20}$ 

 $1.5225^9$   $1.5257^{37}$ 

1.52167

25° 8

2,2,7-Trimethyl-1,2,3,4-tetrahydronaphthalene



B. P., °C @ 760mm 128

140

2348

 $D_{4}^{20}$ 

0.91747

24.5° 48

 $n_{\scriptscriptstyle
m D}^{\rm 20}$ 

1.51635

24.5° 48

 $18.5^{24}$ 

1,6,8-Trimethyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 133-136

# $C_{14}H_{20}$

# x-n-Butyl-1,2,3,4-tetrahydronaphthalene (a)

B. P., °C @ 760mm 272-273 750<sup>39</sup> 269-270 750<sup>39</sup>  $D_4^{20}$ 0.9312<sup>39</sup> 0.9331<sup>29</sup>

 $n_{_{\mathrm{D}}}^{20}$  1.5258 $^{39}$  1.5271 $^{39}$ 

(a) The structure of this compound was not clearly defined in the literature.

# x-sec-Butyl-1,2,3,4-tetrahydronaphthalene (a)

B. P., °C @ 760mm
268-270 759<sup>39</sup>

D<sub>4</sub><sup>20</sup>
0.9356<sup>39</sup>

1.5255<sup>39</sup>

(a) The structure of this compound was not clearly defined in the literature.

# 6-tert-Butyl-1,2,3,4-tetrahydronaphthalene

$$\begin{array}{c} c \\ c \\ \end{array}$$

B. P., °C @ 760mm 262-264 752<sup>10</sup> 138-140 18<sup>10</sup> 135-136 16<sup>10</sup>

# x-tert-Butyl-1,2,3,4-tetrahydronaphthalene (a)

B. P., °C @ 760mm  $265-267 746^{39}$   $129 11^{3}$   $D_{4}^{20} 0.9349^{39}$   $n_{p}^{20} 1.5292^{39}$ 

(a) The structure of this compound was not clearly defined in the literature.

# x-Butyl-1,2,3,4-tetrahydronaphthalene (a)

B. P., °C @ 760mm

265 5-266.5<sup>5</sup>

138

9<sup>5</sup>

D<sub>4</sub><sup>20</sup>

0.9463

15° <sup>5</sup>

n<sub>p</sub><sup>20</sup>

1.5348

15° <sup>5</sup>

(a) The butyl group may be in either the 2- or 6-position.

# 1-Methyl-3-isopropyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 138-139 • 16<sup>18</sup> 5-Methyl-8-isopropyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 135–140

1246

2,2-Diethyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm

448

6,7-Diethyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 150-151

 $18^{22}$ 

1,1-Dimethyl-6-ethyl-1,2,3,4-tetrahydronaphthalene

$$c-c$$

B. P., °C @ 760mm 104-105

441

 $D_4^{20}$  0.9304 25° 41  $n_D^{20}$  1.5220 25° 41

1,1,2,6-Tetramethyl-1,2,3,4-tetrahydronaphthalene (Irene)

B. P., °C @ 760mm 120–125

 $D_4^{20}$ 

0.9379

25° 6

106.7

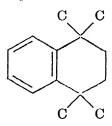
 $n_{\,{\scriptscriptstyle {
m D}}}^{\,20}$ 

 $1.511^{7}$ 

1.5261

25° 6

1,1,4,4-Tetramethyl-1,2,3,4-tetrahydronaphthalene



B. P., °C @ 760mm

24813

82-84

318

 $D_4^{20}$ 

0.9482

27° 18

 $n_{\scriptscriptstyle 
m D}^{20}$ 

1.5278

27° 13

# 1,1,6,7-Tetramethyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 103-104 4<sup>41</sup>

D<sub>4</sub><sup>20</sup>
0.9392 25° 41

n<sub>p</sub><sup>20</sup>
1.5280 25° 41

### $C_{15}H_{22}$

# x-Pentyl-1,2,3,4-tetrahydronaphthalene (a)

B. P., °C @ 760mm

147-150

12<sup>5</sup>

0.9478

1.5332

17° <sup>5</sup>

(a) The pentyl group may be in either the 2- or 6-position.

#### C<sub>16</sub>H<sub>24</sub>

# x-Hexyl-1,2,3,4-tetrahydronaphthalene (a)

thalene (a)

B. P., °C @ 760mm 120.5-122  $0.8^{40}$   $0.9203^{40}$   $n_{\rm p}^{20}$ 

 $n_{\scriptscriptstyle 
m D}^{\scriptscriptstyle 20}$ 

1.517140

(a) The structure of this compound was not clearly defined in the literature.

### C18H28

x-n-Octyl-1,2,3,4-tetrahydronaphthalene (a)

B. P., °C @ 760mm 324-326  $740^{39}$   $D_4^{20}$   $0.9132^{39}$   $n_p^{20}$   $1.5138^{39}$ 

(a) The structure of this compound was not clearly defined in the literature.

# 1,4-Diisobutyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 170-175  $16^{25}$  $D_4^{20}$ 100° 19 0.8619 50° 19 0.8939190 25, 29 0.913710° 25 0.9194 0° 19 0.9259 $n_{\rm p}^{20}$ 18° 25, 29 1.5184

# C22H36

1.540

x,x-Dihexyl-1,2,3,4-tetrahydronaphthalene (a)

B. P., °C @ 760mm 182–184

# C22H36

 $D_4^{20}$ 

0.9303<sup>40</sup> 0.9358

 $D_{15}^{15}$  40

 $n_{\,\mathrm{D}}^{\,20}$ 

1.522040

(a) The structure of this compound was not clearly defined in the literature.

### C26H44

6-n-Hexadecyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 210-215

120

#### C28H48

6-n-Octadecyl-1,2,3,4-tetrahydronaphthalene

M. P., °C

32.5-33<sup>28</sup> 29-30<sup>36</sup>

 $D_4^{20}$ 

0.8656 0.8722

55° 28 45° 28

 $n_{\rm p}^{20}$ 

1.48656

55° 28

1.48280

 $n_{\mathrm{H}lpha}^{55~28}$ 

1.49371

n 55 28

### C29H50

6-Methyl-7-n-octadecyl-1,2,3,4tetrahydronaphthalene

M. P., °C 60–61<sup>20</sup>

## C<sub>32</sub>H<sub>56</sub>

6-Docosyl-1,2,3,4-tetrahydronaphthalene

M. P., °C 43–45<sup>36</sup>

 $D_4^{20}$ 

0.8786

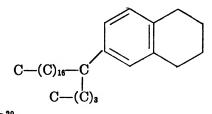
25° 36

 $n_{\scriptscriptstyle 
m D}^{20}$ 

1.4969

25° 36

6-(5'-Docosyl)-1,2,3,4-tetrahydronaphthalene



 $D_4^{20}$  0.881

25° 86

 $n_{\scriptscriptstyle \mathrm{D}}^{20}$ 

1.4723

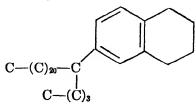
80° 49 25° 86, 49

1.4932

43 C<sub>36</sub>H<sub>64</sub>

#### C<sub>86</sub>H<sub>64</sub>

6-(5'-Hexacosyl)-1,2,3,4-tetrahydronaphthalene



M. P., °C 3536

 $D_4^{20}$ 

0.8762

25° 36

 $n_{\rm p}^{20}$ 

1.4940

25° 36

# References on C<sub>11</sub>H<sub>14</sub> through C<sub>26</sub>H<sub>04</sub> Compounds

- Arbusov, B. A., and E. V. Kuznetsov, Compt. rend. acad. sci. U.R.S.S. 39, 343 1943; Survey For. Petrol. Liter. Feb. 25, 1944.
- Arnold, R. T., and R. Banes, J. Am. Chem. Soc. 65, 2393 1943.
- Barbot, A., Bull. soc. chim. [4] 47, 1314
   1930.
- Barnett, E. deB., and F. G. Sanders, J. Chem. Soc. 1933, 434.
- 5. Boedtker, E., and O. Rambeck, Bull. soc. chim. [4] 35, 631 1924.
- Bogert, M. T., and P. M. Apfelbaum, J. Am. Chem. Soc. 60, 930 1938.
- Bogert, M. T., and P. M. Apfelbaum, Science [2] 79, 280 1934.
- Bogert, M. T., D. Davidson, and P. M. Apfelbaum, J. Am. Chem. Soc. 56, 959 1934.
- Bogert, M. T., and V. G. Fourman, J. Am. Chem. Soc. 55, 4670 1933.
- Bromby, N. G., A. T. Peters, and F. M. Rowe, J. Chem. Soc. 1243, 144.
- Brunner, O., and F. Grof, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 142, Abt. IIb, 682 1933.
- 12. Brunner, O., and F. Grof, Sitzber. Akad.

- Wiss. Wien, Math. naturw. Klasse 144, Abt. IIb, 565 1935.
- Bruson, H. A, and J. W. Kroeger, J. Am. Chem. Soc. 62, 36 1940.
- Clemo, G. R., and H. G. Dickenson, J. Chem. Soc. 1935, 735.
- Clemo, G. R., and H. G. Dickenson, J. Chem. Soc. 1937, 255.
- 16. Coulson, E. A., J. Chem. Soc. 1935, 77.
- 17. Coulson, E. A., J. Chem. Soc. 1938, 1305.
- Darzens, G., and A. Levy, Compt. rend. 201, 152 1935.
- Evans, E. B., J. Inst. Petroleum Tech. 24, 537 1938.
- Ficser, L. F., and J. Cason, J. Am. Chem. Soc. 62, 432 1940.
- Fieser, L. F., and R. N. Jones, J. Am. Chem. Soc. 60, 1940 1938.
- Fleischer, K., and F. Siebert, Ber. 53, 1255 1920.
- Harvey, J., I. M. Heilbron, and D. G. Wilkinson, J. Chem. Soc. 1930, 423.
- IIeilbron, I. M., and D. G. Wilkinson, J. Chem. Soc. 1930, 2537.
- Hugel, G., and M. Lerer, Compt. rend. 195, 249 1932.
- Kipping, F. B., and F. Wild, J. Chem. Soc. 1940, 1239.
- Krollpfeiffer, F., and W. Schäfer, Ber.
   66, 620 1923.
- Larsen, R. G., R. E. Thorpe, and F. A. Armfield, Ind. Eng. Chem. 34, 183 1942.
- 29. Lerer, M., Ann. combustibles liquides 8, 681 1933.
- 30. Levy, G., Ann. chim. [11] 9, 5 1938.
- 31. Levy, G., Compt. rend. 192, 1397 1931.
- Madinaveitia, J., Anales soc. españ. fís. quím. 32, 1100 1934; Chem. Zentr. 1935, I, 3134.
- 33. Madinaveitia, J., Rev. acad. cienc. Madrid 31, 617 1934; C.A. 29, 5438 1935.
- Mair, B. J., and A. J. Streiff, J. Research Natl. Bur. Standards 27, 343 1941.
- Mayer, F., and T. Schulte, Ber. 55, 2164
   1922.
- Mikeska, L. A., Ind. Eng. Chem. 28, 970 1936.
- 37. Müller, A., J. prakt. Chem. [2] 151, 249 1938.

- 38. Musser, D. M., and H. Adkins, J. Am. Chem. Soc. 60, 664 1938.
- Petrov, A. D., and D. N. Andreev, J. Gen. Chem. (U.S.S.R.) 12, 95 1942.
- Pokrovskaya, E. S., and R. Y. Sushchik, J. Gen. Chem. (U.S.S.R.) 11, 170 1941; C.A. 35, 7388 1941.
- 41. Pope, G. W., and M. T. Bogert, J. Org. Chem. 2, 276 1937-1938.
- Price, D., D. Davidson, and M. T. Bogert, J. Org. Chem. 2, 540 1937– 1938.
- Prokopets, E. I., J. Applied Chem. (U.S.S.R.) 10, 126 1937; Chem. Zentr. 1937, II, 1202.

- Roblin, R., Jr., D. Davidson, and M. T.
   Bogert, J. Am. Chem. Soc. 57, 151
   1935.
- Rossini, F. D., Petroleum Engineer 14,
   No. 5, 41 1943.
- 46. Ruzicka, L., and M. Mingazzini, Helv. Chim. Acta 5, 710 1922.
- 47. Schroeter, G., Ber. 54, 2242 1921.
- 48. Sengupta, S. C., J. prakt. Chem. [2] 151, 82 1938.
- Willingham, C. B., J. Research Natl. Bur. Standards 22, 321 1937.
- Ziegler, K., and P. Tiemann, Ber. 55, 3406 1922.

# 3. HIGHER BENZOCYCLANES, C<sub>n</sub>H<sub>2n-8</sub>

### C11H14

### Benzocycloheptane

(Benzosuberane)



B. P., °C @ 760mm

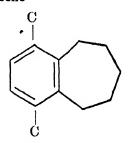
1.54856

B. 1., 0 @	
217	7641, 4
$D_4^{20}$	
0.9693	19° 4
0.9683	16.2° 2
$n_{\rm p}^{20}$	
1.5458	19° 4
1.54364	$n_{\mathrm{II}_{lpha}}^{16.2}$
1.56040	$n_{{ m H}oldsymbol{eta}}^{{f 1}{f 6},{f 2}}$
1.57108	$n_{\mathrm{H}\gamma}^{16~2~2}$

nHe 2

#### C13H18

# 2,3-Cycloheptano-1,4-dimethylbenzene



B. P., °C @ 760mm

$n_{{\scriptscriptstyle \mathrm{D}}}^{20}$	1.5344	20.5° 8
$D_4^{20}$	0.9621	20.5° s
	122	$13^{3}$

References on Higher Benzocyclanes

- 1. Borsche, W., and A. Roth, Ber. 54, 174 1921.
- 2. von Auwers, K., Ber. 60, 2122 1927.
- 3. von Braun, J., and M. Kühn, Ber. 60, 2557 1927.
- von Braun, J., and A. Stuckenschmidt, Ber. 56, 1724 1923.

# II. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA C<sub>n</sub>H<sub>2n-10</sub>

- 1. Indene and Its Alkyl Derivatives
- 2. Indane with One Alkenyl or One Alkylidene Substitution
- 3. Dihydronaphthalene and Its Alkyl Derivatives
- 4. Tetrahydronaphthalene Derivatives of Empirical Formula  $C_nH_{2n-10}$
- 5. Cyclopentanotetrahydronaphthalenes and Their Alkyl Derivatives
- 6. Octahydroanthracenes and Their Alkyl Derivatives
- 7. Octahydrophenanthrenes and Their Alkyl Derivatives
- 8. Miscellaneous Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-10</sub>

# 1. INDENE AND ITS ALKYL DERIVATIVES, $C_nH_{2n-10}$

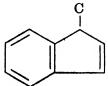
A INDEXE			EEO EE E	$6.5^{12}$
$C_9H_8$			55.0-55.5 35-37	251
Indene		$D_4^{20}$	აე–ა <i>1</i>	2
		$D_4$	0.001540	
	IJ		0.991540	
l~ " <u> </u>		1	0.99244	
			0.9968° 1.000¹	
M. P., °C			0.9232	100° 9
$-1.50 (a)^{38}$		1	0.9692	50° °
-1.678		1	0.984	25° 6
$-1.76^{14}$		1		$D_{25}^{25}$ 29, 30
$-2^{39.40}$		1	0.9906	
B. P., °C @ 760mm		1	1.0121	$D_{25}^{25}$
181		}	0.9925	$D_{20}^{20}$ 29
181-1831, 11			0.9934	$D_{20}^{20\ 30}$
$182.4^{20}$			1.006	$D_{20}^{20\ 21}$
181.8–182.3°		1	1.0152	$D_{ m 20}^{ m 20}$ $^{ m 29, \ 30}$
182.236			0.9975	18° 40
18228, 50			1.008	15° 50
181-18258			1.040	15° 17
181-181.5 <sup>52</sup>			0.9970	$D_{15}^{15\ 30}$
181.0 <sup>29</sup>		1	0.9971	$D_{15}^{15}$ 29
181 (b)			1.0002	$D_{15}^{15}$ 39
$179.5 - 180.5^{29, 30}$ $179.0 - 180.4^{26}$			1.0002	$D_{15}^{15}$ 29, 30
180.029			1.0187	$D_{15}$ $12.7^{\circ}  {}^{29}$
179.5–18056			1.0001	12.7 12° 12
179.5 <sup>28</sup>				$D_{10}^{10}$ 29, 30
178.5-179.530		1	1.0008	
$179^{27, 40}$			1.0225	$D_{10}^{10}$ 29, 30
180.5–181	76430		1.00227	8.2° <sup>29</sup> 8° <sup>40</sup>
182.2–182.4	76139		1.0060	
179.5–180.5	75717	1	1.0050	$D_5^{5-29}$
182.1-182.7	$756.4^{6}$	1	1.0268	$D_5^{5}$ 29
181-181.3	749.613	1	1.0059	4° 29, 30
83.8	30.5 <sup>8</sup>	1	1.0081	4° 13
76	2581		1.0277	4° 29, 80
68	1544		1.0152	00 9
70.8	$12^{16}$	n 20		
<b>62</b> `	1025	l.	1.5642 <sup>1</sup>	

1.57544	
$1.5764^{38}$	
1.57689	
1.5755	25° 25
1.5743	22° 57
1.5769	22° 58
1.5767	19.8° <sup>8</sup> 19° <sup>19</sup>
1.5770	
1.5773	18.5° 39
1.5763	17.6° 16
$1.5738_{7}$	16.1° 7
1.57980	15.3° 30
1.5790	15° 9
1.5804	10° 12
1.57980	8.2° 29
1.56940	$n_{{ m H}lpha}^{ m 20}$
$1.5675_{4}$	$n_{\mathrm{H}lpha}^{16.1}$
1.56454	$n_{\mathrm{H}lpha}^{\mathrm{15.6~30}}$
1.57354	$n_{\mathrm{H}lpha}^{\mathrm{15.3}}$ 80
1.56454	$n_{ m Hlpha}^{12.729}$
1.57354	$n_{\mathrm{H}lpha}^{\mathrm{8.2}}$
$1.5903_{9}$	, $n_{\mathbf{H}oldsymbol{eta}}^{161}$
1.58743	$n_{\mathbf{H}oldsymbol{eta}}^{15~6~30}$
1.59693	$n_{{ m H}oldsymbol{eta}}^{15~3~30}$
1.58743	$n_{{ m H}oldsymbol{eta}}^{12.7}$
1.59693	$n_{{ m H}oldsymbol{eta}}^{ m 8.2~29}$
1.60220	$n_{ m H\gamma}^{ m 15.6-30}$
1.61219	$n_{ m H\gamma}^{15.3}$ 30
1.60220	$n_{\mathrm{H}\gamma}^{12.7~29}$
1.61219	$n_{\mathrm{H}\gamma}^{8~2~29}$
	(c)

- (a) This constant was given as a freezing point in the literature.
- (b) The boiling point 181 is found in references 5, 10, 14, 19, 22, 29, 30, 57.
- (c) Refractive indices at other lines are found in references 29, 30.

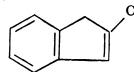
### C10H10

### 1-Methylindene



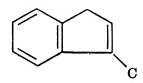
B. P., °C @ 760mm  $198-202^{43}$   $197-200^{24}$   $198.5^{54}$  70  $10^{35}$   $D_4^{20}$   $0.9640^{54}$  0.9708  $25^{\circ 35}$  $n_2^{20}$  1.5595  $25^{\circ 35}$ 

### 2-Methylindene



B. P., °C @ 760mm 184-185 74147 97 - 992415 96 - 972415 94 - 962415 62 - 652047 79 1035  $D_4^{20}$ 0.9734 19° 85 0.9897 14° 47 1.5645 23° 15 23° 15 1.5646 19° 85 1.5650 1.57574 14° 47

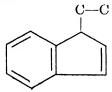
# 3-Methylindene



B. P., °C @ 760m	m
$205-206^2$ , 3, 3	3, 34, 41
91-92	$\mathbf{24^{15}}$
86	1544
81-83	$14^{15}$
77–81	$14^{15}$
81	$13^{32}$
$D_4^{20}$	
$0.975^{44}$	
0.9682	27° ³
$n_{{}^{ { m 20}}}^{ { m 20}}$	
1.56244	
1.5535	27° 15
1.5590	27° 15
1.55907	27° ³
1.5598	23° 15
1.55319	$n_{\mathrm{H}lpha}^{27}$ 3
1.57460	$n_{1\mathbf{I}\boldsymbol{\beta}}^{27-8}$
1.58865	$n_{ m H\gamma}^{27}$ $^3$

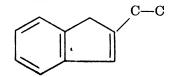
# $C_{11}H_{12}$

# 1-Ethylindene



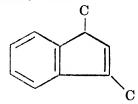
B. P., °C @ 760mm  $\begin{array}{c} 226^{45} \\ 215-216^{54} \\ 116 \\ 92 \\ 12^{35} \\ D_{4}^{20} \\ 0.9645 \\ 0.9732 \\ n_{D}^{20} \\ 1.5543 \\ 22^{\circ 35} \\ 22^{\circ 45} \end{array}$ 

# 2-Ethylindene



B. P., °C @ 760mm 96  $11^{35}$   $D_4^{20}$  0.9619  $22^{\circ 35}$   $n_p^{20}$  1.5525  $22^{\circ 35}$ 

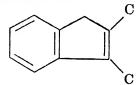
# 1,3-Dimethylindene



B. P., °C @ 760mm  $212-214^{46}$  86-88  $11^{46}$   $D_4^{20}$   $0.9553^{46}$   $n_p^{20}$ 

# 2,3-Dimethylindene

1.5344446

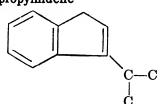


M. P., °C 1141

B. P., °C @ 760mm 111.5-112.5 1941

# $C_{12}H_{14}$

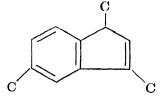
# 3-Isopropylindene



1542

B. P., °C @ 760mm 113

### 1,3,5-Trimethylindene



B. P., °C @ 760mm 103-104

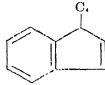
1449

 $n_{\rm D}^{20}$ 

1.547849

C13H16

#### 1-Butylindene



B. P., °C @ 760mm 252-255<sup>45</sup>

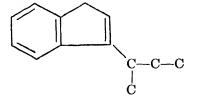
120 1045

 $D_4^{20}$ 

0.9552

23° 45

### 3-sec-Butylindene

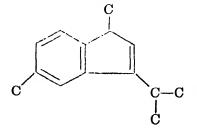


B. P., °C @ 760mm 116-118

1255

#### $C_{14}H_{18}$

# 1,5-Dimethyl-3-isopropylindene (a)



B. P., °C @ 760mm

115-120

1149

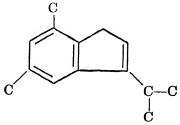
 $D_4^{20}$ 

1.5395

21° 49

(a) The above formula was given for this compound, but the name given in the literature was "1-Isopropyl-3,6-dimethylindene."

## 3-Isopropyl-5,7-dimethylindene



B. P., °C @ 760mm

125-130

 $0.2^{48}$ 

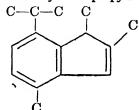
 $n_{\,{\scriptscriptstyle \mathrm{D}}}^{\,20}$ 

1.5426

18.5° 48

#### $C_{15}H_{20}$

# ${\bf 1,2,4\text{-}Trimethyl\text{-}7\text{-}isopropylindene}$



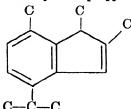
M. P., °C 99.5<sup>53</sup>

B. P., °C @ 760mm

140-145

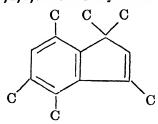
 $10^{53}$ 

## 1,2,7-Trimethyl-4-isopropylindene



B. P., °C @ 760mm  $154-157 17^{4}$   $D_{4}^{20} 0.9574 D_{25}^{25} ^{4}$   $1.5428 25^{\circ} ^{4}$ 

# 1,1,3,4,5,7-Hexamethylindene



M. P., °C 87.5–88.5<sup>37</sup>

#### References on Indene and Its Alkyl Derivatives

- Brown, R. L., Ind. Eng. Chem. 17, 920
   1925.
- 2. Brühl, J. W., Ber. 25, 142 1892.
- 3. Bruhl, J. W., Ber. 25, 151 1892.
- Carter, C. L., and S. N. Slater, J. Chem. Soc. 1938, 546.
- Charlampowicz, B., and L. Marchlewski, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1930A, 376; C.A. 25, 5096 1931; Chem. Zentr. 1931, I, 425.
- Cortese, F., Rec. trav. chim. 48, 564
   1929.
- Cotton, A., and H. Mouton, Ann. chim. phys. [8] 28, 209 1913.
- Dolliver, M. A., T. L. Gresham, G. B. Kistiakowsky, and W. E. Vaughan, J. Am. Chem. Soc. 59, 831 1937.
- Evans, E. B., J. Inst. Petroleum Tech.
   24, 537 1938.
- Fichter, F., and H. Stenzl, Helv. Chim. Acta 22, 425 1939.
- Hall, C. C., J. Soc. Chem. Ind. 54, 208 1935–1936.
- Hayashi, T., Sci. Papers Inst. Phys. Chem. Res. (Tokyo) 23, 274 1934.
- Kipping, F. S., and H. Hall, J. Chem. Soc. 77, 467 1900.

- Klatt, W., Z. physik. Chem. 171A, 454
   1934.
- Koelsch, C. F., and P. R. Johnson, J. Am. Chem. Soc. 65, 567 1943.
- Kohlrausch, K. W. F., and R. Seka, Ber.
   1551 1938.
- Kramer, G., and A. Spilker, Ber. 23, 3276 1890.
- Kraemer, G., and A. Spilker, Muspratt's "Encyclopädischen Handbuch der technischen Chemie" 4, Aufl. Bd. VIII, Braunschweig, 1905.
- 19. Krassilshik, A., Ann. combustibles liquides 10, 923 1935.
- Lecat, M., Ann. soc. sci. Bruxelles 47B, I, 108 1927.
- Lovell, W. G., J. M. Campbell, F. K. Signaigo, and T. A. Boyd, Ind. Eng. Chem. 26, 475 1934.
- Mameli, E., Gazz. chim. ital. 67, 669
   1937.
- 23. Mameli, E., and A. Mossini, Giorn. chim. ind. applicata 15, 161 1933.
- 24. Marckwald, W., Ber. 33, 1504 1900.
- Matsuno, K., and K. Han, Bull. Chem. Soc. Japan 11, 321 1936.
- McVicker, W. H, J. K. Marsh, and A. W. Stewart, J. Chem. Soc. 125, 1743 1924.
- Melamid, M., and E. Rosenthal, Z. angew. Chem. 36, 333 1923.
- Meyer, R., and W. Meyer, Ber. 51, 1571
   1918.
- Perkin, W. H , J. Chem. Soc. 69, 1025
   1896.
- Perkin, W. H., Jr., and G. Rivay, J. Chem. Soc. 65, 228 1894.
- Ramart-Lucas, and J. Hoch, Bull. soc. chim. [5] 2, 327 1935.
- Ramart-Lucas, and J. Hoch, Bull. soc. chim. [5] 5, 848 1938.
- 33. Roser, W., Ann. 247, 129 1888.
- 34. Roser, W., Ann. 247, 157 1888.
- Ruzicka, L., and E. Peyer, Helv. Chim. Acta 18, 676 1935.
- Schwarz, R., D. Pflugmacher, and M. Ruhnke, J. prakt. Chem. [2] 161, 137 1942.
- Smith, L. I., and W. W. Prichard, J. Am. Chem. Soc. 62, 771 1940.
- Smoker, E. H., and P. E. Burchfield,
   Ind. Eng. Chem. Anal. Ed., 15, 128
   1943.

- Spilker, A., and A. Dombrowsky, Ber.
   42, 572 1909.
- Stobbe, H., and E. Farber, Ber. 57, 1838
   1924.
- 41. Stoermer, R., and E. Laage, Ber. 50, 981 1917.
- Thiele, J., and K. Merck, Ann. 415, 257
   1918.
- 43. Thiele, J., and M. Rudiger, Ann. 347, 275 1906.
- 44. von Auwers, K., Ann. 415, 98 1918.
- 45. von Braun, J., Ber. 50, 1658 1917.
- von Braun, J., and G. Kirschbaum, Ber.
   46, 3041 1913.
- von Braun, J., O. Kruber, and E. Danziger, Ber. 49, 2642 1916.
- Wagner-Jauregg, T., H. Arnold, and F. Hüter, Ber. 75, 1293 1942.

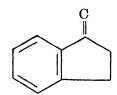
- Wagner-Jauregg, T., and H. Hippchen, Ber. 76, 694 1943.
- 50. Weger, M., Z. angew. Chem. 22, 338 1909.
- Whitby, G. S., and M. Katz, Can. J. Research 4, 344 1931.
- Whitby, G. S., and M. Katz, J. Am. Chem. Soc. 50, 1160 1928.
- Whittleston, W. G., J. Am. Chem. Soc.
   59, 825 1937.
- 54. Wislicenus, W., and W. Hentrich, Ann. 436, 9 1924.
- 55. Wuest, H.-M., Ann. 415, 291 1918.
- Zelinskii, N. D., and P. Borissov, Ber. 57, 2060 1924.
- Zelinskii, N. D., and M. B. Turova-Polyak, Ber. 62, 1658 1929.
- Zelinskiĭ, N. D., and M. B. Turova-Polyak, Ber. 62, 2865 1929.

# 2. INDANE WITH ONE ALKENYL OR ONE ALKYLIDENE SUBSTITUTION, $C_nH_{2n-10}$

### C10H10

# 1-Methyleneindane

(Dihydrobenzofulvene)



B. P., °C @ 760mm 91–93

 $17^{2}$ 

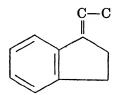
# $\boldsymbol{C_{11}H_{12}}$

# 5-Ethenylindane

B. P., °C @ 760mm 95–100

10<sup>1</sup>

### 1-Ethylideneindane

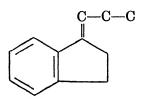


B. P., °C @ 760mm 103-105

 $17^{2}$ 

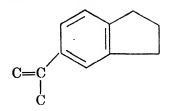
#### $C_{12}H_{14}$

# 1-Propylideneindane



B. P., °C @ 760mm • 113-115 17<sup>2</sup>

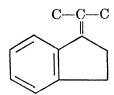
#### 5-Isopropenylindane



B. P., °C @ 760mm 84

21

### 1-Isopropylideneindane

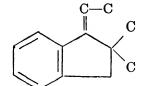


B. P., °C @ 760mm 133-135

 $17^{2}$ 

## $C_{13}H_{16}$

# 1-Ethylidene-2,2-dimethylindane

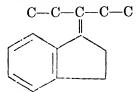


B. P., °C @ 760mm 112-114

 $13^{3}$ 

#### C14H18

# 1-(3'-Pentylidene)-indane



B. P., °C @ 760mm

136-138

 $15^{2}$ 

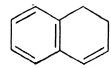
References on Indane with One Alkenyl or One Alkylidene Substitution

- Arnold, R. T., J. Am. Chem. Soc. 61, 1405 1939.
- 2. Courtot, C., Ann chim. [9] 5, 52 1916.
- Ramart-Lucas, and J. Hoch, Bull. soc. chim. [5] 5, 818 1938.

# 3. DIHYDRONAPHTHALENE AND ITS ALKYL DERIVATIVES, $C_nH_{2n-10}$

# $\mathbf{C}_{10}\mathbf{H}_{10}$

# 1,2-Dihydronaphthalene $(\Delta^{1}$ -Dialin)



M. P., °C

$$-4^{26}$$

$$-7--545$$

$$-8 - 744$$

$$-9 - 745$$

$$-9 - 855$$

B. P., °C @ 760mm

211

21210, 25

211-21229

210-2129

 $209-210^{26}$ 

206-20717

206.5-207

89–90 92.5  $\frac{24^{55}}{19^{44}}$ 

76416

84

1748

	93	1645	1.60101	$n_{\mathrm{H}oldsymbol{eta}}^{18.152}$
	89	1652	1.59040	$n_{\mathrm{H}oldsymbol{eta}}^{14.7}$
	84.5	1655	1.59189	$n_{\mathrm{H}oldsymbol{eta}}^{12.7}$
	91	1553	1.59135	n <sub>Hβ</sub> <sup>12.4</sup> 29
	90	15 <sup>45</sup>	1.61720	18.8 52
	82-83	15 <sup>31</sup>		$n_{ m H\gamma}$
	84-85	1244	1.61738	$n_{\mathbf{H}_{\gamma}}$
	83-83.5	1218	Additional Data	100
	89	1047	dD/dt = -0.0007	7473/°C
	78	952	(10 to 27°C)	
	74.0-74.5	612		
	83-84	$5^{26}$	1,4-Dihydronaphthaler	1 <b>e</b>
$D_{\bf 4}^{\bf 20}$				$\overline{}$
	0.9966			1]
	$0.9974^{55}$			IJ
	0.9904	27° 15		
	0.9926	25° 17	M. P., °C	
	0.9931	25° 16	24.5	
	0.9963	21.5° 53	2840	
	0.9976	18.3° 52	25.528	
	0.9982	18.2° 52	2552	
	0.9977	18.15° 52	24.5-25.047	
	0.9983	18.1° 52	24.5-2544	
	0.99448	14.7° 29	24.5-24.843	
	0.99688	12.7° 29	24.53, 18, 46	
	0.99745	12.4° 29	2414, 49	
	1.0031	10.5° 44	23.2419	
$n_{{\scriptscriptstyle \mathrm{D}}}^{20}$			1622	
	$1.5817^{18}$		$15.5^{1}$	
	1.5 <b>782</b>	25° 17	1527, 28	
	1.5789	25° 16	B. P., °C @ 760mm	
	1.58317	18.3° 52	21220. 83	
	1.58326	18.1° 52	208.5-20940	
	1.57399	14.7° 29	2052	
	1.57549	12.7° 29	199-20154	
	1.57494	12.4° 29	211	713 <sup>1</sup>
	1.57637	$n_{\mathrm{H}lpha}^{18352}$	94.5	178, 44
	1.57655	$n_{\mathbf{H}lpha}^{18.1\ 52}$	94	1748
	1.56827	$n_{\mathrm{H}lpha}^{\mathrm{14.7~29}}$	86.5	$12^{18}$ $9.5^{52}$
	1.56981	$n_{{ m H}lpha}^{{ m 12729}}$	84	<b>3.</b> U
	1.56910	$n_{{ m H}lpha}^{12.429}$	$D_4^{20} = 0.9928$	34.1° 52
	1.60088	$n_{\mathrm{H}oldsymbol{eta}}^{1f 8.3}$	0.9935	33.3° 52

	0.9928	32.7° 52
$n_{{\scriptscriptstyle \mathrm{D}}}^{20}$		
	$1.5593^{18}$	
	1.55474	33.3° 52
	1.55489	32.7° 53
	1.54963	$n_{ m H \ v}^{ m 33.3 \ 52}$
	1.54992	$n_{\mathrm{H}lpha}^{\mathrm{32752}}$
	1.56730	$n_{\mathrm{H}eta}^{33~3~52}$
	1.56752	$n_{\mathrm{II}_{oldsymbol{eta}}}^{32752}$
	1.57861	$n_{{ m H}\gamma}^{33\ 3\ 52}$
	1.57866	$n_{\mathrm{H}\gamma}^{32752}$

 $C_{11}H_{12}$ 

### 4-Methyl-1,2-dihydronaphthalene

B. P., °C @ 760mm 2287 107-108  $14^{32}$ 107 1441 92 - 931431 116-117 1 1 50, 51 84 58  $D_4^{20}$ 0.990141  $n_{\rm p}^{20}$ 25° 8 1.5742 21.5° 50, 51 1.5618

# 2-Methyl-x,x-dihydronaphthalene (a)

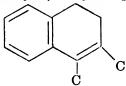
B. P., °C @ 760mm
107-108
14<sup>50</sup>, 51

1,5522<sup>50</sup>, 51

(a) The structure of this compound was not clearly defined in the literature.

#### $C_{12}H_{14}$

# 3,4-Dimethyl-1,2-dihydronaphthalene



B. P., °C @ 760mm

250-251<sup>42</sup>

114-116

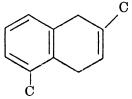
15<sup>42</sup>

0.9885

17° <sup>42</sup>

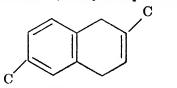
1.576342

### 2,5-Dimethyl-1,4-dihydronaphthalene



B. P., °C @ 760mm 118 10<sup>24</sup> D<sub>4</sub><sup>20</sup> 0.9700 16° <sup>24</sup>

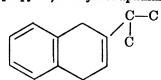
# 2,6-Dimethyl-1,4-dihydronaphthalene



B. P., °C @ 760mm 125–126 15<sup>23</sup>

# $C_{13}H_{16}$

# 2-Isopropyl-1, 4-dihydronaphthalene



B. P., °C @ 760mm 122-127

1235

3-Ethyl-4-methyl-1,2-dihydronaphthalene

B. P., °C @ 760mm 135-145

115

4-Ethyl-6-methyl-1,2-dihydronaphthelene

B. P., °C @ 760mm 135

164

4-Ethyl-8-methyl-1,2-dihydronaphthalene

B. P., °C @ 760mm 130–131

1211

1,2,4-Trimethyl-1,2-dihydronaphthalene

$$\bigcap_{C} \bigcap_{C}$$

B. P., °C @ 760mm 109

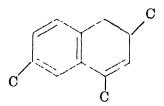
1186

1,4,6-Trimethyl-1,2-dihydronaphthalene

B. P., °C @ 760mm

135-138 122-123 1386 1234

2,4,6-Trimethyl-1,2-dihydronaphthalene



B. P., °C @ 760mm

115-117

1036

3,4,6-Trimethyl-1,2-dihydronaphthalene

B. P., °C @ 760mm

145-1506

130

1336

3,4,8-Trimethyl-1,2-dihydronaphthalene

B. P., °C @ 760mm 130-131

1187

 $D_4^{20}$ 

 $0.9760^{37}$ 

 $n_{\,\mathrm{D}}^{\,20}$ 

1.567237

4,6,8-Trimethyl-1,2-dihydronaphthalene

B. P., °C @ 760mm

1813

# $C_{14}H_{18}$

4-Methyl-6-isopropyl-1,2-dihydronaphthalene

$$C-C$$

B. P., °C @ 760mm 137

1289

### C<sub>15</sub>H<sub>20</sub>

1,4-Dimethyl-6-isopropyl-1,2-dihydronaphthaiene

$$\begin{array}{c} \\ \\ \\ \\ \\ \end{array}$$

B. P., °C @ 760mm 154–155

 $12^{38}$ 

2-Isopropyl-4,8-dimethyl-1,2-dihydronaphthalene

B. P., °C @ 760mm 108

 $0.8^{38}$ 

#### $C_{16}H_{22}$

1-Isobutyl-2,3-dimethyl-1,4-dihydronaphthalene

B. P., °C @ 760mm 150<sup>21</sup>

### C20H30

1,4-Diisobutyl-2,3-dimethyl-1,4-dihydronaphthalene

#### B. P., °C @ 760mm 180<sup>21</sup>

#### References on Dihydronaphthalene and Its Alkyl Derivatives

- Bamberger, E., and W. Lodter, Ber. 20, 1703 1887.
- 2. Berthelot, M., Ann. suppl. 5, 367 1867.
- Bonino, G. B., and P. Cella, Atti accad. Lincei [6] 15, 572 1932.
- Brunner, O., and F. Grof, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 142, Abt. IIb, 682 1933.
- Brunner, O., and F. Grof, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 142, Abt. IIb, 730 1933.
- 6. Brunner, O., H. Hofer, and R. Stein, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 142, Abt. IIb, 289 1933.
- de Pommereau, H., Compt. rend. 172, 1503 1921.
- 8. English, J., Jr., and G. Cavagliari, J. Am. Chem. Soc. 65, 1085 1943.
- 9. Graebe, C., Ber. 5, 677 1872.
- Graebe, C., and P. A. Guye, Ber. 16, 3028
   1883.
- Harvey, J., I. M. Heilbron, and D. G.
   Wilkinson, J. Chem. Soc. 1930, 423.
- 12. Hayashi, T., Sci. Papers Inst. Phys. Chem. Res. (Tokyo) 23, 274 1934.
- Heilbron, I. M., and D. G. Wilkinson, J. Chem. Soc. 1930, 2537.
- Hückel, W., and H. Bretschneider, Ann. 540, 157 1939.
- 15. Joglekar, M. S., Z. Physik 101, 398 1936.
- Kimura, S., Mem. Coll. Sci., Kyoto Imp. Univ. 14A, 173 1931; Chem. Zentr. 1931, II, 3476.

- Kimura, S., Mem. Coll. Sci., Kyoto Imp. Univ. 14, 303 1931.
- Kohlrausch, K. W. F., and R. Seka, Ber.
   1551 1938.
- Küster, F. W., Z. physik. Chem. 8, 577
   1891.
- Leonard, A. G. G., J. Chem. Soc. 97, 1246
   1910.
- 21. Lerer, M., Ann. combustibles liquides 10, 455 1935.
- 22. Mameli, E., and A. Mossini, Giorn. chim. ind. applicata 15, 161 1933.
- Mayer, F., and E. Alken, Ber. 55, 2278
   1922.
- Mayer, F., and T. Schulte, Ber. 55, 2164
   1922.
- Meyer, R., and H. Fricke, Ber. 47, 2765
   1914.
- Morton, R. A., and Λ. J. A. de Gouveia,
   J. Chem. Soc. 1934, 916.
- Padoa, M., Atti accad. Lincei [5] 28, II, 239 1919.
- 28. Padoa, M., Gazz. chim. ital. 48, II, 139 1918.
- 29. Pellini, G., Gazz. chim. ital. 31, I, 1
- 30. Pohl, J., and M. Rawicz, Z. physiol. Chem. 104, 95 1919.
- Ramart-Lucas, and J. Hoeh, Bull. soc. chim. [5] 5, 848 1938.
- Rapson, W. S., and R. Robinson, J. Chem. Soc. 1935, 1533.
- Röckemann, W., Arch. exptl. Path. Pharmakol. 92, 52 1922.
- Rupe, H., and F. Schütz, Helv. Chim. Acta 9, 992 1926.
- Ruzicka, L., and E. Capato, Ann. 453, 62 1927.
- Ruzicka, L., and L. Ehmann, Helv. Chim. Acta 15, 140 1932.
- Ruzicka, L., and J. R. Hosking, Helv. Chim. Acta 13, 1402 1930.
- Ruzicka, L., P. Pieth, T. Reichstein, and L. Ehmann, Helv. Chim. Acta 16, 268 1933.
- Ruzicka, L., and M. Stoll, Helv. Chim. Acta 5, 923 1922.
- 40. Schlenk, W., and E. Bergmann, Ann. 463, 1 1928.
- 41. Schroeter, G., Ber. 58, 713 1925.
- 42. Schroeter, G., L. Lichtenstadt, and D. Irineu, Ber. 51, 1587-1918.
- 43. Straus, F., Ber. 46, 1051 1913.

58

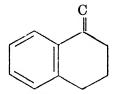
- 44. Straus, F., and L. Lemmel, Ber. 46, 232 1913.
- 45. Straus, F., and L. Lemmel, Ber. 54, 25 1921.
- Straus, F., and A. Rohrbacher, Ber.
   40 1921.
- 47. Straus through W. A. Roth and K. von Auwers, Ann. 407, 145 1914.
- 48. Stuurman, G., Proc. Acad. Sci. Amsterdam 38, 450 1935.
- Timmermans, J, Bull. soc. chim. Belg.
   44, 17 1935.

- Veselý, V., and J. Kapp, Chem. Listy
   24, 97 1930; C.A. 24, 5035 1930; Chem.
   Zentr. 1930, I, 3307.
- Veselý, V., and J. Kapp, Collection Czechoslov. Chem. Commun. 3, 448 1931.
- 52. von Auwers, K., Ber. 46, 2988 1913.
- von Braun, J., and G. Kirschbaum, Ber.
   54, 597 1921.
- 54. von Pechmann, H., Ber. 16, 516 1883.
- Willstatter, R., and V. L. King, Ber.
   46. 527 1913.

# 4. TETRAHYDRONAPHTHALENE DERIVATIVES OF EMPIRICAL FORMULA $C_nH_{2n-10}$

#### $C_{11}H_{12}$

1-Methylene-1,2,3,4-tetrahydronaphthalene



B. P., °C @ 760mm 103

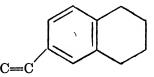
148

 $D_4^{20}$ 

0.98368

#### C12H14

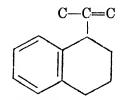
6-Ethenyl-1,2,3,4-tetrahydronaphthalene



B. P., °C @ 760mm 96

25

1-Isopropenyl-1,2,3,4-tetrahydronaphthalene



B. P., °C @ 760mm

259-261

 $745^{2}$ 

1-Isopropylidene-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm

251–253

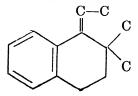
 $771^{2}$ 

245-247

 $744^{2}$ 

C14H18

1-Ethylidene-2,2-dimethyl-1,2,3,4tetrahydronaphthalene



147

#### C16H22

5-Cyclohexyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm

 $0.2^{1}$ 

x-Cyclohexyl-1,2,3,4-tetrahydronaphthalene (a)

B. P., °C @ 760mm 147-149 36 D<sub>4</sub>20

 $0.9891^{6}$  0.9939

 $D_{15}^{15}$  6

 $n_{\,{\scriptscriptstyle D}}^{\,20}$ 

1.5490<sup>6</sup> 1.5420

40° 6

(a) The structure of this compound was not clearly defined in the literature.

#### $C_{32}H_{54}$

6-(5'-Docosen-5'-yl)-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm

263-264

44

D<sub>4</sub><sup>20</sup>

0.890

25° \*

1.5048

25° \*

#### C36H62

6-(5'-Hexacosen-5'-yl)-1,2,3,4tetrahydronaphthalene

 $D_4^{20}$ 

 $0.8790^{2}$ 

 $n_{\,\mathrm{D}}^{\,20}$ 

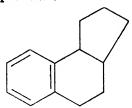
 $1.5030^{3}$ 

References on Tetrahydronaphthalene Derivatives of Empirical Formula  $C_nH_{2n-10}$ 

- Cook, J. W., and C. A. Lawrence, J. Chem. Soc. 1936, 1431.
- Kay, F. W., and A. Morton, J. Chem. Soc. 105, 1565 1914.
- 3. Mikeska, L. A., Ind. Eng. Chem. 28, 970 1936.
- Mikeska, L. A., C. F. Smith, and E. Lieber, J. Org. Chem. 2, 499 1938.
- Newman, M. S., and H. V. Zahm, J. Am. Chem. Soc. 65, 1097 1943.
- Pokrovskaya, E. S., and T. G. Stepanseva, J. Gen. Chem. (U.S.S.R.) 9, 1953 1939; C.A. 34, 4731 1940.
- Ramart-Lucas, and J. Hoch, Bull. soc. chim. [5] 5, 848 1938.
- 8. Schroeter, G., Ber. 58, 713 1925.

## 5. CYCLOPENTANOTETRAHYDRONAPHTHALENES AND THEIR ALKYL DERIVATIVES, C<sub>n</sub>H<sub>2n-10</sub>

C13H16 1,2-Cyclopentano-1,2,3,4-tetrahydronaphthalene



B. P., °C @ 760mm 266-267 739.21, 2 138 197  $D_4^{20}$  $1.0090^{2}$ 

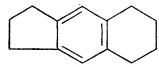
 $1.009^{1}$ 1.0054  $n_{\rm D}^{20}$ 

1.55297

1.55381, 2 19.4° 7

19.4° 7

# 6,7-Cyclopentano-1,2,3,4-tetrahydronaphthalene



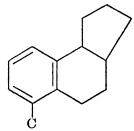
B. P., °C @ 760mm 611, 12 125-126 38 104-106

# C14H18

# 1,2-Cyclopentano-1-methyl-1,2,3,4tetrahydronaphthalene

B. P., °C @ 760mm 128  $12^{7}$ 128-132 119, 10  $D_4^{20}$ 17.8° 7 0.99798  $n_{\rm p}^{20}$ 1.54729, 10 17.8° 7 1.54790

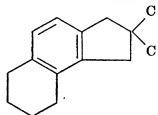
# 1,2-Cyclopentano-5-methyl-1,2,3,4tetrahydronaphthalene



B. P., °C @ 760mm 97-99 18  $n_{\rm p}^{20}$ 21°8 1.5461

# $C_{15}H_{20}$

# 2,2-Dimethyl-4,5-cyclohexanoindane



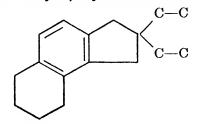
B. P., °C @ 760mm 156 185, 6  $D_{4}^{20}$ 18° 5 0.9685 $n_{\,{\scriptscriptstyle D}}^{\,20}$  $1.5320^{5}$ 

# 2,2-Dimethyl-6,7-cyclopentano-1,2, 3,4-tetrahydronaphthalene

M. P., °C 82<sup>12</sup>

#### C17H24

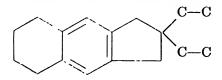
## 2,2-Diethyl-4,5-cyclohexanoindane



B. P., °C @ 760mm 163-165 164-166

 $12^{5.6}$   $11^{13}$ 

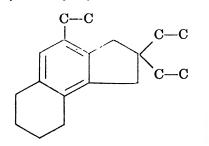
# 2,2-Diethyl-5,6-cyclohexanoindane



M. P., °C 49<sup>5</sup>, 6

#### C19H28

### 2,2,7-Triethyl-4,5-cyclohexanoindane



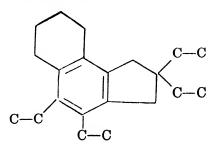
B. P., °C @ 760mm 203-205 194

 $D_4^{20}$  0.96734

 $n_{B}^{20}$  1.53524

#### $C_{21}H_{32}$

#### 2,2,4,5-Tetraethyl-6,7-cyclohexanoindane



B. P., °C @ 760mm 220-222 204

 $D_4^{20}$  0.9647 27° 4

 $n_{\rm p}^{20}$  1.5365 27° 4

#### References on Cyclopentanotetrahydronaphthalenes and Their Alkyl Derivatives

- Denisenko, Y. I., J. Gen. Chem. (U.S.S.R.) 8, 410 1938; C.A. 32, 7899 1938; Chem. Zentr. 1939, II, 630.
- Denisenko, Y. I., and A. D. Naber, Bull. acad. sci. U.R.S.S. 1937, 939; Chem. Zentr. 1939, I, 1540.
- Fieser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 59, 883 1937.
- Fleischer, K., and E. Retze, Ber. 56, 228
   1923.
- 5. Fleischer, K., and F. Siefert, Ann. 422, 272 1921.
- Fleischer, K., and F. Siefert, Ber. 53, 1255 1920.

- 7. Kon, G. A. R., J. Chem. Soc. 1933, 1081.
- 8. McQuillin, F. J., and R. Robinson, J. Chem. Soc. 1938, 1097.
- 9. Newman, M. S., priv. commun., Ohio State University, April 20, 1944.
- 10. Newman, M. S., and R. D. Closson, J. Am. Chem. Soc. 66, 1553 1944.
- 11. Sen-Gupta, S. C., Current Sci. 5,133 1936; C.A. 31, 5789 1937; Chem. Zentr. 1937, I, 76.
- 12. Sen-Gupta, S. C., J. Indian Chem. Soc. 16, 89 1939.
- 13. von Braun, J., G. Kirschbaum, and H. Schumann, Ber. 53, 1155 1920.

### 6. OCTAHYDROANTHRACENES AND THEIR ALKYL DERIVATIVES, $C_nH_{2n-10}$

#### C14H18

# 1,2,3,4,5,6,7,8-Octahydroanthracene

(Octracene)

M. P., °C

73.3

73-7416, 18

72-7412

73.54

 $73 - 73.5^{10}$ 

7315, 21

 $72.0-73.0^{18}$ 

72-731,8

71.611

7117

B. P., °C @ 760mm

 $299^{1}$ 

293-29516

160-163

167

167

 $D_{\scriptscriptstyle A}^{20}$ 

91.3° 19 0.9626 88.8° 19 0.9648 80° 21 0.9703

1.131

0° 22

188

13<sup>1</sup>

1216

 $n_{\rm D}^{20}$ 

88.8° 19 1.53625 20° 21 1.5372

 $n_{\mathrm{H}\alpha}^{88~8~19}$ 1.53230

 $n_{{
m H}m{eta}}^{88~8~19}$ 1.54791

 $n_{\rm H\gamma}^{88.8\ 19}$ 1.55788

#### 1,2,3,4,4a,9,9a,10-Octahydroanthracene

M. P., °C

715, 6, 7, 17

 $63 - 64^3$ 

 $63.5^{14}$ 

B. P., °C @ 760mm 292-2955, 6, 7

#### C15H20

1-Methyl-1,2,3,4,5,6,7,8-octahydroanthracene

M. P., °C 64-6620 2-Methyl-1,2,3,4,5,6,7,8-octahy-droanthracene

B. P., °C @ 760mm 163–165

1320

 $D_4^{20}$ 

0.9917

18° 20

#### C16H22

1,4-Dimethyl-1,2,3,4,5,6,7,8-octahydroanthracene

M. P., °C 90–92<sup>20</sup>

- 9,10-Dimethyl-x<sub>8</sub>-octahydroanthracene (a)
- M. P., °C 140–141.5°
- (a) The structure of this compound was not clearly defined in the literature.

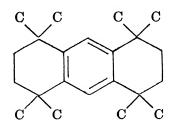
#### C18H26

1,1,4,4-Tetramethyl-1,2,3,4,5,6,7, 8-octahydroanthracene

M. P., °C 90–91²

#### C22H24

1,1,4,4,5,5,8,8-Octamethyl-1,2,3, 4,5,6,7,8-octahydroanthracene



M. P., °C 220–222²

#### References on Octahydroanthracenes and Their Alkyl Derivatives

- 1. Boedtker, E., and O. Rambech, Bull. soc. chim. [4] 35, 631 1924.
- Bruson, H. A., and J. W. Kroeger, J. Am. Chem. Soc. 62, 36 1940.
- 3. Cook, J. W., N. A. McGinnis, and S. Mitchell, J. Chem. Soc. 1944, 286.
- Fries, K., and K. Schilling, Ber. 65, 1494 1932.
- Godchot, M., Ann. chim. phys. [8] 12, 468 1907.
- 6. Godchot, M., Bull. soc. chim. [3] 31, 1339 1904.
- Godchot, M., Compt. rend. 139, 604 1904.
- Kagchira, I., Bull. Chem. Soc. Japan 6, 241 1931.
- Kursanov, D. N., and R. R. Zel'vin, Compt. rend. acad. sci. U.R.S.S. 36, 17 1942; Survey For. Petrol. Liter. Sept. 10, 1943.
- Larsen, R. G., R. E. Thorpe, and F. A. Armfield, Ind. Eng. Chem. 34, 183 1942.
- McVickers, W. II., J. K. Marsh, and A. W. Stewart, J. Chem. Soc. 127, 999 1925.
- Meerwein, H., and A Migge, Ber. 62, 1046 1929.
- Orchin, M., J. Am. Chem. Soc. 66, 535
   1944.

- 14. Prokopets, E. I., J. Applied Chem. (U.S.S.R.) 11, 835 1938.
- Prokopets, E. I., G. E. Gavrilova, and L. A. Klimova, J. Applied Chem. (U.S.S.R.) 11, 823 1938; C.A. 33, 1717 1939.
- 16. Schroeter, G., Ber. 57, 1990 1924.
- 17. Schroeter, G., Ber. 57, 2003 1924.
- 18. Schroeter, G., Ber. 60, 2035 1927.

- 19. von Auwers, K., and F. Krollpfeiffer, Ann. 430, 230 1923.
- von Braun, J., O. Bayer, and L. F. Fieser, Ann. 459, 287 1927.
- Waterman, H. I., J. J. Leendertse, and A. C. Cranendonk, Rec. trav. chim. 58, 83 1939.
- Ziegler, K., and F. Ditzel, Ann. 473, 194
   1929.

# 7. OCTAHYDROPHENANTHRENES AND THEIR ALKYL DERIVATIVES, $C_nH_{2n-10}$

#### C14H18

1,2,3,4,5,6,7,8-Octahydrophenanthrene (Octanthrene)

M. P., °C 16.7<sup>24</sup>, <sup>25</sup>, <sup>27</sup>, <sup>31</sup> 16.6<sup>3</sup> 16.5<sup>26</sup>

B. P., °C @ 760mm 295<sup>24</sup>. <sup>25</sup>. <sup>36</sup> 299 179–180

172–173 208 168 16.5<sup>26</sup> 169 15<sup>25</sup> 168–172 14<sup>25</sup> 167.5 13<sup>24</sup>, <sup>25</sup>, <sup>27</sup> 135–136 68

76627

 $20^{31}$ 

12.8° 35

 $D_4^{20}$   $1.025^{35}$   $1.026^{24}$  1.0313

# 1,2,3,4,4a,9,10,10a-Octahydrophenanthrene

M. P., °C 95 (b)<sup>5</sup>

B. P., °C @ 760mm

280-285³

283-284

744.57

282-284

741²³

146-147

20¹⁴

159

15⁴

149-150

13°

123-124

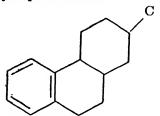
13°

135.5-135.7 (b) (c) 10.5-10.8²²

146-147	1018
138-139	1014
142.6-142.8 (a	) (b) 9.2 <sup>32</sup>
135	92
130	72
135–137	$6.5^{10}$
94-95 (b)	$1.5^{19}$
88-90 (a)	$0.1 - 0.15^{5}$
$D_4^{20}$	
1.006 (b) <sup>8</sup>	
$1.012^{23}$	-
0.997325	32° 2
0.9930	30° 2
0.9828 (b) (c)	25° 32
0.9840 (b)	25° 19
1.0053 (a) (c)	25° 32
1.0067	25° 18
1.0072 (a)	25° <sup>5</sup>
1.0148	17° 7
0.993	15° 3
1.0164 (a)	13° 5
1.006	0° s
$n_{{}^{ m p}}^{20}$	
1.559923	
1.548575	32° 2
1.54940	30° 2
1.5528	25° 9, 18
1.5528 (b)	20.7° 5
1.5549 (a)	20.7° 5
1.5527	19.2° 4
1.5569	17° 7
1.537	15° 8
1.5452 (b)	15° 19
1.5460 (b) (c)	15° 32
1.5586 (a)	12.2° 5
1.5592 (a) (c)	10.6° 82

- (a) This constant was determined on the cis isomer of the compound.
- (b) This constant was determined on the trans isomer of the compound.
- (c) This constant was determined on the dl form of the compound.

C15H20 2-Methyl-1,2,3,4,4a,9,10,10a-octahydrophenanthrene

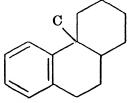


 $6^2$ 

1814

B. P., °C @ 760mm 137

# 4a-Methyl-1,2,3,4,4a,9,10,10aoctahydrophenanthrene



B. P., °C @ 760mm 155-157

98-100

1611 157 1018 145-147  $0.5 - 1.0^{15, 16}$ 

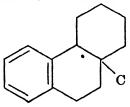
 $D_{4}^{20}$ 

25° 14 1.0025 25° 18 1.0045 18.2° 11 1.0082

 $n_{\scriptscriptstyle \mathrm{D}}^{20}$ 

1.552815, 16 25° 15, 16, 18 1.5508 25° 14 1.5559 1.55437 18.2° 11

## 10a-Methyl-1,2,3,4,4a,9,10,10aoctahydrophenanthrene



 $C_{16}H_{22}$ 

# 5,8-Dimethyl-1,2,3,4,4a,9,10,10a-octahydrophenanthrene

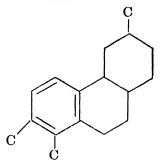
B. P., °C @ 760mm 154-156 6<sup>20</sup>  $n_p^{20}$ 1.5498 25° <sup>20</sup>

# $\mathbf{C}_{17}\mathbf{H}_{24}$

# 4a,5,8-Trimethyl-1,2,3,4,4a,9,10, 10a-octahydrophenanthrene

B. P., °C @ 760mm 155-156  $4^{20}$   $D_{4}^{20}$  0.9975  $25^{\circ 20}$  $n_{p}^{20}$  .

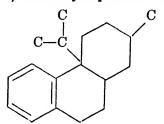
# 3,7,8-Trimethyl-1,2,3,4,4a,9,10, 10a-octahydrophenanthrene



B. P., °C @ 760mm 117-120 0.06<sup>22</sup>

#### C18H26

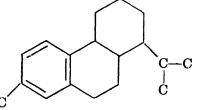
## 2-Methyl-4a-isopropyl-1,2,3,4,4a,9, 10,10a-octahydrophenanthrene



B. P., °C @ 760mm 123–127

217

# 1-Isopropyl-7-methyl-1,2,3,4,4a,9, 10,10a-octahydrophenanthrene



B. P., °C @ 760mm 195-197.5 19<sup>28</sup> 163-165 10<sup>1. 34</sup> 195-198 9<sup>6</sup>

$$D_4^{20}$$
 $0.9578^{1, 34}$ 
 $0.9647$ 
 $15^{\circ 6}$ 
 $n_p^{20}$ 
 $1.53020^{34}$ 
 $1.53023^{1, 34}$ 

#### C19H28

1,4a-Dimethyl-7-n-propyl-1,2,3,4, 4a,9,10,10a-octahydrophenanthrene (Pinabietene)

B. P., °C @ 760mm

191-194
10<sup>33</sup>
191-193
10<sup>33</sup>

D<sub>4</sub><sup>20</sup>
0.9734<sup>33</sup>
0.9740<sup>33</sup>

1,4a-Dimethyl-7-isopropyl-1,2,3,4, 4a,9,10,10a-octahydrophenan threne (Abietene)

B. P., °C @ 760mm

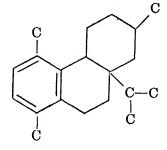
200-202 17<sup>12</sup>
200-202 14.5<sup>13</sup>
180 12<sup>29</sup>

n<sub>p</sub><sup>20</sup>
1.5354 25° <sup>29</sup>

#### C20H30

2,5,8-Trimethyl-10a-isopropyl-1,2,3, 4,4a,9,10,10a-octahydrophenanthrene

(Methyldextropimarin)



B. P., °C @ 760mm 140-145  $0.3^{21}$   $D_4^{20}$  0.9587  $22^{\circ 21}$   $n_p^{20}$  1.5301  $22^{\circ 21}$ 

1,4a,x-Trimethyl-7-isopropyl-1,2,3, 4,4a,9,10,10a-octahydrophenanthrene (a) (Methylabietene)

B. P., °C @ 760mm 194-196  $6^{80}$   $D_4^{20}$  0.9661  $24^{\circ 30}$  $n_D^{20}$  1.53836  $25^{\circ 30}$ 

Additional Data  $[\alpha]_{D}^{24} = +110.2^{\circ 30}$ 

(a) The structure of this compound was not clearly defined in the literature.

 $C_{11}H_{12}$  68

References on Octahydrophenanthrenes and Their Alkyl Derivatives

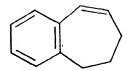
- Adelson, D. E., and M. T. Bogert, Chem. Revs. 24, 135 1939.
- Bardhan, J. C., and S. C. Sengupta, J. Chem. Soc. 1932, 2520.
- 3. Breteau, P., Compt. rend. 140, 942 1905.
- Cook, J. W., and C. L. Hewett, J. Chem. Soc. 1933, 1098.
- Cook, J. W., C. L. Hewett, and A. M. Robinson, J. Chem. Soc. 1939, 168.
- Czerny, O., Bul. Soc. Chim. România,
   7, 91 1925; C.A. 20, 1320 1926; Chem.
   Zentr. 1926, I, 2803.
- Denisenko, Y. I., and V. M. Kotel'nikova, J. Gen. Chem. (U.S.S.R.) 7, 1357 1937; Chem. Zentr. 1937, II, 2990.
- Durland, J. R., and H. Adkins, J. Am. Chem. Soc. 59, 135 1937.
- Durland, J. R., and H. Adkins, J. Am. Chem. Soc. 60, 1501 1938.
- Fulton, J. D., and R. Robinson, J. Chem. Soc. 1933, 1463.
- 11. Kon, G. A. R., J. Chem. Soc. 1933, 1081.
- 12. Levy, P., Ber. 39, 3043 1906.
- 13. Levy, P., Ber. 40, 3658 1907.
- Nenitzescu, C. D., E. Ciorănescu, and M. Maican, Ber. 74, 687 1941.
- 15. Newman, M. S., priv. commun., Ohio State University, April 20, 1944.
- Newman, M. S., and M. D. Farbman,
   J. Am. Chem. Soc. 66, 1550 1944.
- Orcutt, R. M., and M. T. Bogert, J. Org. Chem. 4, 543 1939.

- Perlman, D., D. Davidson, and M. T. Bogert, J. Org. Chem. 1, 288 1936.
- Pinkney, P. S., G. A. Nesty, D. E. Pearson, and C. S. Marvel, J. Am. Chem. Soc. 59, 2666 1937.
- Popa, D., D. Perlman, and M. T. Bogert, J. Am. Chem. Soc. 60, 319 1938.
- Ruzicka, L., and F Balas, Helv. Chim. Acta 7, 875 1924.
- Ruzicka, L., E. Rey, and W. J. Smith, Helv. Chim. Acta 26, 2057 1943.
- Schmidt, J., and R. Mezger, Ber. 40, 4240 1907.
- 24. Schroeter, G., Ber. 57, 1990 1924.
- 25. Schroeter, G., Ber. 57, 2025 1924.
- 26. Schroeter, G., Ber. 60, 2035 1927.
- 27. Schroeter, G., Brenustoff-Chem. 1, 39 1920.
- 28. Schultze, W., Ann. 359, 129 1908.
- 29 Sterling, E. C., and M. T. Bogert, J. Org. Chem. 4, 2028 1939.
- Suzuki, K., Sci. Papers Inst. Phys. Chem. Research (Tokyo) 26, 98 1935.
- van de Kamp, J., and E. Mosettig, J. Am. Chem. Soc. 57, 1107 1935.
- 32. van de Kamp, J., and E. Mosettig, J. Am. Chem. Soc. 58, 1062 1936.
- 33. Virtanen, A. I., Ann. 424, 150 1921.
- 34. Virtanen, A. I., Ber. 53, 1880 1920.
- 35. von Auwers, K., and F. Krollpfeiffer, Ann. 430, 230 1923.
- von Braun, J., and O. Bayer, Ber. 58, 2667 1925.

# 8. MISCELLANEOUS POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-10}$

 $C_{11}H_{12}$ 

1,2-Benzocycloheptene-3



B. P., °C @ 760mm 233.5–234

75711

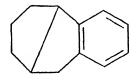
,

 $D_4^{20}$ 

1.009 4° 11

 $C_{12}H_{14}$ 

2,3-Benzobicyclo-[3,3,0]-octane



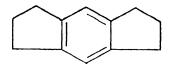
B. P., °C @ 760mm 118-124

 $\theta_3$ 

69 C12H14

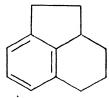
# 5,6-Cyclopentanoindane

(s-Hydrindacene)



M. P., °C 52-541

### 2a,3,4,5-Tetrahydroacenaphthene (Tetraphthene) (a)



B. P., °C @ 760mm

 $254^{16}$ 

 $252^{22}$ 

251-25219

245-24810

249.5

138 - 139248

7192, 12

 $12^{21}$ 115

 $12^{19}$ 112-115

 $D_{4}^{20}$ 

25° 10 1.0065

21°8 1.0290 15° 19 1.018

 $n_{p}^{20}$ 

1.5973719

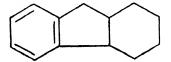
25° 10 1.5550 21°8

1.5777

(a) Reference 21 gives this name and The other references do formula. not clearly define the structure.

#### $C_{13}H_{16}$

# 1,2,3,4,4a,9a-Hexahydrofluorene



В. І	P., °C @ 760mm	
	137	154
	127	155
$D_4^{20}$		
	0.98804	
	1.019	10.9° <sup>5</sup>
$n_{{}^{\scriptscriptstyle D}}^{20}$		
	1.54484	
	1.5572	10.9° 5

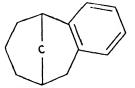
# Cycloheptano-[cd]-indane

(Homotetraphthene)



B. P., °C @ 760mm  $12^{23}$ 127-128  $D_4^{20}$ 17.1° 20 1.0364 15° 28 1.0295  $n_{\rm p}^{20}$ 22° 28 1.5750  $n_{\mathrm{H}\alpha}^{17.1~20}$ 1.57362  $n_{\mathrm{H}\beta}^{17.1~20}$ 1.593501.60645  $n_{\rm He}^{17.1~20}$ 1.57928

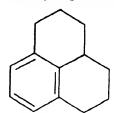
# 2,3-Benzobicyclo-[3,3,1]-nonane



B. P., °C @ 760mm

	123	155
	85	$0.3^{5}$
$D_{f 4}^{20}$ $n_{f p}^{20}$	1.020	. 14.0° •
"D	1.5580	14° 5

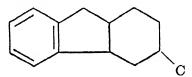
#### 3a, 4, 5, 6-Tetrahydrophenalan



B. P., °C @ 760mm 132-135  $14^{24}$   $D_{4}^{20}$   $23^{\circ 24}$  $n_{D}^{20}$  1.5636  $23^{\circ 24}$ 

#### C14H18

# 3-Methyl-1,2,3,4,4a,9a-hexahydro-fluorene



B. P., °C @ 760mm

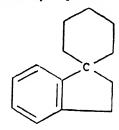
128

14<sup>26</sup>

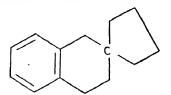
0.99<sup>26</sup>

1.5455<sup>26</sup>

# Spiro[indane-1,1'-cyclohexane]



B. P., °C @ 760mm 135.5-135.7 10<sup>14</sup>  $n_{D}^{20}$ 1.5468 • 25° <sup>14</sup> Spiro[1,2,3,4-tetrahydronaphthalene-2,1'-cyclopentane]

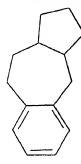


417

B. P., °C @ 760mm 113-114

1,2-Cyclopentano-4,5-benzocycloheptane

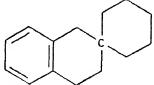
(Octahydrobenzazulene)



M. P., °C 29-30<sup>6</sup>

#### C<sub>15</sub>H<sub>20</sub>

Spiro[1,2,3,4-tetrahydronaphthalene-2,1'-cyclohexane]



B. P., °C @ 760mm

115-117

318

D<sub>4</sub><sup>20</sup>

0.98886

30.4° 18

n<sub>p</sub><sup>20</sup>

1.54314

30° 18

 $C_{16}H_{22}$ 

Spiro[7-methyl-1,2,3,4-tetrahydro-naphthalene-2,1'-cyclohexane]

B. P., °C @ 760mm 155–156

818

 $C_{17}H_{24}$ 

1-Methyl-4-isopropyl-1,2,3,4,4a,9ahexahydrofluorene

B. P., °C @ 760mm

153-155  $10^{25}$  150  $10^{15}$ 

105

0.9843

21° 15

 $0.6^{15}$ 

 $n_{\,\scriptscriptstyle
m D}^{\,\rm 20}$ 

 $D_4^{20}$ 

1.5397 21° 18

Spiro[7-ethyl-1,2,3,4-tetrahydro-naphthalene-2,1'-cyclohexane]

B. P., °C @ 760mm 168-169

20

818

 $D_{4}^{20}$ 

0.972787

24° 18

 $n_{\,{\scriptscriptstyle {
m D}}}^{\,20}$ 

1.53880818

This series continued on next page

#### C22H34

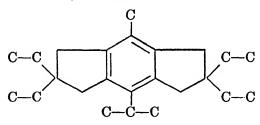
1,4-Dimethyl-2,3,5,6-di-[2',1'-(4',4'-diethylcyclopentano)]benzene

$$\begin{array}{c|c} c-c & \hline \\ c-c & \hline \\ c-c & \hline \end{array}$$

M. P., °C 69-719

#### C24H38

1-Methyl-4-isopropyl-2,3,5,6-di-[2',1'-(4',4'-diethylcyclopentano)]-benzene



M. P., °C 69–71<sup>7</sup>

B. P., °C @ 760mm 215-220

 $D_4^{20}$ 

0.9466

22° 7

127

 $n_{\scriptscriptstyle \mathrm{D}}^{20}$ 

1.52477

References on Miscellaneous Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-10}$ 

- Arnold, R. T., and R. A. Barnes, J. Am. Chem. Soc. 66, 960 1944.
- Bamberger, E., and W. Lodter, Ber. 20, 3073 1887.
- 3. Chatterjee, N. N., J. Indian Chem. Soc. 15, 211 1938.
- Cook, J. W., and C. L. Hewett, J. Chem. Soc. 1933, 1098.

- Cook, J. W., and C. L. Hewett, J. Chem. Soc. 1936, 62.
- Cook, J. W., N. A. McGinnis, and S. Mitchell, J. Chem. Soc. 1944, 286.
- 7. Fleischer, K., and W. W. Melber, Ann. 422, 231 1921.
- Fleischer, K., and F. Siefert, Ann. 422, 272 1921.
- Freund, M., K. Fleischer, and E. Gofferjé, Ann. 414, 1 1918.
- Kagehira, I., Bull. Chem. Soc. Japan
   6, 241 1931.
- Kipping, F. S., and A. E. Hunter, J. Chem. Soc. 83, 246 1903.
- Lebeau, P., and M. Picon, Compt. rend. 159, 70 1914.
- Nenitzescu, C. D., E. Ciorănescu, and M. Maican, Ber. 74, 687 1941.
- Perlman, D., D. Davidson, and M. T. Bogert, J. Org. Chem. 1, 288 1936– 1937.

- Ruzicka, L., and E. Peyer, Helv. Chim. Acta 18, 676 1935.
- Sabatier, P., and J. B. Senderns, Compt. rend. 132, 1254 1901.
- Sen-Gupta, S. C., J. Indian Chem. Soc. 11, 389 1934.
- Sen-Gupta, S. C., J. Indian Chem. Soc. 19, 467 1942.
- Spilker, A., and K. Zerbe, Z. angew. Chem. 39, 1138 1926.
- 20. von Auwers through J. von Braun and E. Rath, Ber. 60, 1182 1927.
- von Braun, J., and G. Kirschbaum, Ber.
   1680 1922.
- 22. von Braun, J., G. Kirschbaum, and H. Schumann, Ber. 53, 1155 1920.
- 23. von Braun, J., and E. Rath, Ber. 60, 1182 1927.
- von Braun, J., and J. Reutter, Ber. 59, 1922 1926.
- 25. Wallach, O., Ann. 305, 261 1899.
- 26. Wallach, O., Ber. 29, 2955 1896.

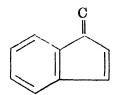
# III. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-13}$

- 1. Indene with One Alkenyl or One Alkylidene Substitution
- 2. Naphthalene and Its Alkyl Derivatives
- 3. Miscellaneous Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-12</sub>

# 1. INDENE WITH ONE ALKENYL OR ONE ALKYLIDENE SUBSTITUTION, $C_nH_{2n-12}$

#### C10H8

#### 1-Methyleneindene



M. P., °C 373, 4

B. P., °C @ 760mm 95–97

\_\_\_\_

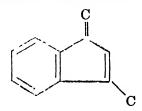
 $17^{8}$ 

## 1-Ethylideneindene

 $C_{11}H_{10}$ 

B. P., °C @ 760mm 119-121

# 1-Methylene-3-methylindene



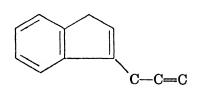
B. P., °C @ 760mm 102-104

118

173

 $C_{12}H_{12}$ 

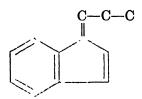
3-(Propen-2'-yl)-indene



B. P., °C @ 760mm

 $27^{7}$ 

#### 1-Propylideneindene



B. P., °C @ 760mm 140-141

 $20^{3}$ 

# 1-Isopropylideneindene

B. P., °C @ 760mm 142

129-130

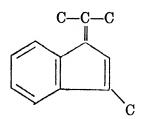
16<sup>6</sup> 8<sup>3</sup>

## C13H14

### 1-sec-Butylideneindene

B. P., °C @ 760mm 151-152 158

# 1-Isopropylidene-3-methylindene

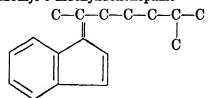


M. P., °C 51<sup>2</sup> 49<sup>1</sup> 48–49<sup>9</sup>

B. P., °C @ 760mm 145-148 11<sup>1</sup> 144-148 11<sup>9</sup> 105 0.2-0.5<sup>9</sup>

# C<sub>14</sub>H<sub>16</sub> 1-(3'-Pentylidene)-indene C-C-C-C-C B. P., °C @ 760mm 140-142 15<sup>3</sup> C<sub>17</sub>H<sub>22</sub>

2-Methyl-6-indenylideneheptane



B. P., °C @ 760mm

120 0.25  $D_4^{20}$ 0.9666 23° 5  $n_p^{20}$ 1.5733 23° 5

References on Indene with One Alkenyl or One Alkylidene Substitution

- Chatterjee, N. N., J. Indian Chem. Soc. 14, 417 1937.
- Chatterjee, N., Science and Culture 1, 584 1936; Chem. Zentr. 1936, II, 1354.
- 3. Courtot, C., Ann. chim. [9] 4, 168 1915.
- Grignard, V., and C. Courtot, Compt. rend. 160, 500 1915.
- Ruzicka, L., and E. Peyer, Helv. Chim. Acta 18, 676 1935.
- Thiele, J., and K. Merck, Ann. 415, 257 1918.
- 7. Wislicenus, W., and W. Hentrich, Ann. 436, 9 1924.
- 8. Wuest, H.-M., Ann. 415, 291 1918.
- 9. Ziegler, K., and F. Crössmann, Ann. 511, 89 1934.

# 2. NAPHTHALENE AND ITS ALKYL DERIVATIVES, $C_nH_{2n-12}$

$\mathbf{C}_{10}\mathbf{H_8}$	80.026, 115, 117, 136,	167, 194, 222	
Naphthalene	80 (e)		
	$79.9 - 80^{220}$		
	$79.5 - 80.0^{209}$		
	$79.5 - 80^{216}$		
	$79.95^{281}$		
~ ~ ~	79.914. 8		
M. P., °C	79.918, 19, 168, 210		
80.01	79.8712, 56, 287		
81 (a)	$79.866^{197}$		
$80.98 \pm 0.15  (b)^{225}$	79.8688		
80.8102, 191, 193	$79.6 - 79.8^{269}$		
80-80.7224	$79.7^{150}$		
80.6126	$79.60 - 79.64^{179}$		
80.544, 114	$79.60^{143}$		
$80.4 - 80.5^{103}$	$79.6^{70}$		
$80.0 - 80.5^{208}$	79.5 (f)		
80-80.545	$79.4^{86, \ 156}$		
80.4162, 263	$79.33^{123}$		
80.3117	$79.32^{123}$		
80.3 (c) <sup>116</sup>	79.3157, 201, 279		
$80.25 - 80.30^{267}$	$79.25^{254}$		
$80.24 \pm 0.01  (c)^{141}$	$79.2^{6, 110, 206}$		
80.2358	79 (g)	atm.	
80.21 <sup>58</sup>	190.1	$3,494 (b)^{280, 281}$	
80.20207	180.1	3,097 (b) <sup>230, 231</sup>	
80.2211, 238, 281	169.8	$2,724 \text{ (b)}^{231}$	
80.15 <sup>27, 140</sup>	170.1	$2,712 (b)^{230}$	
80.12228	159.8	$2,405 \text{ (b)}^{231}$	
80.10-80.12229	150.1	$2,061 (b)^{230}$	
80.1051, 52, 74, 231	148.9	$2,035 (b)^{231}$	
80.1 (d)	138.8	1,719231	
80.05-80.1 (c) <sup>256</sup>	138.9	$1,705 (b)^{281}$	
79.9-80.1139	129.91	$1,450 \text{ (b)}^{231}$	
80.09 <sup>268</sup>	130	1,43011	
80.0867	129.92	1,416231	
80.05-80.0717	125.01	$1,325 (b)^{231}$	
$80.061 \pm 0.002^{165}$	119.92	$1,155^{231}$	
80.0639, 187, 147	120.1	$1,151 (b)^{230}$	
80.0528, 100, 123	115.06	$1,016 (b)^{281}$	
80.04 <sup>261</sup>	110.02	. 853 <sup>281</sup>	
80.0040, 94	110.01	853231	

105.71	$740 \; (b)^{231}$	217.926	
99.98	$561^{231}$	217.8169	
100	56011	217.68100	
95. <b>2</b> 1	$429 (b)^{231}$	269.51	2,149.31
91.14	300%	267.03	2,052.2855
91.10	300 94	264.45	1,957.0255
89.11	$281 (b)^{231}$	263.01	1,912.8655
89.97	278231	259.26	1,784.8355
90.25	$275^{94}$	255.65	1,657.3955
90	$275^{11}$	253.93	$1,600.94^{55}$
89.20	25094	249.92	$1,489.60^{55}$
88.35	22594	247.58	$1,413.80^{55}$
87.45	20094	245.99	$1,378.68^{55}$
86.45	$175^{94}$	243.98	$1,317.86^{55}$
86.01	$173 (b)^{231}$	243.32	$1,306.75^{55}$
85.50	15094	241.34	1,252.2755
84.50	12594	238.79	$1,184.58^{55}$
83.51	$106 (b)^{231}$	236.25	1,111.3355
83.65	10094	231.68	1,086.6355
83	8011	234.50	1,082.1255
82.80	7594	233.94	$1,070.92^{55}$
81.80	5094	230.86	1,002.3155
80.85	2594	228.55	$958.26^{55}$
80.10	$0^{231}$	227.02	$927.75^{55}$
79.95	094	225.39	894.8755
B. P., °C @ 760m	m	224.1	$881.0^{170}$
217.98	•••	223.9	$879.7^{170}$
218.47 (b) <sup>55</sup>		223.8	$878.5^{170}$
218.288, 124		224.45	877.2955
$218.12^{246}$		221.45	$825.2^{175}$
218.0642, 55, 10	00	221.35	816.7855
218.050 (b)89		220	80038
218.05 (h)		219.95	800100
218.047		219.68	795100
$218.0 \pm 0.02$	2243	219.40	790100
218.0170, 195, 2		219.11	785100
218 (i)		218.83	780100
217.8-218.09	3	218.54	$775^{100}$
$217.98 \pm 0.0$		218.25	770100
217.973 66		217.4-217.8	769154
217.9639, 90, 1	47, 242, 249	218.5	$767.63^{54}$
$217.96 \pm 0.0$		217.97	765100
217.9573		218.5	76313
217.9441		218.152 (b)	$762.02^{89}$

78

218.047 (b)	760.3789	206.84	$590^{100}$
218.061 (b)	760.2489	207.00	589.70
217.87	759.2175	206.92	584.01 <sup>55</sup>
217.34	755.8178	206.00	$581.9^{175}$
217.40	755100	206.12	$580^{100}$
217.6	752.2054	205.78	$576.2^{175}$
217	751108	206.05	576.1855
217.6	750.188	205.40	570100
217.11	750100	205.31	564.76
216.4-216.8	747.6110	204.66	560100
216.82	745100	203.91	$550^{100}$
216.89	744155	203.15	$540^{100}$
216.7	741.9187	203.26	$538.37^{55}$
216.53	740100	202.37	530100
216.45	737.6170	202.2	$520.0^{170}$
216.4	737.6170	201.59	520100
216.7	736.9954	202.2	$519.0^{170}$
216.24	735100	202.1	$518.5^{170}$
215.95	730100	200.80	$510^{100}$
215.65	725100	200.50	$505.7^{175}$
215.7	720.3954	200	$502.1^{60}$
215.36	720100	200.00	500100
215.06	715100	200	50038
214.76	710100	199.18	$490^{100}$
215.07	708.9355	198.99	$484.84^{55}$
214.45	705100	198.53	$482.1^{175}$
214.14	700100	198.34	480100
214	70038	197.48	470100
213.51	690100	196.96	$464.1^{175}$
213.30	682.7175	196.60	460100
212.87	680100	196.85	$458.41^{55}$
212.23	670100	196.72	$458.31^{55}$
212.17	667.8175	194.78	440100
212.06	660.1655	193.85	430100
211.58	660100	192.91	420100
210.94	650100	193.05	$417.97^{55}$
210.28	640100	191.96	$410^{100}$
209.60	630100	191.00	400100
209.83	627.8555	191	40038
209.62	625.9555	191.28	398.8955
208.93	620100	191.20	$397.68^{55}$
208.24	610100	190.66	$391.40^{55}$
208	60038	190.02	390100
207.55	600100	189.02	380100

 $C_{10}H_8$  80

188.36	369.7755	115	$32.40^{10}$
187.43	358.9855	113.0	$32.2^{170}$
185.34	$347.6^{175}$	109.65	$28.7^{175}$
184.90	337.4355	110	$27.30^{10}$
184.56	$334.20^{55}$	104.39	$22.9^{175}$
183.89	$326.22^{55}$	105	$22.40^{10}$
181.20	$310.6^{175}$	102	2038
180.80	$300.82^{55}$	100	$18.50^{10}$
180	30038	100.00	$18.5^{175}$
180	299.160	100	$18.5^{8, 60}$
178.44	290.0175	98.1	18.3170
179.39	289.2255	95	$15.50^{10}$
178.92	286.6055	95.00	$15.5^{175}$
178.41	281.5555	95	$15.5^{3}$
178.0	$275.2^{170}$	90	$12.60^{10}$
177.11	$272.34^{55}$	90	$12.6^{3}$
177.3	$269.7^{170}$	87.47	$11.9^{175}$
175.10	260.8175	87.4	$11.9^{170}$
173.14	243.5355	87.2	$11.4^{170}$
172.05	235.2155	87	$10^{38}$
171.23	231.1655	85	$9.80^{10}$
166	20038	85	$9.8^{3}$
163.13	$187.1^{175}$	80	$7.40^{10}$
163.9	181.8 <sup>170</sup>	80	$7.4^{3}$
160	17213	75	$5.43^{3}$
160	168.460	74	538
160.9	$168.1^{170}$	70	438
158.95	$165.1^{175}$	70	$3.95^{3}$
149.7	$117.9^{170}$	70	$3.9^{146}$
148.7	$114.5^{170}$	65	$2.65^{3}$
147.8	$111.4^{170}$	60	1.833
144.45	$103.5^{175}$	59.76	$1.780^{238}$
144	$100^{38}$	55	$1.26^{3}$
140	88.760	55	1146
138	8038	50	$0.815^{221}$
133.72	$71.2^{175}$	50	$0.81^{3}$
132.6	$67.0^{170}$	49.94	$0.801^{238}$
131.1	$63.2^{170}$	44.86	$0.526^{238}$
130	$61.90^{10}$	45	$0.518^{221}$
130.25	$61.2^{170}$	45	$0.51^{3}$
128.06	58.7 <sup>178</sup>	40	$0.35^{146}$
$128.51 \pm 0.10$	57.78141	40	$0.320^{221}$
117.15	38.3175	39.70	$0.320^{288}$
114.70	35.9175	40	0.328

 $C_{10}H_8$ 

$\begin{array}{cccccccccccccccccccccccccccccccccccc$			0.01022	0.0639	100.86° 88
35		34.60	$0.218^{238}$	0.9632	
35					
$\begin{array}{cccccccccccccccccccccccccccccccccccc$				The state of the s	100 147
$\begin{array}{cccccccccccccccccccccccccccccccccccc$					
$\begin{array}{cccccccccccccccccccccccccccccccccccc$				1	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		30		-	99.6 200
$\begin{array}{cccccccccccccccccccccccccccccccccccc$					
$\begin{array}{cccccccccccccccccccccccccccccccccccc$					
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		24.80			
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		25			99.2° 121
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		20		l .	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		15		$0.9634_{5}$ (b)	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		20	$0.06^{153}$	0.96208	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		20	0.0545(b)	0.9621	98.4° 33
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$			10,37	1.0056	$D_{95}^{95}$ 195
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		20	$0.054^{221}$	0.9696	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$		19.86	$0.0500^{238}$	0.9685	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$			$0.047^{3}$	0.9696	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$			$0.0376(b)^{10}$	0.97056 (b)	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$				0.9715	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$				0.9757	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$				$0.9752 \pm 0.0002$	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$				0.9757	85° 255
$\begin{array}{cccccccccccccccccccccccccccccccccccc$				1	$D_{ m 85}^{ m 85}$ $^{195}$
$\begin{array}{cccccccccccccccccccccccccccccccccccc$				t e	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$				1	82.2° 219
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$					82° 147
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$			0.005		81.92° 88
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	<b>20</b>	U	0.00000	•	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$D_{4}^{\circ}$		1000	l .	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$			100	1	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$				j	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$					
$\begin{array}{cccccccccccccccccccccccccccccccccccc$				1	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$					
$\begin{array}{cccccccccccccccccccccccccccccccccccc$					
0.889     187.8° 217     0.9877     66.71° 27       0.8962     177.2° 64     0.9991     50° 68       0.912     164.8° 217     1.1675 (solid)     31.51° 27       0.9212     152° 122     1.149     25° 119       0.9400     127.0° 64     1.1789 (b)     23.9° 87       0.9456     120° 68     1.168     22° 76       1.168     22° 76					
0.8962     177.2° 64     0.9991     50° 68       0.912     164.8° 217     1.1675 (solid)     31.51° 27       0.9212     152° 122     1.149     25° 119       0.9400     127.0° 64     1.1789 (b)     23.9° 87       0.9456     120° 68     1.168     22° 75       1.168     1.168     1.168		$1.175 \text{ (solid)}^{203}$			
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		0.889	187.8° 217	1	50° 68
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		0.8962	$177.2^{\circ}$ 64		
0.9400 127.0° 64 1.1789 (b) 23.9° 87 0.9456 120° 68 1.168 22° 75		0.912	164.8° 217		
0.9456 120° 68 1.168 22° 76		0.9212	152° 122		
0.9450		0.9400	127.0° 64		23.9**
0.95040 (b) 115° 147   1.170 18° 78		0.9456	120° 68		
		0.95040 (b)	115° 147	1.170	18~ 76

1.62295

Olori	5	
	1.1589 (solid)	17° 59
	1.1708 (solid)	15.71° 27
	1.1517	15° 92
	1.15173	15° 254
	1.1788 (solid)	8.31° 27
	1.0373	Oo 68
	1.2355 (solid)	-188° 59
$n_{{\scriptscriptstyle D}}^{20}$		
	1.58214	100°
	1.618120	
	$1.619^{120}$	+ OF OO 917
	1.5330	187.8° <sup>217</sup>
	1.5467	164.8° <sup>217</sup> 124.8° <sup>217</sup>
	1.5695	124.8 <sup>217</sup>
	1.5760 1.58218	99.6° 120, 255
	1.58269	99.6° 258
	1.58291	99.5° 121
	1.58276	99.3° 121
	1.58276	$99.2^{\circ}$ 121
	1.58232	98.4° 174
	1.58987	$85.30^{\circ}$ 273
	1.58996	85.30° 261
	$1.5898 \pm 0.0002$	85° 141
	1.5917	82.8° 217
	1.601	80° 83
	1.57472	$n_{{ m H}lpha}^{99\ 6\ 120,\ 255}$
	1.57509	$n_{\mathrm{H}_{m{lpha}}}^{99}$ 6 256
	1.57535	$n_{\mathrm{H}lpha}^{99~5~121}$
	1.57524	$n_{ m Hlpha}^{99~3~121}$
	1.57520	$n_{{ m H}lpha}^{99.2\ 121}$
	1.57456	$n_{ m H_{lpha}}^{984}$ 174
	1.5746	$n_{\mathrm{H}\alpha}^{984}$ 33
	1.57375	85 255
	1.60328	$n_{ m H_{\odot}}^{ m 99~6~120,~255} \ n_{ m H_{\mathcal{B}}}^{ m 99~6~120,~255}$
		00 8 918
	1.60378	$n_{\mathrm{H}\beta}^{99}$ 5 121
	1.60401	$n_{\mathrm{H}oldsymbol{eta}}^{993}$ 121
	1.60387	$n_{{ m H}m{eta}}$
	1.60386	$n_{\mathrm{H}eta}^{99\ 2\ 121}$
	1.60310	$n_{\mathrm{H}eta}^{98}$ 4 174
	1.62261	$n_{ m H\gamma}^{99.6\ 120,\ 255}$

(j) Additional Data Crit. Temp. (°C) 47484 468.278, 85  $468.0^{265}$ Crit. Pressure (mm Hg) 29,792.085  $t_m = 79.975 + 0.03614 \ p_{alm}$ -0.000001187  $p_{atm}^2$ (1 to 3,500 atm) = 0.00326573 $-0.00042695 \log_{10} p_{mm}$ (900 to 2,150 mm) = 0.00324419 $-0.00041936 \log_{10} p_{mm}$ (300 to 900 mm) = 0.00320848 $-0.00040522 \log_{10} p_{mm}$ (35 to 300 mm) = -0.00077163 [1 + 0.001933 (t -100)]/°C (60 to 190°C) = -0.0005509/°C (80 to 190°C) (a) The melting point 81 is found in references 14, 15, 91, 112, 145, 192, 199, 262. (b) This constant is the average of two or more determinations. (c) This constant was given as a freezing point in the literature. (d) The melting point 80.1 is found in

references 65, 163, 170, 196, 219,

(e) The melting point 80 is found in references 1, 10, 16, 24, 29, 30, 31, 32, 34, 36, 46, 49, 53, 61, 72, 76,

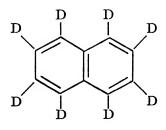
227, 231, 242, 272.

nHy 256

80, 95, 97, 101, 104, 105, 107, 109, 118, 119, 121, 122, 125, 127, 144, 149, 158, 160, 166, 172, 174, 176, 177, 180, 181, 183, 186, 200, 202, 212, 215, 218, 223, 226, 232, 233, 239, 241, 244, 245, 247, 248, 252, 257, 259, 271, 275, 276, 278, 280, 282.

- (f) The melting point 79.5 is found in references 25, 43, 158, 164, 183, 184, 213.
- (g) The melting point 79 is found in references 2, 8, 9, 20, 21, 22, 35, 47, 57, 62, 63, 69, 77, 79, 81, 82, 92, 98, 111, 135, 138, 142, 151, 152, 159, 178, 182, 185, 190, 204, 205, 234, 245, 247, 251, 253, 256, 258, 270, 274, 277.
- (h) The boiling point 218.05 is found in references 7, 128, 129, 130, 131, 132, 133, 240.
- (i) The boiling point 218 is found in references 13, 20, 21, 22, 23, 38, 57, 71, 92, 99, 101, 106, 113, 119, 134, 148, 152, 160, 161, 188, 189, 198, 235, 236, 244, 247, 248, 252, 260, 265, 269.
- (j) Refractive indices at other lines are found in reference 173.

# Octadeuteronaphthalene



M. P., °C 80–81<sup>50</sup> 77,5<sup>48</sup>

#### References on Naphthalene

- 1. Acree, S. F., Am. Chem. J. 29, 588 1903.
- 2. Acree, S. F., Ber. 37, 625 1904.
- Allen, R. W., J. Soc. Chem. Ind. 19, 209 1900.
- 4. Alluard, Ann. 113, 150 1860.
- 5. Alluard, Ann. chim. phys. [3] 57, 438 1859.
- 6. Aronheim, B., Ann. 171, 219 1874.
- 7. Aten, A. H. W., Z. physik. Chem. 78, 1 1912.
- Bamberger, E., and M. Philip, Ber. 19, 1995 1886.
- Bamberger, M., and H. von Klimberg, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 126, Abt. IIb, 325 1917.
- Barker, J. T., Z. physik. Chem. 71, 235 1910.
- Barus, C., Bull. U. S. Geol. Survey, No. 96, 1892.
- 12. Baud, E., Ann. chim. phys. [8] 27, 89 1912.
- Beckmann, E., and O. Liesche, Z. physik. Chem. 89, 111 1915.
- Behr, A., and W. A. van Dorp, Ann.
   172, 263 1874.
- 15. Behr, Λ., and W. Λ. van Dorp, Ber.6, 60 1873.
- Bell, F. K., J. Am. Chem. Soc. 47, 2811 1925.
- 17. Bennett, G. M., and W. G. Philip, J. Chem. Soc. 1928, 1937.
- Berl, E., and A. Kullmann, Ber. 60, 811 1927.
- Bernoulli, A. L., and P. Lotter, Helv. Chim. Acta 16, 246 1933.
- Berthelot, Ann. chim. phys. [4] 12, 1 1867.
- 21. Berthelot, Bull. soc. chim. [2] 7, 218 1367.
- Berthelot, Bull. soc. chim. [2] 7, 274
   1867.
- 23. Berthelot, Bull. soc. chim. [2] 8, 226
- Bhatnagar, S. S., M. R. Verma, and P. L. Kapur, Indian J. Physics 9, 131 1934-1935.
- 25. Bingham, E. C., and J. E. Hatfield, Physics 6, 64 1935.
- Blacet, F. E., P. A. Leighton, and E. P. Bartlett, J. Phys. Chem. 35, 1935 1931.

- 27. Block, H., Z. physik. Chem. 78, 385 1912.
- Bogoyavlenskii, A., Schriften Dorpater Naturfor. Ges. 13, 1 1904; Chem. Zentr. 1905, II, 945.
- Bogoyavlenskii through Tammann, G., "Krystallisieren und Schmelzen," Leipzig, 1903, p. 45.
- Bradley, M. J., and S. W. Parr, Chem.
   Met. Eng. 27, 737 1922.
- 31. Brand, K., and K. Trebing, Ber. 56, 2545 1923.
- 32. Bredt, J., and W. Posth, Ann. 285, 349 1895.
- Brühl, J. W., Z. physik. Chem 1, 307 1887.
- 34. Bruni, G., Atti accad. Lincei 13, II, 373 1904.
- Brunner, H., and R. Brandenburg, Ber. 11, 697 1878.
- 36. Büchner, E. H., Z. physik. Chem. 54, 665 1906.
- 37. Bunte, K., and H. Pippig, Gas-u. Wasserfach 66, 657 1923.
- 38. Burk, F., Gas-u. Wasserfach 67, 523 1924.
- Burriel, F., Anales soc. españ. fis. quím., 29, 89 1931; C. A. 25, 2885 1931; Chem. Zentr. 1931, I, 2732.
- Calcott, W. S., F. L. English, and F. B. Downing, Ind. Eng. Chem. 16, 27 1924.
- 41. Callendar, H. L., and E. H. Griffiths, Chem. News 63, 1 1891.
- Callendar, H. L., and E. H. Griffiths, Phil. Trans. 182A, 119 1891.
- 43. Carnelley, T., J. Chem. Soc. 37, 701 1880.
- Cauwood, J. D., and W. E. S. Turner, J. Chem. Soc. 107, 276 1915.
- Charabot, "Les Principes odorants des Végétaux," O. Doin et Fils, Editeurs, Paris, 1912, pp. 65-67.
- Chuang, C., Y. Tien, and C. Ma, Ber.
   1494 1936.
- 47. Ciamician, G., Ber. 11, 269 1878.
- 48. Clemo, G. R., and A. McQuillen, J. Chem. Soc. 1935, 1325.
- Clemo, G. R., and J. Ormston, J. Chem. Soc. 1933, 352.
- Clemo, G. R., and A. C. Robson, J. Chem. Soc. 1939, 429.
- 51. Cohen, E., W. A. T. de Meester, and

- A. L. T. Moesveld, Rec. trav. chim.42, 779 1923.
- Cohen, E., W. A. T. de Meester, and A. L. T. Moesveld, Z. physik. Chem. 108, 103 1924.
- Cosciug, T., Petroleum 31, No. 41, 5 1935.
- 54. Crafts, J. M., Bull. soc. chim. [2] 39, 277 1883.
- 55. Crafts, J. M., J. chim. phys. 11, 429 1913.
- Dahms, A., Ann. Physik [N.F.] 54, 486 1895.
- Davis, W. C., J. Soc. Chem. Ind. 33, 1120 1914.
- DeBeule, P., Bull. soc. chim. Belg. 40, 195 1931.
- 59. Dewar, J., Chem. News 91, 216 1905.
- 60. Dodge, B. F., J. Ind. Eng. Chem. 14, 569 1922.
- 61. Downs, C. R., and A. L. Dean, Ind. Eng. Chem. 6, 366 1914.
- 62. Dumas, J., Ann. 5, 5 1833.
- Dumas, J., Ann. chim phys. [2] 50, 182 1832.
- Dutoit, P., and L. Friederich, Arch. sci. phys. nat. [4] 9, 105 1900.
- Efremov, N. N., and A. M. Tikhomirova, Ann. inst. anal. phys. chim. (U.S.S.R.) 4, 92 1928; C.A. 23, 3214 1929; Chem. Zentr. 1929, 1, 745.
- Eppley, M., J. Franklin Inst. 205, 383
   1928.
- Essex, H., Z. anorg. allgem. Chem. 88, 189 1914.
- Evans, E. B., J. Inst. Petroleum Tech.
   24, 537 1938.
- Falciola, P., Gazz. chim. ital. 52, I, 175 1922.
- Ferratini, A., and F. Garelli, Gazz. chim. ital. 22, II, 215 1892.
- Ferrero, P., and J. Corbaz, Helv. Chim. Acta 13, 1009 1930.
- Fichter, F., and E. Plüss, Helv. Chim. Acta 15, 236 1932.
- Finck, J. L., and R. M. Wilhelm, J. Am. Chem. Soc. 47, 1577 1925.
- 74. Flumiani, G., Z. Elektrochem. 32, 221 1926.
- 75. Forch, C., Ann. Physik [4] 17, 1012 1905.

85 References

- Füchtbauer, C., Z. physik. Chem. 48, 549 1904.
- 77. Glaser, C., Ann. 135, 40 1865.
- Goldhammer, D. A., Z. physik. Chem.
   71, 577 1910.
- Goldschmiedt, G., and M. von Schmidt, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 83, Abt. IIb, 7 1881.
- Gomberg, M., and C. S. Schoepfle,
   J. Am. Chem. Soc. 41, 1655 1919.
- 81. Graebe, C., Ann. 163, 361 1872.
- Graebe, C., and C. Liebermann, Ann. suppl. 7, 257 1870.
- 83. Grodde, K.-H., Physik. Z. **39,** 772 1938.
- 84. Guldberg, C. M., Z. physik. Chem. 32, 116 1900.
- Guye, P. A., and E. Mallet, Compt. rend. 133, 1287 1901.
- Hatcher, W. H., and F. W. Skirrow,
   J. Am. Chem. Soc. 39, 1939 1917.
- Hendricks, S. B., and M. E. Jefferson,
   J. Optical Soc. Am. 23, 299 1933.
- 88. Heydweiller, A., Ann. Physik [N.F.]
- 61, 527 1897. 89. Holborn, L., and F. Henning, Ann.
- Physik [4] 26, 833 1908. 90. Holborn, L., and F. Henning, Ann. Physik [1] 35, 761 1911.
- 91. Homer, A., Proc. Cambridge Phil. Soc. 16, 65 1912.
- Hubbard, P., and C. N. Draper, Ind. Eng. Chem. 3, 903 1911.
- Huffman, H. M., G. S. Parks, and A. C. Daniels, J. Am. Chem. Soc. 52, 1547 1930.
- 94. Hulett, G. A., Z. physik. Chem. 28, 629 1899.
- Huntenberg, W., J. prakt. Chem. [2]
   145, 23 1936.
- 96. Ineson, W. J., Gas J. 164, 835 1923.
- 97. Innes, W. R., J. Chem. Soc. 113, 410 1918.
- 98. Ipatieff, V. N., and N. Orlov, Ber. **62**, 593 **1929**.
- 99. Jacobsen, O., Ber. 19, 1209 1886.
- 100. Jaquerod, A., and E. Wassmer, Ber. 37, 2531 1904.
- 101. Kaffer, II., Ber. 57, 1261 1924.
- 102. Kempf, R., J. prakt. Chem. [2] 78, 201 1908.

103. Kendall, J., and K. P. Monroe, J. Am. Chem. Soc. 39, 1802 1917.

- Kharasch, M. S., and A. L. Flenner,
   J. Am. Chem Soc. 54, 674 1932.
- 105. Kidd, Berzelius' Jahresber. 3, 185 1824.
- 106. Klages, A., and C. Liecke, J. prakt. Chem. [2] 61, 307 1900.
- Klemm, L., W. Klemm, and G. Schiemann, Z. physik. Chem. 165A, 379 1933.
- 108. Knoevenagel, E., Ann. 297, 113 1897.
- 109. Kollarits, M., and V. Merz, Ber. 6, 536 1873.
- 110. Kopp, II., Ann. 95, 307 1855.
- 111. Kovache, A., and E. Tricot, Chimie & industrie 13, 361, 537 1925.
- 112. Kramers, J. G., Ann. 189, 129 1877.
- 113. Krebs, O., Teer u. Bitumen 27, 254 1929.
- 114. Kremann, R., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 113, Abt. IIb, 865 1904.
- 115. Kremann, R., and E. Janetzky, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 121, Abt. IIb, 709 1912.
- 116. Kremann, R., O. Mauermann, R. Muller II, and W. Rosler, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 131, Abt. IIb, 273 1922.
- 117. Kremann, R., O. Rodinis, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 114, Abt. IIb, 1235 1905.
- 118. Krishnan, K, S., Indian J. Phys. 8, 431 1934.
- 119. Krober, E., Z. physik. Chem. 93, 641 1919.
- 120. Krollpfeiffer, F., Ann. 430, 161 1923.
- 121. Krollpfeiffer, F., Ber. 56, 77 1923.
- 122. Kurnakov, I., D. Krotkov, M. Oksmann, N. Beketov, S. Perelmutter, F. Kanov, and J. Finkel, J. Russ. Phys. Chem. Soc. 47, 558 1915; Z. anorg. allgem. Chem. 135, 81 1924.
- 123. Küster, F. W., Z. physik. Chem. 8, 577 1891.
- 124. Lagerlof, D., J. prakt. Chem. [2] 98, 136 1918.
- 125. Larsen, R. G., R. E. Thorpe, and F. A. Armfield, Ind. Eng. Chem. 34, 183 1942.

- Lassettre, E. N., and R. G. Dickinson,
   J. Am. Chem. Soc. 61, 54 1939.
- Lautsch, W., Z. physik. Chem. 1B, 115 1923.
- Lecat, M., Ann. soc. sci. Bruxelles 45, 169 1926.
- Lecat, M., Ann. soc. sci. Bruxelles 45, 284 1926.
- Lecat, M., Ann. soc. sci. Bruxelles 47,
   1927.
- Lecat, M., Ann. soc. sci. Bruvelles 48B, 113 1928.
- 132. Lecat, M., Ann. soc. sci. Bruxelles 49B, 109 1929.
- 133. Lee, H. H., and J. C. Warner, J. Am. Chem. Soc. 57, 318 1935.
- 134. Leroux, II., Compt. rend. 151, 384 1910.
- 135. Liebermann, C., and C. Riiber, Ber. 35, 2696 1902.
- 36, 2696 1902. 136. Linard, J., Bull. soc. chim. Belg. 34, 363 1925.
- 137. Lippmann, A, Z. Elektrochem. 17, 15 1911.
- 138. Lossen, W., and A. Zander, Ann. 225, 109 1834.
- 139. Mack, E , Compt. rend. 127, 361 1898.
- 140. Maillard, A., Ann. combustibles liquides 9, 1013 1934.
- 141. Mair, B. J., and A. J. Streiff, J. Research Natl. Bur. Standards 24, 395 1940.
- 142. Majima, R., and C. Kuroda, Acta Phytochim. (Japan) 1, 43 1322.
- 143. Mameli, E., and A. Mannessier, Gazz. chim atal. 43, II, 586 1913.
- 144 Mameli, E., and A. Mossini, Giorn. chim. ind applicata 15, 161 1933.
- 145. Manzoni-Ansidei, R., Gazz. chim. ital. 67, 790 1937.
- 146. Marcelin, R., Compt. rend. 158, 1419 1914.
- Marti, F. B., Bull. soc. chim. Belg.
   39, 590 1930.
- 148. Martin, W. H., and S. Lehrmann, J. Phys. Chem. 27, 558 1923.
- 149. Marvel, C. S., and A. E. Broderick, J. Am. Chem. Soc. 47, 3045 1925.
- 150. Mascarelli, L., and V. Babini, Atti accad. Lincei [5] 18, II, 222 1909.
- 151. Mascarelli, L., and U. Pestalozza, Atti accad. Lincei [5] 16, II, 567 1907.

- 152. Mascarelli, L., and U. Pestalozza, Gazz. chim. ital. 38, I, 38 1908.
- 153. Matthews, A., J. Agr. Sci. 14, 1 1924.
- McVicker, W. II., J. K. Marsh, and A. W. Stewart, J. Chem. Soc. 125, 1743 1924.
- 155. Mensching, J., and V. Meyer, Z. physik. Chem. 1, 145 1887.
- 156. Menshutkin, B. N., J. Russ. Phys. Chem. Soc. 44, 1079 1912; C.A. 6, 3403 1912.
- 157. Merz, V., Z. Chem. [N.F.] 4, 393 1868.
- 158. Merz, V., and W. Weith, Ber. 10, 746 1877.
- 159. Merz, V., and W. Weith, Ber. 14, 187 1881.
- 160. Meyer, R., Ber. 45, 1609 1912.
- 161. Meyer, R., and H. Fricke, Ber. 47, 2765 1914.
- 162. Meyer, V., and W. Riddle, Ber. 26, 2443 1893.
- 163. Michel, J, Bull. soc. chim. Belg. 48, 105 1939.
- 164. Miers, H. A., and F. Isaac, J. Chem. Soc. 93, 927 1908.
- 165. Mills, E. J., Phil. Mag. [5] 14, 1 1882.
- 166. Miolati, A., Z. physik. Chem. 9, 649 1892.
- Morris, R. E., and W. A. Cook, J. Am. Chem. Soc. 57, 2403 1935.
- 168. Mortimer, F. S., J. Am. Chem. Soc. 44, 1416 1922.
- 169. Mortimer, F. S., J. Am. Chem. Soc. 44, 1429 1922.
- 170. Mortimer, F. S., and R. V. Murphy, Ind. Eng. Chem. 15, 1140 1923.
- Mukherjee, A., Indian J. Phys. 8, 147 1933.
- 172. Nametkin, S. S., and L. M. Rosenberg, Bull. acad. sci. U.R S.S. 1943, No. 11-12, 3; Survey For. Petrol. Liter. T460, Aug. 18, 1944.
- Narasimham, K. L., Indian J. Phys.
   6, 233 1931.
- 174. Nasini, R., and O. Bernheimer, Gazz. chim. ital. 15, 59 1885.
- 175. Nelson, O., and C. Senseman, Ind. Eng. Chem. 14, 58 1922.
- 176. Nenhaus, A., Ber. 67, 1627 1934.
- Nenitzescu, C. D., and E. Cioranescu, Ber. 69, 1040 1936.
- 178. Norton, L. M., and A. A. Noyes, Am. Chem. J. 8, 362 1886.

87 References

- 179. Noyes, A. A., and C. G. Abbot, Z. physik. Chem. 23, 56 1897.
- Obrelmov, I., and A. Prikhot'ko, Physik. Z. Sowjetunion 1, 203 1932.
- Ogawa, T., and T. Yokota, Bull. Chem. Soc. Japan 5, 266 1930.
- 182. Orlov, N. A., Ber, 60, 1950 1927.
- 183. Orlov, N. A., Ber. 62, 710 1929.
- 184. Orlov, N. A., and M. A. Belopolskii, Ber. 62, 1226 1929.
- 185. Orlov, N. A., and N. D. Likhachev, Ber. 62, 719 1929.
- Orlov, N. A., and N. D. Likhachev, Ber. 63, 2179 1930.
- 187. Orndorff, W. R., and F. K. Cameron, Am. Chem. J. 17, 517 1895.
- 188. Orndorff, W. R., and F. K. Cameron, Am. Chem. J. 17, 658 1895.
- 189. Ormandy, W. R., and E. C. Craven, Chem. Trade J. 70, 41 1922.
- 190. Otto, R. and G. Mories, Ann. 147, 164 1868.
- Padoa, M., Atti accad. Lincei [5] 27,
   II, 59 1913.
- 192. Padoa, M , Atti accad. Lincei [5] 28, II, 239 1919.
- 193. Padoa, M., Gazz. chim. ital. 48, II, 139 1918.
- 194. Parks, G. S., and H. M. Huffman, Ind. Eng. Chem. 23, 1138 1931.
- 195. Perkin, W. II., J. Chem. Soc. 69, 1025 1898.
- 196. Piatti, L., Angew. Chem. 45, 719 1932.
- Pickering, S. U., Proc. Roy. Soc.
   (London) 49, 11 1890-1891.
- 198. Pieters, H. A. J., and H. van den Berge, Brennstoff-Chem. 20, 201 1939.
- 199. Pohl, J., and M. Rawicz, Z. physiol. Chem. 104, 95 1919.
- 200. Prahl, W., Angew. Chem. 52, 481 1939.
- Prins, A., Proc Acad. Sci. Amsterdam
   19, 201 1910.
- 202. Pushin, N. A., and B. Vaić, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 135, Abt. IIb, 503 1926.
- Quercigh, E., Chem. Zentr. 1920, III, 834.
- 204. Quinn, E. L., J. Am. Chem. Soc. 50, 672 1928.
- Raudnitz, H., F. Petru, and F. Haurowitz, Z. physiol. Chem. 209, 103 1932.
- 206. Reingruber, F., Ann. 206, 367 1880.

207. Reissert, A., Ber. 23, 2239 1890.

- Rheinboldt, H., J. prakt. Chem. [2]
   111, 242 1925.
- Rheinboldt, H., and M. Kircheisen, J. prakt. Chem. [2] 113, 199 1926.
- 210. Rhodes, F. II., and F. S. Eisenhauer, Ind. Eng. Chem. 19, 414 1927.
- 211. Rhodes, F. H., and F. E. Hance, J. Phys. Chem. 25, 491 1921.
- 212. Robertson, J. M., Proc. Roy. Soc. (London) 142A, 674 1933.
- 213. Roloff, M., Z. physik. Chem. 17, 325 1895.
- Rossini, F. D., Petroleum Engineer 14,
   No. 5, 41 1943.
- Rudolfi, E., Z. physik. Chem. 66, 705
   1909.
- 216. Ruzicka, L , and C. F. Seidel, Helv. Chim. Acta 19, 424 1936.
- 217. Salceanu, C., Compt. rend. 194, 863 1932.
- Sapozhnikov, A., J. Russ. Phys. Chem. Soc. 35, 1072 1903; Z. physik. Chem. 49, 688 1904.
- 219. Scheuer, O., Z. physik. Chem. 72, 513 1910.
- 220. Schiff, R., Ann. 223, 247 1884.
- 221. Schlumberger, E., J. Gasbeleucht. **55**, 1257 **1912**.
- 222. Schroder, I., Z. physik. Chem. 11, 449 1893.
- 223. Schultz, G., Ber. 9, 517 1876.
- 221. Shorygin, P., Ber. 56, 176 1923.
- 225. Soboleva, V., Z. physik. Chem. 42, 75 1902.
- 226. Speranskii, A., Z. physik. Chem. 46, 70 1903.
- 227. Speyers, C. L., J. Am. Chem. Soc. 18, 146 1896.
- 228. Sunier, A. A., J. Phys. Chem. 34, 2582 1930.
- 229. Sunier, A., and C. Rosenblum, J. Phys. Chem. 32, 1049 1928.
- Tammann, G., Ann. Physik [N.F.] 68, 553 1893.
- 231. Tammann, G., "Krystallisieren und Schmelzen," Leipzig, 1903.
- 232. Tammann, G., Nachr. Ges. Wiss. Göttingen, Jahresber. Geschäftsjahr, Math. physik. Klasse, Fachgruppen III 1913, 335.
- 233. Tammann, G., Z. anorg. allgem. Chem. 157, 321 1926.

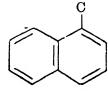
 $C_{11}H_{10}$  88

- 234. Tammann, G., and K. L. Dreyer, Z. physik. Chem. Bodenstein-Festband 1 1931.
- 235. Tausson, W. O., Planta 4, 214 1927.
- 236. Taylor, E. H., and H. S. Taylor, J. Am. Chem. Soc. 61, 503 1939.
- Terres, E., and W. Vollmer, Petroleum Z. 31, No. 19, 1 1935.
- 238. Thomas, J. S. G., J. Soc. Chem. Ind. 35, 506 1916.
- Tiffeneau, M., and A. Orékhov, Bull. soc. chim. [4] 27, 782 1920.
- 240. Timmermans, J., Bull. soc. chim. Belg. 24, 244 1910.
- 241. Timmermans, J., Bull. soc. chim. Belg. 44, 17 1935.
- 242. Timmermans, J., and F. Burriel, Compt. rend congr. chim. ind. 10, 196 1931.
- 243. Timmermans, J, and F. Martin, J. chim. phys. 23, 733 1926.
- 244. Tishenko, D., J. Russ. Phys. Chem. Soc. 62, 1407 1930; Bull. soc. chim. [4] 47, 1137 1930.
- Traube, J., A. Kieke, O. Bartsch, and K. Nishizàwa, Chem. Ztg. 48, 633 1924.
- 246. Travers and Gwyer through A. H. W. Aten, Z. physik. Chem. 78, 1 1912.
- 247. Triebs, W., Ber. 71, 612 1938.
- 248. Trusty, A. W., Petroleum Refiner 22, 95 1943.
- 249. U. S. Bur. Standards, Circ. 35, 3rd ed., July 15, 1918.
- U. S. Bur. Standards through M. Eppley, J. Franklin Inst. 205, 3831 1928.
- 251. van Bijlert, A., Z. physik. Chem. 8, 341 1891.
- 252. van Liempt, J. A. M., Rec. trav. chim. 57, 694 1938.
- 253. Versmann, F., Chem. News 30, 203 1874.
- 254. Vohl, H., J. prakt. Chem. [1] 102, 29 1867.
- 255. von Auwers, K., Ber. 54, 3188 1921.
- 256. von Auwers, K., and A. Fruhling, Ann. 422, 196 1921.
- von Auwers, K., and W. R. Innes, Z. physik. Chem. 18, 595 1895.
- 258. von Baeyer, A., and W. II. Perkin, Jr., Ber. 17, 448 1884.

- 259. von Christiani, A. F., and M. Pailer, Mikrochim. Acta 1, 26 1937.
- 260. von Steiger, A. L., Ber. 53, 666 1920.
- 261. von Steiger, A. L., Ber. 55, 1968 1922.
- 262. von Weimarn, N., Kolloid Z. 54, 296 1931.
- 263. Waidner, C. W., and G. K. Burgess, Chem. News 103, 25 1911.
- Walden, P., Z. physik. Chem. 65, 129
   1909.
- Walden, P., Z. physik. Chem. 66, 385
   1909.
- Ward, H. L., J. Phys. Chem. 30, 1316
   1926.
- 267. Ward, H., J. Phys. Chem. 38, 761 1934.
- 268. Washburn, E. W., and J. W. Read, J. Am. Chem. Soc. 41, 729 1919.
- 269. Weger, M., Z. angew. Chem. 22, 338 1909.
- Weidel, H., and G. L. Ciamician, Ber.
   13, 65 1880.
- 271. Wichelhaus, H., Ber. 36, 2942 1903.
- 272. Williams, J. W., and E. F. Ogg, J. Am. Chem. Soc. 50, 94 1928.
- 273. Willstätter, R., and F. Seitz, Ber. 56, 1388 1923.
- 274. Willstätter, R., and H. Veraguth, Ber. 40, 957 1907.
- 275. Windaus, A., and W. Thiele, Ann. 521, 160 1935.
- 276. Winterstein, A., and K. Schön, Z. physiol. Chem. 230, 146 1934.
- 277. Wolf, K. L., and H. Weghofer, Z. physik. Chem. 39B, 194 1938.
- 278. Wright, R., and N. E. Wallace, J. Chem. Soc. 1936, 1279.
- 279. Zanetti, J. E., and G. Egloff, Ind. Eng. Chem. 9, 474 1917.
- 280. Zelinskii, N., Ber. 57, 264 1924.
- Zelinskil, N., and M. Turova-Pollack, Ber. 58, 1292 1925.
- Zoppellari, I., Gazz. chim. ital. 35,
   I, 355 1905.

#### C11H10

#### 1-Methylnaphthalene



89 C<sub>11</sub>H<sub>10</sub>

	- 0a		1	1 0100 . 0 0001	0.50 41
M. I	P., ℃			$1.0163 \pm 0.0001$	25° 61
	-30.9			$1.0163 \pm 0.0002$	25° 39
	$-30.77 \pm 0.06^{19}$			1.0158	24.7° 107
	$-30.90 \pm 0.01$ (a)	61		1.0181	24.0° 112
	$-31^{25}$			1.0182	23.9° 112
	$-31.9^{19}$			1.0172	23.3° 109
	$-3331^{67}$			1.0170	22.6° 109
	$-3332^{20}$			1.0182	21.6° 109
B. F	., °C @ 760mm			1.0182	21.3° 109
	244.74	;		1.0190	20.9° 109
	$244.8^{80}$			1.0188	$20.6^{\circ}$ 109
	244.7819			1.0192	$20.5^{\circ}$ 109
	$244.78 \pm 0.10^{61}$			1.0192	19.8° 111
	244.655			1.0195	19.8° 109
	$244.4^{25}$			1.0209	$19.7^{\circ}$ 112
	$243 - 244^{16}$			1.0211	19.7° 112
	$242.5 - 243.5^{40}$			1.0218	$19.3^{\circ}$ 112
	$240.4 - 242.6^{35}$			1.0219	$19.2^{\circ}$ 112
	241-24266, 110			1.0218	$18.9^{\circ}$ 112
	244-245	76620		1.02085	$18.2^{\circ}$ 109
	243.5-244.1	76365		1.0216	18.2° 112
	241.5-242	74977		1.0223	18.2° 112
	$151.3 \pm 0.2$	57.861		1.0217	17.8° 109
	148.9	5619		1.0229	$17.6^{\circ}$ 112
	145	$55^2$		1.0212	17.0° 111
	121-123	20108		1.0234	17° 41
	117-120	12-1356		1.0225	$16.8^{\circ}$ 112
	108.6–108.8	1135		1.0250	14.4° 112
	108-109	1041		1.0248	13.6° 110
	100-105	10 <sup>1</sup>		1.0259	$13.6^{\circ}$ 112
	14.6	$0.031^{60}$		1.0267	11.5° 118
$D_4^{20}$	11.0	0.001		1.0347	0° 25
$D_4$	1.0120		$n_{{\scriptscriptstyle m D}}^{20}$		
	1.0120		$n_{\scriptscriptstyle \mathrm{D}}$	$1.6157^{20}$	
	1.02025			1.616777	
	1.0212112	100.2° 109		$1.6172^{25}$	
	0.9606	100.2 100 100.1° 112		1.58077	99.2° 110
	0.9644	100.1 112 100.0° 109		1.5882	40° 116
	0.9604	100.0° 10° 10° 10° 10° 10° 10° 10° 10° 10° 1		$1.6140 \pm 0.0002$	25° 39
	0.9611	100° 23° 100.0° 112		$1.6140 \pm 0.0002$ $1.6145$	25° 25
	0.9632	99.5° 109		$1.61494 \pm 0.00010$	
	0.9606	99.5° 10° 99.4° 10°			18.5° 74
	0.9617			1.6165	18.5 A
	0.9771	78.3° 112		1.6175	17° 4° 15° 25
	0.9979	50° 25		1.6197	19

 $C_{11}H_{10}$  90

1.62124	13.6° 110	1.61051	$n_{{ m H}_{lpha}}^{16.7~112}$
1.57226	$n_{{ m H}lpha}^{100.2\ 109}$	1.61126	$n_{{ m H}lpha}^{15.0\ 112}$
1.57225	$n_{{ m H}lpha}^{100.1\ 112}$	1.61130	$n_{{ m H}lpha}^{14.9\ 112}$
1.57226	$n_{{ m H}lpha}^{100.0\ 109}$	1.61121	$n_{\mathrm{H}lpha}^{\mathrm{14.4\ 112}}$
1.57252	$n_{{ m H}lpha}^{{ m 100.0~112}}$	1.61174	$n_{\mathrm{H}lpha}^{\mathrm{13.6\ 112}}$
1.57244	n <sub>H\alpha</sub> <sup>99 5 109</sup>	1.61300	$n_{\mathrm{H}oldsymbol{lpha}}^{13.6}$ 110
1.57235	$n_{{ m II}lpha}^{99.4\ 109}$	1.60025	$n_{{ m H}m{eta}}^{100.2\ 109}$
1.57353	$n_{{ m H}lpha}^{99.2\ 110}$	1.60018	$n_{{ m H}m{eta}}^{{ m 100.1~112}}$
1.58292	$n_{{ m H}lpha}^{783112}$	1.60034	$n_{{ m H}m{eta}}^{100.0~109}$
$1.60677 \pm 0.00010$		1.60043	$n_{{ m H}m{eta}}^{99.5~109}$
1.60687	$n_{{ m H}lpha}^{24.7\ 107}$	1.60285	$n_{{ m H}m{eta}}^{99.4~109}$
1.60750	$n_{{ m H}lpha}^{24.0\ 112}$	1.60159	$n_{{ m H}m{eta}}^{99.2~110}$
1.60754	$n_{{ m H}_{lpha}}^{23.9\ 112}$	$1.63672 \pm 0.00010$	$n_{\mathrm{H}oldsymbol{eta}}^{25}$
1.60786	$n_{{ m II}lpha}^{23.3\ 109}$	1.63680	$n_{\mathrm{H}oldsymbol{eta}}^{24.7~107}$
1.60811	$n_{{ m H}lpha}^{22.6\ 109}$	1.63740	$n_{{ m H}m{eta}}^{24.0~112}$
1.60846	$n_{{ m H}lpha}^{21.6\ 109}$	1.63744	$n_{{ m H}m{eta}}^{23.9~112}$
1.60846	$n_{{ m H}lpha}^{21.3\ 109}$	1.63776	$n_{\mathrm{H}oldsymbol{eta}}^{23.3}$ 109
1.59156	$n_{{ m H}lpha}^{20~9~109}$	1.63800	$n_{{ m H}m{eta}}^{22.6~109}$
1.60881	$n_{{ m H}lpha}^{20.6\ 109}$	1.63844	$n_{{ m H}m{eta}}^{21.6~109}$
1.60909	$n_{{ m H}lpha}^{20.5\ 109}$	1.63844	$n_{{ m H}m{eta}}^{21.8~109}$
1.60900	$n_{{ m H}lpha}^{20.0\ 112}$	1.61969	$n_{{ m H}m{eta}}^{20.9~109}$
1.60919	$n_{{ m H}lpha}^{19.8\ 111}$	1.63878	$n_{{ m H}m{eta}}^{20.6~109}$
1.60927	$n_{{ m H}lpha}^{19.8\ 109}$	1.63905	$n_{\mathrm{H}oldsymbol{eta}}^{20.5~109}$
1.60909	$n_{{ m H}lpha}^{19.7\ 112}$	1.63895	$n_{{ m H}m{eta}}^{20.0~112}$
1.60944	$n_{{ m H}lpha}^{19.7\ 112}$	1.63931	$n_{{ m H}m{eta}}^{19.8\ 109,\ 111}$
1.60932	$n_{{ m H}lpha}^{19.3\ 112}$	1.63913	$n_{\mathrm{H}oldsymbol{eta}}^{\mathrm{19.7}}$
1.60936	$n_{{ m H}lpha}^{19.2\ 112}$	1.63947	$n_{{ m H}eta}^{19.7112}$
1.60980	$n_{{ m H}\alpha}^{18.9\ 112}$	1.63935	$n_{\mathrm{H}oldsymbol{eta}}^{19.8}$
1.60980	$n_{{ m H}lpha}^{18.2\ 112}$	1.63939	$n_{{ m H}m{eta}}^{19.2\ 112}$
1.60989	$n_{{ m H}lpha}^{18.2\ 112}$	1.63983	$n_{\mathrm{H}oldsymbol{eta}}^{18.9\ 112}$
1.61007	$n_{{ m H}\alpha}^{18.2\ 109}$	1.64000	$n_{{ m H}m{eta}}^{18.2\ 112}$
1.61051	$n_{{ m H}lpha}^{17.8\ 109}$	1.64009	$n_{{ m H}m{eta}}^{18.2\ 109}$
1.61015	$n_{{ m H}lpha}^{17.6\ 112}$	1.64052	$n_{\mathbf{H}\boldsymbol{\beta}}^{17.8 \ 109}$
1.60997	$n_{{ m H}lpha}^{17.2\ 112}$	1.64026	$n_{{ m H}m{eta}}^{17.6\ 112}$
1.61052	17 0 111	1.64008	$n_{\mathrm{H}\beta}^{17.2\ 112}$
1.61042	$n_{\mathrm{H}\alpha}$ $n^{16.9 \ 112}$	1.64060	$n_{{ m H}m{eta}}^{17.0~111}$
	$n_{\mathrm{H}\alpha}$	1.64050	$n_{{ m H}m{eta}}^{16.9}$
1.61051	$n_{\mathrm{H}\alpha}^{16.8 \ 112}$	1.64068	$n_{\mathrm{H}oldsymbol{eta}}^{16.8}$

91  $C_{11}H_{10}$ 

1.64050	$n_{\mathrm{H}oldsymbol{eta}}^{16.7}$
1.64143	$n_{\mathrm{H}oldsymbol{eta}}^{\mathrm{15.0}}$
1.64147	$n_{\mathrm{H}oldsymbol{eta}}^{14.9}$
1.64146	$n_{\mathrm{H}oldsymbol{eta}}^{14.4}$
1.64214	$n_{{ m H}m{eta}}^{13.6112}$
1.64329	$n_{\mathrm{H}oldsymbol{eta}}^{13.6}$
1.62067	$n_{ m H_{m \gamma}}^{ m 99.2\ 110}$
1.65852	$n_{\rm H\gamma}^{21.6~109}$
1.65852	$n_{ m H\gamma}^{21\ 3\ 109}$
1.66024	$n_{ m H\gamma}^{18.2\ 112}$
1.66106	$n_{ m H\gamma}^{168112}$
1.66372	$n_{ m H\gamma}^{13.6\ 110}$
1.5800	$n_{ m He}^{100.2\ 109}$
1.57990	$n_{ m He}^{ m 100.1~112}$
1.5801	$n_{ m He}^{ m 100.0~109}$
1.58027	$n_{ m He}^{100.0~112}$
1.5803	$n_{ m He}^{99\ 5\ 109}$
1.5802	$n_{ m He}^{99~4~109}$
1.59095	$n_{ m He}^{78.3~112}$
1.61527	$n_{ m He}^{24~7~107}$
1.61589	$n_{ m He}^{24~0~112}$
1.61593	$n_{ m He}^{23~9~112}$
1.6162	$n_{ m He}^{23.3~109}$
1.6165	$n_{ m He}^{22.6\ 109}$
1.6168	$n_{\rm He}^{21.6 10 \circ}$
1.6168	70 1 109
1.6172	n <sub>He</sub>
1.6172	n <sub>He</sub>
1.6174	$n_{ m He}$
1.61738	711e
1.61757	$n_{\mathrm{Ho}}$
1.6176	n <sub>He</sub>
1.61747	$n_{\mathrm{He}}$
1.61782 1.61770	n <sub>He</sub>
	n <sub>He</sub>
1.61774 1.61818	n <sub>He</sub> 18.9 112
1.61836	nHe 18.2 112
1.01990	$n_{\mathrm{He}}^{\mathrm{16.2}}$

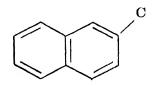
1.61837	$n_{ m He}^{ m 18.2\ 112}$
1.61844	$n_{ m He}^{18.2\ 109}$
1.6188	$n_{ m He}^{17.8~109}$
$\boldsymbol{1.61862}$	$n_{ m He}^{17.6~112}$
1.61844	$n_{ m IIe}^{17.2\ 112}$
1.61898	$n_{ m He}^{ m 17.0~111}$
1.61879	$n_{ m He}^{16.9\ 112}$
1.61907	$n_{ m He}^{16.8112}$
1.61888	$n_{ m He}^{16\ 7\ 112}$
$\boldsymbol{1.61972}$	$n_{ m He}^{ m 15~0~112}$
1.61976	$n_{ m He}^{14.9~112}$
1.61985	$n_{\mathrm{He}}^{\mathrm{14.4~112}}$
1.62038	$n_{ m He}^{13.6~112}$

Additional Data  $dD/dt = -0.0007437/^{\circ}\text{C}$ 

(0 to 50°C)

(a) This constant was given as a freezing point in the literature.

## 2-Methylnaphthalene



M. P., °C  $34.4_4$   $36.6-37.2^{35}$   $37^{27.36}$   $35.5^{40}$   $35.1^{116}$   $35^{38.76}$   $34.5^{25.54}$ , 71  $34.44 \pm 0.01$  (a) <sup>61</sup>  $34.1^{43.75}$   $34^{23.65}$   $32.5-33^{66}$   $32-33^{73.106}$ 

32.556, 96, 98, 115, 119

31.5–32.5<sup>110</sup> 32 (b)

В. Р.	., °C @ 760mm	
	241.14	
	241-24374	
	2426	
	241-24229, 98, 115	
	240-2425, 74	
	241.1419	
	$241.14 \pm 0.05^{61}$	
	241.1 <sup>80</sup>	
	240-241121	77069
	240-242	770°° 759¹¹°
	241-242	
	$148.07 \pm 0.05$	$57.78^{61}$ $50^{38}$
	147	
	110-112	163
•0	110.2-111.4	$13^{35}$
$D_{f 4}^{f 20}$		
	0.883	174° 93
	0.891	170° 37
	0.912	142.5° 93
	0.930	120° 37
	0.932	111.7° 93
	0.9491	$99.4^{\circ}$ 110
	0.961	80° 37
	0.9718	70° 25
	0.985	50° 37
	$0.99045\pm0.00002$	40° 61
	0.9939	39.9° 110
	1.0043	$22^{\circ}$ 78
$n_{{ m D}}^{20}$		
	1.5286	174° 93
	1.551	142.5° 93
	1.569₅	111.7° 93
	1.5750	99.8° 94
	1.57426	$99.4^{\circ}$ 110
	$1.586_{4}$	$78.5^{\circ}$ 93
	1.5864	78.3° 94
	1.5965	$50^{\circ}$ 25
	$1.599_{7}$	49° 93
	1.5997	48.8° 94
	1.6015	$40^{\circ}$ 25
	$1.60192\pm0.00010$	40° 61
	1.6028	40° 116
	1.60263	$39.9^{\circ}$ 110

$1.605_{6}$	34° 93
1.6056	33.8° 94
1.56724	$n_{\mathrm{H}\alpha}^{99.4~110}$
$1.59407\pm0.00010$	$n_{\mathrm{H}lpha}^{\mathrm{40-61}}$
1.59499	$n_{\mathrm{H}lpha}^{39.9~110}$
1.59454	$n_{\mathrm{H}oldsymbol{eta}}^{99.4~110}$
$1.62306\pm0.00010$	$n_{\mathrm{H}oldsymbol{eta}}^{\mathrm{40}}$ 61
1.62395	$n_{\mathrm{H}oldsymbol{eta}}^{39.9~110}$
1.64284	$n_{ m H_{oldsymbol{\gamma}}}^{ m 39.9~110}$

- (a) This constant was given as a freezing point in the literature.
- (b) The melting point 32 is found in references 5, 15, 16, 63, 97, 103, 118.

# $C_{12} H_{12} \\$

# 1-Ethylnaphthalene

 $\begin{array}{c} \text{B. P., °C @ 760mm} \\ 258-259^{45} \\ 257-259^{120} \\ 256-259^{32} \\ 257-259.5 \\ 757.7^{12} \\ 256.5 \\ 247-249 \\ 112-116 \\ 100 \\ 2-3^{12} \\ D_4^{20} \end{array}$ 

 $D_4^{20}$   $1.0123^{31}$  1.0111 1.0123 1.0123 1.0204 1.0221 1.0221 1.0221 1.0221 1.0221 1.0221 1.0221

14.2° 59

 $1.6075^{31}$  1.6089

 $\mathbf{C}_{12}\mathbf{H}_{12}$ 

#### 2-Ethylnaphthalene

B. P., °C @ 760mm 251.5 252<sup>52, 58</sup> 251-252<sup>28, 114</sup>

250-252<sup>14</sup> 251<sup>45, 62</sup> 250-251<sup>9</sup> 249-254

 $D_4^{20}$ 

0.9958

 $D_0^{25 72}$   $D_0^{0 62}$   $D_0^{0 57, 58}$ 

15° 57, 58

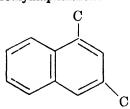
 $n_{\rm p}^{20}$ 1.591
25° 72
1.6028
15° 57. 58

(a) Refractive indices at other lines are found in reference 57.

# 1,2-Dimethylnaphthalene

B. P., °C @ 760mm	
$266-267^{51}$	
265	77022
<b>2</b> 65– <b>2</b> 66	75251
264	74930
144	22110
148-149	1895
139-140	$15^{95}$
135–136	1442
137	1364
131.5	$12^{30}$
120	$0.02^{82}$
$D_4^{20}$	
$1.011^{64}$	
$1.015^{51}$	
$1.019^{30}$	
1.0118	19.0° 64
1.025	16.6° 95
1.0219	16.35° 110
$n_{\rm p}^{20}$	
1.613551	
$1.6142^{64}$	
1.61461	19° 64
1.61795	16.35° 110
1.60691	$n_{\mathrm{H}\alpha}^{19}$
1.61014	$n_{{ m H}lpha}^{1 m 6.35110}$
1.63613	$n_{\mathrm{H}oldsymbol{eta}}^{\mathrm{19}}$ 64
1.63945	$n_{11\beta}^{16.35\ 110}$
1.65976	$n_{ m H\gamma}^{16.35\ 110}$

# 1,3-Dimethylnaphthalene



B. P., °C @ 760mm 248-250<sup>41</sup> 262-264 746<sup>102</sup> 124-126 10<sup>41</sup> 107 1<sup>4</sup>

#### $C_{12}H_{12}$

$D_{4}^{20}$	1.01997	15° 41
$n_{{\scriptscriptstyle \mathrm{D}}}^{20}$	1.6130	15° 41

#### 1,4-Dimethylnaphthalene

M. P., °C 15-17<sup>102, 105</sup> 5.5-6.5<sup>79</sup> B. P., °C @ 760mm

> 264-266<sup>66</sup> 265<sup>68</sup> 264<sup>4</sup>

262-264  $751^{11}$  262.5-263  $742^{10}$  145  $40^{68}$  118  $10^{79}$  110  $6^{34}$ 

 $D_4^{20}$ 

1.0157 1.015734 0.97411 77.7° 68 27.7° 68 1.01058 27.7° 7,8 1.0106  $D_{20}^{20}$  34 1.0176 1.01803 16.4° 68 12° 11 1.0199 0° 11 1.0283

 $n_{\,{\scriptscriptstyle D}}^{\,20}$ 

 1.58656
 77.7° 68

 1.61052
 27.7° 68

 1.61567
 16.4° 68

1.01507 10.4  $^{\circ}$ 1.57901  $n_{\text{H}\alpha}^{77.7 \text{ 68}}$ 1.60250  $n_{\text{H}\alpha}^{27.7 \text{ 68}}$ 

1.6025  $n_{\text{H}\alpha}^{27.7 \text{ 8}}$ 

 1.60765
  $n_{H\alpha}^{16.4}$  68

 1.60710
  $n_{H\beta}^{77.7}$  68

 1.62200
  $n_{H\beta}^{27.7}$  68

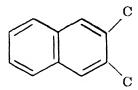
 1.63722
  $n_{H\beta}^{16.4}$  68

 1.65117
  $n_{H\gamma}^{27.7}$  68

Additional Data

 $dD/dt = -0.00072025/^{\circ}C$ (0 to 78°C)

## 2,3-Dimethylnaphthalene



M. P., °C

104 104–105<sup>117</sup>

104-104.595

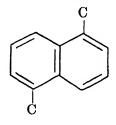
1044, 122

 $102^{\scriptscriptstyle 100}$ 

B. P., °C @ 760mm 265-266

76748

# 1,5-Dimethylnaphthalene



M. P., °C

8249, 50

80-80.51

77-78104

B. P., °C @ 760mm

265.549

265-265.5 76150

95  $C_{12}H_{12}$ 

#### 1,6-Dimethylnaphthalene

M. P., °C 97 99

B. P., °C @ 760mm 262-264<sup>69</sup> 262-263<sup>47, 118</sup> 262.5<sup>13</sup>

 $D_4^{20}$ 

 $n_{\rm p}^{20}$ 

1.0049 16.3° 110 1.0056 15° 118 1.60886 16.3° 110

 $\begin{array}{lll} 1.60119 & n_{\mathrm{H}\alpha}^{16.3 \ 110} \\ 1.63021 & n_{\mathrm{H}\beta}^{16.3 \ 110} \\ 1.64958 & n_{\mathrm{H}\gamma}^{16.3 \ 110} \end{array}$ 

# 1,7-Dimethylnaphthalene

M. P., °C 84-85<sup>101</sup>

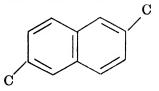
B. P., °C @ 760mm 261-262<sup>52</sup> 258<sup>4</sup> 147-149 15<sup>21</sup>

 $D_4^{20}$ 

1.011552

 $n_{\rm d}^{20}$  1.60831<sup>52</sup>

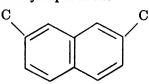
#### 2,6-Dimethylnaphthalene



M. P., °C
110.5
113<sup>69</sup>
111<sup>17</sup>
110-111<sup>17</sup>, <sup>44</sup>, <sup>63</sup>, <sup>113</sup>, <sup>118</sup>
110.5<sup>40</sup>
110<sup>13</sup>, <sup>23</sup>, <sup>53</sup>
109<sup>88</sup>

B. P., °C @ 760mm 261–263<sup>44</sup> 261–262<sup>69</sup> 260–261 767<sup>48</sup> 261–262 762<sup>118</sup>

# 2,7-Dimethylnaphthalene



M. P., °C

97

98.5<sup>83</sup>

98<sup>86</sup>

97-98<sup>90</sup>

96.5-97.5<sup>70</sup>

97<sup>46</sup>, <sup>83</sup>, <sup>89</sup>, <sup>97</sup>

96-97<sup>83</sup>, <sup>91</sup>, <sup>92</sup>, <sup>118</sup>

96.5<sup>24</sup>

96<sup>23</sup>, <sup>84</sup>

94-95<sup>87</sup>

94<sup>69</sup>, <sup>85</sup>

B. P., °C @ 760mm

B. P., °C @ 760mm

265<sup>18</sup>

262

758<sup>118</sup>

128-130

12<sup>85</sup>

# References on C<sub>11</sub>H<sub>10</sub> through C<sub>12</sub>H<sub>12</sub> Compounds

- Anderson, A. R., and W. F. Short, J. Chem. Soc. 1933, 485.
- Arbusov, B. A., and E. V. Kuznetsov, Compt. rend. acad. sci. U.R.S.S.
   39, 343 1943; Survey For. Petrol. Liter. Feb. 25, 1944.
- 3. Barbot, A., Bull. soc. chim. [4] 47, 1314 1930.
- Barnett, E. de B., and F. G. Sanders,
   J. Chem. Soc. 1933, 434.
- 5. Bodroux, F., Bull. soc. chim. [3] 25, 491 1901.
- Bogert, M. T., and V. G. Fourman, J. Am. Chem. Soc. 55, 4670 1933.
- 7. Brühl, J. W., Ann. 235, 1 1886.
- 8. Brühl, J. W., Z. physik. Chem. 1, 307 1887.
- 9. Brunel, O., Ber. 17, 1179 1834.
- 10. Cannizzaro, S., and A. Andreocci, Gazz. chim. ital. 26, I, 13 1896.
- 11. Cannizzaro, S., and G. Carnelutti, Gazz. chim. ital. 12, 393 1882.
- 12. Carnelutti, G., Ber. 13, 1671 1880.
- Charlampowicz, B., and L. Marchlewski, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1930A, 376.
- 14. Ciamician, G., Ber. 11, 1344 1878.
- 15. Cosciug, T, Petroleum. 31, No. 41, 5 1935.
- 16. Cosciug, T., Petroleum. 34, No. 16, 3
- Coscing, T., Petroleum. 34, No. 17, 4 1938.
- 18. Coulson, E. A., J. Chem. Soc. 1935, 77.
- Coulson, E. A., J. Soc. Chem. Ind. 60, No. 5, 123 1941.
- Coulson, E. A., J. Soc. Chem. Ind. 62, 177T 1943.
- Darzens, G., and A. Heinz, Compt. rend. 184, 33 1927.
- 22. Darzens, G., and A. Levy, Compt. rend. 202, 73 1936.
- 23. de Laszlo, H. G., Z. physik. Chem. 118, 369 1925.
- Drake, N. L., and H. M. Duvall, J. Am. Chem. Soc. 58, 1687 1936.
- Evans, E. B., J. Inst. Petroleum Tech.
   24, 537 1938.
- Fieser, L. E., and M. S. Newman, J. Am. Chem. Soc. 58, 2376 1936.

- Fittig, R., and L. Liebmann, Ann.
   255, 257 1889.
- Fittig, R., and J. Remsen, Ann. 155, 112 1870.
- Flegontov, Dissertation through N. A.
  Orlov, K. F. Protyanova, and V. P.
  Flegontov, Khim. Tverdogo Topliva
  7, 748 1936; Refiner Natural Gasoline
  Mfr. 17, 399 1938.
- Fries, K., and A. Küster, Ann. 470, 20 1929.
- Froschl, N., and J. Harlass, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 140, Abt. IIb, 703 1931.
- 32. Gilman, H., and R. E. Hoyle, J. Am. Chem. Soc. 44, 2621 1922.
- 33. Gilman, H., and J. E. Kirby, J. Am. Chem. Soc. **51**, 3475 **1929**.
- 31. Giovannozzi, G., Gazz. chim. ital. 12, 147 1882.
- 35. Gockel, H., Z. physik. Chem. 29B, 79 1935.
- Grimm, II. G., M. Gunther, and H. Tittus, Z. physik. Chem. 14B, 169 1931.
- 37. Grodde, K.-H., Physik. Z. 39, 772 1938.
- 38 Grosse, A. V., and V. N. Ipatieff, J. Org. Chem. 2, 447 1937.
- 39. Grummitt, O., and A. C. Buck, J. Am. Chem. Soc. **65**, 295 **1943**.
- Hall, C. C., J. Soc. Chem. Ind. 54, 208 1935-1936.
- 41. Herzenberg, J., and S. Ruhemann, Ber. 60, 889 1927.
- 42. Hewett, C. L., J. Chem. Soc. 1940, 293.
- Huffman, H. M., G. S. Parks, and M. Barmore, J. Am. Chem. Soc. 53, 3876 1931.
- 41. Kaffer, H., Ber. 57, 1261 1924.
- Kane, H. L., and A. Lowy, J. Am. Chem. Soc. 58, 2605 1936.
- Kariyone, T., and A. Nonaka, J. Pharm. Soc. Japan 57, 20 1937; C.A.
   33, 633 1939; Chem. Zentr. 1937, II, 413.
- 47. Kipping, F. B., and F. Wild, J. Chem. Soc. 1940, 1239.
- 48. Kruber, O., Ber. 62, 3044 1929.
- 49. Kruber, O., Teer u. Bitumen 38, 13 1940.

97 References

- 50. Kruber, O., and A. Marx, Ber. 72, 1970 1939.
- Kruber, O., and W. Schade, Ber. 68, 11 1935.
- Kruber, O., and W. Schade, Ber. 69, 1722 1936.
- Kuhn, R., and A. Winterstein, Ber.
   66, 429 1933.
- Larsen, R. G., R. E. Thorpe, and F. A. Armfield, Ind. Eng. Chem. 34, 183 1942.
- Lecat, M., Ann. soc. sci. Bruxelles 48B, 113 1928.
- 56. Lesser, R., Ann. 402, 1 1913.
- 57. Levy, G., Ann. chim. [11] 9, 5 1938.
- 58. Levy, G., Compt. rend. 192, 1397 1931.
- 59. Levy, G., Compt. rend. 193, 174 1931.
- Linder, E. G., J. Phys. Chem. 35, 531 1931.
- Mair, B. J., and A. J. Streiff, J. Research Natl. Bur. Standards 24, 395 1940.
- 62. Marchetti, C., Gazz. chim. ital. 11, 439 1881.
- Mayer, F., and R. Schiffner, Ber. 67, 67 1934.
- 64. Mayer, F., and A. Sieglitz, Ber. 55, 1835 1922.
- McVicker, W. H., J. K. Marsh, and A. W. Stewart, J. Chem. Soc. 127, 999 1925.
- Meyer, R., and H. Fricke, Ber. 47, 2765 1914.
- Morgan, G. T., and E. A. Coulson, J. Soc. Chem. Ind. 53, 73T 1934.
- 68. Nasini, R., and O. Bernheimer, Gazz. chim. ital. 15, 59 1885.
- Nenitzescu, C. D., D. A. Isacescu, and A. C. Isopescu, Oesterr. Chem. Ztg. 42, 350 1939.
- Noller, C. R., J. Am. Chem. Soc. 56, 1582 1934.
- 71. Olivier, S. C. J., and J. Wit, Rec. trav. chim. 57, 90 1938.
- 72. Orlov, N. A., Ber. 60, 1950 1927.
- Orlov, N. A., and N. D. Likhachev, Ber. 63, 2179 1930.
- Orlov, N. A., K. F. Protyanova, and V. P. Flegontov, Khim. Tverdogo Topliva 7, 748 1936; Refiner Natural Gasoline Mfr. 17, 399 1938.
- 75. Parks, G. S., and H. M. Huffman, Ind. Eng. Chem. 23, 1138 1931.

- Pieters, H. A. J., and H. van den Berge, Brennstoff-Chem. 20, 201 1939.
- Pokrovskaya, E. S., J. Gen. Chem. (U.S.S.R.) 13, 579 1943; Survey For. Petrol. Liter. T472B, Oct. 20, 1944.
- 78. Reingruber, F., Ann. 206, 367 1880.
- 79. Robinson, R., and H. W. Thompson, J. Chem. Soc. **1932**, 2015.
- Rossini, F. D., Petroleum Engineer 14,
   No. 5, 41 1943.
- Roux, L., Ann. chim. phys. [6] 12, 289
   1887.
- Ruzicka, L., and G. Anner, Helv. Chim. Acta 26, 129 1943.
- Ruzicka, L., H. Brüngger, R. Egli, L. Ehmann, M. Furter, and H. Hosli, Helv. Chim. Acta 15, 431 1932.
- Ruzicka, L., H. Brungger, R. Egli, L. Ehmann, and M. W. Goldberg, Helv. Chim. Acta 15, 1496 1932.
- Ruzicka, L., L. Ehmann, and E. Morgelli, Helv. Chim. Acta 16, 311 1933.
- Ruzicka, L., M. Furter, and H. Leuenberger, Helv. Chim. Acta 20, 312 1937.
- Ruzicka, L., and K. Hofmann, Helv. Chim. Acta 19, 114 1936.
- 88. Ruzicka, L., K. Hofmann, and J. Frei, Helv. Chim. Acta 19, 386 1936.
- Ruzicka, L., H. Hosli, and L. Ehmann,
   Helv. Chim. Acta 17, 442 1934.
- Ruzicka, L., H. Schellenberg, and M. W. Goldberg, Helv. Chim. Acta 20, 791 1937.
- Ruzicka, L., F. C. van der Sluys-Veer, and S. L. Cohen, Helv. Chim. Acta 22, 350 1939.
- Ruzicka, L., F. C. van der Sluys-Veer, and O. Jeger, Helv. Chim. Acta 26, 280 1943.
- Salceanu, C., Compt. rend. 193, 161
   1931.
- 94. Salceanu, C., Compt. rend. 194, 863 1932.
- 95. Schroeter, G., L. Lichtenstadt, and D. Irineu, Ber. 51, 1587 1918.
- 96. Schulze, K. E., Ber. 17, 842 1884.
- 97. Sengupta, S. C., J. prakt. Chem. [2] 151, 82 1938.
- Shreve, R. N., and J. H. Lux, Ind. Eng. Chem. 35, 306 1943.

- 99. Späth, E., F. Wessely, and L. Kornfeld, Ber. **65**, 1536 **1932**.
- Thiele, W., and G. Trautmann, Ber.
   2245 1935.
- Veselý, V., and A. Medvedeva, Collection Czechoslov. Chem. Commun.
   3, 440 1931.
- 102. Veselý, V., and F. Štursa, Chem. Listy 26, 495 1932; C.A. 27, 717 1933; Chem. Zentr. 1933, I, 3079.
- 103. Veselý, V., and F. Štursa, Chem. Listy 29, 361 1935; C.A. 30, 8202 1936.
- 104. Vesely, V., and F. Štursa, Collection Czechoslov. Chem. Commun. 3, 430 1931.
- 105. Veselý, V., and F. Štursa, Collection Czechoslov. Chem. Commun. 4, 139 1932.
- 106. Veselý, V., and F. Štursa, Collection Czechoslov. Chem. Commun. 6, 137 1934.
- 107. von Auwers, K., Ann. 499, 123 1932.
- 108. von Auwers, K., Ber. 58, 151 1925.
- 109. von Auwers, K., and F. Bergmann, Ann. 476, 272 1929.
- 110. von Auwers, K., and A. Fruhling, Ann. 422, 196 1921.
- 111. von Auwers, K., and H. Wunderling, Ber. 64, 2748 1931.
- 112. von Auwers, K., and E. Wolter, Nachr. Ges. Wiss. Gottingen, Math. physik. Klasse, Fachgruppen III, 1931, 101.
- 113. von Baeyer, A., Ber. 32, 2429 1899.
- 114. von Braun, J., E. Hahn, and J. Seeman, Ber. **55**, 1687 **1922**.
- 115. Weger, M., Z. angew. Chem. 22, 338 1909.
- 116. Weiss, J. M., and C. R. Downs, Ind. Eng. Chem. 15, 1022 1923.
- 117. Weissgerber, R., Ber. 52, 370 1919.
- 118. Weissgerber, R., and O. Kruber, Ber. **52**, 346 **1919**.
- 119. Wendt, G., J. prakt. Chem. [2] 46, 317 1892.
- 120. Wertyporch, E., Ber. 64, 1369 1931.
- 121. Wichelhaus, H., Ber. 24, 3918 1891.
- 122. Windaus, A, and W. Thiele, Ann. 521, 160 1935.

#### C<sub>13</sub>H<sub>14</sub>

# 1-n-Propylnaphthalene

B. P., °C @ 760mm 275-276<sup>61</sup> 274-275<sup>4, 5</sup>

#### 2-n-Propylnaphthalene

B. P., °C @ 760mm 277-279<sup>4, 5</sup>

B. P., °C @ 760mm

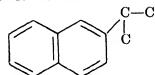
#### 1-Isopropylnaphthalene

264-2669 263-264  $769^{22}$  132 132-134  $10^{38}$ , 39

0.90075  $15^{\circ 88}$   $n_{p}^{20}$ 

1.5728 15° 39 1.5775 15° 38

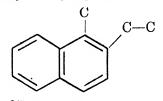
# 2-Isopropylnaphthalene



99  $C_{13}H_{14}$ 

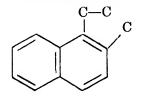
B. P., °C @ 760mm	
26886	
$263 - 265^{22}$	
$262^{86}$	
265	$755^{62}$
145	$25^{62}$
129-130	$14^{2}$
125	$12^{65}$
99-100	$0.2^{8}$
$D_4^{20}$	•
$0.9795^{86}$	
0.974	$D_{\bf 25}^{\bf 25\ 59}$
0.990	0° 62
$n_{p}^{20}$	
$1.5772^{86}$	
$1.5775^{59}$	
$1.5784^{86}$	

#### 1-Methyl-2-ethylnaphthalene



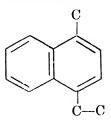
B. P., °C @ 760mm  $12^{45}$ 137.5-140 140-145 1115  $D_{4}^{20}$ 15.4° 45 1.0014  $n_{\rm p}^{20}$ 1.5994515.4° 45 1.60138  $n_{\mathrm{H}lpha}^{15.4}$  45 1.5942215 4 45 1.62151  $n_{\mathrm{H}oldsymbol{eta}}$  $n_{{
m H}\gamma}^{15.4}$  45 1.63995

# 1-Ethyl-2-methylnaphthalene

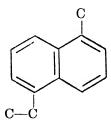


# B. P., °C @ 760mm 135–145

#### 1-Methyl-4-ethylnaphthalene



# 1-Methyl-5-ethylnaphthalene



 $40^{36}$  B. P., °C @ 760mm 140  $12^{77}$  133  $10^{36}$   $D_4^{20}$  1.011  $18^{\circ}$  77

M. P., °C

 $n_{\rm p}^{20}$  1.600  $30^{\circ 36}$  1.6082  $18^{\circ 77}$ 

#### 1-Methyl-6-ethylnaphthalene

$$\begin{array}{c} \\ \\ \\ \\ \end{array}$$

B. P., °C @ 760mm 146 140 12<sup>36</sup>

 $n_{p}^{20}$  1.59836

# 1-Ethyl-6-methylnaphthalene

B. P., °C @ 760mm 135-138 12<sup>14</sup>

# ${\bf 1-Methyl-7-ethylnaphthalene}$

M. P., °C 96-97<sup>20</sup>

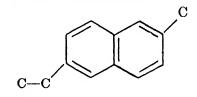
B. P., °C @ 760mm 133-138 16<sup>52</sup> 130-135 16<sup>84</sup> 134.5-136 12<sup>79</sup> 134-136 12<sup>79</sup> 133 12<sup>36</sup>

 $n_{p}^{20}$  1.5970<sup>36</sup>

# 1-Ethyl-7-methylnaphthalene

B. P., °C @ 760mm 128 11<sup>14</sup>

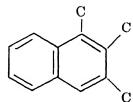
2-Methyl-6-ethylnaphthalene



M. P., °C 44–45<sup>44</sup> B. P., °C @ 760mm

145–150

# 1,2,3-Trimethylnaphthalene



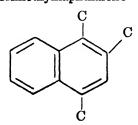
1116

 $12^{66}$ 

M. P., °C 27-2840

B. P., °C @ 760mm 125-130

#### 1,2,4-Trimethylnaphthalene



B.	P.,	$^{\circ}\mathrm{C}$	@	760mm
	1	146		

 $12^{66}$ 

# 1,2,5-Trimethylnaphthalene

# B. P., °C @ 760mm 148-149 16<sup>70</sup> 147-148 11<sup>69</sup> 101-103 113-115 0.5<sup>41</sup> 120 0.2<sup>80</sup>

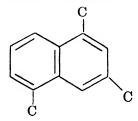
 $\begin{array}{c} 120 & 0.2^{80} \\ D_{4}^{20} & & & \\ & 1.0103 & 22^{\circ 69} \\ & 1.008 & 15^{\circ 70} \\ & 1.011 & 15^{\circ 70} \\ n_{p}^{20} & & & \\ & 1.6093 & 22^{\circ 69} \\ & 1.6093 & 15^{\circ 70} \\ & 1.6093 & 15^{\circ 70} \end{array}$ 

#### 1,2,6-Trimethylnaphthalene

M. P., °C 14<sup>87</sup>

B. P., °C @ 760mm 154-156 15<sup>89</sup> 146  $10^{66}$  119-123  $10^{76}$   $D_4^{20}$   $0.935^{76}$   $n_D^{20}$   $1.521^{76}$ 

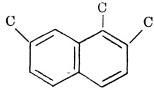
# 1,3,5-Trimethylnaphthalene



M. P., °C 47<sup>37</sup> 43<sup>66</sup> B. P., °C @ 760mm 139.5

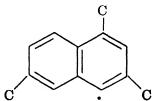
# 1,2,7-Trimethylnaphthalene

(Sapotalene)



B. P., °C @ 760mm 160-163 16<sup>17</sup> 143 13<sup>66</sup> 140-145 12<sup>53</sup> 133 10<sup>18</sup>

# 1,3,6-Trimethylnaphthalene



 $10^{66}$ 

### 1,4,5-Trimethylnaphthalene

M. P., °C 637, 66

B. P., °C @ 760mm

 $12^{66}$ 

1266

#### 2,3,5-Trimethylnaphthalene

M. P., °C 28<sup>7, 81</sup> 25.3<sup>47</sup>

B. P., °C @ 760mm 285<sup>47</sup> 138

# 1,2,8-Trimethylnaphthalene

$$\begin{array}{c|c}
c & c \\
\hline
\end{array}$$

B. P., °C @ 760mm 150 16<sup>66</sup>

## 1,3,7-Trimethylnaphthalene

M. P., °C 13.5<sup>46, 48</sup>

B. P., °C @ 760mm

280 <sup>18</sup> 280 764 <sup>46</sup> 131-133 9<sup>66</sup>

 $D_4^{20}$  1.00746 0.9801

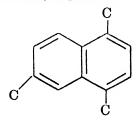
21° 66

 $n_{\rm p}^{20}$ 1.57589<sup>46</sup>
1.5972

15.5° 66

 $15^{66}$ 

# 1,4,6-Trimethylnaphthalene



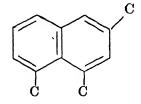
B. P., °C @ 760mm 140–142

# 2,3,6-Trimethylnaphthalene

M. P., °C 102<sup>34</sup>, <sup>46</sup>, <sup>48</sup> 92–93<sup>21</sup>, <sup>23</sup>

# B. P., °C @ 760mm 286<sup>48</sup> 263-264<sup>21</sup> 286 146-148 104-105 1.5<sup>34</sup>

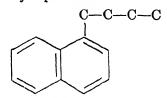
## 2,4,5-Trimethylnaphthalene



M. P., °C 48<sup>37</sup>

# C14H16

#### 1-n-Butylnaphthalene



B. P., °C @ 760mm 281–2834, 5

# 2-n-Butylnaphthalene

B. P., °C @ 760mm 283-285<sup>4, 5</sup> 282 5-283 5

282.5-283.5 745<sup>56</sup> 125-130 13<sup>50</sup>

 $D_4^{20} = 0.9659^{56}$ 

 $n_{\,\mathrm{D}}^{\,20}$  1.5790<sup>56</sup>

# 1-sec-Butylnaphthalene

B. P., °C @ 760mm

111-113 286 105-107 286

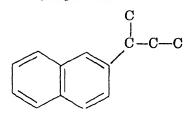
 $D_{\scriptscriptstyle A}^{20}$ 

 $0.9746^{86} \\ 0.9749^{86}$ 

 $n_{\rm p}^{20}$ 

1.569886

# 2-sec-Butylnaphthalene



B. P., °C @ 760mm 147-148 338

 $n_{\,{\scriptscriptstyle \mathrm{D}}}^{\,20}$ 

 $1.5814^{8}$ 

# 1-Isobutylnaphthalene

B. P., °C @ 760mm 136-138 11<sup>27</sup> C14H16

104

#### 2-Isobutylnaphthalene

# 1-tert-Butylnaphthalene

112-113

 $D_{4}^{20}$ 

0.962986

 $n_{\rm p}^{20}$ 

1.572686

# 2-tert-Butylnaphthalene

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ &$$

M. P., °C -490

B. P., °C @ 760mm 274-27686 274-277  $756^{13}$ 272-275  $745^{56}$ 137-139 1713 145  $15^{90}$  142-143 1483 140-145 1359 127-131 932 690115 125 413

 $D_{4}^{20}$ 

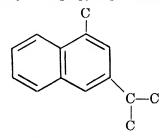
0.9674560968786 0.979

 $D_{25}^{25}$  19

 $n_{\rm p}^{20}$ 

 $1.5685^{59}$  $1.5768^{86}$  $1.5795^{90}$  $1.5812^{56}$ 

#### 1-Methyl-3-isopropylnaphthalene

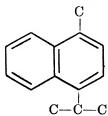


B. P., °C @ 760mm 150

 $11^{26}$ 

166

# 1-Methyl-4-isopropylnaphthalene



M. P., °C 19630

B. P., °C @ 760mm 148

145-148  $12^{60}$  $D_{4}^{20}$ 0.993414.5° 60 14.5° 60 1.5907

# 1-Methyl-7-isopropylnaphthalene (Eudalene)

B. P., °C @ 760mm 2817 73064 280-281 72071 280-281 1824 152 142-143 1271 1271 142 1178 140  $D_4^{20}$ 17° 71 0.9734 17° 78 0.9737 16° 71 0.9747  $n_{\rm p}^{20}$ 17° 71 1.5847 17° 78 1.5849 16° 71 1.5850

# 1-Isopropyl-7-methylnaphthalene (Apocadalene)

B. P., °C @ 760mm

282<sup>7</sup>

139-141

12<sup>72</sup>

D<sub>4</sub><sup>20</sup>

0.9833<sup>72</sup>

n<sub>2</sub><sup>20</sup>

1.5884<sup>72</sup>

#### 1,4-Diethylnaphthalene

M. P., °C 16.5–17<sup>1</sup>

B. P., °C @ 760mm 165  $25^{1}$ 1245 150-152.5 911 136-139  $D_4^{20}$ 0.9933313.1° 45 0.9983  $n_{\rm p}^{20}$ 1.60133 13.1° 45 1.59699 1.59005

# 1-Ethyl-2,6-dimethylnaphthalene

1.61642

1.63423

nH8 45

 $n_{\rm H\gamma}^{13.1\ 45}$ 

B. P., °C @ 760mm 162 23<sup>31</sup>

# 1-Ethyl-2,7-dimethylnaphthalene

B. P., °C @ 760mm 140-142

 $12^{67}$ 

# 1,4-Dimethyl-6-ethylnaphthalene

$$c-c$$

B. P., °C @ 760mm 298–302<sup>35</sup>

 $0.5^{68}$ 

# 1,2,3,4-Tetramethylnaphthalene

M. P., °C 107–108<sup>57</sup> 106.5–107.5<sup>40</sup> 104–105<sup>12</sup>

# 1,3,4,5-Tetramethylnaphthalene

B. P., °C @ 760mm 150

1067

# 1,2,5,6-Tetramethylnaphthalene

M. P., °C 116.4 118<sup>18</sup> 116–117<sup>53, 74</sup> 116–116.5<sup>67, 74</sup> 116<sup>63</sup>

B. P., °C @ 760mm 150-155

 $12^{67}$ 

# 1,2,5,7-Tetramethylnaphthalene

B. P., °C @ 760mm 155-158

 $12^{42}$ 

# 1,2,5,8-Tetramethylnaphthalene

B. P., °C @ 760mm 150

967

## 1,2,6,8-Tetramethylnaphthalene

B. P., °C @ 760mm 166–168

 $15^{67}$ 

#### 1,3,6,8-Tetramethylnaphthalene

M. P., °C 84–85<sup>85</sup>

 $C_{15}H_{18}$ 

# 2-(2'-Pentyl)-naphthalene

B. P., °C @ 760mm 130-131

 $5^{49}$ 

# 1-Isopentylnaphthalene (a)

- B. P., °C @ 760mm 287-289<sup>49</sup>
- (a) The structure of this compound was not clearly defined in the literature.

# 2-Isopentylnaphthalene (a)

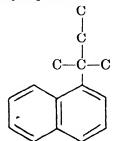
M. P., °C 108-110<sup>54</sup>

B. P., °C @ 760mm

289 -292<sup>54</sup> 140 12<sup>82</sup>

(a) The structure of this compound was not clearly defined in the literature.

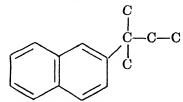
#### 1-tert-Pentylnaphthalene



B. P., °C @ 760mm 301-30386

 $D_4^{20} = 0.9751^{86}$ 

## 2-tert-Pentylnaphthalene



B. P., °C @ 760mm 287-290<sup>86</sup>

 $D_{f 4}^{20}$ 

0.975186

 $n_{\,{\scriptscriptstyle {
m D}}}^{\,20}$ 

1.577286

# 1-Methyl-7-tert-butylnaphthalene

$$C = \begin{pmatrix} C & C & C \\ C & C & C \end{pmatrix}$$

1425

# 1,5-Dimethyl-3-isopropylnaphthalene

B. P., °C @ 760mm 110-112

0.873

#### 1,2-Dimethyl-7-isopropylnaphthalene

$$C-C$$
 $C$ 
 $C$ 
 $C$ 

B. P., °C @ 760mm 149-151

910

# 2,5-Dimethyl-3-isopropylnaphthalene

$$\bigcup_{C \to C} C$$

B. P., °C @ 760mm 154-158

1310

#### 1,3-Dimethyl-7-isopropylnaphthalene

B. P., °C @ 760mm 165–167

1911

#### 1,4-Dimethyl-6-isopropylnaphthalene

$$\begin{array}{c} C \\ C \\ C \\ \end{array}$$

B. P., °C @ 760mm 155–157

1278

## 1,6-Dimethyl-4-isopropylnaphthalene (Cadalene)

B. P., °C @ 760mm

287-29143	
291-292	73064
165	2051
159.5-164	$15^{28}$
159-163	15 <sup>28</sup>
157	1571
153-154	15 <sup>6</sup>
154-159	1329
155-157	1255
155-156	1175
149	10³

 $D_{A}^{20}$ 

0.973155	
0.9716	25° 51
0.9781	19° 28
0.9790	19° 28
0.9798	19° **
0.9824	15° 44

# $n_{P}^{20}$

1.582048	
1.583471	
1.58491	25° 51
1.582	19° 75
1.5850	19° 28
1.5851	19° 28
1.5858	15° 88
1.5869	15° 64

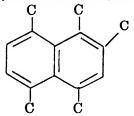
# 1-Isopropyl-4, 6-dimethylnaphthalene

M. P., °C 6060

# 1-Methyl-2,4-diethylnaphthalene

 $D_4^{20}$  0.9870  $13.3^{\circ}$   $^{45}$   $n_p^{20}$  1.59231  $13.3^{\circ}$   $^{45}$  1.58549  $n_{H_6}^{13.3}$   $^{45}$  1.61135  $n_{H_6}^{13.3}$   $n_{H_6}^{13.3}$   $n_{H_6}^{13.3}$   $n_{H_6}^{13.3}$ 

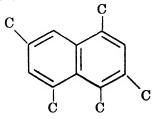
# 1,2,4,5,8-Pentamethylnaphthalene



B. P., °C @ 760mm 150

1067

## 1,3,4,5,7-Pentamethylnaphthalene



M. P., °C 106–107<sup>67</sup>

B. P., °C @ 760mm 150

1067

# References on C<sub>18</sub>H<sub>14</sub> through C<sub>16</sub>H<sub>18</sub> Compounds

- 1. Arnold, R. T., and R. A. Barnes, J. Am. Chem. Soc. 66, 960 1944.
- Barbot, A., Bull. soc. chim. [4] 47, 1314
   1930.
- Bardhan, J. C., and S. K. Banerjee, J. Chem. Soc. 1935, 476.
- 4. Bargellini, G., and G. Melacini, Atti accad. Lincei [5] 17, II, 26 1908.
- 5. Bargellini, G., and G. Melacini, Gazz. chim. ital. 38, II, 567 1908.
- Barnett, E. de B., and J. W. Cook, J. Chem. Soc. 1933, 22.
- Barnett, E. de B., and F. G. Sanders, J. Chem. Soc. 1933, 434.
- Bergmann, F., and A. Weizmann, J. Org. Chem. 9, 352 1944.
- Berry, T. M., and E. E. Reid, J. Am. Chem. Soc. 49, 3142 1927.
- Bradfield, A. E., B. H. Hedge, B. S. Rao, J. L. Simonsen, and A. E. Gillam, J. Chem. Soc. 1936, 667.

- Bradfield, A. E., R. R. Pritchard, and J. L. Simonsen, J. Chem. Soc. 1937, 760.
- Briner, E., W. Plüss, and H. Paillard, Helv. Chim. Acta 7, 1046 1924.
- Bromby, N. G., Λ. T. Peters, and F. M. Rowe, J. Chem. Soc. 1943, 144.
- Brunner, O., and F. Grof, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 142, Abt. IIb, 682 1933.
- Brunner, O., and F. Grof, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 142, Abt. IIb, 730 1933.
- Brunner, O., and F. Grof, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 144, Abt. IIb, 565 1935.
- Brunner, O., H. Hofer, and R. Stein, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 141, Abt. 11b, 671 1932.
- Brunner, O., II. Hofer, and R. Stein, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 142, Abt. IIb, 289 1933.
- Clemo, G. R., and H. G. Dickenson, J. Chem. Soc. 1935, 735.
- Clemo, G. R., and R. D. Haworth, J. Chem. Soc. 1930, 2579.
- 21. Collie, J. N., J. Chem. Soc. 63, 329 1893.
- 22. Cook, J. W., J. Chem. Soc. 1932, 456.
- Coşciug, T., Petroleum. 34, No. 20, 1 1938.
   Darzens, G., and A. Lévy, Compt. rend.
- 24. Darzens, G., and A. Lévy, Compt. rend. 194, 2056 1932.
- Darzens, G., and A. Lévy, Compt. rend. 199, 1426 1934.
- Darzens, G., and A. Lévy, Compt. rend.
   201, 152 1935.
- Darzens, G., and H. Rost, Compt. rend. 146, 933 1908.
- Deussen, E., J. prakt. Chem. [2] 120, 119
   1928.
- Diels, O., and A. Karstens, Ber. 60, 2323
   1927.
- Dziewoński, K., and M. Marusińska, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1938A, 316.
- Dziewoński, K., K. Steo, and P. Zagata, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1938A, 324.
- 32. Fieser, L. F., and C. C. Price, J. Am. Chem. Soc. 58, 1838 1936.
- Frösche, N., and J. Harlass, Monatsh.
   59, 275 1932.

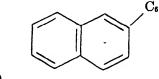
- 34. Gavăt, I., and I. Irimescu, Ber. 75, 820 1942.
- 35. Gucci, P., and G. Grassi-Cristaldi, Gazz. chim. ital. 22, I, 1 1892.
- Harvey, J., I. M. Heilbron, and D. G.
   Wilkinson, J. Chem. Soc. 1930, 423.
- Heilbron, I. M., and D. G. Wilkinson,
   J. Chem. Soc. 1930, 2537.
- Herzenberg, J., and S. Ruhemann, Ber. 60, 889 1927.
- Herzenberg, J., and E. von Winterfeld, Ber. 64, 1036 1931.
- 40. Hewett, C. L., J. Chem. Soc. 1940, 293.
- 41. Hosking, J. R., and C. W. Brandt, Ber. 68, 37 1935.
- Hosking, J. R., and C. W. Brandt, Ber.
   286 1935.
- Kafuku, K., and N. Ichikawa, Bull. Chem. Soc. Japan 8, 371 1933.
- Kon, G. A. R., and W. T. Weller, J. Chem. Soc. 1939, 792.
- 45. Krollpfeiffer, F., Ann. 430, 161 1923.
- 46. Kruber, O., Ber. 72, 1972 1939.
- 47. Kruber, O., Ber. 73, 1174 1940.
- 48. Kruber, O., Teer u. Bitumen 38, 13 1940.
- Larsen, R. G., R. E. Thorpe, and F. A. Armfield, Ind. Eng. Chem. 34, 183 1942.
- Mayer, F., and A. Sieglitz, Ber. 55, 1835
   1922.
- Mousseron, M., R. Granger, and M. Ronayroux, Compt. rend. 208, 1411 1939.
- Nakamura, H., T. Ohta, and G. Hukuti, Proc Imp. Acad. Tokyo 9, 91 1933.
- Noller, C. R., J. Am. Chem. Soc. 56, 1582 1934.
- Oddo, G., and E. Barabini, Gazz. chim. ital. 20, 719 1890.
- Perrottet, E., W. Taub, and E. Briner,
   Helv. Chim. Acta 23, 1260 1940.
- Petrov, A. D., and D. N. Andreev, J. Gen. Chem. (U.S.S.R.) 12, 95 1942.
- 57. Pluss, W., Helv. Chim. Acta 8, 507 1925.
- Pope, G. W., and M. T. Bogert, J. Org. Chem. 2, 276 1937-1938.
- Price, C. C., and J. M. Ciskowski, J. Am. Chem. Soc. 60, 2499 1938.
- Rapson, W. S., and W. F. Short, J. Chem. Soc. 1933, 128.
- Roblin, R., Jr., D. Davidson, and M. T. Bogert, J. Am. Chem. Soc. 57, 151 1935.

- 62. Roux, L., Ann. chim. phys. [6] **12**, 289 **1887**.
- 63. Ruhkopf, II., and P. Mohs, Ber. 69, 1522 1936.
- 64. Ruzicka, L., Fortschr. Chem., Physik u. physik. Chem. 19, No. 5, 1 1928.
- Ruzicka, L., and E. Capato, Ann. 453,
   1927.
- Ruzicka, L., and L. Ehmann, Helv. Chim. Acta 15, 140 1932.
- Ruzicka, L., L. Ehmann, and E. Morgelli, Helv. Chim. Acta 16, 314 1933.
- Ruzicka, L., and E. Eichenberger, Helv. Chim. Acta 13, 1117 1930.
- Chim. Acta 13, 1117 1930.69. Ruzicka, L., and J. R. Hosking, Helv. Chim. Acta 13, 1402 1930.
- Ruzicka, L., H. W. Huyser, M. Pfeiffer, and C. F. Seidel, Ann. 471, 21 1929.
- 71. Ruzicka, L., J. Meyer, and M. Mingazzini, Helv. Chim. Acta 5, 345 1922.
- 72. Ruzicka, L., and M. Mingazzini, Helv. Chim. Acta 5, 710 1922.
- Ruzicka, L., P. Pieth, T. Reichstein, and L. Ehmann, Helv. Chim. Acta 16, 268 1933.
- Ruzicka, L., H. Schellenberg, and M. W. Goldberg, Helv. Chim. Acta 20, 791 1937.
- Ruzicka, L., and C. F. Seidel, Helv. Chim. Acta 5, 369 1922.
- Ruzicka, L., C. F. Seidel, and H. Schinz, Helv. Chim. Acta 16, 1143 1933.
- Ruzicka, L., R. Steiger, and H. Schinz, Helv. Chim. Acta 9, 962 1926.
- Ruzicka, L., and M. Stoll, Helv. Chim. Acta 5, 923 1922.
- Ruzicka, L., and J. A. van Melsen, Helv. Chim. Acta 14, 397 1931.
- Schmid, L., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 148, Abt. IIb, 9 1939.
- 81. Schmid, L., and A. Erdös, Ann. 503, 269 1933.
- Schmid, L., and S. Margulies, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 144, Abt. IIb, 81 1935.
- 83. Simons, J. H., and S. Archer, J. Am. Chem. Soc. **60**, 2953 **1938**.
- Tettweiler, K., O. Engel, and E. Wedekind, Ann. 492, 105 1932.
- Tishenko, D., J. Russ. Phys. Chem. Soc. 62, 1407 1930; Bull. soc. chim. [4] 47, 1137 1930.

- Tsukervanik, I. P., and I. Terent'eva,
   J. Gen. Chem. (U.S.S.R.) 7, 637 1937;
   C. A. 31, 5780 1937; Chem. Zentr. 1938,
   I, 578.
- 87. Veselý, V., and F. Štursa, Collection Czechoslov. Chem. Commun. 4, 21 1932.
- Vesterberg, R., Svensk Kem. Tid. 37, 219 1925.
- 89. von Baeyer, A., Ber. 32, 2429 1899.
- Whitmore, F. C., and W. H. James, J. Am. Chem. Soc. 65, 2088 1943.

#### C<sub>16</sub>H<sub>20</sub>

#### 2-Hexylnaphthalene



 $D_4^{20}$ 

0.9575

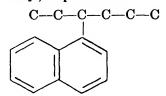
25° 11

 $n_{\scriptscriptstyle D}^{20}$ 

1.5673

25° 11

#### 1-(3'-Hexyl)-naphthalene



B. P., °C @ 760mm 148–158

118

# x-Methyl-x-isopentylnaphthalene (a)

B. P., °C @ 760mm 170-172

158

 $D_4^{20}$ 

0.9607

21.5° 8

1,6-Diisopropylnaphthalene

$$\begin{array}{c} c - c - c \\ c \\ c \\ \end{array}$$

M. P., °C 52¹⁰

# 2,7-Diisopropylnaphthalene

B. P., °C @ 760mm 278-280<sup>20</sup>

D<sub>4</sub><sup>20</sup> 0.9683<sup>20</sup>

 $n_{\rm p}^{20}$ 

1.570120

x,x-Diisopropylnaphthalene (a)

M. P., °C 3810

B. P., °C @ 760mm 317-319<sup>10</sup>

(a) The structure of this compound was not clearly defined in the literature.

## 1,6,7-Trimethyl-4-isopropylnaphthalene

M. P., °C 39.5-40.03

#### C17H22

2-(x'-Naphthyl)-heptane (a)

B. P., °C @ 760mm 162-165 6<sup>19</sup>

 $D_{4}^{20}$ 

0.933

26.5° 19

 $n_{\scriptscriptstyle \mathrm{D}}^{20}$ 

1.5448

24.4° 19

(a) The structure of this compound was not clearly defined in the literature.

# 1,2,6,7-Tetramethyl-4-isopropylnaphthalene

$$c$$
 $c$ 
 $c$ 
 $c$ 
 $c$ 
 $c$ 
 $c$ 

M. P., °C 102-103<sup>3</sup>

# 1-Isopropyl-2,4,6,7-tetramethylnaphthalene

M. P., °C 96.5–97³

x<sub>4</sub>-Tetramethyl-x-isopropylnaphthalene (a)

M. P., °C

1118

#### C<sub>18</sub>H<sub>24</sub>

#### x-n-Octylnaphthalene (a)

B. P., °C @ 760mm

335-337

 $750^{14}$ 

 $D_{4}^{20}$ 

 $0.9366^{14}$ 

 $n_{\rm p}^{20}$ 

 $1.5510^{14}$ 

(a) The structure of this compound was not clearly defined in the literature.

# x,x-Di-sec-butylnaphthalene (a)

B. P., °C @ 760mm

144 -145

 $2^{20}$ 

140-141

 $2^{20}$ 

 $D_{4}^{20}$ 

 $0.9529^{20}$ 

 $0.9517^{20}$ 

 $n_{\rm p}^{20}$ 

1.559620

1.560120

(a) The structure of this compound was not clearly defined in the literature.

# 1,4-Diisobutylnaphthalene

 $D_{4}^{20}$ 

0.9310 0.93316 0.8784 100° 4 0.9104 50° 4 0.9424 0° 9  $n_{\rm p}^{20}$ 

1.5380

25° 16

1.5410

19° 9

Additional Data

 $dD/dt = -0.0006561/^{\circ}C$ 

(0 to 100°C)

#### x,x-Di-tert-butylnaphthalene (a)

M. P., °C

148 (b)17

146-1471, 6

145-146 (b)7

 $143^{2}$ 

14210

 $132^{20}$ 

82-83 (b)<sup>7</sup>

80-81 (b)17

B. P., °C @ 760mm

 $320 (b)^7$ 

319 (b)<sup>7</sup> 180–185

1315

- (a) The structure of this compound was not clearly defined in the literature.
- (b) These constants were determined on isomeric forms.

#### $C_{19}H_{26}$

# x,x,x-Triisopropylnaphthalene (a)

B. P., °C @ 760mm

 $265-270^{15}$ 

148-152

220

 $D_{4}^{20}$ 

 $0.9591^{20} \\ 0.946$ 

 $D_{25}^{25}$  15

 $n_{\rm p}^{20}$ 

 $1.5566^{15}$ 

 $1.5605^{20}$ 

#### $C_{20}H_{28}$

#### x,x-Di-tert-pentylnaphthalene (a)

M. P., °C

 $154 - 155^{20}$ 

(a) The structure of this compound was not clearly defined in the literature.

#### $C_{22}H_{32}$

#### 2,3-Dihexylnaphthalene

$$\bigcap_{C_6}$$

 $n_{\scriptscriptstyle
m D}^{20}$ 

1.5434

25° 11

#### x<sub>4</sub>-Tetraisopropylnaphthalene (a)

M. P., °C 128<sup>2</sup> 127<sup>10</sup>

(a) The structure of this compound was not clearly defined in the literature.

# $\mathbf{C}_{28}\mathbf{H}_{44}$

#### 2-Octadecylnaphthalene

M. P., °C

 $51 - 52^{11}$ 

 $D_4^{20}$ 

0.906

25° 11

0.912

 $D_{15.6}^{15.6}$  13

 $n_{\,{\scriptscriptstyle \mathrm{D}}}^{\,20}$ 

1.5206 (a)

25° 11

(a) This constant is an extrapolated value.

# x,x,x-Trihexylnaphthalene (a)

 $D_4^{20}$ 

0.908 25° 11 0.911 25° 11

 $0.914 \\ 0.917$ 

 $D_{15.6\ 13}^{15.6\ 13} \ D_{15.6\ 15.6}^{15.6\ 13}$ 

 $n_{\,\mathrm{D}}^{\,20}$ 

1.5276 25° <sup>11</sup> 1.5297 25° <sup>11</sup>

(a) The structure of this compound was not clearly defined in the literature.

#### $C_{32}H_{52}$

# 2-Docosylnaphthalene

M. P., °C

 $59.77^{13}$ 

56-5811

 $D_{4}^{20}$ 

0.8987  $25^{\circ 11}$  0.905  $D_{15.6}^{15}$  6 13

# 5-(2'-Naphthyl)-docosane

M. P., °C 38<sup>12</sup>

B. P., °C @ 760mm 200-201

 $2^{12}$ 

 $D_4^{20}$ 

0.898 (a)  $25^{\circ 11}$ 0.904  $D_{15.6}^{15.6 18}$   $n_{p}^{20}$  1.5170 25° 11

(a) This constant is an extrapolated value.

#### $C_{34}H_{56}$

#### 1-(2'-Naphthyl)-2-butyleicosane

$$C_{4}$$
 $C_{-}$ 
 $C_{-$ 

C36H60

#### 5-(2'-Naphthyl)-hexacosane

 $D_4^{20}$ 

0.885

 $D_{15.6}^{15.6}$  13

# 3-(2'-Naphthyl)-4-n-butyldocosane

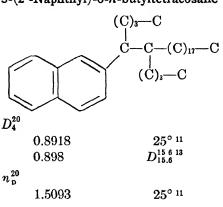
0.8890 0.895

 $25^{\circ 11} \\ D_{15.6}^{15.6 13}$ 

 $n_{\rm p}^{20}$  1.5090 25° 11

#### C38H64

#### 5-(2'-Naphthyl)-6-n-butyltetracosane



# References on C<sub>16</sub>H<sub>20</sub> through C<sub>38</sub>H<sub>64</sub> Compounds

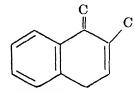
- Bromby, N. G., A. T. Peters, and F. M. Rowe, J. Chem. Soc. 1943, 114.
- Calcott, W. S., J. M. Tinker, and V. Weinmayr, J. Am. Chem. Soc. 61, 1010 1939.
- Campbell, W. P., and M. D. Soffer, J. Am. Chem. Soc. 64, 417 1942.
- Evans, E. B., J. Inst. Petroleum Tech. 24, 537 1938.
- Farmer, E. H., and R. C. Pitkethly, J. Chem. Soc. 1938, 287.
- Fieser, L. F., and C. C. Price, J. Am. Chem. Soc. 58, 1838 1936.
- 7. Gump, W., J. Am. Chem. Soc. **53**, 380 **1931**.
- 8. Hugel, G., Chimie & industrie **26**, 1282 **1931**.
- 9. Lerer, M., Ann. combustibles liquides 8, 681 1933.
- Meyer, II., and K. Bernhauer, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 138, Abt. IIb, 721 1929.
- Mikeska, L. A., Ind. Eng. Chem. 28, 970 1936.
- Mikeska, L. A., C. F. Smith, and E. Lieber, J. Org. Chem. 2, 499 1938.
- Nelson, W. L., Oil Gas J. 35, No. 44, 46 1937.

- 14. Petrov, A. D., and D. N. Andreev, J. Gen. Chem. (U.S.S.R.) 12, 95 1942.
- Price, C. C., and J. M. Ciskowski, J. Am. Chem. Soc. 60, 2499 1938.
- Rossini, F. D., Refiner Natural Gasoline Mfr. 14, 268 1935.
- Simons, J. II., and S. Archer, J. Am. Chem. Soc. 60, 2953 1938.
- Spiegler, L., and J. M. Tinker, J. Am. Chem. Soc. 61, 1002 1939.
- Tilicheev, M. D., Khim. Tverdogo Topliva 9, 181 1938.
- Tsukervanik, I. P., and I. Terent'eva,
   J. Gen. Chem. (U.S.S.R.) 7, 637 1937;
   C. A. 31, 5780 1937; Chem. Zentr. 1938,
   I, 578.

# 3. MISCELLANEOUS POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-12}$

#### $C_{12}H_{12}$

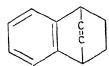
1-Methylene-2-methyl-1,4-dihydronaphthalene



B. P., °C @ 760mm

1517

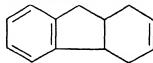
1,4-Endoethene-1,2,3,4-tetrahydronaphthalene



M. P., °C 63.5<sup>32</sup>

#### $C_{13}H_{14}$

1,4,4a,9a-Tetrahydrofluorene



B. P., °C @ 760mm 121 12<sup>49</sup> 116-118 11<sup>2</sup>

 $D_{4}^{20}$ 

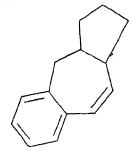
1.018 15° 49

 $n_{\scriptscriptstyle \mathrm{D}}^{20}$ 

 $1.58924^{49}$ 

C14H16

1,2-Cyclopentano-5,6-benzocycloheptene-3

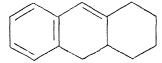


M. P., °C 29-35<sup>16</sup>

B. P., °C @ 760mm 130-135

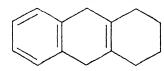
116

1,2,3,4,4a,10-Hexahydroanthracene



M. P., °C 63-6616

1,2,3,4,9,10-Hexahydroanthracene



- M. P., °C 66.5<sup>28</sup>. <sup>24</sup>, <sup>25</sup>, <sup>46</sup> 63<sup>38</sup> 62-63<sup>36</sup> 60-62<sup>9</sup>
- B. P., °C @ 760mm 303-306<sup>23</sup>, <sup>24</sup>, <sup>25</sup>

#### x<sub>5</sub>-Hexahydroanthracene (a)

M. P., °C 78<sup>50</sup>

B. P., °C @ 760mm 290<sup>26</sup>

 $63^{26}$ 

(a) The structure of this compound was not clearly defined in the literature.

## x<sub>6</sub>-Hexahydrophenanthrene (a)

M. P., °C -37

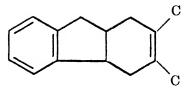
B. P., °C @ 760mm  $305-307^{7, 8}$  289-290  $737^{43}$  179-180  $23^{19}$  165-167  $13^{7}$  125-126  $2.5^{13}$   $D_4^{20}$ 

1.043 15° 7 1.053 0° 7 1.5704<sup>48</sup> 1.586 15° 7 1.5810 14.5° <sup>13</sup>

(a) The structure of this compound was not clearly defined in the literature.

#### $C_{15}H_{18}$

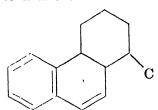
2,3-Dimethyl-1,4,4a,9a-tetrahydro-fluorene



B. P., °C @ 760mm 146-148

 $11^{2}$ 

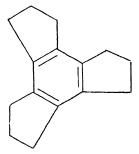
1-Methyl-1,2,3,4,4a,10a-hexahydrophenanthrene



B. P., °C @ 760mm 124

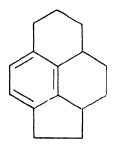
 $0.3^{27}$ 

4,5,6,7-Dicyclopentanoindane



M. P., °C 9741

Cyclopentano-[gh]-3a,4,5,6-tetrahy-drophenalan

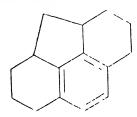


M. P., °C 29<sup>58</sup>

B. P., °C @ 760mm 130-132

 $12^{58}$ 

# Cyclopentano-[def]-1,2,3,4,5,6,7,8-octahydrophenanthrene



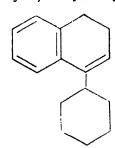
M. P., °C 47<sup>58</sup>

B. P., °C @ 760mm 238<sup>58</sup> 130-132

 $15^{58}$ 

# C16H20

# 4-Cyclohexyl-1,2-dihydronaphthalene



B. P., °C @ 760mm 172 140

 $D_4^{20}$ 

1.021

12.2° 14

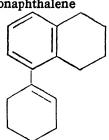
1014

114

 $n_{\scriptscriptstyle 
m D}^{20}$ 

1.5762 12.2° 14

5-(Cyclohexen-1'-yl)-1,2,3,4-tetrahydronaphthalene

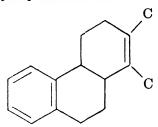


B. P., °C @ 760mm

181

 $15^{14}$ 

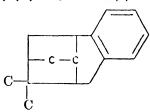
1,2-Dimethyl-3,4,4a,9,10,10a-hexahydrophenanthrene



B. P., °C @ 760mm 105-107

 $0.02^{6}$ 

2,3-Benzo-6,6-dimethyltricyclo-[3,2,1,1<sup>7,9</sup>]-decane (a)



B. P., °C @ 760mm 121.5-122.5 5³ 110.5-111 0.3³

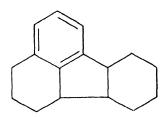
 $D_4^{20}$  1.0192

 $D_0^{16\ 3}$ 

 $n_{\rm p}^{20} = 1.5540$ 

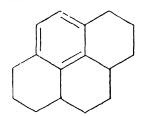
16° 3

# Benzo-[jk]-1,2,3,4,4a,4b,5,6,7,9a-decahydrofluorene



B. P., °C @ 760mm 174-176  $13^{57}$  181-183  $12^{57}$   $D_{\bullet}^{20}$ 1.0207  $29^{\circ 57}$ 1.043  $22^{\circ 57}$ 

# 1,2,3,3a,4,5,5a,6,7,8-Decahydropyrene



M. P., °C 34<sup>58</sup>

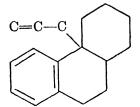
B. P., °C @ 760mm 160 5<sup>35</sup> D<sub>4</sub><sup>20</sup> 1.0612<sup>58</sup> 1.0553 25° <sup>35</sup>

 $n_{\rm D}^{20}$   $1.5806^{58}$  1.5782

25° 35

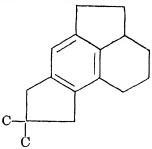
#### $C_{17}H_{22}$

4a-(Propen-2'-yl)-1,2,3,4,4a,9,10, 10a-octahydrophenanthrene



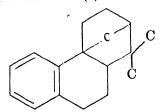
B. P., °C @ 760mm 118 0.2<sup>28</sup>

6,7-(4',4'-Dimethylcyclopentano)-2a,3,4,5-tetrahydroacenaphthene



B. P., °C @ 760mm 173-175  $13^{21}$   $D_4^{20}$  0.9884  $22^{\circ 21}$   $n_p^{20}$  1.5399  $22^{\circ 21}$ 

1,1-Dimethyl-2,4a-endomethylene-1,2,3,4,4a,9,10,10a-octahydrophenanthrene (a)



B. P., °C @ 760mm 132.5-133.5 1<sup>3</sup>  $D_4^{20}$ 1.0252  $D_0^{19}$  <sup>3</sup>  $n_{\rm p}^{20}$ 

1.5578

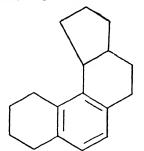
19° 3

(a) The structure of this compound was not clearly defined in the literature.

Spiro[6,7-cyclopentano-1,2,3,4tetrahydronaphthalene-2,1'cyclopentane]

M. P., °C 64-65<sup>48</sup>

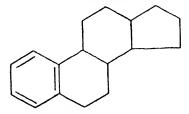
3,4-Cyclopentano-1,2,3,4,5,6,7,8-octahydrophenanthrene



B. P., °C @ 760mm 172-173

587

1,2-Cyclopentano-1,2,3,4,4a,9,10, 10a-octahydrophenanthrene

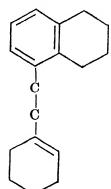


B. P., °C @ 760mm 201

1680

C18H24

5-[2'-(Cyclohexen-1"-yl)-ethyl]-1,2,3,4-tetrahydronaphthalene



B. P., °C @ 760mm

140-150

 $0.05^{10}$ 

1-Methyl-7-isopropyl-x<sub>5</sub>-hexahydrophenanthrene (a)

B. P., °C @ 760mm

 $355^{38}$ 

175-177

101, 53

 $D_4^{20}$ 

 $0.9802^{1, 53}$ 

 $0.997^{38}$ 

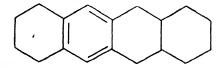
 $n_{\rm p}^{20}$ 

1.547051, 53

 $1.54770^{53}$ 

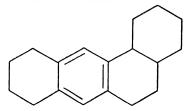
(a) The structure of this compound was not clearly defined in the literature.

1,2,3,4,4a,5,7,8,9,10,12,12a-Dodecahydronaphthacene



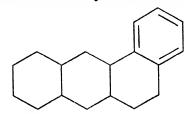
M. P., °C 82-83<sup>55</sup>

# 1,2-Cyclohexano-1,2,3,4,5,6,7,8-octahydroanthracene



M. P., °C 87–88<sup>4</sup>

#### 1,2-Benzododecahydroanthracene



M. P., °C 87-88<sup>11</sup> 71-72<sup>4</sup>, <sup>12</sup> 71<sup>16</sup> 68-70<sup>15</sup>

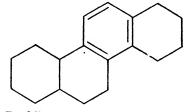
# Dodecahydro-x,x-benzoanthracene (a)

 $D_{4}^{20}$ 

1.12331

(a) The structure of this compound was not clearly defined in the literature.

## 1,2,3,4,5,6,6a,7,8,9,10,10a-Dodecahydrochysene



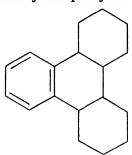
M. P., °C 93–9439

83-84<sup>40</sup> 55-57<sup>56</sup>

B. P., °C @ 760mm 224-226

1256

#### 1,2,3,4,4a,4b,5,6,7,8,8a,12b-Dodecahydrotriphenylene



B. P., °C @ 760mm 200-201.5 20<sup>44</sup> D<sub>4</sub><sup>20</sup>

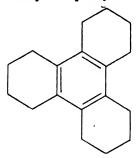
0.9468

19.2° 44

 $n_{\,{\scriptscriptstyle D}}^{\,20}$ 

 $\begin{array}{cccc} 1.50432 & n_{\mathrm{H}\alpha}^{20} & 4\\ 1.51382 & n_{\mathrm{H}\beta}^{20} & 4\\ 1.5959 & n_{\mathrm{H}\gamma}^{20} & 4\\ 1.50693 & n_{\mathrm{H}\alpha}^{20} & 4\\ \end{array}$ 

1,2,3,4,5,6,7,8,9,10,11,12-Dodecahydrotriphenylene



M. P., °C

232

232-23318, 84, 45, 59

2325

231-23259

226-22852 22651

 $D_4^{20}$ 

1.141

0° 60

0° 60 1.148

## C19H26

6,7-(4',4'-Diethylcyclopentano)-2a,3,4,5-tetrahydroacenaphthene

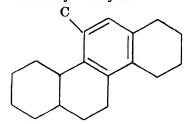
$$C-C$$

B. P., °C @ 760mm 190-195

1621, 22

- 4-Methyldodecahydro-1,2-benzoanthracene (a)
- M. P., °C  $92.5 - 93.5^{17}$
- (a) The structure of this compound was not clearly defined in the literature.
- 4-Methyl-1,2,3,4,5,6,6a,7,8,9,10, 10a-dodecahydrochrysene

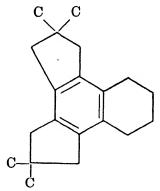
M. P., °C 87-87.540 11-Methyl-1,2,3,4,5,6,6a,7,8,9,10a-dodecahydrochrysene



M. P., °C  $98.8 - 99.8^{20}$ 

#### C20H28

5,6,7,8-Di-(4',4'-dimethylcyclopentano)-1,2,3,4-tetrahydronaphthalene



M. P., °C 105-10621, 22

# $C_{22}H_{32}$

x,x-Dicyclohexyl-1,2,3,4-tetrahydronaphthalene (a)

M. P., °C  $-4^{42}$ 

B. P., °C @ 760mm 198-203  $D_{\scriptscriptstyle A}^{20}$ 0.9978  $D_{15}^{15\ 42}$ 1.0145

1.561842

40° 42 1.5548

 $3^{42}$ 

40° 42

(a) The structure of this compound was not clearly defined in the literature.

#### C26H40

1,2-[3'-(5'',6''-Dimethyl-sec-heptyl)cyclopentano]-2-methyl-1,2,3,4, 5.6-hexahydrophenanthrene (Neoergostatetraene)

M. P., °C  $63 - 64^{29}$ 

References on Miscellaneous Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-12}$ 

- 1. Adelson, D. E., and M. T. Bogert, Chem. Revs. 24, 135 1939.
- 2. Alder, K, and H. F. Rickert, Ber. 71, 379 **1938**.
- 3. Arbusov, B. A., J. Gen. Chem. (U.S. S.R.) 9, 239 1939; Chem. Zentr. 1939, II, 1284.
- 4. Bachmann, W. E., J. W. Cook, A. Dansi, C. G. M. de Worms, G. A. D. Haslewood, C. L. Hewett, and A. M. Robinson, Proc Roy. Soc. 123B, 343 1937.
- 5. Bergmann, E., and O. Blum-Bergmann, J. Am. Chem. Soc. 59, 1441 1937.
- 6. Bergmann, E., and A. Weizmann, J. Org. Chem. 4, 266 1939.
- 7. Breteau, P., Compt. rend. 140, 942 1905.
- 8. Brissemoret, A., and A. Jospin, Compt. rend. 151, 1151 1910.
- 9. Clemmensen, E., Ber. 47, 681 1914.
- 10. Cohen, A., J. W. Cook, and C. L. Hewett, J. Chem. Soc. 1935, 1633.
- 11. Cook, J. W., and G. A. D. Haslewood, J. Chem. Soc. 1935, 767.

- 12. Cook, J. W., and C. L. Hewett, J. Chem. Soc. 1934, 365.
- 13. Cook, J. W., C. L. Hewett, and C. A. Lawrence, J. Chem. Soc. 1936, 71.
- 14. Cook, J. W., and C. A. Lawrence, J. Chem. Soc. 1936, 1431.
- 15. Cook, J. W., and C. A. Lawrence, J. Chem. Soc. 1937, 817.
- 16. Cook, J. W., N. A. McGinnis, and S. Mitchell, J. Chem. Soc. 1944, 286.

- 17. Cook, J. W., A. M. Robinson, and F. Gouldon, J. Chem. Soc. 1937, 393.
- 18. Favorskii, A., and V. Boshovskii, Ann. **390.** 122 **1912**.
- 19. Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 57, 2192 1935.
- 20. Fieser, L. F., and L M. Joshel, J. Am. Chem. Soc. 62, 1211 1940.
- 21. Fleischer, K, and F. Siefert, Ann. 422, 272 **1921.**
- 22. Fleischer, K., and F. Siefert, Ber. 53, 1255 **1920**.
- 23. Godchot, M., Ann. chim. phys. [8] 12, 468 1907.
- 24. Godchot, M., Bull. soc. chim. [4] 1, 710 1907.
- 25. Godchot, M., Compt. rend. 142, 1202 1906.
- 26. Graebe, C., and C. Liebermann, Ann. suppl. 7, 257 1870.
- 27. Grewe, R., Ber. 72, 785 1939.
- 28. Grewe, R., Ber. 76, 1072 1943.
- 29. Haslewood, G. A. D., and E. Roe, J. Chem. Soc. 1935, 465.
- 30. Hawthorne, J. R., and R. Robinson, J. Chem. Soc. 1936, 763.
- 31. Iball, J., Chemistry & Industry 1935, 716; C.A. 29, 7737 1935.
- 32. Kasanskii, B. A., and L. G. Vol'fson, J.

- Gen. Chem. (U.S.S.R.) 8, 1685 1938; C.A. 33, 4966 1939.
- 33. Liebermann, C., Ann. 212, 1 1882.
- 34. Mannich, C., Ber. 40, 153 1907.
- Matsuno, K., and K. Han, Bull. Chem. Soc. Japan 11, 321 1936.
- Meyer, H., R. Bondy, and A. Eckert, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 121, Abt. IIb, 1011 1912.
- Nenitzescu, C. D., E. Ciorănescu, and M. Maican, Ber. 74, 687 1941.
- 38. Olsson, J., Ing. Vetenskaps Akad., Handl. 1931, No. 111.
- Peak, D. A., and R. Robinson, J. Chem. Soc. 1936, 759.
- Peak, D. A., and R. Robinson, J. Chem. Soc. 1937, 1581.
- Perkin, W. H., Jr., and S. G. P. Plant, J. Chem. Soc. 127, 1138 1925.
- Pokrovskaya, E. S., and T. G. Stepanseva, J. Gen. Chem. (U.S.S.R.) 9, 1953 1939.
- Schmidt, J., and R. Mezger, Ber. 40, 4240 1907.
- 44. Schrauth, W., Z. angew. Chem. 36, 571 1923.
- 45. Schroeter, G., Ber. 57, 1990 1924.
- 46. Schroeter, G., Ber. 57, 2003 1924.

- Schroeter, G., L. Lichtenstadt, and D. Irineu, Ber. 51, 1587 1918.
- Sen-Gupta, Science and Culture 3, 56
   1937; C.A. 31, 7868 1937; Chem. Zentr.
   1938, I, 4621.
- Spilker, A., and K. Zerbe, Z. angew. Chem. 39, 1138 1926.
- Sugino, K., and K. Outi, J. Chem. Soc. Japan 62, 401 1941; C.A. 37, 4726 1943.
- 51. Treibs, W., Ber. 61, 683 1928.
- van de Kamp, J., A. Burger, and E. Mosettig, J. Am. Chem. Soc. 60, 1321 1938.
- 53. Virtanen, A. J., Ber. 53, 1880 1920.
- 54. von Braun, J., and O. Bayer, Ann. 472, 90 1929.
- von Braun, J., O. Bayer, and L. F. Fieser, Ann. 459, 287 1927.
- von Braun, J., G. Irmisch, Ber. 65, 883
   1932.
- 57. von Braun, J., and G. Manz, Ber. 63, 2608 1930.
- von Braun, J., and E. Rath, Ber. 61, 956 1928.
- Ward, J. J., W. R. Kirner, and H. C. Howard, J. Am. Chem. Soc. 67, 246 1945.
- Ziegler, K., and F. Ditzel, Ann. 473, 194 1929.

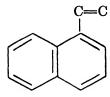
# IV. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA C<sub>n</sub>H<sub>2n-14</sub>

- 1. Naphthalene Derivatives of Empirical Formula C<sub>n</sub>II<sub>2n-14</sub>
- 2. Cyclanonaphthalenes and Their Alkyl Derivatives
- 3. Tetrahydroanthracenes, Tetrahydrophenanthrenes, and Their Alkyl Derivatives
- 4. Miscellaneous Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-14</sub>

# 1. NAPHTHALENE DERIVATIVES OF EMPIRICAL FORMULA $C_nH_{2n-14}$

#### $C_{12}H_{10}$

# 1-Ethenylnaphthalene



B. P., °C @ 760mm

126-128	$15^{16}$
124-125	$15^{32}$
116-117	829
105-106	634
103-104	484

115-116

 $D_4^{20} \\ 1.034^{20} \\ 1.0656^{34}$ 

1.0439  $D_{24}^{24}^{29}$  1.0361  $18^{\circ 32}$  1.0774  $0^{\circ 34}$ 

 $n_{\rm D}^{20}$ 

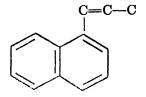
1.6388<sup>34</sup> 1.6436<sup>20</sup>, <sup>32</sup> 1.6425

24° 29

3-432

#### $C_{13}H_{12}$

# 1-(Propen-1'-yl)-naphthalene



B. P., °C @ 760mm

275-278<sup>23</sup> 147-149 135-136

15<sup>33</sup> 12<sup>22</sup> 137–138 112–113

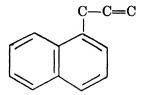
 $D_4^{20}$ 

 $1.0282^{84}$  1.0429

 $n_{\rm D}^{20}$ 

 $1.6335^{34}$ 

#### 1-(Propen-2'-yl)-naphthalene



 $10^{26}$ 

 $6^{34}$ 

0° 34

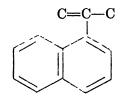
B. P., °C @ 760mm

265–267<sup>33</sup> 130–131 12<sup>22</sup> 127.5–128.5 8<sup>9</sup>

 $n_{\rm p}^{20}$ 

1.6089 25° 9

# 1-Isopropenylnaphthalene



B. P., °C @ 760mm

107–109 4<sup>34</sup> 97 0.1<sup>28</sup>

$D_4^{20}$		
	$1.027^{34}$	
	1.0078	25°30
	1.0143	9° 10, 11
	1.0208	0° 10, 11
$n_{{\scriptscriptstyle D}}^{20}$		
	1.614234	
	1.6070	25° 28
	1.6068	25° 30
	1.6040	16° 13
	1.61435	9° 10, 11

#### 2-Isopropenylnaphthalene

 $\begin{array}{c} \text{M. P., °C} \\ 56^2 \\ 46\text{--}47^{10, \ 11} \\ 45\text{--}47^{12} \end{array}$ 

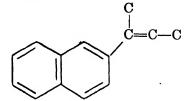
B. P., °C @ 760mm 126-127 12<sup>27</sup> 155 11<sup>2</sup> 138-140 7<sup>12</sup>

# $C_{14}H_{14}$

# 2-(1'-Naphthyl)-butene-2

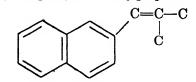
B. P., °C @ 760mm 113.5-115  $5^{34}$   $D_{4}^{20}$   $1.0533^{34}$  $n_{p}^{20}$   $1.6137^{34}$ 

## 2-(2'-Naphthyl)-butene-2



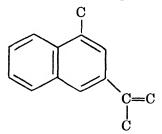
B. P., °C @ 760mm 108-112 0.15<sup>2</sup>  $n_p^{20}$  1.6202<sup>2</sup>

#### 1-(2'-Naphthyl)-2-methylpropene-1



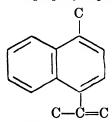
B. P., °C @ 760mm 287-288<sup>15</sup>

# 1-Methyl-3-isopropenylnaphthalene



B. P., °C @ 760mm 154 12<sup>7</sup>

# 1-Methyl-4-isopropenylnaphthalene



B. P., °C @ 760mm • 168

#### C<sub>15</sub>H<sub>16</sub>

# 2-(1'-Naphthyl)-pentene-2

B. P., °C @ 760mm  

$$158-160$$
  $20^{31}$   
 $D_4^{20}$   
 $1.0094$   $22^{\circ 81}$   
 $n_P^{20}$ 

# 3-(1'-Naphthyl)-pentene-2

1.59206

22° 31

B. P., °C @ 760mm 119-120  $6^{34}$   $D_4^{20}$   $1.034^{34}$  $n_p^{20}$ 

 $1.597^{34}$ 

#### 2-Methyl-3-(1'-naphthyl)-butene-2

$$\begin{array}{c|c} & & & \\ \hline \end{array}$$

B. P., °C @ 760mm 165-166  $23^{21}$   $D_4^{20}$  1.0039  $19^{\circ 31}$   $n_p^{20}$  . 1.59343  $19^{\circ 31}$ 

# 1,6-Dimethyl-4-isopropenylnaphthalene

(Dehydrocadalene)

B. P., °C @ 760mm

154
155-157
12<sup>21</sup>

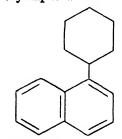
0.9731<sup>21</sup>

n<sub>P</sub><sup>20</sup>

#### C16H18

#### 1-Cyclohexylnaphthalene

 $1.5572^{21}$ 



B. P., °C @ 760mm

118-120

0.36

D<sub>4</sub><sup>20</sup>

1.044

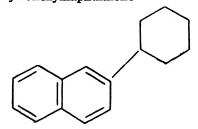
15° 6

1.6099

1.6000

11° 6

# 2-Cyclohexylnaphthalene



# 1-Methyl-x-cyclopentylnaphthalene (a)

B. P., °C @ 760mm 156-158 1<sup>23</sup>

 $D_4^{20}$ 

 $1.0353^{23}$  1.0419

 $D_{15}^{15}$  23

 $n_{\,{\scriptscriptstyle D}}^{\,20}$ 

 $1.6088^{23}$ 

(a) The structure of this compound was not clearly defined in the literature.

#### $C_{17}H_{20}$

# 2-(2'-Naphthyl)-5-methylhexene-x (a)

B. P., °C @ 760mm

175–178

10<sup>10</sup>, <sup>12</sup>

D<sub>4</sub><sup>20</sup>

0.9728

0.9808

0° 10, <sup>12</sup>

n<sub>p</sub><sup>20</sup>

1.59124

9° 10, <sup>12</sup>

(a) The double bond may be in either the 1- or 2-position.

#### CaoH46

#### 5-(1'-Naphthyl)-eicosene-5

 $D_4^{20}$ 

 $n_{\rm p}^{20}$ 

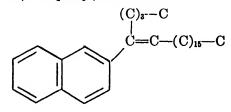
 $0.9107^{17}$ 

1.53019<sup>17</sup> 1.52536 1.54290

 $n_{{
m H}lpha}^{20\ 17} \ n_{{
m H}lpha}^{20\ 17}$ 

#### C39H50

#### 5-(2'-Naphthyl)-docosene-5



 $D_4^{20}$ 0.9081
0.915

 $25^{\circ}$  18  $D_{15.6}^{15.6}$  19

1.5300

25° 18

#### $C_{36}H_{58}$

## 5-(2'-Naphthyl)-hexacosene-5

 $\begin{array}{cccc} D_{4}^{20} & & & & & & \\ & 0.8830 & & & 25^{\circ} \, ^{18} \\ & 0.889 & & & D_{15.6}^{15.6} \, ^{19} \\ n_{_{D}}^{20} & & & & & \\ & & & 1.5128 & & 25^{\circ} \, ^{18} \end{array}$ 

### 3-(2'-Naphthyl)-4-n-butyldocosene-3

$$C-C$$
 $C=C-(C)_{17}-C$ 
 $(C)_3-C$ 
 $(C)_3-C$ 
 $(C)_3-C$ 

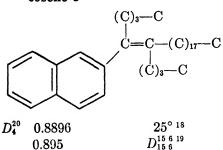
n<sub>p</sub><sup>20</sup>
1.5170 25° 18

0.902

#### C<sub>38</sub>H<sub>62</sub>

 $D_{15.6}^{15.6}$ 

#### 5-(2'-Naphthyl)-6-n-butyltetracosene-5



n<sub>p</sub><sup>20</sup> 1.5161 25° 18

References on Naphthalene Derivatives of

- Empirical Formula C<sub>n</sub>H<sub>2n-1</sub>
  1. Barnett, E. de B., and J. W. Cook, J. Chem. Soc. 1933, 22.
- Bergmann, F., and A. Weizmann, J. Org Chem. 9, 352 1944.
- Bergmann, E., and A. Weizmann, Trans. Faraday Soc. 32, 1327 1936.
- 4. Bodroux, D., Ann. chim. [10] 11, 511 1929.
- 5. Cook, J. W., J. Chem. Soc. 1932, 456.
- Cook, J. W., and C. A. Lawrence, J. Chem. Soc. 1936, 1431.
- Darzens, G., and A. Levy, Compt. rend. 201, 152 1935.
- Dziewoński, K., and M. Marusińka, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1938A, 316.

- Ficser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 60, 1658 1938.
- Grignard, V., Ann. chim. phys. [7] 24, 433 1901.
- 11. Grignard, V., Ann. Univ. Lyon 1901, 1.
- 12. Grignard, Bull. soc. chim. [3] 25, 497 1901.
- Herzenberg, J., and E. von Winterfeld, Ber. 64, 1036 1931.
- 14 Kay, F. W., and A. Morton, J. Chem. Soc. 105, 1565 1914.
- 15. Klinckhard, T., Ann. 379, 362 1911.
- Koton, M. M., J. Gen. Chem. (U.S.S.R.)
   9, 1626 1939.
- I.arsen, R. G., R. E. Thorpe, and F. A. Armfield, Ind. Eng. Chem. 34, 183 1942.
- 18. Mikeska, L. A., Ind. Eng. Chem. 28, 970 1936.
- Nelson, W. L., Oil Gas J. 35, No. 44, 46 1937.
- Palfray, L., S. Sabetay, and D. Sontag, Compt. rend. 194, 2065 1932.
- Perrottet, E., W. Taub, and E. Briner,
   Helv Chim. Acta 23, 1260 1940.
- Pestemer, M., and F. Manchen, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 145, Abt. Hb, 312 1936.
- Pokrovskaya, E. S., J. Gen. Chem. (U.S.S.R.) 13, 579 1943; Survey For. Petrol Liter. T172B, Oct. 20, 1944.
- 24 Pokrovskaya, E. S., and T. G. Stepanseva, J. Gen. Chem. (U.S.S.R.) 9, 1953 1939; C.A. 34, 4731 1940.
- Price, C. C., and J. M. Ciskowski, J. Am. Chem. Soc 60, 2499 1938.
- 26. Rousset, L., Bull. soc. chim. [3] 17, 812
- 27. Ruzicka, L., and E. Capato, Ann. 453, 62 1927.
- Sapiro, R. H., R. P. Linstead, and D. M. Newitt, J. Chem. Soc. 1937, 1784.
- Shorygin, P. P., and N. V. Shorygina, J. Gen. Chem. (U.S.S.R.) 5, 555 1935;
   C.A. 29, 6885 1935.
- Shurakovski, E., J. Russ. Phys. Chem.
   Soc. 41, 1687 1909; C.A. 5, 1097 1911;
   Chem. Zentr. 1910, I, 1144.
- Soniss, S. A., J. Gen. Chem. (U.S.S.R.)
   119 1939; Chem. Zentr. 1939, II, 3276.

131 C12H10

- 32. Sontag, D., Ann. chim. [11] 1, 359 1934. | 34. Zalkind, Y. S., and S. A. Zonis, J. Gen. 33. Tiffeneau and Daudel, Compt. rend. 147, 678 1908.
  - Chem. (U.S.S.R.) 6,988 1936; C.A. 31, 676 1937; Chem. Zentr. 1937, I, 1934.

#### CYCLANONAPHTHALENES AND THEIR ALKYL DERIVATIVES, 2. $C_n H_{2n-14}$

	$\mathcal{O}_n$	112n-14		
$\mathbf{C}_{12}\mathbf{H}_{10}$			287.0	$932.3^{55}$
Acenaphthene			286.8	930.355
	1		275.4	$733.4^{55}$
			275.3	$733.4^{55}$
$\downarrow$	L		264.4	573.5
			264.4	$573.1^{55}$
1_ 11			252.5	$434.7^{55}$
			252.4	$434.5^{55}$
M. P., °C	~		247.0	382.955
95.4			246.6	$377.9^{55}$
10030, 80			246.2	$374.6^{55}$
$99-100^{29}$			233.2	$272.0^{55}$
9937, 72			229.5	$250^{64}$
$96.5^{46}$			227.2	234.655
$96.2^{19, 20, 21}$			210.4	$149.0^{55}$
$94-96^{2}$ , 31			210.2	$148.0^{55}$
$95.5^{18}$			182.4	63.6
95.039, 65			147	$20^{48}$
95(a)			147.2	$19.2^{55}$
$94.1^{55}$		l	144	$12^{62}$
$94.0^{38}$			20.0	< .0245
$94^{28}$		$D_4^{20}$		
93-9434			1.220 (solid) (b)	56
$93.5^{51}$ , $61$ , $78$			0.962	178.7° 18
93.354, 77			1.003	128.6° 18
934, 5, 6			1.0300	103° 73
92-9358, 59			1.0685	$D_{100}^{100}^{64}$
$92.5^{27}$			1.0242	99.2° 81
$92.1^{16}$			1.0331	98.8° 15
B. P., °C @ 760mm			1.0687	$D_{95}^{95}$ 64
277 .3			1.2195	24.8° 32
27930, 62, 83		20		
27884		$n_{\scriptscriptstyle  m D}^{20}$	1 00050	
277.944			1.60658	99.2° 81
277.51, 3, 41, 68			1.60482	98.8° 15
277.355			1.59877	$n_{{ m H}_{m{lpha}}}^{99.281}$
287.8	943.055		1.59772	$n_{\mathrm{H}lpha}^{98.8~15}$

 $n_{{
m H}\beta}^{99.2~81} \ n_{{
m H}\beta}^{99.2~81} \ 1.62705 \ n_{{
m H}\beta}^{98.8~15}$ 

Additional Data

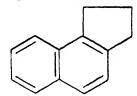
Crit. Temp. (°C) 530.0<sup>82</sup>

 $\frac{1}{T_b} = 0.0028359 - 0.0003538 \log_{10} p_{mm}$ (60 to 945mm)

- (a) The melting point 95 is found in references 1, 3, 7, 15, 24, 33, 35, 47, 52, 53, 57, 60, 62, 63, 67, 68, 69, 74, 79, 81, 83, 84.
- (b) The temperature of this determination was not given.

### $C_{13}H_{12}$

#### 1,2-Cyclopentanonaphthalene



B. P., °C @ 760mm

<b>296–296.6</b>	$757.5^{17}$
294-295	75740
170	$15^{14}$
118	$0.5^{36}$

 $D_4^{20}$ 

$1.066^{40}$	
1.0569	19.6° 36
1.0671	17.5° 17
1.629040	

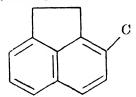
 $n_{\,\scriptscriptstyle
m D}^{\,20}$ 

1.6323	19.7° 14
1.62649	19.6° 36
1.6325	17.5° 17

## 2,3-Cyclopentanonaphthalene

# M. P., °C 9475, 76

### 3-Methylacenaphthene

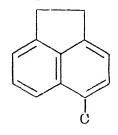


M. P., °C

B. P., °C @ 760mm 130–135

311

### 5-Methylacenaphthene

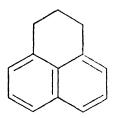


M. P., °C 88<sup>11</sup>

B. P., °C @ 760mm 125

12<sup>11</sup>

#### Phenalan



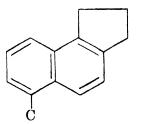
M. P., °C 65.4 68-69<sup>42</sup> 66-67<sup>26</sup>

 $65.1 - 65.4^{22}$ 

 $65^{12}$ 

 $C_{14}H_{14}$ 

## 1,2-Cyclopentano-5-methylnaphthalene



M. P., °C 44<sup>50</sup>

## 1-Ethylacenaphthene

M. P., °C 34.8-35.1<sup>23</sup> 30<sup>13</sup>

## 5-Ethylacenaphthene

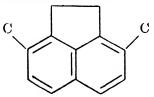
M. P., °C 42.5-43.0<sup>70</sup> 42.5-43<sup>26</sup>

B. P., °C @ 760mm

310<sup>18</sup>
166 20<sup>48</sup>
160 20<sup>70</sup>
158 13<sup>26</sup>

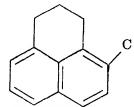
 $D_{4}^{20}$  1.040726  $n_{D}^{20}$  1.611726

## 3,8-Dimethylacenaphthene



M. P., °C 68<sup>11</sup>

## 4-Methylphenalan



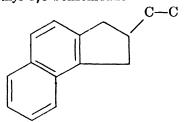
B. P., °C @ 760mm 160–170

 $C_{15}H_{16}$ 

79

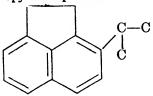
1449

# 2-Ethyl-4,5-benzoindane



B. P., °C @ 760mm 157–160

# 3-Isopropylacenaphthene



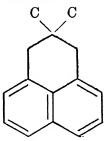
 $0.8^{11}$ 

## 2-Ethylphenalan

B. P., °C @ 760mm 167–168

 $16^{49}$ 

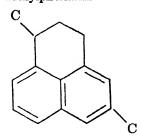
## 2,2-Dimethylphenalan



B. P., °C @ 760mm 168-170

178

## 1,5-Dimethylphenalan



M. P., °C 43-44<sup>10</sup>

B. P., °C @ 760mm 150-160

1.210

## 1,6-Dimethylphenalan

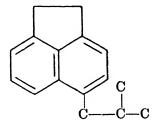
M. P., °C 75<sup>10</sup>

B. P., °C @ 760mm 150

 $0.8^{10}$ 

#### C16H18

## 5-Isobutylacenaphthene



B. P., °C @ 760mm 138-139

0.548

## 3-tert-Butylacenaphthene

M. P., °C 73-74<sup>66</sup>

B. P., °C @ 760mm

170-174

766

155-160

0.811

# 5,6-Diethylacenaphthene

$$C-C$$
  $C-C$ 

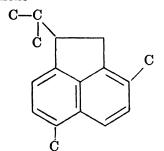
M. P., °C 10-11<sup>26</sup>, <sup>70</sup>

B. P., °C @ 760mm

1426, 70

#### C17H20

### 1-Isopropyl-3, 6-dimethylacenaphthene

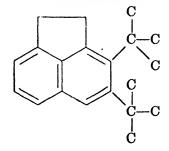


B. P., °C @ 760mm

 $0.05^{71}$ 

#### C20H26

## 3,4-Di-tert-butylacenaphthene



M. P., °C 162–16366

B. P., °C @ 760mm 358-360

75866

#### References on Cyclanonaphthalenes and Their Alkyl Derivatives

- Behr, A., and W. A. van Dorp, Ann. 172, 263 1874.
- Behr, A., and W. A. van Dorp, Ber. 6, 60 1873.
- 3. Behr and van Dorp through C. Graebe and F. Bossel, Ann. 290, 207 1896.
- 4. Berthelot, M., Ann. suppl. 5, 367 1867.
- 5. Berthelot, Bull. soc. chim. [2] 8, 226 1867.
- 6. Berthelot, Compt. rend. 65, 507 1867.
- Billows, E., Riv. min. crist. ital. 26, 5 1901.
- Buu-Hoi and P. Cagniant, Compt. rend.
   216, 346 1943; C.A. 38, 3377 1944.
- 9. Buu-Hoi and P. Cagniant, Rev. sci. 79, 614 1941; C.A. 38, 3612 1944.
- Buu-Hoi and P. Cagniant, Rev. sci. 80, 130 1942; C.A. 38, 3612 1944.
- Buu-Hoi and P. Cagniant, Rev. sci. 80, 176 1942; C.A. 38, 4933 1944; Chem. Zentr. 1943, I, 2680.
- 12. Clar, E., Ber. 76, 458 1943.
- Cook, J. W., G. A. D. Haslewood, and A. M. Robinson, J. Chem. Soc. 1935, 667.
- 14. Cook, J. W., and C. L. Hewett, J. Chem. Soc. 1933, 1098.
- Crompton, H, and W. R. Smyth, J. Chem. Soc. 103, 1302 1913.
- Crompton, H., and M. Walker, J. Chem. Soc. 101, 958 1912.
- Denisenko, Y. I., J. Gen. Chem. (U.S.-S.R.) 8, 410 1938; C.A. 32, 7899 1938;
   Chem. Zentr. 1939, II, 630.
- Dutoit, P., and L. Friederich, Arch. sci. phys. nat. [4] 9, 105 1900.
- Efremov, N. N., J. Russ. Phys. Chem. Soc. 50, 429 1920; Beil. 5, 275 1930.
- Efremov, N. N., and A. M. Tikhomirova, Ann. inst. anal. phys. chim. (U.S.S.R.) 4, 92 1928; C.A. 23, 3214 1929; Chem. Zentr. 1929, I, 745.
- Efremov, N. N., and A. N. Tikhomirova, J. Russ. Phys. Chem. Soc. 59, 373 1927; C.A. 22, 2508 1928; Chem. Zentr. 1928, I, 188.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 60, 1658 1938.
- Fieser, L. F., and G. W. Kilmer, J. Am. Chem. Soc. 62, 1354 1940.

- Fieser, L. F., and M. A. Peters, J. Am. Chem. Soc. 54, 4373 1932.
- Fleischer, K., and E. Retze, Ber. 55, 3280 1922.
- Fleischer, K., and P. Wolff, Ber. 53, 925 1920.
- Gina, M., Gazz. chim. ital. 46, I, 289
   1916.
- Goldschmiedt, G., and M. von Schmidt, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 83, Abt. IIb, 7 1881.
- 29. Graebe, C., Ann. 163, 361 1872.
- 30. Graebe, C., and F. Bossel, Ann. 290, 206 1896.
- 31. Hahn, D. A., and H. E. Holmes, Ind. Eng. Chem. 13, 822 1921.
- 32. Hendricks, S. B., and M. E. Jefferson, J. Optical Soc. Am. 23, 299 1933.
- 33. Hertel, E., and H. Kleu, Z. physik. Chem. 11B, 59 1930.
- 34. Kagehira, I., Bull. Chem. Soc. Japan 6, 241 1931.
- 35. Kebler, J. T., and T. H. Norton, Am. Chem. J. 10, 217 1888.
- 36. Kon, G. A. R., J. Chem. Soc. 1933, 1081.
- 37. Kopp, A., Mon. sci. [3] 8, 1147 1878.
- Kremann, R., E. Honigsberg, and O. Mauermann, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 132, Abt. IIb, 51 1923.
- Kremann, R., and H. Strzelba, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 130, Abt. Hb, 173 1921.
- 40. Kruber, O., Ber. 65, 1382 1932.
- Lagerlöf, D., J. prakt. Chem. [2] 98, 136 1918.
- Langstein, E., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 119, Abt. IIb, 675 1910.
- Larsen, R. G., R. E. Thorpe, and F. A. Armfield, Ind. Eng. Chem. 34, 183 1942.
- 44. Lecat, M., Ann. soc. sci. Bruxelles 49B, 109 1929.
- 45. Linder, E. G., J. Phys. Chem. 35, 531 1931.
- 46. Mameli, E., and A. Mossini, Giorn. chim. ind. applicata 15, 161 1933.
- 47. Manzoni-Ansidei, R., Gazz. chim. ital. 67, 790 1937.
- Mayer, F., and W. Kaufmann, Ber. 53, 289 1920.

- Mayer, F., and A. Sieglitz, Ber. 55, 1835
   1922.
- McQuillin, F. J., and R. Robinson, J. Chem. Soc. 1938, 1097.
- McVicker, W. H., J. K. Marsh, and A. W. Stewart, J. Chem. Soc. 127, 999 1925.
- Meyer, R., and K. Taeger, Ber. 53, 1261
   1920.
- Meyer, R., and A. Tanzen, Ber. 46, 3183
   1913.
- Mortimer, F. S., J. Am. Chem. Soc. 44, 1416 1922.
- Mortimer, F. S., and R. V. Murphy, Ind. Eng. Chem. 15, 1140 1923.
- Mukherjee, A., Indian J. Phys. 8, 147
   1933.
- 57. Nenhaus, A., Ber. 67, 1627 1934.
- Oliveri-Mandalâ, E., Atti accad. Lincei
   [5] 21, I, 779 1912.
- Oliveri-Mandalâ, E., Gazz. chim. ital.
   68, 321 1938.
- Orlov, N. A., K. F. Protyanova, and V. P. Flegontov, Refiner and Natural Gasoline Mfr. 17, 399 1938, translated from Khim. Tverdogo Topliva 7, 748 1936.
- Paillard, H., and R. Duckert, Helv. Chim. Acta 16, 773 1933.
- Paillard, H., and P. Favarger, Helv. Chim. Acta 16, 614 1933.
- 63. Pellini, G., Gazz. chim. ital. 31, I, 1 1901.
- 64. Perkin, W. H., J. Chem. Soc. **69**, 1025 **1896**.
- Pestemer, M., and F. Manchen, Sitzber.
   Akad. Wiss. Wien, Math. naturw.
   Klasse 145, Abt. IIb, 312 1936.
- 66. Peters, A. T., J. Chem. Soc. 1942, 562.
- 67. Pfeiffer, P., Ann. 440, 241 1924.
- 68. Quincke, F., Ber. 20, 609 1887.
- 69. Reingruber, F., Ann. 206, 367 1880.
- Rowe, F. M., J. Ind. Eng. Chem. 14, 80
   1922.
- Ruzicka, L., and E. Rey, Helv. Chim. Acta 26, 2136 1943.
- 72. Sabatier, P., and J. B. Senderens, Ann. chim. phys. [8] 4, 319 1905.
- 73. Schiff, R., Ann. 223, 247 1884.
- 74. Schönberg, A., Ber. 54, 2838 1921.
- 75. Sen-Gupta, S. C., Current Sci. 5, 133 1936; C. A. 31, 5789 1937.

C<sub>14</sub>H<sub>14</sub>

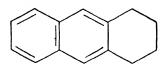
- Sen-Gupta, S. C., J. Indian Chem. Soc. 16, 89 1939.
- 77. Skau, E. L., J. Phys. Chem. 39, 761 1935.
- Speyers, C. L., J. Am. Chem. Soc. 18, 146 1896.
- 79. Terisse, A., Ann. 227, 133 1885.
- 80. Versmann, F., Chem. News 30, 203 1874.
- von Auwers, K., and A. Fruhling, Ann.
   422, 196 1921.
- 82. Walden, P., Z. physik. Chem. 65, 129 1909.
- 83. Waser, E., Helv. Chim. Acta 8, 758 1925.
- 84. Weger, M., Z. angew. Chem. 22, 338 1909.

### 3. TETRAHYDROANTHRACENES, TETRAHYDROPHENAN-THRENES, AND THEIR ALKYL DERIVATIVES, C<sub>n</sub>H<sub>2n-14</sub>

137

#### C14H14

# 1,2,3,4-Tetrahydroanthracene (Tethracene)



M. P., °C

103.5

106-10730

104-10521

103-10517, 28

10433

103-10418

1037, 16

 $100.5 - 101.5^{23}$ 

10110, 12

B. P., °C @ 760mm

170-173

1418

#### x,x,9,10-Tetrahydroanthracene (a)

M. P., °C

898, 9, 11, 28

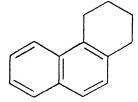
7488

B. P., °C @ 760mm

309-3138, 9, 11

(a) The structure of this compound was not clearly defined in the literature.

## 1,2,3,4-Tetrahydrophenanthrene (Tetanthrene)



M. P., °C

33 .5

 $34 - 35^6$ 

33-3420, 29, 81

32.5-33.51

3324

B. P., °C @ 760mm

173

11<sup>31</sup> 10<sup>6</sup>

 $D_4^{20}$  170–171

1.0601

40° 31

#### x<sub>4</sub>-Tetrahydrophenanthrene (a)

M. P., °C

703

 $14^{35}$ 

 $-4--3^{27}$ 

-5-427

B. P., °C @ 760mm

310<sup>4</sup>, 18, 14, 19 302–302.5<sup>25</sup>

30214

306–308 737<sup>27</sup> 307 737<sup>27</sup>

302-303 73727

300-304 7232

147  $18^{25}$ 

145-149 926

 $D_4^{20}$ 

1.08027	
1.08527	
1.0707	24.0° 85
1.02887	16.8° 25
1.02934	12.3° 25
1.067	10.2° 13

 $n_{\rm p}^{20}$ 

1.0707	24.0
1.02887	16.8° 25
1.02934	12.3° 25
1.067	10.2° 13
1.582027	
1.57909	16.8° 25
1.58112	12.3° 25
1.61880	$n_{\mathrm{H}lpha}^{24~0~35}$
1.57373	$n_{\mathrm{H}lpha}^{16.825}$
1.57560	$n_{{ m H}lpha}^{12325}$
1.64684	$n_{ ext{II}oldsymbol{eta}}^{24~0~35}$
1.59369	$n_{\mathbf{H}oldsymbol{eta}}^{16.8}$
1.59550	$n_{\mathbf{H}oldsymbol{eta}}^{12.3}$
1.62660	$n_{ m He}^{24.0~35}$
The standtune	of this comm

(a) The structure of this compound was not clearly defined in the The data probably literature. represent more than one compound.

## C16H18

9-Ethyl-1,2,3,4-tetrahydrophenanthrene

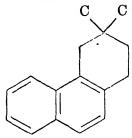
M. P., °C 23-251

2,2-Dimethyl-1,2,3,4-tetrahydrophenanthrene

B. P., °C @ 760mm 161-163

 $6^{32}$ 

3,3-Dimethyl-1,2,3,4-tetrahydrophenanthrene



B. P., °C @ 760mm 155-157

 $7^{32}$ 

## C<sub>17</sub>H<sub>20</sub>

2,2,9-Trimethyl-1,2,3,4-tetrahydrophenanthrene

M. P., °C 90 - 9182

## $C_{18}H_{22}$

1-Methyl-3-isopropyl-1,2,3,4-tetrahydrophenanthrene

B. P., °C @ 760mm 208

## 1-Methyl-7-isopropyl-x<sub>4</sub>-tetrahydrophenanthrene (a)

B. P., °C @ 760mm 360–366<sup>22</sup>

280 502

180–183

 $10^{34}$ 

235

 $D_4^{20}$ 

1.005784

1.01122

 $n_{\rm p}^{20}$ 

1.5606134

(a) The structure of this compound was not clearly defined in the literature.

## 5,6,7,8-Tetramethyl-1,2,3,4-tetrahydroanthracene

M. P., °C 127.5–128<sup>15</sup>

References on Tetrahydroanthracenes, Tetrahydrophenanthrenes, and Their Alkyl Derivatives

 Bachmann, W. E., and W. S. Struve, J. Org. Chem. 4, 472 1939.

- 2. Bamberger, E., and W. Lodter, Ber. 20, 3073 1887.
- 3. Breteau, P., Bull. soc. chim. [4] 9, 515 1911.
- 4. Breteau, P., Bull. soc. chim. [4] 9, 764 1911.
- Darzens, G., and A. Levy, Compt. rend. 201, 152 1935.
- Durland, J. R., and H. Adkins, J. Am. Chem. Soc. 60, 1501 1938.
- Fries, K., and K. Schilling, Ber. 65, 1494 1932.
- 8. Godehot, M., Ann. chim. phys. [8] 12, 468 1907.
- 9. Godchot, M, Bull. soc. chim. [3] 31, 1339 1904.
- Godchot, M., Bull. soc. chim. [4] 1, 719 1907.
- Godchot, M., Compt. rend. 139, 604 1904.
- 12. Godchot, M., Compt. rend. 142, 1202 1906.
- 13. Graebe, C., Ber. 8, 1054 1875.
- Grinbaum, R., and L. Marchlewski, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1937A, 171.
- 15. Hewett, C. L., J. Chem. Soc. 1940, 293.
- 16. Hugel, G., and J. Friess, Ann. combustibles liquides 6, 1109 1931.
- Ipatieff, V., V. Yakovlev, and L. Rakitin, Ber. 41, 996 1908.
- Klepper, J., Compt. rend. Congr. chim. ind. 8, 261 1929.
- Lebeau, P., and M. Picon, Compt. rend. 159. 70 1914.
- 20. Mameli, E., and A. Mossini, Giorn. chim. ind. applicata 15, 161 1933.
- Meerwein, H., and A. Migge, Ber. 62, 1046 1929.
- Olsson, J., Ing. Vetenskaps Akad., Handl. 1931, No. 111, 27 pp.; Chem. Zentr. 1931, II, 3420.
- Orchin, M., J. Am. Chem. Soc. 66, 535
   1944.
- 24. Orlov, N. A., Ber. 62, 710 1929.
- 25. Pellini, G., Gazz. chim. ital. 31, I, 1 1901.
- 26. Schlenk, W., and E. Bergmann, Ann. 463. 1 1928.
- 27. Schmidt, J., and R. Mezger, Ber. 40, 4240 1907.

- 28. Schroeter, G., Ber. 57, 2003 1924.
- 29. Schroeter, G., Ber. 57, 2025 1924.
- 30. Schroeter, G., Ber. 60, 2035 1927.
- 31. Schroeter, G., H. Müller, and J. Y. S. Huang, Ber. 62, 645 1929.
- 32. Sengupta, S. C., J. prakt. Chem. [2] 152, 9 1939.
- Sugino, K., and K. Outi, J. Chem. Soc.
   Japan 62, 401 1941; C. A. 37, 4726
   1943.
- 34. Virtanen, A. J., Ber. 53, 1880 1920.
- von Auwers, K., and R. Kraul, Ann. 443,
   181 1925.

# 4. MISCELLANEOUS POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-14}$

#### $C_{13}H_{12}$

#### x,x-Dihydrofluorene (a)

M. P., °C

1095, 7, 8

- B. P., °C @ 760mm 251-254<sup>5, 7, 8</sup>
- (a) The structure of this compound was not clearly defined in the literature.

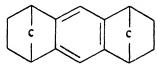
#### C<sub>16</sub>H<sub>18</sub>

## x<sub>8</sub>-Octahydrofluoranthene (a)

(Octahydroidryl)

- B. P., °C @ 760mm 309-3113
- (a) The structure of this compound was not clearly defined in the literature.

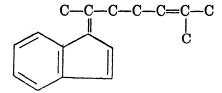
# 1,4,5,8-Diendomethylene-1,2,3,4,5,6,7,8-octahydroanthracene



M. P., °C 157<sup>2</sup>

#### C17H20

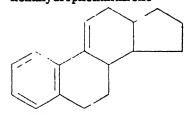
2-(1'-Indenylidene)-6-methylheptene-5



B. P., °C @ 760mm

130  $0.2^{\circ}$   $D_{4}^{20}$  0.9680  $24^{\circ}$   $n_{p}^{20}$  1.5942  $24^{\circ}$ 

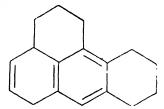
1,2-Cyclopentano-1,2,3,9,10,10ahexahydrophenanthrene



B. P., °C @ 760mm 164-165

34

4,5-Cyclohexano-7,9a-dihydrophenalan



B. P., °C @ 760mm 213-214

 $13^{10}$ 

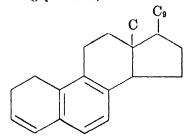
 $C_{20}H_{26}$ 

2a,3,4,5,6b,7,8,9,10,10a,11,12-Dodecahydrocholanthrene

M. P., °C 62-63<sup>1</sup>

#### C27H40

1,2-(3'-Nonylcyclopentano)-2methyl-1,2,3,4,5,6-hexahydrophenanthrene (Ergopentaene)



M. P., °C 89–90<sup>11</sup>

Additional Data

$$[\alpha]_{D}^{19} = +69.5^{\circ 11}$$
  
 $[\alpha]_{D}^{19} = +66.7^{\circ 11}$ 

#### C34H54

Tolylcholestane (a)

M. P., °C

202 (b)<sup>6</sup> 120 (b)<sup>6</sup>

- (a) The structure of this compound was not clearly defined in the literature.
- (b) These constants were determined on isomeric forms.

References on Miscellancous Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-14}$ 

- Cook, J. W., and G. A. D. Haslewood, J. Chem. Soc. 1935, 767.
- Diels, O., and K. Alder, Ann. 460, 98 1928.
- Goldschmidt, G., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 81, Abt. I, 415 1880.
- 4. Hawthorne, J. R., and R. Robinson, J. Chem. Soc. 1936, 763.
- Hofmann, F., and P. Damm, Brennstoff-Chem. 3, 73, 81 1922.
- Kawasaki, C., J. Pharm. Soc. Japan 57, 713 1937; C. A. 32, 187 1938; Chem. Zentr. 1938, II, 81.
- 7. Pictet, A., Ann. chim. [9] 10, 249 1918.
- Pictet, Λ., L. Ramseyer, and O. Kaiser, Compt. rend. 163, 358 1916.
- Ruzicka, L., and E. Peyer, Helv. Chim. Acta 18, 676 1935.
- von Braun, J., and O. Bayer, Ber. 58, 2667 1925.
- 11. Windaus, A., and M. Deppe, Ber. 70, 76 1937.

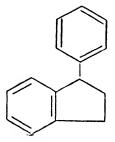
# V. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA C<sub>n</sub>H<sub>2n-16</sub>

- 1. Indane with One Phenyl Substitution
- 2. Naphthalene Derivatives of Empirical Formula  $C_nH_{2n-16}$
- 3. Tetrahydronaphthalene with One Phenyl Substitution
- 4. Biphenylene, Fluorene, and Their Alkyl Derivatives
- 5. Dihydroanthracenes, Dihydrophenanthrenes, and Their Alkyl Derivatives
- 6. Miscellaneous Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-16</sub>

## 1. INDANE WITH ONE PHENYL SUBSTITUTION, C<sub>n</sub>H<sub>2n-16</sub>

#### C15H14

#### 1-Phenylindane

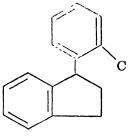


B. P., °C @ 760mm 148-150

 $13^{4}$ 

C16H16

#### 1-o-Tolylindane

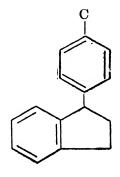


M. P., °C 579

B. P., °C @ 760mm 160-162

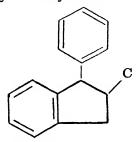
149

## 1-p-Tolylindane



B. P., °C @ 760mm 168-170 14°  $D_4^{20}$ 1.0455°  $n_D^{20}$ 1.5878°

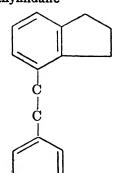
## 1-Phenyl-2-methylindane



B. P., °C @ 760mm 181

10³

# $C_{17}H_{18}$ 4-Phenethylindane



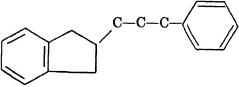
B. P., °C @ 760mm 115-120 0.0001<sup>5</sup> D<sub>4</sub><sup>20</sup> 25° <sup>5</sup> n<sub>p</sub><sup>20</sup> ...

1.5640 29° <sup>5</sup>

143

#### C<sub>18</sub>H<sub>20</sub>

#### 1-Phenyl-3-(2'-indanyl)-propane (a)



B. P., °C @ 760mm

197

C18H20

138

 $D_4^{20}$ 

1.0128 18° 8

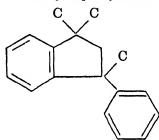
 $n_{\scriptscriptstyle 
m D}^{20}$ 

1.5667

18° 8

(a) This compound was given as 2γphenylpropylhydrindene in the literature, but no structural formula was given.

#### 1,1,3-Trimethyl-3-phenylindane



M. P., °C 53<sup>2</sup> 52<sup>10</sup>

B. P., °C @ 760mm

307-3106

166-167

 $25^2$ 

165-166

 $24^2$ 

155

122

 $D_4^{20}$ 

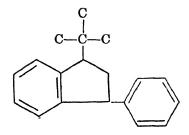
1.00096

 $n_{\rm p}^{20}$ 

1.568096

#### C19H22

#### 1-tert-Butyl-3-phenylindane

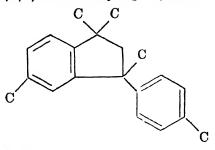


M. P., °C 181–182¹

144

#### C20H24

# 1,1,3,5-Tetramethyl-3-p-tolylindane



M. P., °C 37.5–38.5<sup>7</sup>

B. P., °C @ 760mm

328.5-329

7597

References on Indane with One Phenyl Substitution

- Althausen, D., and C. S. Marvel, J. Am. Chem. Soc. 54, 1174 1932.
- 2. Bergmann, E., Ber. 64, 1493 1931.
- 3. Blum-Bergmann, O., Ber. 65, 109 1932.
- Mayer, F., A. Sieglitz, and W. Ludwig, Ber. 54, 1397 1921.
- Natelson, S., and S. P. Gottfried, J. Am. Chem. Soc. 58, 1432 1936.
- Puranene, N., Ann. Acad. Sci. Fennicae 37A, No. 10, 1 1933; Chem. Zentr. 1933, II, 856.
- Puranene, N., Suomen Kemistilehti 5, 42B 1932; Chem. Zentr. 1933, I, 934.
- von Braun, J., and G. Manz, Ann. 468, 258 1929.
- von Braun, J., G. Manz, and E. Reinsch, Ann. 468, 277 1929.
- Welsh, L. H., and N. L. Drake, J. Am. Chem. Soc. 60, 59 1938.

# 2. NAPHTHALENE DERIVATIVES OF EMPIRICAL FORMULA $C_nH_{2n-16}$

145

### $C_{12}H_{8}$

## 1-Ethynylnaphthalene

B. P., °C @ 760mm 143-44

2515, 17

 $2^{22}$ 

 $1^{22}$ 

 $D_{4}^{20}$ 

1.05717

#### 2-Ethynylnaphthalene

$$\text{C}\!\!=\!\!\text{C}$$

M. P., °C 36<sup>16</sup>, <sup>22</sup>

B. P., °C @ 760mm 104–107 104–107

#### C13H10

#### 1-(Propyn-x'-yl)-naphthalene (a)

B. P., °C @ 760mm

154

154

154

1.056

1.056

13° 4

1.066

13° 4

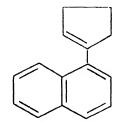
1.630

13° 4

(a) The double bond may be in either the 1- or 2-position.

#### C15H14

# 1-(Cyclopenten-1'-yl)-naphthalene



M. P., °C -2 (a)<sup>1</sup>

B. P., °C @ 760mm 165–168 1<sup>1</sup> 115 0.04<sup>2</sup>

 $D_4^{20}$ 

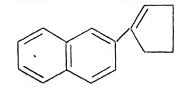
1.0611

24° 2

 $n_{\rm p}^{20}$  1.6285<sup>2</sup>

(a) This constant was given as a freezing point in the literature.

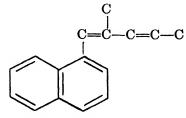
## 2-(Cyclopenten-1'-yl)-naphthalene



M. P., °C 85-86¹

#### C16H16

## 1-(1'-Naphthyl)-2-methylpentadiene-1,3



125

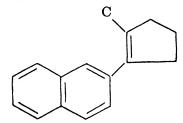
B. P., °C @ 760mm 178–181

$$D_4^{20}$$
0.9801 25° 5
 $n_D^{20}$ 
1.56967 25° 5

1-(1'-Naphthyl)-2-methylcyclopentene-1

B. P., °C @ 760mm 165-168 1<sup>1</sup>

1-(2'-Naphthyl)-2-methylcyclopentene-1



B. P., °C @ 760mm 180–182

1-(Cyclohexen-1'-yl)-naphthalene

11

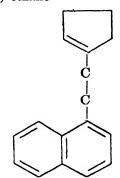
M. P., °C 46<sup>10</sup> 44<sup>10</sup>

3628

B. P., °C @ 760mm 332<sup>23</sup>

C<sub>17</sub>H<sub>18</sub>

1-(Cyclopenten-1'-yl)-2-(1"-naphthyl)-ethane



B. P., °C @ 760mm

164

4-58

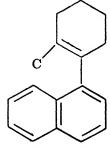
 $D_{4}^{20}$ 

 $1.0298^{8}$ 

 $n_{\scriptscriptstyle 
m D}^{20}$ 

1.60348

1-(1'-Naphthyl)-2-methylcyclohexene-1



M. P., °C 55-56<sup>18</sup>

B. P., °C @ 760mm 125

1.036

 $D_{ullet}^{20}$ 

15.5°11

 $0.3^{11}$ 

 $n_{\scriptscriptstyle \mathrm{D}}^{20}$ 

1.6111 15° 11

#### 1-(1'-Naphthyl)-4-methylcyclohexene-1

B. P., °C @ 760mm 142

 $0.15^{3}$ 

 $n_{\rm p}^{20}$ 

1.6040

14° 3

## 1-(1'-Naphthyl)-x-methylcyclohexene-1 (a)

B. P., °C @ 760mm 131

 $0.2^{3}$ 

 $n_{\rm p}^{20}$ 

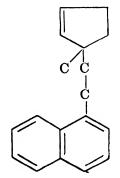
1.6054

14° 3

(a) The methyl group may be in either the 3- or 5-position.

#### C18H20

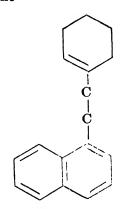
## 1-(1'-Methylcyclopentene-2'-yl)-2-(1''-naphthyl)-ethane



B. P., °C @ 760mm 168-172

 $0.5^{14}$ 

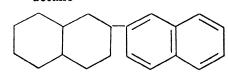
### 1-(Cyclohexen-1'-yl)-2-(1''-naphthyl)ethane



B. P., °C @ 760mm 167-168  $3-4^{9}$  135  $0.1^{7}$   $D_{4}^{20}$  1.0254  $23.5^{\circ 9}$  1.0158  $8^{\circ 7}$   $n_{D}^{20}$  1.5993  $23.6^{\circ 9}$ 1.5992  $8^{\circ 7}$ 

#### $C_{20}H_{24}$

3-(2'-Naphthyl)-bicyclo-[4,4,0]-decane



M. P., °C 67 (a)<sup>13</sup> 62 (a)<sup>13</sup>

(a) These constants were determined on isomeric forms.

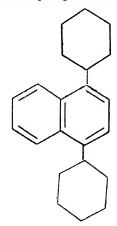
#### C22H28

1-(2'-Methyl-5'-isopropylcyclohexen-1'-yl)-2-(1''-naphthyl)-ethane

B. P., °C @ 760mm 160–165

 $0.15^{12}$ 

#### 1,4-Dicyclohexylnaphthalene



M. P., °C 83–83.5<sup>21</sup>

## x,x-Dicyclohexylnaphthalene (a)

M. P., °C

151 - 1526

150-15119

B. P., °C @ 760mm 215–225

 $7^{20}$ 

(a) The structure of this compound was not clearly defined in the literature.

References on Naphthalene Derivatives of Empirical Formula  $C_nH_{2n-16}$ 

- Bachmann, W. E., and M. C. Kloetzel, J. Am. Chem. Soc. 60, 2204 1938.
- Bergmann, F., and F. Bergmann, J. Am. Chem. Soc. 59, 1443 1937.
- Bergmann, F., and A. Weizmann, J. Org. Chem. 9, 352 1944.
- 4. Bert, L., and C. Dorier, Bull. soc. chim. [4] 37, 1600 1925.
- 5. Bjelouss, E., Ber. 45, 625 1912.
- 6. Bodroux, D., Ann. chim. [10] 11, 511 1929.
- Cook, J. W., G. A. D. Haslewood, and A. M. Robinson, J. Chem. Soc. 1935, 667.
- Cook, J. W., and C. L. Hewett, J. Chem. Soc. 1933, 1098.
- Cook, J. W., and C. L. Hewett, J. Chem. Soc. 1934, 365.
- Cook, J. W., C. L. Hewett, and C. A. Lawrence, J. Chem. Soc. 1936, 71.
- Cook, J. W., and C. A. Lawrence, J. Chem. Soc. 1936, 1431.
- Cook, J. W., and A. M. Robinson, J. Chem. Soc. 1938, 505.
- 13. Gysin, E., Helv. Chim. Acta 9, 59 1926.
- Kon, G. A. R., and E. S. Narracott, J. Chem. Soc. 1938, 672.
- Leroy, A. J., Bull. soc. chim [3] 6, 385
   1891.
- Leroy, J. A., Bull. soc. chim. [3] 7, 644
   1892.
- Leroy, J. A., Compt. rend. 113, 1056 1891.
- Mayer, F., and R. Schieffner, Ber. 65, 1337 1932.
- Pokrovskaya, E. S., and T. G. Stepanseva, J. Gen. Chem. (U.S.S.R.) 9, 1953 1939; C. A. 34, 4731 1940.
- Price, C. C., and J. M. Ciskowski, J. Am. Chem. Soc. 60, 2499 1938.
- Price, C. C., H. M. Shafer, M. F. Huber, and C. Bernstein, J. Org. Chem. 7, 517 1942.
- 22. Robin, J., Ann. chim. [10] 16, 421 1931.
- Weiss, R., and K. Woidich, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 134, Abt. IIb, 453 1925.

# 3. TETRAHYDRONAPHTHALENE WITH ONE PHENYL SUBSTITUTION, $C_nH_{2n-16}$

## $C_{16}H_{16}$

## 2-Phenyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 180-181

- . -

 $D_{4}^{20}$ 

1.0579 18° 10

 $n_{\,{\scriptscriptstyle \mathrm{D}}}^{\,20}$ 

1.5980

18° 10

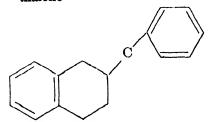
 $13^{10}$ 

## x-Phenyl-1,2,3,4-tetrahydronaphthalene (a)

- B. P., °C @ 760mm 184-187 14<sup>7</sup>
- (a) The phenyl group may be in either the 2- or 6-position.

### $C_{17}H_{18}$

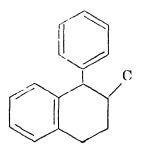
## 2-Benzyl-1,2,3,4-tetrahydronaphthalene



B. P., °C @ 760mm 195 194–195 193 134

D<sub>4</sub><sup>20</sup> 1.0428 22.5° 4

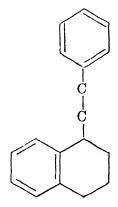
# 1-Phenyl-2-methyl-1,2,3,4-tetrahy-dronaphthalene



M. P., °C 46 -47<sup>2</sup>

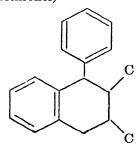
#### C18H20

## 1-Phenethyl-1,2,3,4-tetrahydronaphthalene



B. P., °C @ 760mm 204-208 14° D<sub>4</sub><sup>20</sup> 1.028 18°° n<sub>D</sub><sup>20</sup> 1.5770 18°°

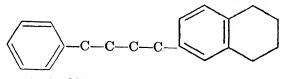
## 1-Phenyl-2,3-dimethyl-1,2,3,4tetrahydronaphthalene (Methronol)



B. P., °C @ 760mm 322–323² This series continued on next page

#### $C_{20}H_{24}$

# $1\hbox{-}Phenyl\hbox{-}4\hbox{-}[6'\hbox{-}(1',2',3',4'\hbox{-}tetrahydronaphthyl)]-butane$



B. P., °C @ 760mm 236–237

136

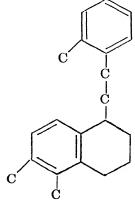
 $D_4^{20}$ 

1.0172

22.5° 6

#### $C_{21}H_{26}$

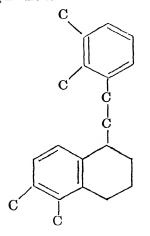
## 1-(2'-Methylphenethyl)-5,6-dimethyl-1,2,3,4-tetrahydronaphthalene



M. P., °C 53-54<sup>5</sup>

 $C_{22}H_{28}$ 

1-(2',3'-Dimethylphenethyl)-5,6dimethyl-1,2,3,4-tetrahydronaphthalene



B. P., °C @ 760mm 165

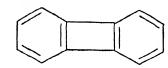
15

References on Tetrahydronaphthalene with One Phenyl Substitution

- 1. Borsche, W., P. Hofmann, and H. Kuhn, Ann. 554, 23 1943.
- 2. Erdmann, H., Ann. 227, 247 1885.
- 3. Hewett, C. L., J. Chem. Soc. 1936, 596.
- 4. Hugel, G., Chimie & industrie 26, 1282 1931.
- Ruzicka, L., A. Grob, and G. Anner, Helv. Chim. Acta 26, 254 1943.
- 6. Schroeter, G., Ber. 57, 1990 1924.
- Spath, E., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 121, Abt. IIb, 683 1912.
- 8. von Braun, J., Ber. 61, 441 1928.
- von Braun, J., and G. Koehendorfer, Ber. 56, 2172 1923.
- von Braun, J., and G. Manz, Ann. 468, 258 1929.

# 4. BIPHENYLENE, FLUORENE, AND THEIR ALKYL DERIVATIVES, $C_nH_{2n-16}$

# Biphenylene

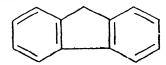


C12H8

M. P., °C 111-112<sup>66</sup> 109-110<sup>43</sup> 74.5-75<sup>22</sup> 74-75<sup>59</sup>

 $C_{13}H_{10}$ 

#### Fluorene



M. P., °C 114 116-117<sup>4, 49</sup> 116<sup>16, 76, 77, 81</sup> 115-11680

115.0 - 115.861

115-115.577

 $115.4^{37}$ 

11520, 35, 44, 46, 67, 75

 $114 - 115^{78}$ 

113-11533

114.89

 $114.5^{56}$ 

 $114 - 114.5^{15}$ 

113.5-114.519, 77

114.255

11425, 68, 69, 72

113-11426, 30, 42, 48

113-113.577

113.354

113(a)

112-113(b)

112.539, 40

112.323, 24

112.240

112.040

11247, 51

 $C_{18}H_{10}$  152

B. P., °C @ 760mm 297.2 30512, 14, 29, 74 298.05529632 29552, 76 294-29528, 63 293-29549, 75  $294^{62}$ 812.355 300.4801.355 299.8 738.855295.7 295.6738.855490.755277.1487.555 276.6 208.455 241.4  $205.5^{55}$ 240.4 $70.5^{55}$ 203.0 69.555202.52665 170 18.055 161.0

 $D_4^{20}$ 

1.181(c) (solid)41

1.200(c) (solid)<sup>57</sup> 1.207(c) (solid)<sup>34</sup>

 $n_{\,{\scriptscriptstyle D}}^{\,20}$ 

(d)

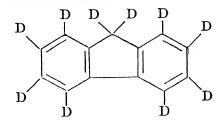
Additional Data

$$\frac{1}{T_b} = 0.0027238$$

 $-0.0003369 \log_{10} p_{mm}$  (200 to 813 mm)

- (a) The melting point 113 is found in references 1, 2, 3, 8, 10, 11, 12, 13, 14, 17, 27, 32, 36, 38, 45, 50, 52, 53, 60, 62, 65, 70, 71, 73, 74.
- (b) The melting point 112-113 is found in references 5, 6, 7, 21, 28, 31, 63, 64, 79.
- (c) The temperature of this determination was not given.
- (d) Refractive indices at other lines are found in reference 58.

#### Decadeuterofluorene



M. P., °C 115–117<sup>18</sup>

References on Biphenylene and Fluorene

- 1. Adam, P., Ann. chim. phys. [6] 15, 224 1888.
- Adam, P., Bull. soc. chim. [2] 47, 686
   1887.
- 3. Anschutz, R., and G. Schultz, Ann. 196, 32 1879.
- 4. Askew, F. A., J. Chem. Soc. 1935, 512.
- Bamberger, E., and S. C. Hooker, Ann. 229, 102 1885.
- Bamberger, E., and S. C. Hooker, Ber. 18, 1030 1885.
- Barbier, P., Ann. chim. phys. [5] 7, 479 1876.
- 8. Barth, L., and G. Goldschmidt, Ber. 11, 846 1878.
- 9. Berl, E., and A. Kullman, Ber. 60, 811 1927.
- Berthelot, Ann. chim. phys. [4] 12, 173
   1867.
- 11. Berthelot, Ann. chim. phys. [4] 12, 193
- 12. Berthelot, M., Ann. suppl. 5, 367 1867.
- Berthelot, Bull. soc. chim.[2] 8, 226
   1867.
- 14. Berthelot, Compt. rend. 65, 465 1867.
- Capper, N. S., and J. K. Marsh, J. Chem. Soc. 129, 724 1926.
- Charlampowicz, B., and L. Marchlewski, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1930A, 376;
   C. A. 25, 5096 1931; Chem. Zentr. 1931, I, 425.
- Charrier, G., and E. Ghigi, Gazz. chim. ital. 63, 685 1933.
- Clemo, G. R., and A. C. Robson, J. Chem. Soc. 1939, 429.
- Cook, J. W., and C. L. Hewett, J. Chem. Soc. 1936, 62.

153 References

- 20. Delacre, Bull. soc. chim. [3] 27, 875 1902.
- Dimroth, O., and B. Kerkovius, Ann. 399, 120 1913.
- Dobbie, J. J., J. J. Fox, and A. J. H. Gauge, J. Chem. Soc. 97. 683 1911.
- Efremov, N. N., and A. M. Tikhomirova, Ann. inst. anal. phys. chim. (U.S.S.R.) 4, 92 1928; C. A. 23, 3214 1929; Chem. Zentr. 1929, I, 745.
- Efremov, N. N., and A. N. Tikhomirova, J. Russ. Phys. Chem. Soc. 59, 373 1927; C. A. 22, 2508 1928.
- Elagina, N. V., and N. D. Zelinskii, Compt. rend. acad. sci. U.R.S.S. 23, No. 8, 799 1939.
- 26. Fittig, R., Ber. 6, 187 1873.
- 27. Fittig, R., and H. Liepmann, Ann. 200, 1 1880.
- 28. Fittig, R., and A. Schmitz, Ann. 193, 134 1878.
- Goldschmiedt, G., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 84, Abt. IIb, 305 1881.
- 30. Graebe, C., Ann. 174, 177 1874.
- 31. Graebe, C., Ber. 7, 1623 1874.
- 32. Graebe, C., and C. Aubin, Ann. 247, 257 1888.
- Graebe, C., and B. von Mantz, Ann. 290, 238 1896.
- Hengstenberg, J., and H. Mark, Z. Krist. 70, 283 1929.
- Jeanes, A., and R. Adams, J. Am. Chem. Soc. 59, 2608 1937.
- 36. Kempf, R., J. prakt. Chem. [2] 78, 201
- 37. Kirby, W., J. Soc. Chem. Ind. 59, 168 1940.
- 38. Kopp, A., Mon. sei. [3] 8, 1147 1878.
- Kremann, R., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 120, Abt. IIb, 329 1911.
- Kremann, R., E. Honigsberg, and O. Mauermann, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 132, Abt. IIb, 51 1923.
- 41. Krishnan, K.S., and S. Banerjee, Trans. Roy. Soc. (London) 234, 265 1935.
- Larsen, R. G., R. E. Thorpe, and F. A. Armfield, Ind. Eng. Chem. 34, 183 1942.
- Lothrop, W. C., J. Am. Chem. Soc. 63, 1187 1941.

- 44. Mameli, E., Gazz. chim. ital. 67, 669 1937.
- 45. Manchot, W., and P. Krische, Λnn. 337, 170 1904.
- Manzoni-Ansidei, R., Gazz. chim. ital.
   67, 790 1937.
- McVicker, W. H., J. K. Marsh, and A. W. Stewart, J. Chem. Soc. 127, 999 1925.
- 48. Menczel, S., Z. physik. Chem. 125, 161 1927.
- Meyer, H., and A. Hoffmann, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 125/126, Abt. IIb, 449 1916-17.
- 50. Meyer, R., Ber. 45, 1609 1912.
- Meyer, R., and K. Taeger, Ber. 53, 1261 1920.
- Meyer, R., and H. Wesche, Ber. 50, 422 1917.
- Morgenstern, O., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 120, Abt. IIb, 523 1911.
- Mortimer, F. S., J. Am. Chem. Soc. 44, 1416 1922.
- Mortimer, F. S., and R. V. Murphy, Ind. Eng. Chem. 15, 1140 1923.
- Mortimer, S. P., J. Am. Chem. Soc. 45, 633 1923.
- Mukherjee, A., Indian J. Phys. 8, 147 1933.
- Narasimham, K. L., Indian J. Phys.
   233 1931.
- 59. Nierenstein, M., Ann. 386, 318 1912.
- Nierenstein, M., and C. W. Webster,
   J. Am. Chem. Soc. 67, 691 1945.
- Orchin, M., J. Am. Chem. Soc. 67, 499 1945.
- 62. Pictet, A., Ann. chim. [9] 10, 249 1918.
- Pictet, A., and L. Ramseyer, Arch. sci. phys. nat. [4] 34, 234 1912.
- Pictet, A., and L. Ramseyer, Ber. 44, 2486 1911.
- Ramart-Lucas and J. Hoch, Bull. soc. chim. [5] 2, 1376 1935.
- Rapson, W. S., R. G. Shuttleworth, and J. N. van Niekerk, J. Chem. Soc. 1943, 326.
- 67. Sadikov, W. S., and A. K. Michailov, Ber. 61, 1792 1928.
- 68. Schlenk, W., Jr., Ber. 64, 739 1931.
- Schmidt, J., and E. Fischer, Ber. 41,
   4227 1908.

 $C_{14}H_{12}$  154

70. Schmidt, J., and R. Mezger, Ber. 40, 4566 1907.

- Schwarz, R., D. Pflugmacher, and M. Ruhnke, J. prakt. Chem. [2] 161, 137 1942.
- Staudinger, H., and R. Endle, Ber. 46, 1437 1913.
- 73. Staudinger, H., and A. Gaule, Ber. 49, 1951 1916.
- 74. Versmann, F., Chem. News 30, 203 1874.
- 75. Waser, E., Helv. Chim. Acta 8, 758 1925.
- Weger, M., Z. angew. Chem. 22, 338
   1909.
- 77. Weger, M., and K. Döring, Ber. 36, 878 1903.
- 78. Wieland, H., C. Schoff, and W. Hermsen, Ann. 444, 40 1925.
- 79. Zeidler, O., Ann. 191, 285 1878.
- 80. Zelinskii, N., Ber. 57, 264 1924.
- Zelinskii, N. D., I. Tits, and M. Gaverdovskaya, Ber. 59, 2590 1926.

#### C14H12

## 2,7-Dimethylbiphenylene

M. P., °C 112<sup>17</sup>

#### 2-Methylfluorene

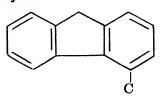
M. P., °C 104-105<sup>21</sup>· <sup>22</sup> 104<sup>14</sup> 102-103<sup>11</sup>

B. P., °C @ 760mm 317-319<sup>14</sup>

#### 3-Methylfluorene

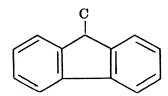
M. P., °C 88–89<sup>20</sup>. <sup>23</sup> 88<sup>33</sup>

#### 4-Methylfluorene



M. P., °C 71.5-72.5<sup>29</sup> 63<sup>19</sup>

#### 9-Methylfluorene



M. P., °C
45.5
48<sup>12</sup>
46-47<sup>35</sup>
46<sup>25</sup>. <sup>37</sup>
45-46<sup>5</sup>. <sup>34</sup>
45<sup>2</sup>, <sup>30</sup>, <sup>32</sup>, <sup>36</sup>
44.5<sup>4</sup>
44<sup>31</sup>

B. P., °C @ 760mm 157

> 141-144 16<sup>82</sup> 154-156 15<sup>86</sup>

1827

 $D_4^{20}$ 

1.070<sup>18</sup> 1.0263 66.2° <sup>82</sup>

$n_{{}_{ m D}}^{20}$	
$1.631^{18}$	
1.61009	$66.2^{\circ}$ 32
1.60233	$n_{ m Hlpha}^{ m 66.2~32}$
1.63189	$n_{\mathrm{H}\beta}^{66.2~32}$

#### C15H14

### 9-Ethylfluorene

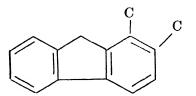
M. P., °C 108<sup>37</sup> 107–108<sup>35</sup> 106–107<sup>28</sup>

 $105^{27}$ 

B. P., °C @ 760mm 306-310<sup>25</sup>

 $\begin{array}{r}
 170 & 18^{27} \\
 169-171 & 14^{36} \\
 165-166 & 13^{35}
 \end{array}$ 

## 1,2-Dimethylfluorene (a)



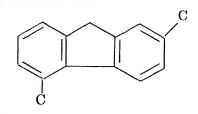
M. P., °C 107-108<sup>10</sup>

(a) The structure of this compound was not clearly defined in the literature.

## 2,3-Dimethylfluorene

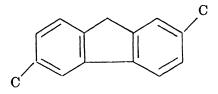
## M. P., °C 125-126<sup>1</sup> 125<sup>1</sup>, <sup>9</sup>

## 2,5-Dimethylfluorene



M. P., °C 58-59<sup>24</sup>

## 2,6-Dimethylfluorene

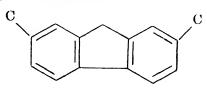


M. P., °C 66–67<sup>15, 16</sup>

# 3,5-Dimethylfluorene

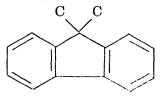
M. P., °C 81–82<sup>16</sup>

# 2,7-Dimethylfluorene



M. P., °C 114-115<sup>16</sup>

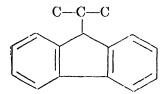
## 9,9-Dimethylfluorene



M. P., °C 70<sup>2</sup>

#### $C_{16}H_{16}$

## 9-Isopropylfluorene



M. P., °C 54-55<sup>4</sup> 53-55<sup>18</sup>

## $C_{17}H_{18}$

#### 9-n-Butylfluorene

M. P., °C 101<sup>28</sup>

B. P., °C @ 760mm 192–195

#### x-Isobutylfluorene (a)

(a) The structure of this compound was not clearly defined in the literature.

 $33^{28}$ 

#### 9-tert-Butylfluorene (a)

M. P., °C 101–102<sup>6</sup> 98–99<sup>26</sup>

(a) The structure of this compound was not clearly defined in the literature.

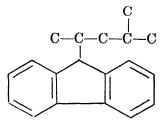
#### $C_{19}H_{22}$

#### 2-Methyl-2-(9'-fluoryl)-pentane (a)

M. P., °C 84-85<sup>6</sup>, <sup>7</sup>

(a) The structure of this compound was not clearly defined in the literature.

## 2-Methyl-4-(9'-fluoryl)-pentane



M. P., °C 101–103<sup>18</sup>

#### 9,9-Di-n-propylfluorene

M. P., °C 37–38³

#### References on C<sub>14</sub>H<sub>12</sub> through C<sub>19</sub>H<sub>22</sub> Compounds

- Alder, K., and H. F. Rickert, Ber. 71, 379 1938.
- 2. Blum-Bergmann, O., Ann. 484, 26 1930.
- Bradsher, C. K., and S. T. Amore, J. Am. Chem. Soc. 65, 2016 1943.
- Brown, W. G., and B. A. Bluestein, J. Am. Chem. Soc. 65, 1082 1943.
- Ferrer, J., Anales soc. españ. fís. quím.
   459 1922; Chem. Zentr. 1923,
   III, 1161.
- France, H., P. Maitland, and S. II. Tucker, J. Chem. Soc. 1937, 1739.
- France, H., S. H. Tucker, and J. Forrest, J. Chem. Soc. 1945, 7.
- Freund, M., K. Fleischer, and E. Stemmer, Ann. 414, 1 1918.
- 9. Fujise, S., Ber. 71, 2461 1938.
- 10. Ghigi, E., Ber. 70, 2469 1937.
- Hinkel, L. E., E. E. Ayling, and J. H. Beynon, J. Chem. Soc. 1936, 339.
- Hugel, G., and Gidaly, Bull. soc. chim.
   [4] 51, 639 1932.
- 13. Krollpfeiffer, F., Ann. 430, 161 1923.
- 14. Kruber, O., Ber. 65, 1382 1932.
- Longo, B., Atti accad. sci. Torino, Classe sci. fis., mat. nat. 73, 434 1938.
- Longo, B., Atti congr. intern. chim. Roma, 10th Congress 3, 239 1939.

- Lothrop, W. C., J. Am. Chem. Soc. 63, 1187 1941.
- Maitland, P., and S. H. Tucker, J. Chem. Soc. 1929, 2559.
- Mascarelli, L., and A. Angeletti, Atti accad. sci. Torino, Classe sci. fis., mat. nat. 72, 115 1936.
- Mascarelli, L., and B. Longo, Atti accad. Lincei [6] 26, 292 1937.
- Mascarelli, L., and B. Longo, Atti accad. sci. Torino, Classe sci. fis., mat. nat. 72, 109 1936.
- Mascarelli, L., and B. Longo, Gazz. ehim. ital. 67, 812 1937.
- Mascarelli, L., and B. Longo, Gazz. chim. ital. 68, 121 1938.
- 24. Mascarelli, L., and B. Longo, Gazz. chim. ital. 71, 289 1941.
- 25. Mayer, F., Ber. 46, 2579 1913.
- McEwen, W. K., J. Am. Chem. Soc. 58, 1124 1936.
- 27. Meerwein, II., Ann. 396, 200 1913.
- Miller, H. F., and J. Bachman, J. Am. Chem. Soc. 57, 766 1935.
- Orchin, M., and E. O. Woolfolk, J. Am. Chem. Soc. 67, 122 1945.
- Sieglitz, A., and H. Jassoy, Ber. 55, 2032 1922.
- Vansheidt, A., and B. Moldavski, Ber. 64, 917 1931.
- von Auwers, K., and A. Fruhling, Ann.
   422, 196 1921.
- Vorlander, D, and Λ. Pritzsche, Ber. 46, 1793 1913.
- Wicland, H., F. Reindel, and J. Ferrer, Ber. 55, 3313 1922.
- Wislicenus, W., and A. Deusch, Ber. 35, 759 1962.
- Wislicenus, W., and W. Mocker, Ber. 46, 2772 1913.
- 37. Zelinskii, N. D., and M. V. Gaverdovskaya, Ber. 61, 1049 1928.

# 5. DIHYDROANTHRACENES, DIHYDROPHENANTHRENES, AND THEIR ALKYL DERIVATIVES, C<sub>n</sub>H<sub>2n-16</sub>

#### $\mathbf{C}_{14}\mathbf{H}_{12}$

### 1,2-Dihydroanthracene

M. P., °C 150<sup>79</sup>

#### 9,10-Dihydroanthracene

M. P., °C 108

11118

108-11015

108.0-109.858

108-10927

107-10971

108.59, 52, 71, 82

108 (a)

107-10837, 45, 48

106-10863

10743, 74, 80

106-10714

 $106 - 106.5^{62}$ 

10625, 26, 66

 $105^{10}$ 

104-10553, 61

B. P., °C @ 760mm

31582

 $313^{43}$ 

 $305^{25}$ 

 $302 - 303^{26}$ 

165-170

 $D_{4}^{20}$ 

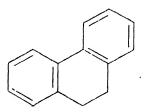
0.88951

10.3° 62

1337

(a) The melting point 108 is found in references 32, 40, 41, 47, 49, 70, 75, 80, 81.

#### 9,10-Dihydrophenanthrene



M. P., °C

9678

94-9568, 69

9431, 60

 $35^{4}$ 

34.5-3565

33.8-34.419

 $32 - 33^{22}$ 

## B. P., °C @ 760mm

 $314^{31}$ 

312-314 740<sup>68</sup> 312-314 739<sup>69</sup>

212-213 6012

176-178 2019

158 1165

154 7.5<sup>22</sup>

 $D_4^{20}$ 

140-142

1.0942

1.0940 24.5° 78

619

24.5° 78

 $n_{D}^{20}$ 

1.6406 25° 19

1.62093  $n_{\text{H}\alpha}^{24.5}$  78

1.62128  $n_{\text{H}\alpha}^{24.578}$ 

1.65269  $n_{\mathrm{H}\beta}^{24.5}$  78

1.65312  $n_{H\beta}^{24.578}$ 

1.62979  $n_{\text{He}}^{24.5 78}$ 

1.63015  $n_{\text{He}}^{24.5}$  78

## x,x-Dihydrophenanthrene (a)

M. P., °C

 $34.5 - 35^{72}$ 

B. P., °C @ 760mm

168–170 15<sup>67</sup> 168–169 15<sup>72</sup>

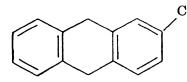
 $D_4^{20}$ 

1.0757 40° 72 1.0953 14° 72

(a) The structure of this compound was not clearly defined in the literature.

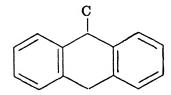
#### C15H14

### 2-Methyl-9, 10-dihydroanthracene



M. P., °C 51<sup>23</sup>

## 9-Methyl-9, 10-dihydroanthracene



M. P., °C 61.5–62<sup>78</sup>

## 1-Methyl-x,x-dihydroanthracene (a)

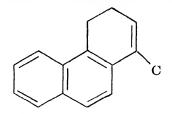
M. P., °C

3024

B. P., °C @ 760mm 314-315 740<sup>24</sup>

(a) The structure of this compound was not clearly defined in the literature.

### 1-Methyl-3,4-dihydrophenanthrene



M. P., °C 86.0-86.5<sup>5</sup>

#### C16H16

#### 9-Ethyl-x,x-dihydroanthracene (a)

B. P., °C @ 760mm 320–323<sup>45</sup> 320<sup>46</sup>

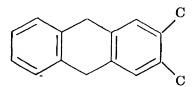
 $D_4^{20}$ 

1.049

D18 45, 46

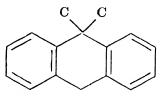
(a) The structure of this compound was not clearly defined in the literature.

#### 2,3-Dimethyl-9,10-dihydroanthracene



M. P., °C 1228

### 9,9-Dimethyl-9,10-dihydroanthracene



M. P., °C 56<sup>28</sup>

9,10-Dimethyl-9,10-dihydroanthracene

M. P., °C 181 181–182<sup>3</sup> 181–181.5<sup>2</sup> 181<sup>16</sup> 179–181<sup>17</sup> 178–179<sup>1</sup>, <sup>11</sup>

## 1-Ethyl-3,4-dihydrophenanthrene

M. P., °C 42–43<sup>5</sup>

#### 1,2-Dimethyl-3,4-dihydrophenanthrene

$$\begin{array}{c} \\ \\ \\ \\ \end{array}$$

M. P., °C 99–100<sup>30</sup>

 $C_{17}H_{18}$ 

9-n-Propyl-9, 10-dihydroanthracene

B. P., °C @ 760mm 175-176

1173

# 1-Isopropyl-3,4-dihydrophenanthrene

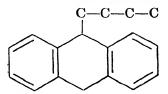
M. P., °C 66-67<sup>5</sup>

## 1-Ethyl-2-methyl-3,4-dihydrophenanthrene

M. P., °C 77-78<sup>30</sup>

## $C_{18}H_{20}$

# 9-n-Butyl-9,10-dihydroanthracene



1178

B. P., °C @ 760mm 191–192

## 9,9-Diethyl-9,10-dihydroanthracene

M. P., °C 210<sup>36</sup>

#### 9, 10-Diethyl-9, 10-dihydroanthracene

 $D_4^{20}$ 1.014<sup>64</sup>
1.0132

21° 44

# x,x-Diethyl-x,x-dihydroanthracene (a)

 $D_4^{20}$ 

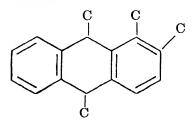
0.9634  $100^{\circ 21}$  0.9949  $50^{\circ 21}$  1.0264  $0^{\circ 21}$ 

(a) The structure of this compound was not clearly defined in the literature.

## 2,3,6,7-Tetramethyl-9,10-dihydroanthracene

B. P., °C @ 760mm 217-219<sup>54</sup>

#### 1,2,9,10-Tetramethyl-9,10-dihydroanthracene



M. P., °C 100–101<sup>7</sup>

## 1,3,5,7-Tetramethyl-x,x-dihydroanthracene (a)

M. P., °C 132–133 83

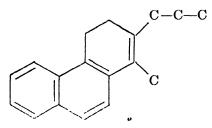
(a) The structure of this compound was not clearly defined in the literature.

## x<sub>4</sub>-Tetramethyl-x,x-dihydroanthracene (a)

M. P., °C 171-171.5³ 170.5-171²

(a) The structure of this compound was not clearly defined in the literature.

### 1-Methyl-2-n-propyl-3,4-dihydrophenanthrene



B. P., °C @ 760mm 208–210

1339

### 1-Isopropyl-7-methyl-3,4-dihydrophenanthrene

B. P., °C @ 760mm 150 2<sup>59</sup>

## 1-Methyl-7-isopropyl-x,x-dihydrophenanthrene (a)

M. P., °C 64-65<sup>38, 55, 76</sup>

B. P., °C @ 760mm 375–376<sup>57</sup> 188–190 10<sup>76</sup>

 $D_4^{20}$ 

 $1.027^{57}$ 

(a) The structure of this compound was not clearly defined in the literature.

### C19H22

## 9-Isopentyl-9, 10-dihydroanthracene

B. P., °C @ 760mm
200
201-205
23<sup>77</sup>
17<sup>77</sup>
D<sub>4</sub><sup>20</sup>

 $D_{f 4}^{z0}$  1.016 $^{77}$  1.025 $^{77}$  1.0022 .0.9940

45.1° 77 44.4° 77  $n_{\rm p}^{20}$ 1.573677 1.577177 45.1°77 1.56584 44.4°77 1.56261  $n_{\mathrm{H}\alpha}^{45.1}$  77 1.56091 $n_{\mathrm{H}lpha}^{44~4~77}$ 1.55791 45.1 77 1.58047  $n_{{
m H}eta}$ 44.4 77 1.57651  $n_{{
m H}eta}$ 44.4 77  $n_{
m H\gamma}$ 1.58868

# x-Pentyl-x,x-dihydroanthracene (a)

B. P., °C @ 760mm 350<sup>46</sup> 291–292 570<sup>46</sup>

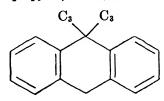
 $D_4^{20}$ 

1.031 18° 48

(a) The structure of this compound was not clearly defined in the literature.

#### C20H24

# 9,9-Dipropyl-9,10-dihydroanthracene



M. P., °C 46–47<sup>29</sup>

## x,x-Diisopropyl-9,10-dihydroanthracene (a)

M. P., °C 90<sup>20</sup>

(a) The structure of this compound was not clearly defined in the literature.

# 1-Methyl-4-ethyl-7-isopropyl-x,x-dihydrophenanthrene (a)

M. P., °C 51-52.5<sup>55</sup> 51-52<sup>55</sup>

5156

B. P., °C @ 760mm 223-225

 $7.5^{56}$ 

(a) The structure of this compound was not clearly defined in the literature.

#### C22H28

### 9,9-Diisobutyl-9,10-dihydroanthracene

M. P., °C 97-98<sup>50</sup>

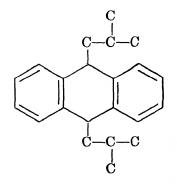
B. P., °C @ 760mm 140-143

 $0.01^{50}$ 

 $n_{\,{\scriptscriptstyle D}}^{20}$ 

 $1.5595^{50}$ 

#### 9,10-Diisobutyl-9,10-dihydroanthracene



B. P., °C @ 760mm 142 0.5<sup>42</sup>

 $D_4^{20}$ 

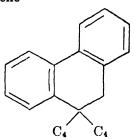
 $0.9899^{42} \\ 0.9902^{51}$ 

 $n_{\rm p}^{20}$ 

1.5633642

 $\begin{array}{r}
 1.55830 & n_{\pi_d}^{20} \\
 1.57600 & n_{\pi_d}^{20}
 \end{array}$ 

9,9-Dibutyl-9,10-dihydrophenanthrene



M. P., °C 76<sup>13</sup>

x,x-Diisobutyl-x,x-dihydrophenanthrene (a)

 $D_{4}^{20}$ 

0.9305

23° 34

- (a) The structure of this compound was not clearly defined in the literature.
- 1-Methyl-x,x-diethyl-7-isopropyl-x,x-dihydrophenanthrene (a)

B. P., °C @ 760mm 238-241

9 55

 $D_4^{20}$ 

1.005555

(a) The structure of this compound was not clearly defined in the literature.

## 1,2,3,4,5,6,7,8-Octamethyl-9,10dihydroanthracene

$$\begin{array}{c|c} c & c & c \\ \hline \\ c & c & c \\ \hline \end{array}$$

M. P., °C 283–284<sup>6</sup>

### C24H32

## 9,10-Di-(3'-methylbutyl)-9,10dihydroanthracene

 $D_{4}^{20}$ 0.9720  $0.972^{64}$ 100° 21 0.9130 50° 21 0.94990.9713 21° 35 0° 35 0.9868  $n_{\rm p}^{20}$ 25° 64 1.5640 10° 35 1.5670

Additional Data

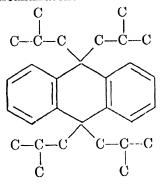
$$\frac{dD}{dt} = -0.0007380/^{\circ}\text{C}$$
(0 to 100 °C)

#### C30H44

# 2,3-Dioctyl-9,10-dihydroanthracene

 $D_4^{20}$   $0.948^{64}$  0.9016  $100^{\circ 21}$  0.9331  $50^{\circ 21}$  0.9646  $0^{\circ 21}$ 

## 9,9,10,10-Tetraisobutyl-9,10-dihydroanthracene



M. P., °C 177.5<sup>42</sup>

 $n_{\rm p}^{20}$ 

1.548 (a)  $n_{\text{H}\alpha}^{50}$ 1.603 (a)  $n_{\text{H}\gamma}^{50}$ 

(a) The temperature of this determination was not given.

## $C_{34}H_{52}$

# x.-Tetraisoamyl-x,x-dihydroanthracene (a)

M. P., °C 168–170<sup>33</sup>

(a) The structure of this compound was not clearly defined in the literature.

165 References

#### References on Dihydroanthracenes, Dihydrophenanthrencs, and Their Alkyl Derivatives

- Angelbis, A., and R. Anschütz, Ber. 17, 165 1884.
- 2. Anschütz, R., Ann. 235, 299 1886.
- Anschütz, R., and E. Romig, Ber. 18, 662 1885.
- 4. Askew, F. A., J. Chem. Soc. 1935, 512.
- Bachmann, W. E., and A. I. Wilds, J. Am. Chem. Soc. 60, 624 1938.
- Backer, H. J., J. Strating, and L. H. H. Huisman, Rec. trav. chim. 58, 761 1939.
- Badger, G. M., F. Goulden, and F. H. Warren, J. Chem. Soc. 1941, 18.
- Baker, W., R. Banks, D. R. Lyon, and F. G. Mann, J. Chem. Soc. 1945, 27.
- Bamberger, E., and W. Lodter, Ber. 20, 3073 1887.
- Blum-Bergmann, O., Ann. 484, 26 1930.
- Boeseken, J., and M. C. Bastet, Rec. trav. chim. 32, 184 1913.
- Burger, A., and E. Mosettig, J. Am. Chem. Soc. 58, 1857 1936.
- Chu, E. J. II., and Z.-I. Shen, J. Chinese Chem. Soc. 10, 116 1943; C. A. 38, 2053 1943.
- 14. Clemmensen, E., Ber. 47, 681 1914.
- 15. Cook, J. W., J. Chem. Soc. 1926, 1677.
- Cook, J. W., and V. J. Chambers, J. Am. Chem. Soc. 43, 334 1921.
- Davidson, J. M., and A. Lowy, J. Am. Chem. Soc. 51, 2978 1929.
- Dufraisse, C., L. Velluz, and Mme. L. Velluz, Bull. soc. chim. [5] 5, 1073 1938.
- Durland, J. R., and H. Adkins, J. Chem. Soc. 59, 135 1937.
- Errera, G., Gazz. chim. ital. 14, 277
   1884.
- Evans, E. B., J. Inst. Petroleum Tech.
   24, 537 1938.
- Fieser, L. F., and W. S. Johnson, J. Am. Chem. Soc. 61, 168 1939.
- Fischer, O., and K. Reinkober, J. prakt. Chem. [2] 92, 49 1915.
- Fischer, O., and H. Ziegler, J. prakt. Chem. [2] 86, 289 1912.
- Graebe, C., and C. Liebermann, Ann. suppl. 7, 257 1870.

- Graebe, C., and C. Liebermann, Ber. 1,
   49, 104, 186 1868.
- Grimm, H. G., M. Günther, and H. Tittus, Z. physik. Chem. 14B, 169 1931.
- 28. Hallgarten, F., Ber. 21, 2508 1888.
- 29. Hallgarten, F., Ber. 22, 1069 1889.
- 30. Haworth, R. D., C. R. Mavin, and G. Sheldrick, J. Chem. Soc. 1934, 454.
- 31. Henstock, H., J. Chem. Soc. 119, 1461 1921.
- 32. Hugel, G., and J. Friess, Ann. combustibles liquides 6, 1109 1931.
- Hugel, G., and M. Lerer, Bull. soc. chim. [4] 53, 1498 1933.
- Hugel, G., and M. Lerer, Bull. soc. chim. [4] 53, 1502 1933.
- Hugel, G., and M. Lerer, Compt. rend. 195, 249 1932.
- Kehrmann, F., R. Monier, and M. Ramm, Ber. 56, 169 1923.
- Klepper, J., Compt. rend. Congr. chim. ind. 8, 261 1929.
- Komppa, G., and H. P. Fogelberg, J. Am. Chem. Soc. 54, 2900 1932.
- Kon, G. A. R., E. S. Narracott, and C. Reid, J. Chem. Soc. 1938, 778.
- Kovache, A., and E. Tricot, Chimie & industrie 13, 361, 537 1925.
- 41. Krollpfeiffer, F., Ann. 430, 161 1923.
- Larsen, R. G., R. E. Thorpe, and F. A. Armfield, Ind. Eng. Chem. 34, 183 1942.
- 43. Lebeau, P., and M. Picon, Compt. rend. 159, 70 1914.
- 44. Lerer, M., Ann. combustibles liquides 8, 681 1933.
- 45. Liebermann, C., Ann. 212, 1 1882.
- Liebermann, C., and L. Landshoff, Ber.
   455 1881.
- 47. Liebermann, C., and Topf, Ber. 9, 1201 1876.
- 48. Mameli, E., and A. Mossini, Giorn. chim. ind. applicata 15, 161 1933.
- Manzoni-Ansidei, R., Gazz. chim. ital.
   67, 790 1937.
- 50. Martin, E., Ann. combustibles liquides 12, 97 1937.
- Martin, E., and M. G. Hugel, Bull. soc. chim. [4] 53, 1500 1933.

Michael, A., and A. Thal, Ber. 47, 473
 1914.

53 Miller, H. F., and J. Bachman, J. Am. Chem. Soc. 57, 766 1935.

 Morgan, G. T., and E. A. Coulson, J. Chem. Soc. 1931, 2323.

Nyman, G. A., Ann. Acad. Sci. Fennicae 41, No. 5, 1935, 74 pp.; Chem. Zentr. 1936, I, 2348.

Nyman, G. A., Ann. Acad. Sci. Fennicae, 48A, No. 6, 1938, 32 pp.; C. A.
 33, 8192 1939; Chem. Zentr. 1939, II, 3118.

 Olsson, J., Ing. Vetenskaps Akad. Handl. 1931, No. 111, 27 pp.; Chem. Zentr. 1931, II, 3420.

58. Orchin, M., J. Am. Chem. Soc. 66, 535 1944.

 Orcutt, R. M., and M. T. Bogert, J. Am. Chem. Soc. 63, 127 1941.

Padoa, M., Atti accad. Lincei [5] 28,
 11, 239 1919.

61. Padova, R., Ann. chim. phys. [8] 19, 353 1910.

62. Pellini, G., Gazz. chim. ital. 31, I, I 1901.

63. Perkin, A. G., J. Chem. Soc. 59, 634 1891.

 Rossini, F. D., Refiner Natural Gasoline Mfr. 14, 268 1935.

 Ruggli, P., and A. Staub, Helv. Chim. Acta 20, 37 1937. 66. Sachse, H., Ber. 21, 2512 1888.

67. Schlenk, W., and E. Bergmann, Ann. 463, 1 1928.

 Schmidt, J., and E. Fischer, Ber. 41, 4225 1908.

 Schmidt, J., and R. Mezger, Ber. 40, 4240 1907.

70. Scholl, R., and C. Seer, Ber. 55, 330 1922.

71. Schroeter, G., Ber. 57, 2003 1924.

 Schroeter, G., H. Müller, and J. Y. S. Huang, Ber. 62, 645 1929.

 Sieglitz, A., and R. Marx, Ber. 56, 1619 1923.

 Sugino, K., and K. Outi, J. Chem. Soc. Japan 62, 401 1941; C. A. 37, 4726 1943.

75. Ullman, F., Ann. 291, 17 1896.

76. Virtanen, A. J., Ber. 53, 1880 1920.

77. von Auwers, K., Ber. 53, 941 1920.

78. von Auwers, K., and R. Kraul, Ann. 443, 181 1925.

von Braun, J., and O. Bayer, Ann. 472,
 90 1929.

80. von Braun, J., and O. Bayer, Ber. 58, 2667 1925.

81, von Perger, H. R., J. prakt. Chem. [2] 23, 137 1881.

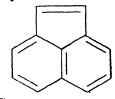
82. Waser, E., Helv. Chim. Acta 8, 758 1925.

83. Weiler, N., Ber. 33, 464 1900.

# 6. MISCELLANEOUS POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA C<sub>n</sub>H<sub>2n-16</sub>

#### $C_{12}H_8$

#### Acenaphthylene



M. P., ℃

93

93-9430

9314, 29

92-93 (a)

9231, 44

B. P., °C @ 760mm

27044

 $236^{27}$ 

 $175.5 - 176.5^{35}$ 

 $D_4^{20}$ 

0.88469

15.05° 56

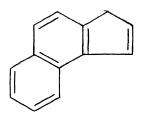
Additional Data

Sublimation Temp. (°C) 86.5° 38

(a) The melting point 92-93 is found in references 5, 6, 9, 13, 32, 33, 52, 55, 56.

#### C13H10

#### 4,5-Benzoindene

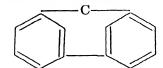


M. P., °C 48.5–50<sup>49</sup>

B. P., °C @ 760mm 173

3349

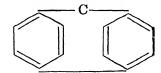
#### 3,3'-Diphenylenemethane



M. P., °C 118<sup>15</sup>

B. P., °C @ 760mm 295<sup>15</sup>

# 4,4'-Diphenylenemethane

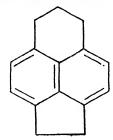


M. P., °C 208<sup>15</sup>, <sup>54</sup>

B. P., °C @ 760mm 320<sup>54</sup>

#### C<sub>15</sub>H<sub>14</sub>

# Cyclopentano-[gh]-phenalan

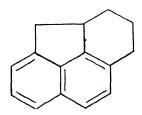


M. P., °C 122<sup>12</sup>

B. P., °C @ 760mm 168-170

1-312

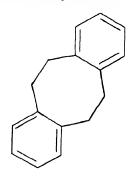
#### Cyclopentano-[def]-1,2,3,4-tetrahydrophenanthrene



M. P., °C 54.5-55.51

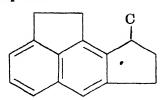
# C<sub>16</sub>H<sub>16</sub>

# 1,2,5,6-Dibenzocyclooctane



M. P., °C 108.5³

# 3,4-(3'-Methylcyclopentano)-acenaphthene

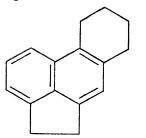


M. P., °C 38–38.5<sup>34</sup>

B. P., °C @ 760mm 143–148

134

Cyclopentano-[jk]-1,2,3,4-tetrahydrophenanthrene



M. P., °C 92.5<sup>37</sup>

Cycloheptano-[fg]-acenaphthene

M. P., °C 138<sup>36</sup>

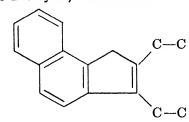
1,2,3,6,7,8-Hexahydropyrene

M. P., °C
133<sup>41</sup>
131–132<sup>47</sup>
129–130<sup>51</sup>
129<sup>57</sup>
127<sup>42</sup>, 45

1,2,3,9,10,10a-Hexahydropyrene

M. P., °C 106<sup>26</sup> 105–105.5<sup>22</sup>

2,3-Diethyl-6,7-benzoindene



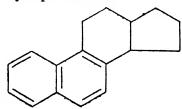
C17H18

B. P., °C @ 760mm 205-207 16<sup>53</sup>

2,3-Diethylphenalene

B. P., °C @ 760mm 185 13<sup>53</sup>

1,2-Cyclopentano-1,2,3,4-tetrahydrophenanthrene



M. P., °C 133-134.5<sup>20</sup>

B. P., °C @ 760mm 160-161 (a)

3-420

 $D_4^{20}$ 

1.0859 (a)20

 $n_{\rm p}^{20}$ 

1.6256 (a)

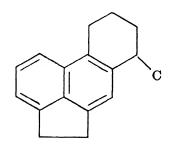
20.2° 20

(a) This constant was determined on the *trans* isomer of the compound.

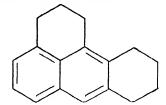
#### 2,3-Cyclopentano-1,2,3,4-tetrahydrophenanthrene

M. P., °C 119–1218

# Cyclopentano-[jk]-1-methyl-1,2,3,4-tetrahydrophenanthrene



M. P., °C 133<sup>27</sup> 4,5-Cyclohexanophenalan

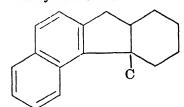


B. P., °C @ 760mm 228–230

1659

C<sub>18</sub>H<sub>20</sub>

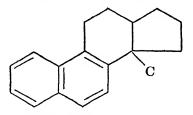
4a-Methyl-5,6-benzo-1,2,3,4,4a,9ahexahydrofluorene



B. P., °C @ 760mm 200-205

717

1-Methyl-1,2-cyclopentano-1,2,3,4tetrohydrophenanthrene



B. P., °C @ 760mm

155

 $0.4^{46}$ 

 $D_{f 4}^{20}$ 

 $1.0692^{50}$ 

1.04552

18.9° 46

 $n_{\,{\scriptscriptstyle \mathrm{D}}}^{\,\mathrm{20}}$ 

1.617650

1.61363

18.9° 46

Spiro[4,5-benzoindane-1,1'-cyclohexane] 170

M. P., °C 56-57<sup>16</sup>

B. P., °C @ 760mm 140

 $0.1^{16}$ 

x<sub>8</sub>-Octahydro-1,2-benzoanthracene
(a)

M. P., °C 124.5-125.5<sup>24</sup>

- (a) The structure of this compound was not clearly defined in the literature.
- 1,2,3,4,7,8,9,10-Octahydronaphthacene

M. P., °C 174<sup>60</sup>

1,2,3,4,7,8,9,10-Octahydrochrysene

M. P., °C 138-14Q<sup>61</sup> 136-138<sup>16</sup> B. P., °C @ 760mm 180-181

0.261

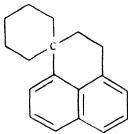
1,2,3,4,4a,11,12,12a-Octahydrochrysene

M. P., °C 114-114.5 (a)<sup>21</sup> 78-79 (b)<sup>21</sup>

- (a) This constant was determined on the *trans* isomer of the compound.
- (b) This constant was determined on the cis isomer of the compound.
- 1,2,3,4,5,6,7,8-Octahydrotriphenylene

M. P., °C 117.5–119.5<sup>2</sup>

Spiro[phenalan-1,1'-cyclohexane]



M. P., °C 
$$_{55-56^{21}}$$
B. P., °C @ 760mm  $_{176-177^{21}}$ 
 $D_{4}^{20}$ 
 $1.0809$ 
 $23.5^{\circ 21}$ 
 $n_{p}^{20}$ 
 $1.6197$ 
 $23.6^{\circ 21}$ 

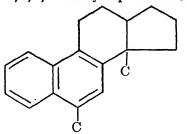
C19H22

### Cyclopentano-[gh]-2,2-diethylphenalan

M. P., °C 93-95<sup>39</sup>, 40

# 1,2-Cyclopentano-1,3-dimethyl-1,2,3,4-tetrahydrophenanthrene

B. P., °C @ 760mm 160 C.4<sup>46</sup>  $D_4^{20}$  1.04203  $18.2^{\circ}$  <sup>46</sup>  $n_D^{20}$ 1.60681  $18.2^{\circ}$  <sup>46</sup> 1,2-Cyclopentano-1,9-dimethyl-1,2,3,4-tetrahydrophenanthrene



B. P., °C @ 760mm 160 0.5<sup>42</sup>

Spiro[x-methyl-5,6-benzoindane-1,1'-cyclohexane] (a)

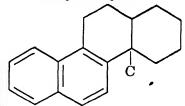
M. P., °C 109-110<sup>23</sup>

(a) The structure of this compound was not clearly defined in the literature.

Spiro[9-methyl-1,2,3,4-tetrahydro-phenanthrene-2,1'-cyclopentane]

M. P., °C 69-70<sup>58</sup>

4a-Methyl-1,2,3,4,4a,11,12,12aoctahydrochrysene



B. P., °C @ 760mm 145

 $0.1^{19}$ 

#### C20H24

1,2-(3'-Methylcyclopentano)-1,9dimethyl-1,2,3,4-tetrahydrophenanthrene

B. P., °C @ 760mm 170

0.642

#### C21H26

1,2-(3'-Isopropylcyclopentano)-4methyl-1,9,10,10a-tetrahydrophenanthrene

$$C-C-C$$

B. P., °C @ 760mm 190

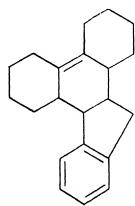
 $0.1^{10}$ 

1,2-Cyclohexano-5-(propen-2'-yl)-1,2,3,4,7,8-hexahydroanthracene

B. P., °C @ 760mm

 $0.1^{18}$ 

9,10-(1',2'-Indano)-1,2,3,4,5,6,7, 8,8a,9,10,10a-dodecahydrophenanthrene



B. P., °C @ 760mm 185-190 (a)<sup>7</sup> 180-185 (a)<sup>7</sup>

(a) These constants were determined on isomeric forms.

## $C_{22}H_{28}$

x-Methyl-x-isopropyl-x<sub>8</sub>-octahydrochrysene (a)

M. P., °C 108<sup>25</sup>

(a) The structure of this compound was not clearly defined in the literature.

#### x<sub>14</sub>-Tetradecahydropicene (a)

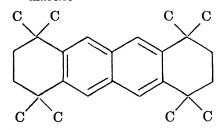
M. P., °C

2854

(a) The structure of this compound was not clearly defined in the literature.

#### C26H36

1,1,4,4,7,7,10,10-Octamethyl-1,2, 3,4,7,8,9,10-octahydronaphthacene



M. P., °C 319-320<sup>11</sup>

#### C<sub>33</sub>H<sub>50</sub>

#### 3-Phenyl-5-cholestene (a)

M. P., °C 15248

(a) The structure of this compound was not clearly defined in the literature.

# Phenylcholestane (a)

M. P., °C 153<sup>48</sup>

(a) The structure of this compound was not clearly defined in the literature.

#### $C_{35}H_{54}$

#### 3-Phenylcholestene-3 (a)

M. P., °C

 $94 - 95^{28}$ 

(a) The structure of this compound

was not clearly defined in the literature.

References on Miscellaneous Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-16</sub>

- Bachmann, W. E., and J. C. Sheenan,
   J. Am. Chem. Soc. 63, 204 1941.
- Bachmann, W. E., and W. S. Struve,
   J. Org. Chem. 4, 472 1939.
- Baker, W., R. Banks, D. R. Lyon, and F. G. Mann, J. Chem. Soc. 1945, 27.
- 4. Bamberger, E., and F. D. Chattaway, Ann. 284, 52 1895.
- Behr, A., and W. A. van Dorp, Ann. 172, 263 1874.
- Behr, A., and W. A. van Dorp, Ber. 6, 753 1873.
- Bergmann, E., and F. Bergmann, J. Am. Chem. Soc. 60, 1805 1938.
- Bergmann, E., and O. Blum-Bergmann,
   J. Am. Chem. Soc. 59, 1572 1937.
- 9. Billows, E., Riv. min. crist. ital. 26, 5 1901.
- Birch, A. J., and R. Robinson, J. Chem. Soc. 1944, 503.
- Bruson, H. A., and J. W. Kroeger, J. Am. Chem. Soc. 62, 36 1940.
- Buu-Hoi, and P. Cagniant, Compt. rend. 214, 493 1942; C. A. 37, 2370 1943.
- Campbell, A. W., N. H. Cromwell, and J. J. Hager, J. Am. Chem. Soc. 58, 1051 1936.
- 14. Campbell, B., J. Chem. Soc. 107, 918 1915.
- Carnelley, T., J. Chem. Soc. 37, 701 1880.
- Cohen, A., J. W. Cook, and C. L. Hewett, J. Chem. Soc. 1935, 1633.
- Cook, J. W., A. Dansi, C. L. Hewett,
   J. Iball, W. V. Mayneord, and E.
   Roe, J. Chem. Soc. 1935, 1319.
- Cook, J. W., and G. A. D. Haslewood,
   J. Chem. Soc. 1935, 767.
- Cook, J. W., G. A. D. Haslewood, and A. N. Robinson, J. Chem. Soc. 1935, 667
- Cook, J. W., and C. L. Hewett, J. Chem. Soc. 1933, 1098.

- 21. Cook, J. W., and C. L. Hewett, J. Chem. Soc. 1934, 365.
- Cook, J. W., C. L. Hewett, and I. Hieger, J. Chem. Soc. 1933, 395.
- Cook, J. W., C. L. Hewett, and A. M. Robinson, J. Chem. Soc. 1939, 168.
- Cook, J. W., and C. A. Lawrence, J. Chem. Soc. 1937, 817.
- Cook, J. W., and A. M. Robinson, J. Chem. Soc. 1938, 505.
- Coulson, E. A., J. Chem. Soc. 1937, 1298.
- 27. Criegee, R., Ann. 522, 75 1936.
- 28. Dansi, A., Gazz. chim. ital. 68, 273 1938.
- 29. Dziewoński, K., and L. Gizler, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1937 A, 441; Chem. Zentr. 1938, I, 1352.
- Dziewoński, K., and Z. Leyko, Bull. intern. acad. sci. Cracovie 1916A, 159.
- Dziewoński, K., and G. Rapalski, Ber. 45, 2491 1912.
- Dziewoński, K., and T. Stolyhwo, Ber. 57, 1531 1924.
- Dziewoński, K., and T. Stolyhwo, Bull. intern. acad. sci. Cracovie 1924A, 159.
- Fieser, L. F., and E. B. Hershberg,
   J. Am. Chem. Soc. 61, 1272 1939.
- Fieser, L. F., and M. S. Newman, J. Am. Chem. Soc. 57, 1602 1935.
- Fieser, L. F., and M. A. Peters, J. Am. Chem. Soc. 54, 4347 1932.
- Fieser, L. F., and M. A. Peters, J. Am. Chem. Soc. 54, 4373 1932.
- Fischer, F., H. Schrader, and W. Meyer, Ges. Abhandl. Kenntnis Kohle 5, 413 1920.
- 39. Fleischer, K., and F. Seifert, Λnn. 422, 317 1921.
- 40. Fleischer, K., and P. Wolff, Ber. 53, 925 1920.

- 41. Fromherz, H., L. Thaler, and G. Wolf, Z. Elektrochem. 49, 387 1943.
- 42. Gamble, D. J. C., and G. A. R. Kon, J. Chem. Soc. 1935, 443.
- 43. Goldschmiedt, G., Ann. 351, 218 1907.
- Goswami, M. N., Compt. rend. 179, 1269 1924.
- 45. Graebe, C., Ann. 158, 285 1871.
- Harper, S. H., G. A. R. Kon, and F. C. J. Ruzicka, J. Chem. Soc. 1934, 124.
- Kagehira, I., Bull. Chem. Soc. Japan 6, 241 1931.
- Kawasaki, C., and J. Yamamura, J. Pharm. Soc. Japan 58, 606 1938;
   C. A. 32, 8432 1938; Chem. Zentr. 1939, II, 646.
- Koelsch, C. F., and R. A. Scheiderbauer,
   J. Am. Chem. Soc. 65, 2311 1943.
- Kon, G. A. R., and E. S. Narracott,
   J. Chem. Soc. 1938, 672.
- Langstein, E., Sitzber. Akad. Wiss.
   Wien, Math. naturw. Klasse, Abt.
   IIb, 119, 675 1910.
- 52. Mameli, E., and A. Mossini, Giorn. chim. ind. applicata 15, 161 1933.
- Mayer, F., and A. Sieglitz, Ber. 55, 1835 1922.
- 54. Meyer, H., and A. Hoffmann, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 125-126, Abt. IIb, 449 1916-1917.
- Monti, L., Gazz. chim. ital. 68, 608
   1938.
- Pellini, G., Gazz. chim. ital. 31, I, 1 1901.
- Pestemer, M., and F. Manchen, Sitzber.
   Akad. Wiss. Wien, Math. naturw.
   Klasse 145, Abt. IIb, 312 1936.
- Sen-Gupta, S. C., Current Sei. 5, 295
   1936; C. A. 31, 2587 1937.
- von Braun, J., and O. Bayer, Ber. 58, 2667 1925.
- von Braun, J., O. Bayer, and L. F. Fieser, Ann. 459, 287 1927.
- 61. von Braun, J., and G. Irmisch, Ber. 65, 883 1932.

# VI. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA

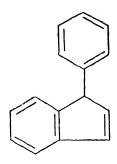
 $C_n H_{2n-18}$ 

- 1. Indene with One Phenyl Substitution
- 2. Anthracene and Its Alkyl Derivatives
- 3. Phenanthrene and Its Alkyl Derivatives
- 4. Miscellaneous Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-18</sub>

#### 1. INDENE WITH ONE PHENYL SUBSTITUTION, C<sub>n</sub>H<sub>2n-18</sub>

#### $C_{15}H_{12}$

## 1-Phenylindene

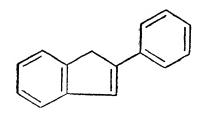


B. P., °C @ 760mm 200–201 2919

 $D_{f 4}^{f 20}$ 

1.0829 27° 19

#### 2-Phenylindene



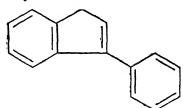
M. P., °C 167<sup>20</sup> 166–167<sup>6</sup>

B. P., °C @ 760mm 167-170 12<sup>15</sup> 162-163 10<sup>20</sup>

 $D_{\bullet}^{20}$   $10^{20}$   $D_{\bullet}^{20}$  1.0821  $16^{\circ 20}$ 

 $n_{\rm p}^{20}$  1.5955  $16^{\circ 20}$ 

#### 3-Phenylindene



B. P., °C @ 760mm 164-167 12<sup>5</sup> 167-171 10<sup>6</sup>

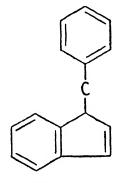
#### x-Phenylindene (a)

B. P., °C @ 760mm 183-187 174

(a) The structure of this compound was not clearly defined in the literature.

#### C16H14

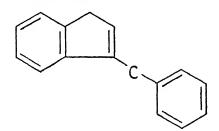
#### 1-Benzylindene



B. P., °C @ 760mm

230–235 15<sup>14</sup> 175–177 14° 173–175 12°

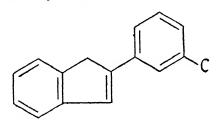
#### 3-Benzylindene



M. P., °C 35<sup>22</sup> 34<sup>9,10</sup> 33–34<sup>21</sup>

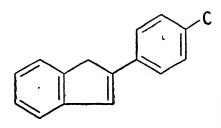
B. P., °C @ 760tnm 185 18<sup>21</sup> 175–177 14<sup>10</sup> 183–185 13<sup>17</sup> 179 11<sup>22</sup>

# 2-m-Tolylindene



M. P., °C 99–100<sup>20</sup>

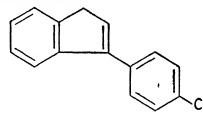
# 2-p-Tolylindene



M. P., °C 183–184<sup>20</sup>

#### 3-p-Tolylindene

177

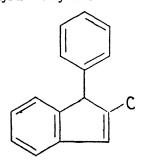


 $11^{20}$ 

1415

B. P., °C @ 760mm 184-188

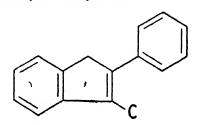
# 1-Phenyl-2-methylindene



M. P., °C 57.5<sup>15</sup> 56.5<sup>23</sup>

B. P., °C @ 760mm 177

# 2-Phenyl-3-methylindene



M. P., °C 76–78<sup>6</sup> 75–76<sup>1</sup>

C<sub>17</sub>H<sub>16</sub>

3-Phenethylindene

B. P., °C @ 760mm 186

97

 $n_{\,{\scriptscriptstyle D}}^{\,20}$ 

1.5987 18.7° 7

# (3-Indenyl)-phenylmethylmethane

B. P., °C @ 760mm

196 161  $15^{18}$   $5^{18}$ 

# 1-(4'-Methylbenzyl)-indene

M. P., °C 27-29³

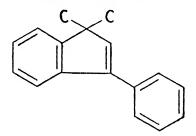
B. P., °C @ 760mm 218–222

128

#### 1,1-Dimethyl-2-phenylindene

M. P., °C 61-62<sup>12</sup> 51<sup>11</sup>

# 1,1-Dimethyl-3-phenylindene



M. P., °C  $50-51^{2, 13}$ 

B. P., °C @ 760mm 184–185

2713

# 1,2-Dimethyl-3-phenylindene

M. P., °C 68-69<sup>16</sup>

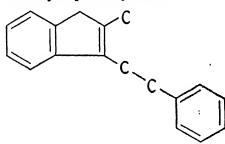
B. P., °C @ 760mm 145-153

#### 1-Phenyl-2,3-dimethylindene

M. P., °C 68–69<sup>13</sup>

#### C18H18

#### 2-Methyl-3-phenethylindene

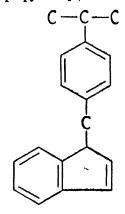


B. P., °C @ 760mm 170

18

#### $C_{19}H_{20}$

#### 1-(4'-Isopropylbenzyl)-indene



#### M. P., °C 32³

#### References on Indene with One Phenyl Substitution

- Allen, C. F. H., C. G. Eliot, and A. Bell, Can. J. Research 17B, 75 1939.
- Bergmann, E., H. Taubadel, and H. Weiss, Ber. 64, 1493 1931.
- 3. Bernthsen, W., Ann. 415, 274 1918.
- 4. Blum-Bergmann, O., Ann. 484, 26 1930.
- Blum-Bergmann, O., Ann. 492, 277 1932.
- 6. Blum-Bergmann, O., Ber. 65, 109 1932.
- Cook, J. W., and C. L. Hewett, J. Chem. Soc. 1934, 365.
- Cook, J. W., C. L. Hewett, W. V. Mayneord, and E. Roe, J. Chem. Soc. 1934, 1727.
- 9. Courtot, C., Ann. chim. [9] 5, 52 1916.
- Courtot, C., Compt. rend. 160, 523
   1915.
- Earl, J. C., and C. A. Smythe, J. Proc. Roy. Soc. N. S. Wales 64, 90 1930.
- Koelsch, C. F., and P. R. Johnson.
   J. Am. Chem. Soc. 65, 565 1943.
- Koelsch, C. F., and P. R. Johnson, J. Am. Chem Soc. 65, 567 1943.
- 14. Marckwald, W., Ber. 33, 1501 1900.
- Mayer, F., A. Sieglitz, and W. Ludwig, Ber. 54, 1397 1921.
- Smith, L. I., and L. I. Hanson, J. Am. Chem. Soc. 57, 1326 1935.
- 17. Thiele, J., and A. Bühner, Ann. 347, 249 1906.
- Thiele, J., and K. Merck, Ann. 415, 257 1918.
- 19. von Braun, J., Ber. 50, 1658 1917.
- 20. von Braun, J., and G. Manz, Ber. 62, 1059 1929.
- 21. Weissgerber, R., Ber. 44, 1436 1911.
- 22. Wislicenus, W., and W. Hentrick, Ann. 436, 9 1924.
- 23. Ziegler, K., and P. Tiemann, Ber. 55, 3406 1922.

# 2. ANTHRACENE AND ITS ALKYL DERIVATIVES, C<sub>n</sub>H<sub>2n-1</sub>,

$\mathbf{C}_{14}\mathbf{H}_{10}$		340.58	$738.9^{106}$
Anthracene		337.70	$697.6^{106}$
Antimacenc	^	337.14	$692.7^{106}$
		335.52	$674.4^{106}$
		329.47	$612.2^{106}$
	<u> </u>	328.0	$584.0^{104}$
$\sim$	<b>&gt;</b>	327.9	$583.4^{104}$
M. P., °C		327.4	$577.4^{104}$
216.6		325.47	$555.5^{106}$
219147, 148, 149		321.29	$515.0^{106}$
21818, 23, 67, 136		320	$502.9^{36}$
$217.6^{146}$		318.20	$488.1^{106}$
217-217.598		313.4	437.0104
217 (a)		313.2	$436.2^{104}$
$216-217^{57, 66}$		312.8	$432.9^{104}$
$216 - 216.8^{140}$		310.31	$415.6^{106}$
$216.4 - 216.7^{14, 15}$		304.35	$363.9^{106}$
216.6 <sup>104</sup>		300.6	$334.2^{104}$
$216.55^{116}$		300	332.036
216.528, 97, 112, 113		300.1	$331.3^{104}$
$216.0 - 216.5^{117}$		299.9	$330.4^{104}$
216-216.5%		298.22	$319.1^{106}$
$216.4^{103}$		297.38	$317.7^{106}$
$216.18^{92}$		295.37	$303.0^{106}$
$216.1^{128}$		292.15	$279.0^{106}$
216.0 <sub>5</sub> <sup>26</sup>		286.66	$247.7^{106}$
216 (b)		282.1	$219.8^{104}$
B. P., °C @ 760mm		282.0	$219.6^{104}$
341.4		280	211.036
342.0104		277.50	$\boldsymbol{203.2^{106}}$
342114		277.66	$198.2^{106}$
340.3644		271.75	$175^{106}$
$340^{67}$		270.70	$169.3^{108}$
$339.9^{26}$ , $92$ , $128$		263.74	$142.5^{106}$
339.8744		261.76	$135.5^{106}$
339.7725		260.3	$130.2^{104}$
343.25	$778.2^{106}$	259.50	$129.5^{106}$
342.05	761.2106	260	$129.1^{36}$
341.70	757.3106	259.8	$128.3^{104}$
341.57	756.4106	259.4	$127.3^{104}$
340.6	742.2104	<b>254.8</b> 6	$113.4^{106}$
340.5	742.2104	244.6	$86.5^{104}$

C14H10 181

$86.5^{106}$
$85.9^{104}$
$76.0^{36}$
$69.6^{106}$
$62.1^{106}$
$60.2^{106}$
$55.0^{104}$
$52.7^{106}$
$52.7^{106}$
$48.0^{104}$

 $1.250^{14}$ 

 $1.252^{14}$ 

 $1.26^{126}$ 

254.7° 92 0.9457 254.3° 92 0.9456 246.5° 92 0.9516 27° 109 1.250 25.00°99  $1.2477_{6}$ 25° 68 1.252 25° 68 1.253

22° 19 1.252 17° 14 1.251

1.303

 $n_{\rm p}^{20}$ 1.59480

90.35° 135

 $-195^{\circ 19}$ 

#### Additional Data

$$\frac{1}{T_b} = 0.0025640$$

$$- 0.0003252 \log_{10} p_{mm}$$
(100 to 780mm)

- (a) The melting point 217 is found in references 31, 56, 61, 62, 69, 118, 119, 129, 139.
- (b) The melting point 216 is found in references 17, 70, 72, 90, 91, 96, 108, 110, 111, 127.

## C15H12

#### 1-Methylanthracene

M. P., °C 86132 85-8647, 48, 49, 65, 77 83-8589

 $D_4^{20}$ 99.4° 76, 77 1.0471  $n_{\scriptscriptstyle \mathrm{D}}^{20}$ 99.4° 76, 77 1.68027 1.66692 1.72106

#### 2-Methylanthracene

M. P., °C 206.5

> 208-210187 209-209.594

 $207 - 207.5^{72}$ 

207 (a) 206-207122

205-20775

 $205^{142}$ 

204-20574

204.582 204115

203-20447, 55, 71

202-20421, 22

20320, 70, 98

202-20364

20229, 34, 115, 133, 134

B. P., °C @ 760mm 293-29746

(a) The melting point 207 is found in references 35, 45, 73, 76, 87, 89, 102, 141.

#### 9-Methylanthracene

M. P., °C 81.5<sup>125</sup> 79-80<sup>76, 78</sup> 78-79<sup>4</sup>

 $D_4^{20}$  1.0657

76.3-77.843

 $99.4^{\circ}~^{76}$ 

 $n_{\,\mathrm{D}}^{\,20}$ 

 $\begin{array}{ccc} 1.69589 & 99.4^{\circ \ 76} \\ 1.68166 & n_{\rm H\alpha}^{99 \ 4 \ 76} \\ 1.74075 & n_{\rm H\beta}^{99.4 \ 76} \end{array}$ 

C<sub>16</sub>H<sub>14</sub>

#### 1-Ethylanthracene

M. P., °C 33–34<sup>138</sup>

# 2-Ethylanthracene

M. P., °C 150–151<sup>188</sup>

#### 9-Ethylanthracene

 $\begin{array}{c} \text{M. P., °C} \\ & 60-61^{\,84,\ 85} \\ & 59^{\,75,\ 77,\ 78,\ 125} \\ & 56-58^{43} \end{array}$ 

 $D_{4}^{20}$ 

182

1.0413

99.2° 76, 77

 $n_{\,{\scriptscriptstyle \mathrm{D}}}^{\,20}$ 

1.67621 1.66282 1.71845

 $99.2^{\circ}$  76, 77  $n_{\mathrm{H}\alpha}^{99.2}$  76, 77  $n_{\mathrm{H}\beta}^{99.2}$  76, 77

# 1,2-Dimethylanthracene

M. P., °C 85.5–86<sup>7</sup>

B. P., °C @ 760mm 180

 $0.4^{7}$ 

# 1,3-Dimethylanthracene

M. P., °C 204<sup>38</sup> 202-203<sup>55</sup>, 86 202<sup>71</sup> 200<sup>130</sup>

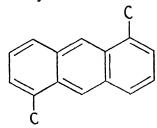
#### 1,4-Dimethylanthracene

 $\begin{array}{c} \text{M. P., °C} \\ 76^{11} \\ 74^{132} \\ 71^{88} \end{array}$ 

# 2,3-Dimethylanthracene

 $\begin{array}{c} \text{M. P., °C} \\ 252^{12,\ 102} \\ 246^{38,\ 40} \\ 244-246^{24} \end{array}$ 

# 1,5-Dimethylanthracene



M. P., °C 139-140<sup>58</sup>

# 1,6-Dimethylanthracene

M. P., °C 240<sup>22</sup>, 20, 82, 83

# 2,6-Dimethylanthracene

M. P., °C 245

250100

 $248^{102}$ 

 $243^{6, 123}$ 

242-24350

# 2,7-Dimethylanthracene

M. P., °C

242

244.5 82, 83

243-2442, 81

241100

 $239^{102}$ 

 $235^{73}$ 

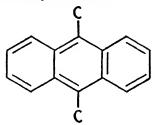
# 2,9-Dimethylanthracene

M. P., °C 858

# 2,10-Dimethylanthracene

M. P., °C 858

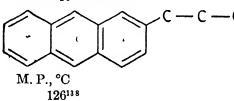
# 9,10-Dimethylanthracene



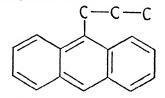
M. P., °C
181
182.8-183.8<sup>16</sup>
181-181.5<sup>79</sup>
181<sup>13</sup>
180.5-181<sup>3</sup>
180-181<sup>54</sup>

#### C<sub>17</sub>H<sub>16</sub>

#### 2-n-Propylanthracene

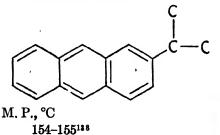


#### 9-n-Propylanthracene



M. P., °C 69-70<sup>125</sup>

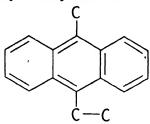
#### 2-Isopropylanthracene



#### 9-Isopropylanthracene

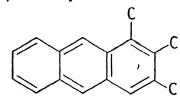
M. P., °C 76<sup>13</sup>

# 9-Methyl-10-ethylanthracene



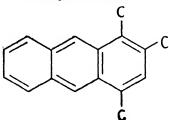
M. P., °C 144<sup>145</sup> 143.2–144<sup>16</sup> 143–144<sup>144</sup>

# 1,2,3-Trimethylanthracene



M. P., °C 243<sup>143</sup> 236<sup>148</sup>

# 1,2,4-Trimethylanthracene



M. P., °C 244<sup>29</sup> 243<sup>55</sup>, 102 236<sup>86</sup>

## 1,3,6-Trimethylanthracene

M. P., °C 222<sup>39, 102</sup>

#### 1,4,6-Trimethylanthracene

M. P., °C 227<sup>37, 39, 41, 102</sup>

#### 2,3,6-Trimethylanthracene

 $\begin{array}{c} \text{M. P., °C} \\ 255^{101} \\ 252^{102} \end{array}$ 

## 1,3,10-Trimethylanthracene

M. P., °C 10010

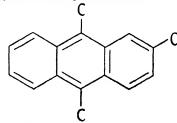
# 1,4,9-Trimethylanthracene

M. P., °C 81<sup>11</sup>

# 2,3,9-Trimethylanthracene

M. P., °C 125<sup>12</sup>

#### 2,9,10-Trimethylanthracene



M. P., °C 100-101 (a)<sup>3</sup> 95-96 (a)<sup>3</sup>

(a) These constants were determined on different crystalline forms.

## C<sub>18</sub>H<sub>18</sub>

# 9-n-Butylanthracene

$$c-c-c-c$$

49-50125

#### 9-Isobutylanthracene

M. P., °C 5784

# 9-Methyl-10-n-propylanthracene

M. P., °C 97.8-98.6<sup>16</sup>

# 9,10-Diethylanthracene

$$c-c$$

M. P., °C 146-147<sup>8</sup> 146<sup>107</sup> 145.5<sup>60</sup>

# 1,2,3,4-Tetramethylanthracene

M. P., °C 135.5–136.5<sup>59</sup>

# 1,3,5,7-Tetramethylanthracene

M. P., °C 163-164<sup>124</sup> 162-163<sup>52</sup>

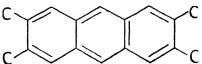
# 1,3,6,8-Tetramethylanthracene

M. P., °C 286-287<sup>51</sup> 281-283<sup>124</sup> 280<sup>1, 32, 33</sup>

# 1,4,5,8-Tetramethylanthracene

M. P., °C 270<sup>42</sup>

# 2,3,6,7-Tetramethylanthracene



#### 1,2,9,10-Tetramethylanthracene

#### $C_{19}H_{20}$

# 9-Pentylanthracene (a)

M. P., °C 61<sup>76, 78</sup> 59<sup>84, 85</sup> 58–59<sup>63</sup> 58<sup>131</sup>

 $D_4^{20}$  0.9812  $99.4^{\circ}$  0.9982 0.9987 1.0017  $71.1^{\circ}$  131  $n_{10}^{20}$ 

 1.0017
  $71.1^{\circ}$  131

 1.62529
  $99.4^{\circ}$  76

 1.63475
  $76.5^{\circ}$  131

 1.63636
  $71.1^{\circ}$  131

 1.61423
  $n_{H\alpha}^{99.4}$  76

 1.62353
  $n_{H\alpha}^{76.5}$  131

$$\begin{array}{lll} 1.62491 & n_{\mathrm{H}\alpha}^{71.1} & ^{131} \\ 1.66023 & n_{\mathrm{H}\beta}^{99.4} & ^{76} \\ 1.67008 & n_{\mathrm{H}\beta}^{76.5} & ^{131} \\ 1.67152 & n_{\mathrm{H}\beta}^{71.1} & ^{131} \end{array}$$

(a) This compound was named *Iso*-amylanthracene in references 53, 76, 78, 131.

# 9-Methyl-10-n-butylanthracene

$$\begin{array}{c} c \\ c \\ \end{array}$$

M. P., °C 78.2–78.8<sup>16</sup>

#### $C_{20}H_{22}$

# 9-Methyl-10-n-pentylanthracene

$$\begin{array}{c} c \\ c \\ -c \\ -c \\ -c \\ -c \\ \end{array}$$

M. P., °C 71-71.8<sup>16</sup>

# Diisopropylanthracene (a)

B. P., °C @ 760mm 202-206 0.2<sup>27</sup>

(a) The structure of this compound was not clearly defined in the literature.

C21H24

9-Methyl-10-n-hexylanthracene

M. P., °C 65.8-66.5<sup>16</sup>

C22H26

9,10-Di-n-butylanthracene

M. P., °C 105-10660

9,10-Diisobutylanthracene

$$\begin{array}{c} c \\ c \\ -c \\ -c \\ \end{array}$$

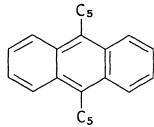
M. P., °C 137-138<sup>80</sup>. <sup>93</sup> 132-133<sup>60</sup>. <sup>98</sup> 1,2,3,4,5,6,7,8-Octamethylanthracene

$$\begin{array}{c|c} C & C & C \\ \hline C & C & C \\ \hline \end{array}$$

M. P., °C 299–300<sup>5</sup>

#### $C_{24}H_{30}$

9,10-Dipentylanthracene



M. P., °C 132–137<sup>63</sup> 134.5–135 (a)<sup>60</sup>

(a) This compound was named Diiso-amylanthracene.

#### C26H34

x,x-Di-(ethylbutyl)-anthracene (a)

B. P., °C @ 760mm 240-256 3<sup>27</sup>

(a) The structure of this compound was not clearly defined in the literature.

#### C54H90

x<sub>5</sub>-Penta-(diethylbutyl)-anthracene (a)

M. P., °C 89.2–101<sup>27</sup>

(a) The structure of this compound was not clearly defined in the literature.

189 References

# References on Anthracene and Its Alkyl Derivatives

- 1. Anschütz, R., Ann. 235, 150 1886.
- 2. Anschütz, R., Ann. 235, 299 1886.
- 3. Bachmann, W. E., and J. M. Chemerda, J. Org. Chem. 4, 583 1939.
- Bachmann, W. E., and M. C. Kloetzel,
   J. Org. Chem. 3, 55 1938-1939.
- Backer, H. J., J. Strating, and L. H. H. Huisman, Rec. trav. chim. 58, 761 1939.
- Baddeley, G., and J. Kenner, Ber. 69, 902 1936.
- Badger, G. M., J. W. Cook, and F. Goulden, J. Chem. Soc. 1940, 16.
- 8 Barnett, E. de B., and N. F. Goodway, J. Chem. Soc. 1929, 1754.
- Barnett, E. de B., N. F. Goodway, and J. W. Watson, Ber. 66, 1876 1933.
- Barnett, E. de B., and C. L. Hewett, Ber. 64, 1572 1931.
- Barnett, E. de B., and J. A. Low, Ber.
   49 1931.
- Barnett, E. de B., and F. C. Marrison, Ber. 64, 535 1931.
- Barnett, E. de B., and M. A. Mathews, Ber. 59, 1429 1926.
- Baxter, G. P., and A. H. Hale, J. Am. Chem. Soc. 58, 510 1936.
- Baxter, G. P., and A. H. Hale, J. Am. Chem. Soc. 59, 506 1937.
- Berliner, E., J. Am. Chem. Soc. 66, 533 1944.
- 17. Berton, A., Compt. rend. 208, 1898 1939.
- Bhatnagar, S. S., M. R. Verma, and P. L. Kapur, Indian J. Phys. 9, 131 1934.
- Biltz, W., W. Fischer, and E. Wünnenberg, Z. physik. Chem. 151A, 13 1930.
- 20. Bornstein, E., Ber. 15, 1820 1882.
- 21. Bornstein, E., Ber. 39, 1238 1906.
- Bornstein, E., H. Schliewiensky, and G. V. Szszesny-Heyl, Ber. 59, 2812 1926.
- Bradley, G., and J. K. Marsh, J. Chem. Soc. 1933, 650.
- Bradley, M. J., and S. W. Parr, Chem. & Met. Eng. 27, 737 1922.

- Burgstaller, S., Abhandl. Deutschen nat. Ver. für Böhmen "Lotos" in Prag. 3, 83 1912; Chem. Zentr. 1912, II, 1525.
- Burriel, F., Anales soc. españ. fís. quím. 29, 89 1931; C. A. 25, 2885
   1931; Chem. Zentr. 1931, I, 2732.
- Calcott, W. S., J. M. Tinker, and V. Weinmayr, J. Am. Chem. Soc. 61, 1010 1939.
- Capper, N. S., and J. K. Marsh, J. Am. Chem. Soc. 47, 2847 1925.
- Clar, E., F. John, and B. Hawran, Ber.
   62, 940 1929.
- Cook, J. W., and V. J. Chambers, J. Am. Chem. Soc. 43, 334 1921.
- 31. Dermer, O. C., and J. King, J. Am. Chem. Soc. **63**, 3232 **1941**.
- Dewar, J., and H. O. Jones, J. Chem. Soc. 85, 212 1904.
- 33. Dewar, J., and H. O. Jones, Proc. Chem. Soc. 20, 6 1904.
- 34. Dieterle, H., and A. Salomon, Arch. Pharm. 271, 177 1933.
- Dieterle, H., A. Salomon, and E. Nosseck, Ber. 64, 2086 1931.
- Dodge, B. F., J. Ind. Eng. Chem. 14, 569 1922.
- Elbs, K., J. prakt. Chem. [2] 35, 465
   1887.
- 38. Elbs, K., J. prakt. Chem. [2] 41, 1 1890.
- 39. Elbs, K., J. prakt. Chem. [2] 41, 121 1890.
- 40. Elbs, K., and H. Eurich, Ber. 20, 1361 1887.
- 41. Elbs, K., and G. Olberg, Ber. 19, 408 1886.
- Ellison, H., and D. H. Hey, J. Chem. Soc. 1938, 1847.
- Fieser, L. F., and J. L. Hartwell, J. Am. Chem. Soc. 60, 2555 1938.
- Finck, J. L., and R. M. Wilhelm, J. Am. Chem. Soc. 47, 1577 1925.
- 45 Fischer, O., J. prakt. Chem. [2] 79, 555 1909.
- Fischer, O., and L. Costner, J. prakt. Chem. [2] 82, 280 1910.
- Fischer, O., F. Falco, and H. Gross, J. prakt. Chem. [2] 83, 208 1911.
- 48. Fischer, O., and A. Sapper, J. prakt. Chem. [2] **83**, 201 **1911**.
- Fischer, O., and H. Ziegler, J. prakt. Chem. [2] 86, 289 1912.

- Flumiani, G., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 133, Abt. IIb, 43 1924.
- Frankforter, G. B., and V. R. Kokatnur, J. Am. Chem. Soc. 36, 1529 1914.
- Friedel, C., and J. M. Crafts, Ann. chim. phys. [6] 11, 263 1887.
- 53. Gerry, II. T., through R. L. Wakeman, Thesis, Mass. Inst. Tech. 1930.
- Gibson, C. S., and J. D. A. Johnson,
   J. Chem. Soc. 1931, 753.
- 55. Gresly, L., Ann. 234, 234 1886.
- Grigor'ev, S. M., Coke and Chem. (U.S.S.R.) 7, No. 6, 16 1937; Chem. Zentr. 1938, II, 463.
- Grimm, H. G., M. Günther, and H. Tittus, Z. physik. Chem. 14B, 169 1931.
- Haworth, R. D., and G. Sheldrick, J. Chem. Soc. 1934, 1950.
- 59. Hewett, C. L., J. Chem. Soc. 1940, 293.
- Hugel, G., and M. Lerer, Bull. soc. chim. [4] 53, 1497 1933.
- 61. Il'inskii, M. A., and B. I. Afremov, Ber. 69, 1824 1936.
- 62. Il'inskii, M., and V. Afremov, Compt. rend. acad. sci. U R.S.S. 3, 135 1936.
- 63. Jungermann, E., Ber. 38, 2868 1905.
- Kafuku, K., and K. Sebe, Bull. Chem. Soc. Japan 7, 114 1932.
- Keimatsu, S., I. Hirano, and K. Yoshimi, J. Pharm. Soc. Japan 50, 95 1930; C.A. 24, 5037 1930; Chem. Zentr. 1930, II, 2384.
- Kempf, R., J. prakt. Chem. [2] 78, 201 1908.
- 67. Kirby, W., J. Soc. Chem. Ind. 40, 274T 1921.
- Klemm, W., W. Tilk, and S. von Mullenheim, Z. anorg. allgem. Chem. 176, 1 1928.
- Klepper, J., Compt. rend. Congr. chim. ind. 8, 261 1929.
- 70. Kogl, F., and H. Erxleben, Ann. 479, 11 1930.
- 71. Kögl, F., and I. Y. Postovskii, Ann. 440, 19 1924.
- Koller, G., and H. Russ, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 146, Abt. IIb, 54 1937.
- Kraemer, G., A. Spilker, and P. Eberhardt, Ber. 23, 3269 1890.

- Krasovskif, N., J. Russ. Phys. Chem. Soc. 40, 1510 1908; C.A. 4, 2263 1910; Chem. Zentr. 1909, I, 772.
- Krasovskii, N., J. Russ. Phys. Chem. Soc. 46, 1067 1914; Chem. Zentr. 1915. I. 999.
- 76. Krollpfeiffer, F., Ann. 430, 161 1923.
- 77. Krollpfeiffer, F., Ber. 56, 77 1923.
- 78. Krollpfeiffer, F., and F. Branscheid, Ber. 56, 1617 1923.
- Kursanov, D. N., and R. R. Zel'vin, Compt. rend. acad. sci. U.R.S.S.
   36, 17 1942; Survey For. Petrol. Liter. Sept. 10, 1943.
- Larsen, R. G., R. E. Thorpe, and F. A. Armfield, Ind. Eng. Chem. 34, 183 1942.
- 81. Lavaux, J., Compt. rend. 141, 354 1905.
- Lavaux, P., Ann. chim. phys. [8] 20, 436 1910.
- 83. Lavaux. P., Ann. chim. phys. [8] 21, 131 1910.
- 84. Liebermann, C., Ann. 212, 1 1882.
- 85. Liebermann, C., and G. Tobias, Ber. 14, 795 1881.
- 86. Liebermann, C., and S. von Kostanecki, Ann. 240, 245 1887.
- 87. Limpricht, H., and O. Weigand, Ann. 311, 178 1900.
- 88. Louise, E., Bull. soc. chim. [2] 44, 177 1885.
- 89. Majima, R., and C. Kuroda, Acta Phytochim. (Japan) 1, 43 1922.
- 90. Mameli, E., and A. Mossini, Giorn. chim. ind. applicata 15, 161 1933.
- 91. Manzoni-Ansidei, R., Gazz. chim. ital. 67, 790 1937.
- 92. Marti, F. B., Bull. soc. chim. Belg. 39, 590 1930.
- 93. Martin, E., Ann. combustibles liquides 12, 97 1937.
- Martin, E. L., J. Am. Chem. Soc. 58, 1438 1936.
- Marx, W., and H. Sobotka, J. Org. Chem. 1, 275 1936.
- McVicker, W. H., J. K. Marsh, and A. W. Stewart, J. Chem. Soc. 127, 999 1925.
- Miller, H. F., and J. Bachman, J. Am. Chem. Soc. 57, 766 1935.
- Mitter, P. C., and A. K. Sen, J. Indian Chem. Soc. 5, 631 1928.

191 References

- 99. Miyaka, S., Mem. Coll. Eng. Kyushu Imp. Univ. 6, 1 1931-1932.
- 100. Morgan, G. T., and E. A. Coulson, J. Chem. Soc. 1929, 2203.
- Morgan, G. T., and E. A. Coulson, J. Chem. Soc. 1929, 2551.
- 102. Morgan, G. T., and E. A. Coulson, J. Soc. Chem. Ind. 53, 71T 1934.
- 103. Mortimer, F. S., J. Am. Chem. Soc. 44, 1416 1922.
- 104. Mortimer, F. S., and R. V. Murphy, Ind. Eng. Chem. 15, 1140 1923.
- 105. Mukherjee, A., Indian J. Phys. 8, 147 1933.
- Nelson, O., and C. Senseman, Ind. Eng. Chem. 14, 58 1922.
- 107. Nenitzescu, C. D., and D. A. Isacescu, Ber. 66, 1100 1933.
- 108. Obreĭmov, I., and A. Prikhot'ko, Physik. Z. Sowjetunion 1, 203 1932.
- Physik. Z. Sowjetunion 1, 203 1932. 109. Orndorff, W. R., and F. K. Cameron,
- Am. Chem. J. 17, 658 1895. 110. Padoa, M., Atti accad. Lincei [5] 27, II, 59 1918.
- 111. Padoa, M., Gazz. chim. ital. 48, II, 139 1918.
- 112. Parks, G. S., and H. M. Huffman, Ind. Eng. Chem. 23, 1138 1931.
- 113. Pascal, P., Bull. soc. chim. [4] 29, 644 1921.
- 114. Pieters, H. A. J., and H. van den Burge, Brennstoff-Chem. 20, 201 1939.
- Raistrick, H., R. Robinson, and A. R. Todd, J. Chem. Soc. 1937, 80.
- 116. Reissert, A., Ber. 23, 2239 1890.
- 117. Rheinboldt, H., J. prakt. Chem. [2] 111, 242 1925.
- 118. Robertson, J. M., Proc. Roy. Soc. 140 A, 79 1933.
- 119. Robertson, J. M., Z. Krist. **84**, 321 1933.
- 120. Rudolfi, E., Z. physik. Chem. 66, 705 1909.
- Sandin, R. B., R. Kitchen, and L. F. Fieser, J. Am. Chem. Soc. 65, 2018 1943.
- 122. Seholl, R., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 127, Abt. IIb, 33 1918.
- 123. Seer, C., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 120, Abt. IIb, 3 1911.

- 124. Seer, C., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 120, Abt. IIb, 957 1911.
- 125. Sieglitz, A., and R Marx, Ber. 56, 1619 1923.
- 126. Skraup, S., and M. Eisemann, Ann. 449, 1 1926.
- 127. Tamman, G., Z. anorg. allgem. Chem. 157, 321 1926.
- 128. Timmermans, J., and F. Burriel, Compt. rend. Congr. chim. ind. 10, 196 1931.
- 129. Trusty, A. W., Petroleum Refiner 22, 95 1943.
- 130. van Dorp, A., Ber. 5, 674 1872.
- 131. von Auwers, K., Ber. 53, 941 1920.
- 132. von Braun, J., and O. Bayer, Ber. 59, 914 1926.
- 133. von Braun, J., O. Bayer, and L. F. Fieser, Ann. 459, 287 1927.
- 134. von Niementowski, S., Ber. 33, 1629 1900.
- 135. von Steiger, A. L., Ber. 56, 998 1923.
- 136. von Weimarn, N., Kolloid Z. 54, 296 1931.
- 137. Wachendorff, C., and T. Zincke, Ber. 10, 1481 1877.
- 138. Waldmann, H., and E. Marmorstein, Ber. 70, 106 1937.
- 139. Waser, E., Helv. Chim. Acta 8, 758 1925.
- 140. Waterman, H. I., J. J. Leendertse, and A. C. Cranendonk, Rec. trav. chim. 58, 83 1939.
- 141. Weger, M., Z. angew. Chem. 22, 338 1909.
- 142. Weigert, F., and O. Krüger, Z. physik. Chem. 85, 579 1913.
- 143. Wende, H., Ber. 20, 867 1887.
- 144. Willemart, A., Bull. soc. chim. [5] 5, 556 1938.
- 145. Willemart, A., Compt. rend. 205, 866 1937.
- Williams, J. W., and E. F. Ogg, J. Am. Chem. Soc. 50, 94 1928.
- 147. Winterstein, A., and K. Schon, Naturwissenschaften 22, No. 15, 237 1934.
- 148. Winterstein, A., and K. Schön, Z. physiol. Chem. 230, 146 1934.
- 149. Winterstein, A., K. Schön, and H. Vetter, Z. physiol. Chem. 230, 158 1934.

# 3. PHENANTHRENE AND ITS ALKYL DERIVATIVES, C<sub>n</sub>H<sub>2n-18</sub>

C14H10	9815, 29, 88, 183	
Phenanthrene	97.5-9835, 52	
r nenanun ene	97-982, 19, 20, 27, 33, 72	
	97.880	
] ]	97.749	
$\wedge$ $\wedge$ $\vee$	$97 - 97.5^{181}$	
	$96.5 - 97.5^{38}$	
	97.318	
	$97.2^{10}$	
M. P., °C	$96-97^{20}$ , 84, 92	
100.3	96.399	
103.559	96.2537, 89	
103.05108	96.274	
103.059	$96.1^{121}$	
10360, 75	9645, 56, 68, 93, 180	
102.5–10351	95-96129	
101-102118	95.8578	
10117, 18, 48, 53	95.5712	
100.7-10132	B. P., °C @ 760mm	
100-101107	339.6	
100.0-100.9 <sup>24</sup>	340 (d)	
100.5113	338.5 82	
100.4119	337.881	
$100.35^{126}$	346.8	883.982
$100.3^{122}$	345.7	$870.9^{82}$
100(a)	345.1	$864.4^{82}$
$99.6 - 100^{66}$	340.59	$764.2^{86}$
99 5-100 <sup>187</sup>	340.41	$762.3^{86}$
99-100(b)	339.91	$754.3^{86}$
$99.8^{22}$	337.1	$741.0^{82}$
$99.6^{132}$	333.08	664.686
$99.5^{67,101}$	332.11	$654.4^{86}$
$99-99.5^{28}, 79$	330	$629.4^{28}$
$98.8 - 99.5^{89}$	325.4	$592.6^{82}$
$98.7 - 99.5^{1}$	324.9	$588.7^{82}$
$98.5 - 99.5^{5}$	324.5	$584.7^{82}$
$99.0^{80}$	321.24	531.586
99 (c)	320	$518.8^{28}$
98_9955, 83, 114, 115, 124, 125	319.27	$510.7^{86}$
98.8100	312.62	446.086
98.2–98.8103	309.44	$420.3^{86}$
98.5 <sup>57, 94, 97</sup>	308.64	414.886
98.082	306.5	399.7 82

193  $C_{14}H_{10}$ 

306.4	$\mathbf{399.2^{82}}$
300.91	$350.9^{86}$
300	$343.7^{28}$
299.88	$340.2^{86}$
295.37	$311.4^{86}$
294.57	$307.4^{86}$
293.2	$\boldsymbol{299.9^{82}}$
293.2	$299.8^{82}$
293.1	$299.7^{82}$
282.73	$234.1^{86}$
281.33	226.186
280	$220.0^{28}$
271.5	$183.0^{82}$
271.5	$182.9^{82}$
269.89	$172.4^{86}$
264.73	$151.4^{86}$
260	$135.5^{28}$
249.14	$101.5^{86}$
246.1	$94.6^{82}$
<b>2</b> 46.59	$94.5^{86}$
246.0	$94.3^{82}$
240	$79.0^{28}$
<b>233</b> .8	$67.5^{82}$
233.54	$65.4^{86}$
232.34	$62.2^{86}$
203.6	$27.2^{82}$
$1.172 \text{ (solid)}^{61}$	
1.174 (solid) 84	
1.175 (solid) <sup>47</sup>	
1.182 (solid)106	
1.018	170° 110
1.031	150° 110
1.030	149.8° 111
1.0412	131.1° 68
1.046	130° 110
1.046	129.8° 111
1.058	110° 110
1.0630	100.5° 118
1.058	99.8° 111
1.0395	97.03° 12
1.0418	93.53° 12
1.0459	89.43° 12
1 0 400	OE 02º 12

 $D_{4}^{20}$ 

1.0483

	1.0514	81.82° 12
	1.0540	78.32° 12
	1.0590	$71.62^{\circ}$ 12
	1.0620	67.53° 12
	1.0671	60.83° 12
	1.0688	57.33° 12
	1.1648 (solid)	47.12° 12
	1.1752 (solid)	28.91° 12
	1.179	25° 62
	1.1816 (solid)	19.46° 12
$n_{\rm p}^{20}$		
_	1.629	170° 110
	$1.639_{5}$	150° 110
	1.6395	149.8° 111
	1.650	130° 110
	1.6502	129.8° 111
	1.65671	$129.6^{\circ}$ 63
	1.660	110° 110
	1.6600	99.8° 111
	1.64646	$n_{{ m H}lpha}^{130.6}$
	(e)	
Add	itional Data	

$$\frac{1}{T_b} = 0.0025694 - 0.0003254 \log_{10} p_{mm}$$
(100 to 885mm)

- (a) The melting point 100 is found in references 3, 4, 11, 14, 16, 18, 21, 40, 41, 43, 44, 46, 64, 70, 77, 86, 87, 90, 95, 96, 98, 102, 108, 109, 112, 114, 115, 120, 128.
- (b) The melting point 99-100 is found in references 7, 25, 36, 50, 71, 116, 134, 136.
- (c) The melting point 99 is found in references 8, 9, 26, 31, 37, 42, 58, 69, 73, 76, 91, 104, 117, 119, 134, 135, 136.
- (d) The boiling point 340 is found in references 6, 16, 38, 54, 65, 69, 117, 123, 127, 129, 130.
- (e) Refractive indices at other lines are found in reference 85.

#### References on Phenanthrene

- Bachmann, W. E., J. Am. Chem. Soc. 57, 555 1935.
- Bachmann, W. E., and M. C. Kloetzel,
   J. Am. Chem. Soc. 60, 2204 1938.
- Bardhan, J. C., Chemistry & Industry 55, 879 1936.
- Bardhan, J. C., and S. C. Sengupta,
   J. Chem. Soc. 1932, 2520.
- Chem. Soc. 1302, 2320.
   Behrend, R., Z. physik. Chem. 10, 265 1892.
- 6. Beilstein through E. Krober, Z. physik. Chem. 93, 641 1919.
- Bergell, P., and R. Pschorr, Z. physiol. Chem. 38, 16 1903.
- 8. Berger, II., J. prakt. Chem. [2] 133, 331 1932.
- Bergmann, E., J. Am. Chem. Soc. 60, 1798 1938.
- Bernoulli, A. L., and P. Lotter, Helv. Chim. Acta 16, 246 1933.
- Bhatnagar, S. S., M. R. Verma, and P. L. Kapur, Indian J. Phys. 9, 131 1934-1935.
- Block, H., Z. physik. Chem. 78, 385
   1912.
- 13. Bradley, G., and J. K. Marsh, J. Chem. Soc. 1933, 650.
- 14. Buchner, E. II., Z. physik. Chem. 54, 665 1906.
- Capper, N. S., and J. K. Marsh, J. Chem. Soc. 129, 724 1926.
- 16. Carnelley, T., J. Chem. Soc. 32, 653 1877.
- Carnelley, T., J. Chem. Soc. 37, 701 1880.
- Centnerszwer, M., Z. physik. Chem. 46, 427 1903.
- Charrier, G., and A. Beretta, Gazz. chim. ital. 54, 765 1924.
- Charrier, G., and E. Ghigi, Ber. 69, 2211 1936.
- Chatterjee, N. N., J. Indian Chem. Soc. 13, 659 1936.
- Clark, J. M., J. Ind. Eng. Chem. 11, 205 1919.
- Cohen, F. L., and U. Cormier, J. Am. Chem. Soc. 52, 4363 1930.
- Davis, W. W., M. E. Krahl, and G. H.
   A. Clewes, J. Am. Chem. Soc. 62, 3080 1940.

- Denisenko, Y. I., and V. M. Kotel'nikova, J. Gen. Chem. (U.S.S.R.)
   7, 2819 1937; Chem. Zentr. 1938, II, 3081.
- Dewar, J., and J. Read, J. Soc. Chem. Ind. 55, 347T 1936.
- 27. Dieterle, H., and H. Rochelmeyer, Arch. Pharm. 273, 532 1935.
- 28. Dodge, B. F., J. Ind. Eng. Chem. 14, 569 1922.
- Durland, J. R., and H. Adkins, J. Am. Chem. Soc. 60, 1501 1938.
- Efremov, N. N., and A. N. Tikhomirova, Ann. inst. anal. phys. chim. (U.S.S.R.) 4, 92 1928; C.A. 23, 3214 1929; Chem. Zentr. 1929, I, 745.
- Efremov, N. N., and A. Tikhomirova,
   J. Russ. Phys. Chem. Soc. 59, 373
   1927; C.A. 22, 2508 1928.
- Fieser, L. F., M. Fieser, and E. B. Hershberg, J. Am. Chem. Soc. 58, 1463 1936.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 57, 1508 1935.
- Fieser, L. F., and H. L. Holmes, J. Am. Chem. Soc. 58, 2319 1936.
- Fieser, L. F., R. P. Jacobsen, and C. C. Price, J. Am. Chem. Soc. 58, 2163 1936.
- 36. Fittig, R., and E. Ostermeyer, Ann. 166, 361 1873.
- 37. Flumiani, G., Z. Elektrochem. 32, 221 1926.
- 38. Fries, K., R. Walter, and K. Schilling, Ann. 516, 248 1935.
- 39. Garelli, F., and Λ. Ferratini, Atti accad. Lincei [5] 2, I, 275 1893.
- 40. Gavăt, I., and I. Irimescu, Ber. 75, 820 1942.
- Goldschmiedt, G., and M. von Schmidt, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 83, Abt. IIb, 7 1881.
- 42. Graebe, C., Ber. 37, 4145 1904.
- 43. Grewe, R., Ber. 72, 785 1939.
- Grinbaum, R., and L. Marchlewski Bull. intern. acad. polon. sci., Classe sci. math. nat. 1937A, 171; Chem. Zentr. 1938, I, 1102.
- 45. Hayduck, M., Ann. 167, 177 1873.
- 46. Heller, G., Ber. 45, 665 1912.
- 47. Hengstenberg, J., and H. Mark, Z. Krist. 70, 283 1929.

References

- 48. Henstock, H., J. Chem. Soc. 121, 2124 1922.
- Huffman, H. M., G. S. Parks, and M. Barmore, J. Am. Chem. Soc. 53, 3876 1931.
- Il'inskii, M. A., and R. B. Rozal, Compt. rend. acad. sci. (U.R.S.S.)
   [N.S.] 17, 120 1937.
- Juch, V., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 114, Abt. IIb, 411 1905.
- Khadzhinov, V. N., Coke and Chem. (U.S.S.R.) 7, No. 8, 45 1937; Chem. Zentr. 1938, I, 492.
- Kirby, W., J. Soc. Chem. Ind. 40, 274T 1921.
- Knovenagel, E., and A. Tomasczewski, Ber. 36, 2829 1903.
- Kögl, F., H. Erxleben, and L. Janecke, Ann. 482, 105 1930.
- 56. Kopp, A., Mon. sci. [3] 8, 1147 1878.
- 57. Krafft, F., and H. Weiland, Ber. 29, 2240 1896.
- 58. Kramers, J. G., Ann. 189, 129 1877.
- Kremann, R., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 117, Abt. IIb, 569 1908.
- Kremann, R., and F. Hofmeier, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 119, Abt. Hb, 121 1910.
- Krishnan, K. S., and S. Banerjee, Trans. Roy. Soc. (London) 234, 265 1935.
- 62. Kröber, E., Z. physik. Chem. 93, 641 1919.
- 63. Krollpfeiffer, F., Ann. 430, 161 1923.
- 64. Krollpfeisser, F., Ber. 56, 77 1923.
- Lagerlof, D., J. prakt. Chem. [2] 98, 136 1918.
- Larsen, R. G., R. E. Thorpe, and F. A. Armfield, Ind. Eng. Chem. 34, 183 1942.
- Lettré, H., H. Barnbeck, and W. Lege, Ber. 69, 1151 1936.
- 68. Limpricht, H., Ber. 6, 532 1873.
- 69. Lippmann, A., Z. Elektrochem. 17, 15 1911.
- 70. Mameli, E., and A. Mossini, Giorn. chim. ind. applicata 15, 161 1933.
- Manzoni-Ansidei, R., Gazz. chim. ital.
   67. 790 1937.

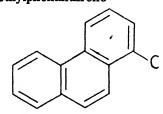
- Marvel, C. S., R. Mozingo, and E. C. Kirkpatrick, J. Am. Chem. Soc. 61, 2003 1939.
- Marx, W., and H. Sobotka, J. Org. Chem. 1, 275 1936.
- McVicker, W. H., J. K. Marsh, and A. W. Stewart, J. Chem. Soc. 127, 999 1925.
- Meyer, II., and A. Hoffmann, Sitzber.
   Akad. Wiss. Wien, Math. naturw.
   Klasse 125-126, Abt. IIb, 449 1916-1917.
- 76. Meyer, R., and K. Taeger, Ber. 53, 1261 1920.
- Meyer, R., and A. Tanzen, Ber. 46, 3183 1913.
- 78. Miolati, A., Z. physik. Chem. 9, 649 1892.
- Morgenstern, O., Sitzber. Akad. Wiss.
   Wien, Math. naturw. Klasse 120,
   Abt. IIb, 523 1911.
- Mortimer, F. S., J. Am. Chem. Soc. 44, 1416 1922.
- 81. Mortimer, F. S., J. Am. Chem. Soc. 44, 1429 1922.
- Mortimer, F. S., and R. V. Murphy, Ind. Eng. Chem. 15, 1140 1923.
- 83. Mosettig, E., and J. van de Kamp, J. Am. Chem. Soc. **52**, 3704 **1930**.
- Mukherjee, A., Indian J. Phys. 8, 147
   1933.
- Narasimham, K. L., Indian J. Phys.
   233 1931.
- 86. Nelson, O., and C. Senseman, Ind. Eng. Chem. 14, 58 1922.
- 87. Nenhaus, A., Ber. 67, 1627 1934.
- 88. Nenitzescu, C. D., E. Ciorănescu, and M. Maican, Ber. 74, 687 1941.
- Newman, M. S., and M. D. Farbman,
   J. Am. Chem. Soc. 66, 1550 1944.
- Obreĭmov, I., and A. Prikhot'ko, Physik. Z. Sowjetunion 1, 203 1932.
- Ochiachi, E., J. Pharm. Soc. Japan 1924, No. 503, 3; Chem. Zentr. 1927, I, 2321.
- 92. Orlov, N. A., and N. O. Likhachev, Ber. **62**, 719 **1929**.
- 93. Ostermayer, E., Ber. 7, 1089 1874.
- Padoa, M., Atti accad. Lincei [5] 13,
   I, 329 1904.
- Padoa, M., Atti accad. Lincei [5] 27,
   II, 59 1918.

- 96. Padoa, M., Atti accad. Lincei [5] 28, II, 239 1919.
- Padoa, M., Gazz. chim. ital. 24, II, 232 1904.
- 98. Padoa, M., Gazz. chim. ital. 48, II, 139 1918.
- Parks, G. S., and H. M. Huffman, Ind. Eng. Chem. 23, 1138 1931.
- 100. Pascal, P., Bull. soc. chim. [4] 29, 644 1921.
- Perlman, D., D. Davidson, and M. T. Bogert, J. Org. Chem. 1, 288 1936.
- 102. Pestemer, M., and J. Cecelsky, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 140, Abt. IIb, 541 1931.
- 103. Price, C. C., J. Am. Chem. Soc. 58, 1834 1936.
- 104. Pschorr, R., Ber. 29, 496 1896.
- 105. Reissert, A., Ber. 23, 2239 1890.
- 106. Rudolfi, E., Z. physik. Chem. 66, 705 1909.
- Ruggli, P., and A. Staub, Helv. Chim. Acta 19, 1288 1936.
- Ruggli, P., and A. Staub, Helv. Chim. Acta 20, 37 1937.
- Ruzicka, L., and H. Hösli, Helv. Chim. Acta 17, 470 1934.
- 110. Salceanu, C., Compt. rend. 193, 161
- 111. Salceanu, C., Compt. rend. **194**, 863 **1932**.
- 112. Schiedt, B., Ber. 71, 1248 1938.
- 113. Schiff, R., Ann. 223, 247 1884.
- 114. Schmidt, E., J. prakt. Chem. [2] 9, 241 1874.
- 115. Schmidt, G. A., Ber. 12, 1153 1879.
- 116. Schmidt, J., Ber. 33, 3251 1900.
- 117. Schmidt, J., and R. Mezger, Ber. 40, 4240 1907.
- 118. Schroeter, G., Ber. 57, 1990 1924.
- 119. Speyers, C. L., J. Am. Chem. Soc. 18, 146 1896.
- Stohmann, F., C. Kleber, and H. Langbein, J. prakt. Chem. [2] 40, 77 1899.
- 121. Tammann, G., Nachr. Ges. Wiss. Gottingen, Jahresber. Geschäftsjahr, Math. physik. Klasse, Fachgruppen III 1913, 335.
- Terres, E., and W. Vollmer, Petroleum Z. 31, No. 19, 1 1935.
- 123. Versmann, F., Chem. News 30, 203 1874.

- 124. von Gerichten, E., Ber. 31, 3198 1898.
- 125. von Gerichten, E., and E. Schrötter, Ann. 210, 396 1881.
- von Narbutt, J., Z. physik. Chem. 53, 697 1905.
- 127. von Steiger, A. L., Ber. 53, 666 1920.
- 128. von Weimarn, N., Kolloid Z. 54, 296 1931.
- 129. Waser, E., Helv. Chim. Acta 8, 758 1925.
- 130. Weger, M., Z. angew. Chem. 22, 338 1909.
- Weger, M., and K. Döring, Ber. 36, 878
   1903.
- 132. Weiss, J. M., and C. R. Downs, Ind. Eng. Chem. 15, 1022 1923.
- 133. Weitzenboch, R., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 121, Abt. IIb, 1227 1912.
- 134. Werner, A., and K. Rekner, Ann. 321, 273 1902.
- 135. Wolf, K. L., and H. Weghofer, Z. physik. Chem. 39 B, 194 1938.
- 136. Zanetti, J. E., and G. Egloff, Ind. Eng. Chem. 9, 474 1917.
- 137. Zelinskiĭ, N. D., I. Tits, and M. Gaverdovskaya, Ber. 59, 2590 1926.

#### C15H12

#### 1-Methylphenanthrene



M. P., °C

121.3

 $123.5^{32}$ 

123100

122 - 122.566

120-1219

12054

119-12028, 53

11968, 82

11857.59

B. P., °C @ 760mm

354-355

· 76382

#### 2-Methylphenanthrene

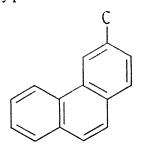
M. P., °C 55.3  $56-56.5^{98}$  $56^{95}$  $55 - 56^{31}$ , 67, 124 $54 - 55^{11}$ 5383 $52 - 53^{76}$ 

B. P., °C @ 760mm 155-160

 $3^{95}$ 

 $769^{82}$ 

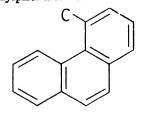
#### 3-Methylphenanthrene



M. P., °C  $85^{124}$ 65 82, 100 62 - 6357 $61 - 62^{5}$ 

B. P., °C @ 760mm 350

# 4-Methylphenanthrene

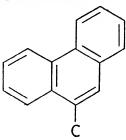


M. P., °C  $117^{103}$  $116^{34}$ 49-5057, 67

B. P., °C @ 760mm 160

234

#### 9-Methylphenanthrene

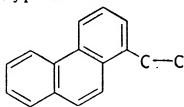


M. P., °C 94  $99.5^{17}$ 94142  $91.5 - 92.5^{26}$ 92<sup>82</sup>91-9286 90-914, 62, 84, 139 88-89138 88140

#### B. P., °C @ 760mm 354 - 35582

## C16H14

#### 1-Ethylphenanthrene



M. P., °C  $109 - 110^{101}$  $108^{52}$ 63.5-64.09

 $62 - 63^{55}$ 

 $62.5^{64}$ 

#### 2-Ethylphenanthrene

$$c-c$$

M. P., °C 172-173<sup>101</sup> 67-68<sup>91</sup> 65-66<sup>6</sup>

#### 9-Ethylphenanthrene

64 - 6563

M. P., °C

63

66<sup>16</sup>

63.5-64.5<sup>8</sup>

62.5-63<sup>91</sup>

61-63<sup>102</sup>

61<sup>137</sup>

58-60<sup>86</sup>

1.64834

1.68588

B. P., °C @ 760mm

220<sup>16</sup>

198-200

11<sup>102</sup>

D<sub>4</sub><sup>20</sup>

1.0603

77.5° 81

n<sub>p</sub><sup>20</sup>

1.65818

77.5° 81

# 1,2-Dimethylphenanthrene

M. P., °C

142-143<sup>64</sup>

140<sup>29</sup>

139-140<sup>13</sup>

B. P., °C @ 760mm

B. P., °C @ 760mm 115–120

218

# 1,3-Dimethylphenanthrene

M. P., °C 76–77<sup>24</sup> 75–76<sup>64</sup>

# 1,4-Dimethylphenanthrene

M. P., °C 77<sup>11</sup> 50-51<sup>2</sup> 49.5-50.0<sup>99</sup>

## 2,3-Dimethylphenanthrene

M. P., °C 79-80<sup>65</sup> 78-78.5<sup>48</sup>. <sup>50</sup> 77-78<sup>49</sup> 65-66<sup>64</sup>

# 1,5-Dimethylphenanthrene

M. P., °C 57–58<sup>64</sup>

# 2,4-Dimethylphenanthrene

M. P., °C 111<sup>64</sup>

# 1,6-Dimethylphenanthrene

M. P., °C 87–88<sup>64</sup>

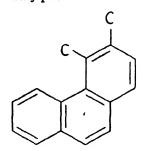
# 2,5-Dimethylphenanthrene

1564

M. P., °C 46–47<sup>64</sup>

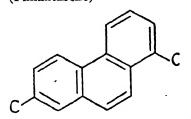
B. P., °C @ 760mm 204-205

# 3,4-Dimethylphenanthrene



M. P., °C 62-63<sup>64</sup>

# 1,7-Dimethylphenanthrene (Pimanthrene)



M. P., °C 85.7 86<sup>11, 27, 114, 119, 120</sup> 85–86<sup>23, 61, 106</sup> 85<sup>75</sup> 79–81<sup>70</sup> 45<sup>88</sup>

# B. P., °C @ 760mm

34088

190

 $0.2^{118}$ 

158-160

0.169

# 2,6-Dimethylphenanthrene

M. P., °C 33-34<sup>64</sup>

# 1,8-Dimethylphenanthrene

M. P., °C 191-192<sup>64</sup>

# 2,7-Dimethylphenanthrene

M. P., °C 101–102<sup>64</sup>

# 3,6-Dimethylphenanthrene

M. P., °C 141<sup>64</sup>

# 1,9-Dimethylphenanthrene

M. P., °C 88<sup>35</sup> 87–88<sup>62</sup>

# 2,9-Dimethylphenanthrene

M. P., °C 56-57<sup>64</sup> 55-56<sup>124</sup>

### 9,10-Dimethylphenanthrene

М. Р., °C 142.5-143<sup>25A</sup> 139<sup>148</sup>

## x,x-Dimethylphenanthrene (a)

M. P., °C 126-127<sup>46</sup>

(a) The structure of this compound was not clearly defined in the literature.

#### $C_{17}H_{16}$

## 1-n-Propylphenanthrene

$$\begin{array}{c} \\ \\ \\ \\ \end{array}$$

M. P., °C 34–35<sup>64</sup> 32–33<sup>9</sup>

## 2-n-Propylphenanthrene

35-367

#### 9-n-Propylphenanthrene

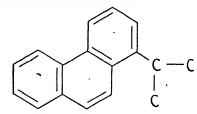
201

M. P., °C 74<sup>89</sup> 58.5–59.5<sup>6</sup> 59<sup>3</sup>

B. P., °C @ 760mm 265-270

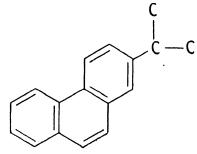
2289

## 1-Isopropylphenanthrene



M. P., °C 88-88.5° 85-8664

## 2-Isopropylphenanthrene



M. P., °C 44-45<sup>64</sup>

# 1-Methyl-2-ethylphenanthrene

M. P., °C 100<sup>64</sup>

## 1-Ethyl-2-methylphenanthrene

$$c-c$$

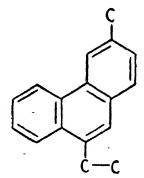
M. P., °C 80<sup>64</sup>

## 1-Ethyl-7-methylphenanthrene

## 3-Methyl-6-ethylphenanthrene

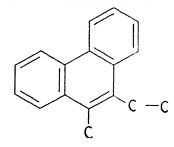
### M. P., °C 47–48<sup>5</sup>

## 3-Methyl-9-ethylphenanthrene



M. P., °C 47–48<sup>5</sup>

## 9-Methyl-10-ethylphenanthrene



M. P., °C 85<sup>25</sup>

## 1,2,6-Trimethylphenanthrene

M. P., °C 128.5–129<sup>111</sup> 128–129<sup>113</sup>

### 2,3,4-Trimethylphenanthrene

M. P., °C 62.8 - 63.847

## 1,2,7-Trimethylphenanthrene

M. P., °C 120-12160

## 1,2,8-Trimethylphenanthrene

M. P., °C

144

212-213122

146-147112

144-14562, 106

14471

142-143108

 $141 - 142^{38}$ 

B. P., °C @ 760mm

175-180

 $0.5^{71}$ 

160-165

 $0.2^{108}$ 

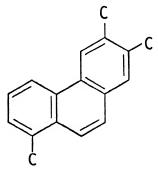
#### 1,3,7-Trimethylphenanthrene

M. P., °C  $68 - 69^{60}$ 

## 1,4,7-Trimethylphenanthrene

M. P., ℃  $72 - 73^{61}$ 

## 2,3,8-Trimethylphenanthrene



M. P., °C 123-12460

## x,x,x-Trimethylphenanthrene (a)

M. P., °C

81105

(a) The structure of this compound was not clearly defined in the literature.

 $C_{18}H_{18}$ 

#### 1-n-Butylphenanthrene

#### 9-n-Butylphenanthrene

M. P., °C 78.5–79<sup>3</sup> 58<sup>89</sup>

B. P., °C @ 760mm 282–284

2089

## 2-tert-Butylphenanthrene

М. Р., °С 99–100<sup>51</sup>

#### 3-tert-Butylphenanthrene

M. P., °C 54-55<sup>51</sup>

## 1-Methyl-2-n-propylphenanthrene

M. P., °C 54<sup>78</sup>

## 1-n-Propyl-2-methylphenanthrene

$$c-c-c$$

M. P., °C 65<sup>31</sup>

## 1-Methyl-3-isopropylphenanthrene

#### 1-Methyl-4-isopropylphenanthrene

# 1-Methyl-6-isopropylphenanthrene (a) (Isoretene)

$$c-c$$

M. P., °C 86-87<sup>14</sup>

(a) The structure of this compoundwas not clearly defined in the literature.

# 1-Methyl-7-isopropylphenanthrene (Retene)

$$c-c$$

B. P., °C @ 760mm

 $D_4^{20}$ 

 $n_{\rm D}^{20}$ 

7., ~ C @ 760mm	
394	
400131	
$394^{123}$	
390-39393	
39022	
216	
208-210	
135	

216 11<sup>133</sup>
208–210 10<sup>1, 134</sup>
135 0<sup>80</sup>

 1.063
 98.2–98.8° 4³

 1.077
 98.2–98.8° 4³

 1.067 (solid)
 90° 4³

 1.093 (solid)
 90° 4³

 1.0841
 25.6° 1³5

 1.0908
 17.5° 1³5

 $\begin{array}{cccc} 1.61783 & n_{\mathrm{H}\alpha}^{25.6 \ 135} \\ 1.62132 & n_{\mathrm{H}\alpha}^{17.5 \ 135} \\ 1.64921 & n_{\mathrm{H}\beta}^{25.6 \ 135} \\ 1.65286 & n_{\mathrm{H}\phi}^{17.5 \ 135} \\ 1.62662 & n_{\mathrm{He}}^{25.6 \ 135} \\ 1.63009 & n_{\mathrm{He}}^{17.5 \ 185} \end{array}$ 

(a) The melting point 98-99 is found in references 12, 30, 61, 74, 77, 79, 85, 110, 132, 136.

- (b) The melting point 98.5 is found in references 1, 10, 24, 44, 134, 135.
- (c) The melting point 98 is found in references 28, 92, 96, 97, 117, 120, 125, 129, 133.

#### 1-Methyl-9-isopropylphenanthrene

$$C - C - C$$

B. P., °C @ 760mm 204–205

1436

# 2-Methyl-8-isopropylphenanthrene (Scianthrene)

M. P., °C 86-87<sup>130</sup>

9,10-Diethylphenanthrene

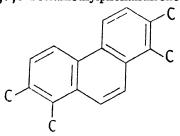
$$c-c$$

M. P., °C 90-91<sup>137</sup>

## 1,2,3,4-Tetramethylphenanthrene

M. P., °C 92-93<sup>66</sup>

### 1,2,7,8-Tetramethylphenanthrene



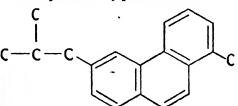
M. P., °C 126-127<sup>107</sup>

### $C_{19}H_{20}$

#### 9-n-Pentylphenanthrene

M. P., °C 69–70<sup>25A</sup>

## 1-Methyl-6-isobutylphenanthrene



#### 9-Ethyl-10-n-propylphenanthrene

M. P., °C 69<sup>25</sup>

#### 1,4-Dimethyl-7-isopropylphenanthrene

M. P., °C 61-62<sup>61</sup>

## 1,x-Dimethyl-7-isopropylphenanthrene (a)

(Methylretene)

M. P., °C 81-82<sup>109</sup> 79<sup>110</sup> 78.5<sup>127</sup>

(a) The structure of this compound was not clearly defined in the literature.

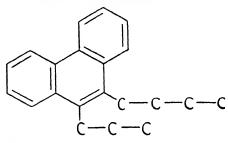
#### C20H22

## 1-Methyl-x-ethyl-7-isopropylphenanthrene (a)

M. P., °C 54-55<sup>23</sup> (a) The structure of this compound was not clearly defined in the literature.

#### $C_{21}H_{24}$

#### 9-n-Propyl-10-n-butylphenanthrene



M. P., °C 7.4<sup>25</sup>

#### 1-Methyl-7,x-diisopropylphenanthrene (a) (Isopropylretene)

M. P., °C 52–53 92

B. P., °C @ 760mm 231-233 9.592

 $D_4^{20}$ 

 $1.0125^{92}$ 

(a) The structure of this compound was not clearly defined in the literature.

## References on C15H12-C21H24 Compounds

- Adelson, D. E., and M. T. Bogert, Chem. Revs 24, 135 1939.
- Akin, R. B., G. S. Stamatoff, and M. T. Bogert, J. Am. Chem. Soc. 59, 1268 1937.
- Bachman, G. B., and R. I. Hoaglin, J. Am. Chem. Soc. 63, 621 1941.
- Bachmann, W. E., J. Am. Chem. Soc. 56, 1363 1934.
- Bachmann, W. E., and G. D. Cortes, J. Am. Chem. Soc. 65, 1329 1943.
- Bachmann, W. E., and M. W. Cronyn, J. Org. Chem. 8, 456 1943.
- Bachmann, W. E., and W. S. Struve,
   J. Org. Chem. 4, 456 1939.

- 8. Bachmann, W. E., and W. S. Struve, 2 32. Darzens, G., and A. Lévy, Compt. J. Org. Chem. 4, 472 1939.
- 9. Bachmann, W. E., and A. L. Wilds, J. Am. Chem. Soc. 60, 624 1938.
- 10. Bamberger, E., and S. C. Hooker, Ann. **229,** 102 **1885**.
- 11. Bardhan, J. C., and S. C. Sengupta, J. Chem. Soc. 1932, 2520.
- 12. Bardhan, J. C., and S. C. Sengupta, J. Chem. Soc. 1932, 2798.
- 13. Barker, R. L., and G. R. Clemo, J. Chem. Soc. 1940, 1277.
- 14. Beath, G. B., J. Soc. Chem. Ind. 52, 338T 1933.
- 15. Bergmann, E., J. Am. Chem. Soc. 60, 1798 **1938**.
- 16. Bergmann, E., and F. Bergmann, J. Am. Chem. Soc. 59, 1443 1937.
- 17. Bergmann, E., and F. Bergmann, J. Chem. Soc. 1939, 1019.
- 18. Bergmann, E., and A. Weizmann, J. Org. Chem. 4, 266 1939.
- 19. Berthelot, Ann. chim. phys. [4] 12, 173 **1867**.
- 20. Berthelot, Ann. chim. phys. [4] 12, 193 **1867.**
- 21. Berthelot, Bull. soc. chim. [2] 7, 43 1867.
- 22. Berthelot through D. E. Adelson and M. T. Bogert, Chem. Revs. 24, 135 1939.
- 23. Bogert, M. T., and T. Hasselström, J. Am. Chem. Soc. 53, 3462 1931.
- 24. Bogert, M. T., and G. S. Stamatoff, Rec. trav. chim. 52, 584 1933.
- 25. Bradsher, C. K., and S. T. Amore, J. Am. Chem. Soc. 65, 2016 1943.
- 25A. Bradsher, C. K., and S. T. Amore, J. Am. Chem. Soc. 66, 1280 1944.
- 26. Bradsher, C. K., and R. W. H. Tess, J. Am. Chem. Soc. 61, 2184 1939.
- 27. Brandt, C. W., New Zealand J. Sci. Tech. 20, 8B 1938; Chem. Zentr. 1939, II, 856.
- 28. Brandt, C. W., and L. G. Neubauer, J. Chem. Soc. 1939, 1031.
- 29. Butenandt, A., H. A. Weidlich, and H. Thompson, Ber. 66, 601 1933.
- 30. Cheung, L. M., Bull. inst. pin. 1929, 108.
- 31. Cohen, A., J. W. Cook, and C. L. Hewett, J. Chem. Soc. 1935, 1633.

- rend. 200, 2187 1935.
  - 33. Darzens, G., and A. Lévy, Compt. rend. 201, 152 1935.
- 34. Darzens, G., and A. Lévy, Compt. rend. 201, 730 1935.
- 35. Darzens, G., and A. Lévy, Compt. rend. 202, 427 1936.
- 36. Darzens, G., and A. Lévy, Compt. rend. 203, 669 1936.
- 37. Diels, O., and A. Karstens, Ber. 60, 2323 **1927**.
- 38. Drake, N. L., and W. T. Haskins, J. Am. Chem. Soc. 58, 1684 1936.
- 39. Easterfield, T. H., and G. Bagley, J. Chem. Soc. 85, 1238 1904.
- 40. Easterfield, T. H., and G. Bagley, Proc. Chem. Soc. 20, 112 1904.
- 41. Efremov, N. N., and A. M. Tikhomirova, Ann. inst. anal. phys. chim. (U.S.S.R.) 4, 92 1928; C.A. 23, 3214 1929; Chem. Zentr. 1929, I, 745.
- 42. Efremov, N. N., and A. N. Tikhomirova, J. Russ. Phys. Chem. Soc. 59, 373 **1927**; C.A. **22**, 2508 **1928**.
- 43. Ekstrand, A. G., Ann. 185, 75 1877.
- 44. Ekstrand, A. G., Ber. 9, 855 1876.
- 45. Ekstrand, A. G., Ber. 17, 692 1884.
- 46. Elderfield, R. C., and W. A. Jacobs, J. Biol. Chem. 107, 143 1934.
- 47. Fieser, L. F., and W. H. Daudt, J. Am. Chem. Soc. 63, 782 1941.
- 48. Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 57, 1508 1935.
- 49. Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 57, 2192 1935.
- 50. Fieser, L. F., and H. L. Holmes, J. Am. Chem. Soc. 58, 2319 1936.
- 51. Fieser, L. F., and C. C. Price, J. Am. Chem. Soc. 58, 1838 1936.
- 52. Gadamer, J., M. Oberlin, and A. Schoeler, Arch. pharm. 263, 81 1925.
- 53. Grewe, R., Ber. 72, 785 1939.
- 54. Grewe, R., Ber. 72, 1314 1939.
- 55. Grewe, R., Ber. 76, 1072 1943.
- 56. Hasselstrom, T., J. Am. Chem. Soc. 63, 1164 1941.
- 57. Haworth, R. D., J. Chem. Soc. 1932,
- 58. Haworth, R. D., J. Chem. Soc. 1932, 2717.

209 References

- Haworth, R. D., and R. L. Barker, J. Chem. Soc. 1939, 1299.
- Haworth, R. D., and F. M. Bolam, J. Chem. Soc. 1932, 2248.
- Haworth, R. D., B. M. Letsky, and C. R. Mavin, J. Chem. Soc. 1932, 1784.
- Haworth, R. D., and C. R. Mavin, J. Chem. Soc. 1932, 2720.
- Haworth, R. D., and C. R. Mavin, J. Chem. Soc. 1933, 1012.
- Haworth, R. D., C. R. Mavin, and G. Sheldrick, J. Chem. Soc. 1934, 454.
- Haworth through L. F. Fieser and E. B. Hershberg, J. Am. Chem. Soc. 57, 2192 1935.
- Howett, C. L., and R. M. Martin, J. Chem. Soc. 1940, 1396.
- Hill, P., W. F. Short, and A. Higginbottom, J. Chem. Soc. 1936, 317.
- 68. Hoch, J., Compt. rend. 205, 65 1937.
- 69. Hosking, J. R., Ber. 69, 780 1936.
- Hosking, J. R., Rec. trav. chim. 49, 1036 1930.
- Hosking, J. R., and C. W. Brandt, Ber. 68, 37 1935.
- Hosking, J. R., and W. T. McFadyen,
   J. Soc. Chem. Ind. 53, 195T 1934.
- Jezierski, T. W., Roczniki Chem. 13, 720 1933; C.A. 28, 6048 1934; Chem. Zentr. 1934, I, 2622.
- Keimatsu, S., and T. Ishiguro, J. Pharm. Soc. Japan 55, 99, 186 1935;
   C.A. 29, 7323 1935; Chem. Zentr. 1935, 11, 3663.
- Keimatsu, S., T. Ishiguro, and G. Fukui, J. Pharm. Soc. Japan 57, 92 1937; Chem. Zentr. 1937, II, 596.
- 76. Klinckhard, T., Ann. 379, 362 1911.
- 77. Komppa, G., and E. Wahlforss, J. Am. Chem. Soc. **52**, 5000 **1930**.
- 78. Kon, G. A. R., E. S. Narracott, and C. Reid, J. Chem. Soc. 1938, 778.
- Kono, M., and R. Maruyama, J. Agr. Chem. Soc. Japan 12, 512 1936; C. A. 30, 6747 1936.
- 80. Krafft, F., and H. Weilard, Ber. 29, 2240 1896.
- 81. Krollpfeiffer, F., Ann. 430, 161 1923.
- 82. Kruber, O., and A. Marx, Ber. 71, 2478 1938.
- 83. Laucht, F., Z. physiol. Chem. 237, 236 1935.

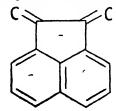
- Levitz, M., and M. T. Bogert, J. Org. Chem. 8, 253 1943.
- Manzoni-Ansidei, R., Gazz. chim. ital.
   67, 790 1937.
- Marvel, C. S., D. E. Pearson, and R. V.
   White, J. Am. Chem. Soc. 62, 2741
   1940.
- McVicker, W. H., J. K. Marsh, and A. W. Stewart, J. Chem. Soc. 127, 999 1925.
- 88. Meyer, H., Z. angew. Chem. 37, 796 1924.
- Miller, H. F., and J. Bachman, J. Am. Chem. Soc. 57, 766 1935.
- Morgenstern, O., Sitzber. Akad. Wiss.
   Wien, Math. naturw. Klasse 120,
   Abt. IIb, 523 1911.
- 91. Mosettig, E., and J. van de Kamp, J. Am. Chem. Soc. 55, 3442 1933.
- Nyman, G. A., Ann. Acad. Sci. Fennicae 41A, No. 5, 1935; C.A. 30, 2958
   1936; Chem. Zentr. 1936, I, 2348.
- Olsson, J., Ing. Vetenskaps Akad. Handl. 1931, No. 111.
- 94. Orcutt, R. M., and M. T. Bogert, J. Am. Chem. Soc. 63, 127 1941.
- Orcutt, R. M., and M. T. Bogert, J. Org. Chem. 4, 543 1939.
- 96. Pascal, P., Bull. soc. chim. [4] 29, 644
- Paul, L., Seifensieder-Ztg. 42, 393, 412,
   434 1915; Chem. Zentr. 1915, II, 188.
- Perlman, D., and M. T. Bogert, J. Am. Chem. Soc. 59, 2534 1937.
- Popa, D., D. Perlman, and M. T. Bogert, J. Am. Chem. Soc. 60, 319 1938.
- 100. Psehorr, R., Ber. 39, 3106 1906.
- 101. Pschorr, R., Ber. 39, 3124 1906.
- 102. Pschorr, R., Ber. 39, 3128 1906.
- Radeliffe, C. B., I. R. Sherwood, and W. F. Short, J. Chem. Soc. 1931, 2293.
- 104. Ruzicka, L., R. G. R. Bacon, R. Lukas, and J. D. Rose, Helv. Chim. Acta 21, 583 1938.
- Ruzicka, L., and F. Balas, Helv. Chim. Acta 7, 875 1924.
- 106. Ruzicka, L., L. L. Engel, and W. H. Fischer, Helv. Chim. Acta 21, 364 1938.

- Ruzicka, L., M. Goldberg, and K. Hofmann, Helv. Chim. Acta 20, 325 1937.
- 108. Ruzicka, L., and J. R. Hosking, Helv. Chim. Acta 14, 203 1931.
- Ruzicka, L., and H. Jacobs, Rec. trav. chim. 57, 509 1938.
- Ruzicka, L., and J. Meyer, Helv. Chim. Acta 5, 581 1922.
- Ruzicka, L., E. Rey, and W. J. Smith, Helv. Chim. Acta 26, 2057 1943.
- 112. Ruzicka, L., E. Rey, and M. Spillmann, Helv. Chim. Acta 25, 1375 1942.
- 113. Ruzicka, L., and W. J. Smith, Chem. and Ind. 16, II, 1210 1938.
- 114. Ruzicka, L., R. Steiger, and H. Schinz, Helv. Chim. Acta 9, 962 1926.
- 115. Ruzicka, L., and H. Waldmann, Helv. Chim. Acta 16, 842 1933.
- 116. Ruzieka, L., and E. Waldmann, Helv. Chim. Acta 18, 611 1935.
- 117. Ruzicka, L., H. Waldmann, P. J. Meier, and H. Hösli, Helv. Chim. Acta 16, 169 1933.
- 118. Schmid, L., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 148, Abt. IIb, 9 1939.
- 119. Schmid, L., and E. Erdos, Ann. 503, 269 1933.
- 120. Schmid, L., and W. Hosse, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 147, Abt. IIb, 366 1939.
- 121. Schultze, W., Ann. 359, 129 1908.
- 122. Schulze, H., Z. physiol. Chem. 238, 35 1936.
- 123. Schweitzer, R., Ann. 264, 193 1891.
- 124. Sengupta, S. C., J. prakt. Chem. [2] 152, 9 1939.

- 125. Soltys, A., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 138 suppl., Abt. IIb, 175 1929.
- 126. Sterling, E. C., and M. T. Bogert, J. Org. Chem. 4, 20 1939.
- 127. Suzucki, K., Sci. Papers Inst. Phys. Chem. Research (Tokyo) 26, 98 1935.
- 128. Trost, F., Ann. chim. applicata 25, 496 1935.
- 129. Tschirch, A., and F. Koritschoner, Arch. pharm. 240, 568 1902.
- Uota, H., J. Dept. Agr. Kyushu Imp. Univ. 5, 117 1937.
- 131. Versmann, F., Chem. News 30, 203 1874.
- 132. Vesterberg, A., Ber. 36, 4200 1903.
- 133. Vesterberg, K. A., and E. Borge, Ann. 440, 305 1924.
- 134. Virtanen, A. J., Ber. 53, 1880 1920.
- 135. von Auwers, K., and R. Kraul, Ann. 443, 181 1925.
- 136. Wahlforss, A., Z. Chem. 12 [5], 73 1869.
- Willgerodt, C., and B. Albert, J. prakt. Chem. [2] 84, 383 1911.
- 138. Windaus, A., Ann. 439, 59 1924.
- 139. Windaus, A., H. Jensen, and A. Schramme, Ber. 57, 1875 1924.
- 140. Windaus, A., and H. Schiele, Nachr. Ges. Wiss. Göttingen, Jahresber. Geschäftsjahr 1923, 17; Chem. Zentr. 1923, III, 673.
- 141. Zeidler, O., Ann. 191, 285 1878.
- 142. Zelinskii, N. D., and M. V. Gaverdovskaya, Ber. 61, 1049 1928.
- 143. Zincke, T., and W. Tropp, Ann. 362, 242 1908.

# 4. MISCELLANEOUS POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-18}$

# $C_{14}H_{10}$ 1.2-Dimethyleneacenaphthene



- M. P., °C 184 (a)<sup>41</sup> 158 (b)<sup>41</sup>
- (a) This constant was determined on the *trans* isomer of the compound.
- (b) This constant was determined on the cis isomer of the compound.

#### 9-Methylenefluorene

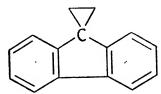
M. P., °C 104-106<sup>39</sup> 53<sup>23</sup>, <sup>78</sup> 46-48<sup>59</sup>

#### $C_{15}H_{12}$

#### 9-Ethylidenefluorene

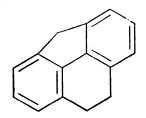
M. P., °C 104<sup>18</sup>, <sup>20</sup>, <sup>64</sup>, <sup>66</sup>, <sup>67</sup> 102<sup>51</sup> 100<sup>42</sup>

## Spiro[fluorene-9,1'-cyclopropane]



M. P., °C 73–73.5<sup>77</sup>

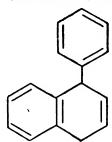
### Cyclopentano-[def]-9, 10-dihydrophenanthrene



### M. P., °C 140.5–141.2<sup>26</sup>

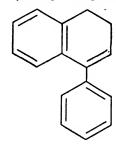
#### C16H14

## 1-Phenyl-1,4-dihydronaphthalene



M. P., °C 50<sup>63</sup>

## 4-Phenyl-1,2-dihydronaphthalene



B. P., °C @ 760mm 302<sup>75</sup>

175-177 186-188

 $12^{76} 10^{68}$ 

 $D_4^{20}$ 

1.0784

27° 70

## 1,2-Diethylideneacenaphthene

M. P., °C 140<sup>40</sup>

#### 9-(Propen-2'-yl)-fluorene

$$c-c=c$$

B. P., °C @ 760mm 174-176

 $15^{79}$ 

#### 9-Isopropylidenefluorene

M. P., °C

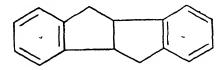
113-11780, 88

113 (a)55

89 (a)18

(a) These constants were determined on stereoisomers.

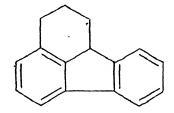
# 2,3,6,7-Dibenzobicyclo-[3,3,0]-octane



M. P., °C 102<sup>5</sup>

10053

## Cyclohexano-[jk]-fluorene



M. P., ℃

75

7636

 $75^{35}$ 

74-75<sup>74</sup> 73-74<sup>73</sup>

B. P., °C @ 760mm

363-365 203-205  $749^{35} 12^{74}$ 

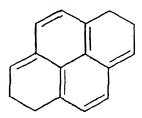
200-205

 $12^{73}$ 

155

0.769

## 1,2,6,7-Tetrahydropyrene



M. P., °C

## x4-Tetrahydropyrene (a)

M. P., °C 84<sup>45</sup>

(a) The structure of this compound was not clearly defined in the literature.

### $C_{17}H_{16}$

## x-Phenethenylindane (a)

B. P., °C @ 760mm 198–201

1860

(a) The structure of this compound was not clearly defined in the literature.

## 1-Methyl-4-phenyl-1,2-dihydronaphthalene

M. P., °C 48<sup>70</sup>

B. P., °C @ 760mm 185–188

1470

 $D_4^{20}$ 

1.0513

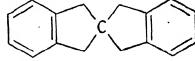
22° 70

### 1-(Propen-2'-yl)-3, 4-dihydrophenanthrene

$$c-c=c$$

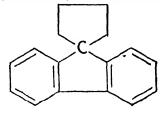
M. P., °C 129-130.5<sup>56</sup>

## Spiro[indane-2,2'-indane]



M. P., °C 66-67<sup>37</sup>

## Spiro[fluorene-9,1'-cyclopentane]

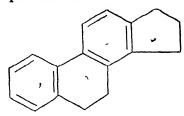


M. P., °C 91<sup>2</sup>

## x<sub>4</sub>-Tetrahydro-x,x-benzofluorene (a)

M. P., °C 129<sup>34</sup>

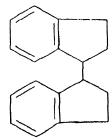
- (a) The structure of this compound was not clearly defined in the literature.
- 1,2-Cyclopentano-9,10-dihydrophenanthrene



M. P., °C 65-69<sup>11</sup>

#### C18H18

# 1,1'-Biindanyl



B. P., °C @ 760mm

194-195

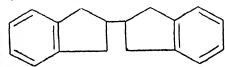
 $14^{62}$ 

 $D_4^{20}$ 

1.0669

24° 62

## 2,2'-Biindanyl



M. P., °C 165-166<sup>24</sup>, <sup>25</sup>

B. P., °C @ 760mm

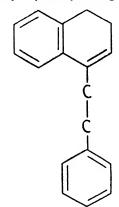
**239–24**1

5061

225-226

2061

#### 4-Phenethyl-1,2-dihydronaphthalene



M. P., °C

1615

209-212 165-168

1254

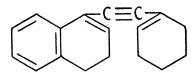
## x-Phenethenyl-1,2,3,4-tetrahydronaphthalene (a)

B. P., °C @ 760mm 216–218

1860

(a) The structure of this compound was not clearly defined in the literature.

# Cyclohexen-1-yl-[4'-(1',2'-dihydro-naphthyl)]-ethyne



B. P., °C @ 760mm

170-172

246

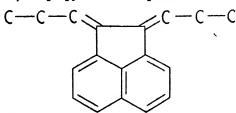
 $D_4^{20}$ 

1.044346

 $n_{\rm D}^{20}$ 

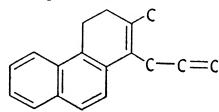
1.618646

#### 1,2-Dipropylideneacenaphthene



M. P., °C 10141

1-(Propen-2'-yl)-2-methyl-3,4-dihy-drophenanthrene

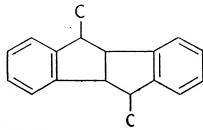


B. P., °C @ 760mm

177-179

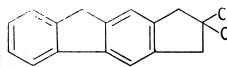
 $0.6^{12}$ 

2,3,6,7-Dibenzo-4,8-dimethylbicy-clo-[3,3,0]-octane



M. P., °C 947

# 2,3-(4',4'-Dimethylcyclopentano)fluorene



M. P., °C 135–137 (a)<sup>32</sup>

128-129 (a)<sup>32</sup>

(a) These constants were determined on different forms.

### 1,2-Benzo-9a-methyl-3,4,4a,9atetrahydrofluorene

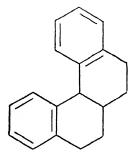
B.P., °C @ 760mm

 $0.8^{16}$ 

# 1,2-Cyclohexano-9,10-dihydroan-thracene

M. P., °C 69.3-69.9<sup>29</sup>

# 3,4-Benzo-1,2,4a,9,10,10a-hexahy-drophenanthrene



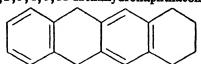
M. P., °C 47.5-48\*\*

B. P., °C @ 760mm

148-153

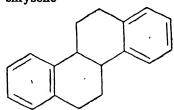
0.233

1,2,3,4,6,11-Hexahydronaphthacene



M. P., °C 128<sup>71</sup>

4b,5,6,10b,11,12-Hexahydrochrysene



M. P., °C

140 (a)4

115 (a)50

 $112 (a)^{49}$ 

79 (a)49

76.8-77.8 (b)44

75 (b)4, 50

B. P., °C @ 760mm

223 (a)

1250

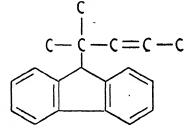
**208** (b)

 $12^{50}$ 

- (a) These constants were determined on the *trans* isomer of the compound.
- (b) These constants were determined on the *cis* isomer of the compound.

## C19H20

2-Methyl-2-(9'-fluoryl)-pentene-3



9-Cyclohexylfluorene

M. P., °C 115-116<sup>9</sup> 102-103<sup>13</sup>

1-(Buten-3'-yl)-2-methyl-3,4-dihy-drophenanthrene

$$c-c-c=c$$

B. P., °C @ 760mm 155-160

 $0.3^{12}$ 

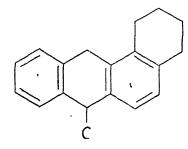
Spiro[2,2,4-trimethylcyclobutane-1,9'-fluorene]

M. P., °C 92-94<sup>31</sup>

Benzo-[lm]-2,4,4-trimethyl-2,3,4,4a-tetrahydrofluorene

M. P., °C 103-104<sup>31</sup>

1,2-Cyclohexano-10-methyl-9,10dihydroanthracene



M. P., °C 155.5-156<sup>29</sup>

#### C20H22

x-Naphthyl-2,2-dimethyl-3-methylenebicyclo-[2,2,1]-heptane (a)

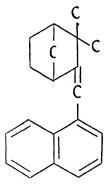
M. P., °C 92-93\*

B. P., °C @ 760mm 210

(a) The structure of this compound was not clearly defined in the literature.

168

2,2-Dimethylbicyclo-[2,2,1]-heptyliden-3-yl-(1'-naphthyl)-methane



B. P., °C @ 760mm 186.5–188

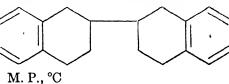
 $D_4^{20}$ 

1.0525

 $D_0^{20}$ 

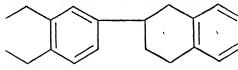
11

Bi-2,2'-(1,2,3,4-tetrahydronaphthyl)



M. P., °C 113<sup>72</sup>

Bi-2,6'-(1,2,3,4-tetrahydronaphthyl)



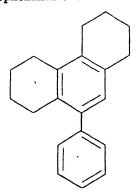
M. P., °C 53-54<sup>57</sup>

1,2-Diisobutylideneacenaphthene

B. P., °C @ 760mm 215

1541

9-Phenyl-1,2,3,4,5,6,7,8-octahydrophenanthrene

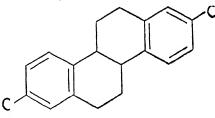


M. P., °C 94<sup>22</sup>

B. P., °C @ 760mm 186

 $0.2^{22}$ 

2,8-Dimethyl-4b,5,6,10b,11,12-hexahydrochrysene



M. P., °C 140 (a)<sup>49</sup> 108 (a)<sup>49</sup>

(a) These constants were determined on different forms.

4b, 10b-Dimethyl-4b, 5, 6, 10b, 11, 12hexahydrochrysene

M. P., °C

144 (a)<sup>49</sup> 105 (a)<sup>49</sup>

(a) These constants were determined on different forms.

Benzo-[fg]-1,1-dimethyl-2,4a-methylene-1,2,3,4,4a,9,9a-heptahydroanthracene (a)

B. P., °C @ 760mm

185-186

11

 $D_{ullet}^{20}$ 

1.0991

 $D_0^{20\ 1}$ 

 $n_{\rm p}^{20}$ 

 $1.6205^{1}$ 

(a) The structure of this compound was not clearly defined in the literature.

Cyclohexano-[g]-cyclopentano-[jk]-1,2,3,4-tetrahydrophenanthrene



M. P., °C 148.6–149.0<sup>28</sup>

 $C_{21}H_{24}$ 

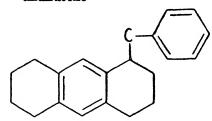
1-(3'-Acenaphthyl)-2-(2''-methylcyclohexen-1''-yl)-ethane

B. P., °C @ 760mm

182-185

 $0.2 - 0.3^{14}$ 

1-Benzyl-1,2,3,4,5,6,7,8-octahydroanthracene



B. P., °C @ 760mm

244-246

1458

2-Benzyl-1,2,3,4,5,6,7,8-octahydroanthracene

M. P., °C

65-6658

B. P., °C @ 760mm

248-251

1258

2,3-(3',3',6',6'-Tetramethylcyclohexano)-fluorene

M. P., °C 93–9410

#### C22H26

- 1,1-Di-[x'-(x'<sub>4</sub>-tetrahydronaphthyl)]ethane (a)
- B. P., °C @ 760mm 384<sup>52</sup> 261–263

1652

- (a) The structure of this compound was not clearly defined in the literature.
- 3,4,7,8-Dibenzo-2,6-di-n-propylbicyclo-[3,3,0]-octane

(9,12-Di-n-propyldiphensuccindane)

$$\begin{array}{c} C - C - C \\ \hline \\ C - C - C \end{array}$$

M. P., °C 98–996

3,4,7,8-Dibenzo-2,6-diisopropylbicyclo-[3,3,0]-octane

(9,12-Diisopropyldiphensuccindane)

$$\begin{array}{c} c-c-c \\ \hline \\ c-c-c \end{array}$$

M. P., °C 80–81• 1-Isopropyl-6,7-cyclohexano-8methyl-9,10-dihydrophenanthrene

M. P., °C 44.8–46<sup>27</sup>

3,6b-Dimethyl-6b,7,8,9,10,10a,11, 12-octahydrocholanthrene

M. P., °C 132-134<sup>13</sup>

### C28HJ8

x,x,x-Tricyclohexylnaphthalene (a)

M. P., °C 121-12247

(a) The structure of this compound was not clearly defined in the literature.

#### C<sub>80</sub>H<sub>42</sub>

1,2,3,4,5,6-Tri-(3',2'-octahydronaphtho)-benzene

M. P., °C 360–362<sup>48</sup>

#### $C_{33}H_{48}$

## 3-Phenylcholestadiene-2,4 (a)

M. P., °C

174 - 17565

Additional Data

$$[\alpha]_{\rm p}^{29} = -133^{\circ} \, ^{65}$$

(a) The structure of this compound was not clearly defined in the literature.

## 3-Phenylcholestadiene-3,5

M. P., °C 158–159³

**Additional Data** 

$$[\alpha]_{\rm p}^{25} = -132^{\circ}$$

## $C_{35}H_{52}$

## 3-Phenethylcholestadiene-2,4

·M. P., °C 94–9519

#### C<sub>36</sub>H<sub>54</sub>

#### Phenylfriedelene (a)

M. P., °C 269-271<sup>21</sup>

(a) The structure of this compound was not clearly defined in the literature.

References on Miscellaneous Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-18}$ 

- Arbusov, B. A., J. Gen. Chem. (U.S. S.R.) 9, 239 1939; Chem. Zentr. 1939, II, 1284.
- Bachmann, W. E., and M. C. Kloetzel, J. Am. Chem. Soc. 59, 2207 1937.
- 3. Bergmann, W., and F. Hirschmann, J. Org. Chem. 4, 40 1939.
- Bernal, J. D., Chemistry & Industry
   52, 288 1933.
- 5. Brand, K., and K. O. Müller, Ber. 55, 601 1922.
- Brand, K., and T. Sasaki, Ber. 58, 2546
   1925.

References

- Brand, K., and F. Schläger, Ber. 56, 2541 1923.
- Bredt, J., J. prakt. Chem. [2] 98, 96 1918.
- Brown, W. G., and B. A. Bluestein, J. Am. Chem. Soc. 65, 1082 1943.
- Bruson, H. A., and J. W. Kroeger, J. Am. Chem. Soc. 62, 36 1940.
- Burger, A., and E. Mosettig, J. Am. Chem. Soc. 59, 1302 1937.
- Cohen, A., J. W. Cook, and C. L. Hewett, J. Chem. Soc. 1935, 1633.
- Cook, J. W., and G. A. D. Haslewood,
   J. Chem. Soc. 1934, 428.
- Cook, J. W., G. A. D. Haslewood, and A. M. Robinson, J. Chem. Soc. 1935, 667.
- Cook, J. W., and C. L. Hewett, J. Chem. Soc. 1934, 365.
- Cook, J. W., C. L. Hewett, W. V. Mayneord, and E. Roe, J. Chem. Soc. 1934, 1727.
- 17. Coulson, E. A., J. Chem. Soc. 1937, 1298.
- 18. Courtot, C., Ann. chim. [9] 4, 168 1915.
- Dansi, A., Gazz. chim. ital. 68, 273
   1938.
- Daufresne, Bull. soc. chim. [4] 1, 1233
   1907.
- Drake, N. L., and W. P. Campbell, J. Am. Chem. Soc. 58, 1681 1936.
- Eschinazi, H. E., and F. Bergmann, J. Am. Chem. Soc. 65, 1411 1943.
- Ferrer, J., Anales soc. españ. fis. quím.
   459 1922; C. A. 17, 3177 1923; Chem.
   Zentr. 1923, III, 1161.
- 24. Fichter, F., and H. Stenzl, Helv. Chim. Acta 22, 425 1939.
- Fichter, F., and H. Stenzl, Roczniki Chem. 18, 510 1938; Chem. Zentr. 1939, II, 3053.
- Fieser, L. F., and J. Cason, J. Am. Chem. Soc. 62, 1293 1940.
- Fieser, L. F., and R. C. Clapp, J. Am. Chem. Soc. 63, 319 1941.
- Fieser, L. F., M. Fieser, and E. B. Hershberg, J. Am. Chem. Soc. 58, 1463 1936.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 59, 2502 1937.
- France, H., P. Maitland, and S. H.
   Tucker, J. Chem. Soc. 1937, 1739.
- France, H., S. H. Tucker, and J. Forrest,
   J. Chem. Soc. 1945, 7.

- Freund, M., K. Fleischer, and J. Stemmer, Ann. 414, 44 1918.
- 33. Hewett, C. L., J. Chem. Soc. 1936, 596.
- 34. Koelsch, C. F., J. Am. Chem. Soc. 55, 3885 1933.
- 35. Krüber, O., Ber. 64, 84 1931.
- 36. Krüber, O., Ber. 67, 1000 1934.
- 37. Leuchs, H., and L. Lock, Ber. 48, 1432 1915.
- Maitland, P., and S. H. Tucker, J. Chem. Soc. 1929, 2559.
- Manchot, W., and P. Krische, Ann. 337, 170 1904.
- Maxim, N., Bull. soc. chim. [4] 43, 769
   1928.
- 41. Maxim, N., Bull. soc. chim. [4] **45**, 1137 **1929**.
- 42. Mayer, F., Ber. 46, 2579 1913.
- 43. Miller, H. F., and J. Bachman, J. Am. Chem. Soc. 57, 766 1935.
- Newman, M. S., J. Am. Chem. Soc. 60, 2947 1938.
- Pestemer, M., and F. Manchen, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 145, Abt. IIb, 312 1936.
- Pinkney, P. S., and C. S. Marvel, J. Am. Chem. Soc. 59, 2669 1937.
- Pokrovskaya, E. S., and T. G. Stepanseva, J. Gen. Chem. (U.S S R.) 9, 1953 1939; C.A. 34, 4731 1940.
- Pummerer, R, A. Luttringhaus, R. Fick, S. Pfaff, G. Riegelbauer, and E. Rosenhauer, Ber. 71, 2569 1938.
- Ramage, G. R., J. Chem. Soc. 1938, 397.
- Ramage, G. R., and R. Robinson, J. Chem. Soc. 1933, 607.
- Ramart-Lucas, and J. Hoch, Bull. soc. chim. [5] 2, 1376 1935.
- Reilly, J. A., and J. A. Nieuwland, J. Am. Chem. Soc. 50, 2564 1928.
- 53. Roser, W., Ann. 247, 129 1888.
- Ruzicka, L., and H. Hosli, Helv. Chim. Acta 17, 470 1934.
- Schlenk, W., and E. Bergmann, Ann.
   463, 1 1928.
- Schlenk, W., O. Bergmann, and E. Bergmann, J. Soc. Chem. Ind. 52, 209T 1933.
- 57. Schroter, G., Ber. 57, 1990 1924.
- 58. Schroter, G., Ber. 57, 2003 1924.
- 59. Sieglitz, A., and II. Jassoy, Ber. 55, 2032 1922.
- Spilker, A., and W. Schade, Ber. 65, 1686 1932.

- 61. Stobbe, H., and E. Färber, Ber. 57, 1838 1924.
- Straus, F., R. Kühnel, and R. Haensel, Ber. 66, 1847 1933.
- Thiele, J., and J. Meisenheimer, Ann. 306, 225 1899.
- Ullmann, F., and R. von Wurstemberger, Ber. 38, 4105 1905.
- Urushibara, Y., T. Ando, H. Araki, and A. Ozawa, Bull. Chem. Soc. Japan 12, 353 1937.
- Vansheidt, A., and B. Moldavski, Ber.
   917 1931.
- Vansheidt, A., and B. Moldavski, J. Gen. Chem. (U.S.S.R.) 1, 304 1931;
   Chem. Zentr. 1931, II, 3208.
- Veselý, V., and F. Štursa, Chem. Listy
   29, 257 1935; Chem. Zentr. 1936, I,
   1221.
- von Braun, J., and E. Anton, Ber. 62, 145 1929.

- von Braun, J., and E. Anton, Ber. 67, 1051 1934.
- von Braun, J., O. Bayer, and L. F. Fieser, Ann. 459, 287 1927.
- von Braun, J., and G. Kirschbaum, Ber.
   54, 597 1921.
- von Braun, J., and G. Manz, Ann. 488,
   111 1931.
- 74. von Braun, J., and G. Manz, Ber. 63, 2608 1930.
- 75. Wagner-Jauregg, T., Ann. 491, 1 1931.
- Weiss, R., and K. Woidich, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 134, Abt. IIb, 453 1925.
- 77. Wieland, H., and O. Probst, Ann. 530, 274 1937.
- Wieland, H., F. Reindel, and J. Ferrer, Ber. 55, 3313 1922.
- 79. Wislicenus, W., and W. Mocker, Ber. 46, 2772 1913.

# VII. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-20}$

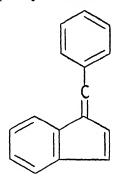
- 1. Indene Derivatives of Empirical Formula C<sub>n</sub>H<sub>2n-20</sub>
- 2. Naphthalene Derivatives of Empirical Formula C<sub>n</sub>H<sub>2n-20</sub>
- 3. Cyclanoanthracenes and Their Alkyl Derivatives
- 4. Cyclanophenanthrenes and Their Alkyl Derivatives
- 5. Miscellaneous Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-20</sub>

## 1. INDENE DERIVATIVES OF EMPIRICAL FORMULA C<sub>n</sub>H<sub>2n-20</sub>

#### C16H12

## 1-Benzylideneindene

(Phenylbenzofulvene)



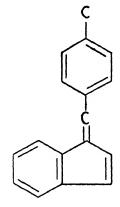
M. P., °C 88-89<sup>7</sup>

> 88<sup>2</sup>· <sup>4</sup> 87<sup>3</sup>

CT

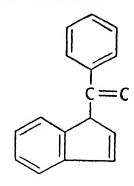
## $C_{17}H_{14}$

## 1-(4'-Methylbenzylidene)-indene



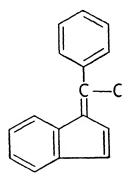
M. P., °C 91-92¹

## 1-Phenyl-1-indenylethene



M. P., °C 80<sup>2</sup>

## 1-Phenyl-1-indenylidene-ethane



M. P., °C 70<sup>2</sup> 68-69<sup>6</sup>

B. P., °C @ 760mm 204-205 166 178-179 56

## 1-Benzylidene-3-methylindene

M. P., °C 43–44<sup>5</sup>, 9

# $$C_{18}H_{16}$$ 1-Benzylidene-3-ethylindene

M. P., °C 588

## 1-Isopropylidene-3-phenylindene

M. P., °C 99.5–100<sup>10</sup>

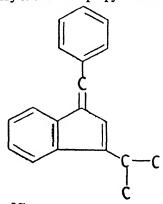
 $C_{10}H_{18} \\$  1-(4'-Isopropylbenzylidene)-indene

M. P., °C 60-61<sup>1</sup>

#### 1-Benzylidene-3-n-propylindene

M. P., °C 1328

## 1-Benzylidene-3-isopropylindene



M. P., °C 1016 References on Indene Derivatives of Empirical Formula  $C_nH_{2n-20}$ 

- 1. Bernthsen, W., Ann. 415, 274 1918.
- 2. Courtot, C., Ann. chim. [9] 4, 168 1915.
- 3. Thiele, J., Ber. 33, 851 1900.
- 4. Thiele, J., Ber. 33, 3395 1900.
- Thiele, J., and S. Bühner, Ann. 347, 249 1906.
- Thiele, J., and K. Merck, Ann. 415, 257 1918.
- Whitby, G. S., and M. Katz, J. Am. Chem. Soc. 50, 1160 1928.
- 8. Wislicenus, W., and W. Hentrick, Ann. 436, 9 1924.
- 9. Wuest, H. M., Ann. 415, 291 1918.
- Ziegler, K., and F. Crössmann, Ann.
   89 1934.

# 2. NAPHTHALENE DERIVATIVES OF EMPIRICAL FORMULA C<sub>n</sub>H<sub>2n-20</sub>

#### C16H12

Cyclopentadien-2,4-ylidene-(x'-naphthyl)-methane (a)

(a) The structure of this compound was not clearly defined in the literature.

## 1-Phenylnaphthalene

B. P., °C @ 750mm

336-337<sup>29</sup>

324-326<sup>30</sup>

324-325<sup>31</sup>

334

770<sup>61</sup>

194

18<sup>29</sup>

192-195

16-1

192–195 16–17<sup>51</sup> 190 12<sup>42</sup> 189–190 12<sup>60</sup> 187–189 12<sup>56</sup> 186-188  $10^{52}$  134-135  $2^{60}$   $D_4^{20}$ 

#### 2-Phenylnaphthalene

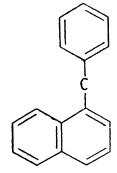
 $1.103^{56}$ 

M, P., °C 102 104-1051 104<sup>27</sup> 103-10410  $103^{23}$ 102.2-102.718  $102-102.5^{2}$ 101.5-1024 101-102 (a)  $101.5^{11, 32}$ 101-101.59 10149 98.5-10013 98-9919 97-9885

B. P., °C @ 760mm 357-358<sup>28</sup> 346-347<sup>54</sup> 345-346<sup>9, 11</sup> (a) The melting point 101-102 is found in references 7, 22, 31, 45, 46, 47, 50, 54, 55, 57.

#### C17H14

#### 1-Benzylnaphthalene



M. P., °C

58.6

59 (a)

59 (b)58

58-5943

 $58.5^{17}$ 

5833

57.5-588

57-5841

B. P., °C @ 760mm

35058

200-205

937

 $D_{4}^{20}$ 

1.165

00 88, 58

- (a) The melting point 59 is found in references 3, 14, 21, 25, 38, 44, 53.
- (b) This constant was given as a freezing point in the literature.

#### 2-Benzylnaphthalene

M. P., °C 58<sup>8</sup> 57<sup>3</sup> 55.5<sup>16</sup> 55-55.5<sup>53</sup> 39-40<sup>41</sup> 35.5<sup>28</sup>

 $D_4^{20}$ 

1.176

00 \$8, 58

#### x-Benzylnaphthalene (a)

M. P., °C 64<sup>20</sup> 58.6<sup>30</sup> 58<sup>28</sup>

B. P., °C @ 760mm

168.5-169

 $D_4^{20}$ 

1.077724

1.166

17° 80

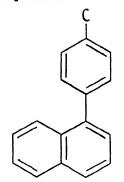
1.524

(a) The structure of this compound was not clearly defined in the literature.

### 1-o-Tolylnaphthalene

M. P., °C 67.5-68.5<sup>18</sup> 63<sup>26</sup>

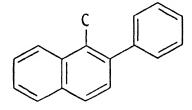
#### 1-p-Tolylnaphthalene



B. P., °C @ 760mm 148-150

 $0.15^{5}$ 

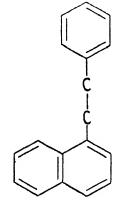
## 1-Methyl-2-phenylnaphthalene



M. P., °C 84<sup>49</sup>

 $C_{18}H_{16}$ 

## 1-Phenethylnaphthalene



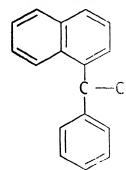
B. P., °C @ 760mm 175

512

## 2-Phenethylnaphthalene

1-Phenyl-1-(1'-naphthyl)-ethane

99-10012

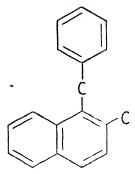


M. P., °C 6948

B. P., °C @ 760mm 220-222

1548

## 1-Benzyl-2-methylnaphthalene



B. P., °C @ 760mm 221–222

1718

2-Methyl-7-p-tolylnaphthalene

M. P., °C 140–141<sup>55</sup>

 $C_{20}H_{20}$ 

1-Phenyl-4-(1'-naphthyl)-butane

M. P., °C 80.5–82<sup>40</sup>

 $C_{21}H_{22}$ 

1,2-Dimethyl-5-(2'-methylphenethyl)naphthalene

M. P., °C 6039

 $C_{22}H_{24}$ 

1,2-Dimethyl-5-(2',3'-dimethyl-phenethyl)-naphthalene

M. P., °C 90.5-91.5<sup>39</sup>

C31H42

1-Methyl-x<sub>4</sub>-tetracyclopentylnaphthalene (a)

M. P., °C 9734

(a) The structure of this compound was not clearly defined in the literature.

C34H48

x4-Tetracyclohexylnaphthalene (a)

M. P., °C 269<sup>35</sup>

(a) The structure of this compound was not clearly defined in the literature.

References on Naphthalene Derivatives of Empirical Formula  $C_nH_{2n-20}$ 

 Bamberger, E., and C. Burgdorf, Ber. 23, 2433 1890.

- Bamberger, E., and F. Chattaway, Ber. 26, 1745 1893.
- Barrett, J. W., and R. P. Linstead, J. Chem. Soc. 1936, 611.
- Baumgarten, P., and J. Olshausen, Ber. 64, 925 1931.
- Bergmann, F., and A. Weizmann, J. Org. Chem. 9, 352 1944.
- Blicke, F. F., and L. D. Powers, J. Am. Chem. Soc. 51, 3378 1929.
- Bodroux, D., Ann. chim. [10] 11, 511 1929.
- Borsche, W., P. Hofmann, and H. Kuhn, Ann. 554, 23 1943.
- 9. Breuer, A., and T. Zincke, Ber. 11, 1403 1878.
- Carter, H. E., and E. J. Van Loon, J. Am. Chem. Soc. 60, 1077 1938.
- Chattaway, F. D, and W. H. Lewis,
   J. Chem. Soc. 65, 869 1894.
- Cook, J. W., and C. L. Hewett, J. Chem. Soc. 1934, 365.
- Cook, J. W., and C. A. Lawrence, J. Chem. Soc. 1936, 1431.
- 14 Dziewoński, K., and S. Dziecielewski, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1927A, 273.
- Dziewoński, K., and E. Ritt, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1927A, 181.
- Dziewoński, K., and S. Wodelski, Roczniki Chem. 12, 366 1932; C. A. 27, 2145 1933.
- Elbs, K., J. prakt. Chem. [2] 35, 465 1887.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 60, 940 1938.
- 19. Fittig, Ber. 5, 806 1872.
- 20. Froté, C., Jahresber. 27, 390 1873.
- Gasopoulos, I., Praktika Akad. Athenon 7, 47 1932; C. A. 28, 1672 1934;
   Chem. Zentr. 1933, II, 43.
- Hey, D. H., and S. E. Lawton, J. Chem. Soc. 1940, 374.
- Kruber, O., and A. Marx, Ber. 71, 2478
   1938
- Larsen, R. G., R. E. Thorpe, and F. A. Armfield, Ind. Eng. Chem. 34, 183 1942.
- Lorriman, F. R., J. Am. Chem. Soc. 47, 211 1925.
- Mayer, F., and R. Schiffner, Ber. 65, 1337 1932.

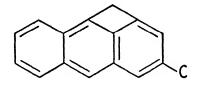
- Mayer, F., and R. Schiffner, Ber. 67, 67 1934.
- McKenzie, A., and W. S. Dennler, J. Chem. Soc. 125, 2105 1924.
- Michael, A., and J. E. Bucher, Am. Chem. J. 20, 89 1898.
- Miguel, P., Bull. soc. chim. [2] 26, 2
   1876.
- Möhlau, R., and R. Berger, Ber. 26, 1196 1893.
- 32. Möhlau, R., and R. Berger, Ber. 26, 1994 1893.
- 33. Nenitzescu, C. D., D. A. Isăcescu, and C. D. Ionescu, Ann. 491, 210 1931.
- Pokrovskaya, E. S., J. Gen. Chem. (U.S.S.R.) 13, 579 1943; Survey For. Petrol. Liter. T472B Oct. 20, 1944.
- Pokrovskaya, E. S., and T. G. Stepanzeva, J. Gen. Chem. (U.S.S.R.) 9, 1953 1939; C. A. 34, 4731 1940.
- Postovskii, I., N. Apollov, and B. Lugovkin, Ann. inst. polytech. Oural
   241 1927; C.A. 22, 4787 1928.
- Price, C. C., and J. M. Ciskowski, J. Am. Chem. Soc. 60, 2499 1938.
- Roux, L., Ann. chim. phys. [6] 12, 289
   1887.
- Ruzicka, L., A. Grob, and G. Anner, Helv. Chim. Acta 26, 254 1943.
- 40. Schroeter, G., Ber. 57, 1990 1924.
- 41. Sharma, V. N., and S. Dutt, J. Indian Chem. Soc. 12, 774 1935.
- Sherwood, I. R., W. F. Short, and R. Stansfield, J. Chem. Soc. 1932, 1832.
- 43. Shima, G., Mem. Coll. Sci., Kyoto Imp. Univ. 13A, 315 1930.
- Skraup, S., and K. Böhm, Ber. 59, 1007
   1926.
- 45. Smith, W., Ber. 12, 1396 1879.
- 46. Smith, W., and T. Takamatsu, J. Chem. Soc. 39, 546 1881.
- Späth, E., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 121, Abt. IIb, 683 1912.
- 48. Spilker, A., and W. Schade, Ber. 65, 1686 1932.
- 49. Spring, F. S., J. Chem. Soc. 1934, 1332.
- Stoermer, R., and O. Kippe, Ber. 36, 3992 1903.
- 51. Straus, F., Ann. 342, 190 1905.
- Veselý, V., and F. Štursa, Collection Czechoslov. Chem. Commun. 5, 343 1933.

- Vincent, C., and L. Roux, Bull. soc. chim. [2] 40, 163 1883.
- 54. Volhard, J., Ann. 296, 1 1897.
- von Auwers, K., and G. Keil, Ber. 36, 3902 1903.
- von Braun, J., and E. Anton, Ber. 67, 1051 1934.
- 57. von Braun, J., and G. Manz, Ann. 468, 258 1929.
- von Meyer, E., J. prakt. Chem. [2]
   538 1910.
- 59. Wagner-Jauregg, T., Ann. 491, 1 1931.
- 60. Weiss, R., Org. Syntheses 24, 84 1944.
- 61. Weiss, R., and K. Woidich, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 134, Abt. IIb, 453 1925.

# 3. CYCLANOANTHRACENES AND THEIR ALKYL DERIVATIVES, C<sub>n</sub>H<sub>2n-20</sub>

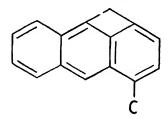
#### $C_{16}H_{12}$

Cyclobutano-[mn]-3-methylanthracene



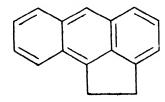
M. P., °C 85<sup>2</sup>, 4

Cyclobutano-[mn]-4-methylanthracene



M. P., °C 63<sup>2</sup>

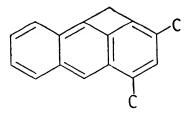
Cyclopentano-[de]-anthracene (Aceanthrene)



M. P., °C 113°

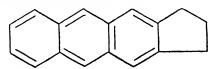
#### C<sub>17</sub>H<sub>14</sub>

Cyclobutano-[mn]-2,4-dimethylanthracene



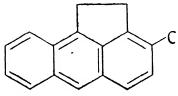
M. P., °C . 643

2,3-Cyclopentanoanthracene



M. P., °C 242-243<sup>10</sup>

Cyclopentano-[mn]-2-methylanthracene



M. P., °C 122¹

#### C18H18

## 1,2-Cyclopentano-5-methylanthracene

M. P., °C 131–132<sup>8</sup>

#### Cyclopentano-[fg]-1,2-dimethylanthracene

M. P., °C 206–207<sup>8</sup>

#### 1,2-Cyclohexanoanthracene

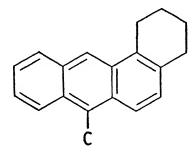
M. P., °C 104.5–105<sup>7</sup>

### C19H18

## 1,2-Cyclohexano-4-methylanthracene

M. P., °C 82.3-82.9<sup>8</sup>

#### 1,2-Cyclohexano-10-methylanthracene



M. P., °C 117.3-117.8<sup>7</sup>

### $C_{20}H_{20}$

#### 1,2-Cyclohexano-4,9-dimethylanthracene

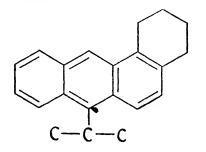
M. P., °C 62.4-62.8\*

1,2-Cyclohexano-4,10-dimethylan-thracene

M. P., °C 105–105.5<sup>8</sup>

#### C21H22

#### 1,2-Cyclohexano-10-isopropylanthracene



#### M. P., °C 81.9-82.5°

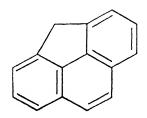
233

# References on Cyclanoanthracenes and Their Alkyl Derivatives

- 1. Dansi, A., and A. Sempronj, Gazz. chim. ital. 66, 182 1936.
- 2. Elbs, K., J. prakt. Chem. [2] 41, 1 1890.
- 3. Elbs, K., J. prakt. Chem. [2] 41, 121 1890.
- Elbs, K., and M. Günther, Ber. 20, 1364 1887.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 59, 394 1937.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 59, 2331 1937.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 59, 2502 1937.
- Fieser, L. F., and R. N. Jones, J. Am. Chem. Soc. 60, 1940 1938.
- Fieser, L. F., and M. A. Peters, J. Am. Chem. Soc. 54, 4373 1932.
- von Braun, J., G. Kirschbaum, and H. Schumann, Ber. 53, 1155 1920.

# 4. CYCLANOPHENANTHRENES AND THEIR ALKYL DERIVATIVES, $C_nH_{2n-20}$

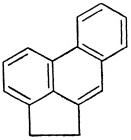
 $$C_{15}H_{10}$$  Cyclopentano-[def]-phenanthrene



M. P., °C 116<sup>35</sup> 114.6–115.3<sup>18</sup> 114.0–115.0<sup>7</sup>

B. P., °C @ 760mm 353<sup>25</sup>

# $$C_{16}H_{12}$$ Cyclopentano-[jk]-phenanthrene



M. P., °C 106<sup>25</sup>

#### $C_{17}H_{14}$

1,2-Cyclopentanophenanthrene

M. P., °C

135

137-138<sup>21</sup>

135-136<sup>6</sup>, <sup>32</sup>

134.4-135.8<sup>38</sup>

135<sup>8</sup>, <sup>28</sup>, <sup>32</sup>

134.5-135<sup>17</sup>

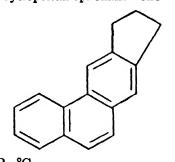
134-135<sup>40</sup>

133-134<sup>15</sup>, <sup>40</sup>

132-133<sup>5</sup>

130-131<sup>5</sup>

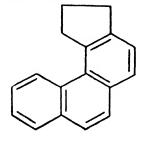
### 2,3-Cyclopentanophenanthrene



M. P., °C 85 85–85.5<sup>11</sup>, <sup>36</sup> 84–84.5<sup>16</sup> 83.5–84.5<sup>6</sup>

8441

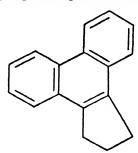
## 3,4-Cyclopentanophenanthrene



M. P., °C 73.5-75 (a)<sup>5</sup> 71.5-72<sup>6</sup>

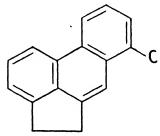
(a) This compound remelts at 60.

## 9,10-Cyclopentanophenanthrene



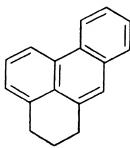
M. P. °C 154<sup>44, 45</sup> 150–151<sup>14</sup> 149–150<sup>5</sup>

#### Cyclopentano-[jk]-1-methylphenanthrene



M. P., °C 159<sup>25</sup>

## Cyclohexano-[jk]-phenanthrene



#### C18H16

#### 1,2-(3'-Methylcyclopentano)-phenanthrene

(Diels' Hydrocarbon)

M. P., °C

126

128-129 (a)19

126.5-12781

126-12712, 30, 38

12637

125-12622, 27, 29, 43

125 (a)20

124-12510, 42

123-12439

B. P., °C @ 760mm

162-165

 $0.1^{29}$ 

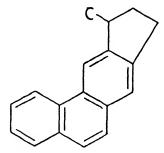
(a) Diels assigned the structure of 1,2- (cyclopenten-3'-o) -7,8- (3''ethylcyclopenten-3''-o) - naphthalene to this compound, but it was later proved to be the above compound.

#### 1,2-(4'-Methylcyclopentano)-phenanthrene

#### 1,2-(5'-Methylcyclopentano)-phenanthrene

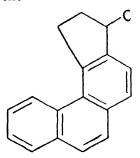
M. P., °C 76-77<sup>40</sup>

#### 2,3-(3'-Methylcyclopentano)-phenanthrene



M. P., °C 75-76<sup>30</sup>

#### 3,4-(5'-Methylcyclopentano)-phenanthrene



M. P., °C 28-2930

B. P., °C @ 760mm 172-173

 $0.05^{30}$ 

1,2-Cyclopentano-7-methylphenanthrene

M. P., °C 132<sup>34</sup>

1,2-Cyclopentano-9-methylphenanthrene

M. P., °C 109-110<sup>26</sup>

x,x-Cyclopentano-x-methylphenanthrene (a)

M. P., °C 147°

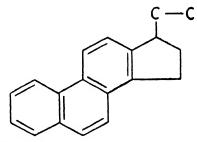
(a) The structure of this compound was not clearly defined in the literature.

## 2,3-Cyclohexanophenanthrene

M. P., °C 88.5–89.5<sup>24</sup>

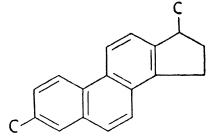
 $C_{19}H_{18}$ 

1,2-(3'-Ethylcyclopentano)-phenanthrene



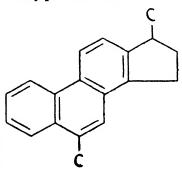
M. P., °C 85–86<sup>38</sup>

1,2-(3'-Methylcyclopentano)-7methylphenanthrene



M. P., °C 139-140<sup>34</sup>

1,2-(3'-Methylcyclopentano)-9methylphenanthrene



## x,x-Dimethyl-x,x-cyclopentanophenanthrene (a)

M. P., °C 165<sup>28</sup>

(a) The structure of this compound was not clearly defined in the literature.

## 2,3-(4'-Methylcyclohexano)-phenanthrene

M. P., °C 114-115.54

## 2,3-Cyclohexano-1-methylphenanthrene

M. P., °C 73.7-74.5<sup>18</sup> 73.9-74.4<sup>24</sup>

 $C_{20}H_{20}$ 

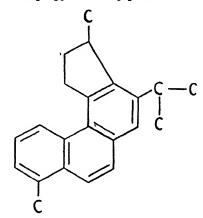
1,2-(3'-Isopropylcyclopentano)phenanthrene

$$c-c-c$$

M. P., °C 97.6-98.4<sup>38</sup> 96<sup>13</sup>

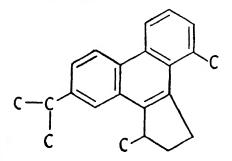
## $C_{22}H_{24}$

3,4-(5'-Methylcyclopentano)-2isopropyl-8-methylphenanthrene



M. P., °C 74.5–75.5³

9,10-(5'-Methylcyclopentano)-1methyl-7-isopropylphenanthrene



## M. P., °C 74.5–75.5<sup>1</sup>

# 3,4-Cyclohexano-2-isopropyl-8-methylphenanthrene

M. P., °C 88–89²

## References on Cyclanophenanthrenes and Their Alkyl Derivatives

- Adelson, D. E., and M. T. Bogert, J. Am. Chem. Soc. 59, 399 1937.
- Adelson, D. E., and M. T. Bogert, J. Am. Chem. Soc. 59, 1776 1937.
- 3. Adelson, D. E., and M. T. Bogert, Proc. Nat. Acad. Sci. U. S. 23, 117 1937.
- Bachmann, W. E., and J. M. Chemerda,
   J. Org. Chem. 6, 36 1941.
- Bachmann, W. E., and M. C. Kloetzel,
   J. Am. Chem. Soc. 59, 2207 1937.
- Bachmann, W. E., and M. C. Kloetzel,
   J. Am. Chem. Soc. 60, 2204 1938.
- Bachmann, W. E., and J. C. Sheenan,
   J. Am. Chem. Soc. 63, 204 1941.
- 8. Bardhan, J. C., J. Chem. Soc. 1936, 1848.
- 9. Barger, G., and H. L. Fraenkel-Conrat,
- J. Chem. Soc. 1936, 1537.10. Bergmann, E., and F. Bergmann, J. Chem. Soc. 1939, 1019.
- Bergmann, E., and O. Blum-Bergmann, J. Am. Chem. Soc. 59, 1572 1937.
- Bergmann, E., and H. Hillemann, Ber.
   1302 1933.
- Birch, A. J., and R. Robinson, J. Chem. Soc. 1944, 503.

- 14. Bradsher, C. K., J. Am. Chem. Soc. 61, 3131 1939.
- Burger, A., and E. Mosettig, J. Am. Chem. Soc. 59, 1302 1937.
- Clar, E., and F. Furnari, Ber. 65, 1420
   1932.
- Cook, J. W., and C. L. Hewett, J. Chem. Soc. 1933, 1098.
- Davis, W. W., M. E. Krahl, and G. H. A. Clowes, J. Am. Chem. Soc. 62, 3080 1940.
- Diels, O., W. Gädke, and P. Körding, Ann. 459, 1 1927.
- Diels, O., and A. Karstens, Ann. 478, 129 1930.
- Diels, O., and H. Klare, Ber. 67, 113 1934.
- Elderfield, R. C., and W. A. Jacobs, J. Biol. Chem. 107, 143 1934.
- Farmer, S. N., and G. A. R. Kon, J. Chem. Soc. 1937, 414.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 59, 2502 1937.
- Ficser, L. F., and M. A. Peters, J. Am. Chem. Soc. 54, 4373 1932.
- Gamble, D. J. C., and G. A. R. Kon,
   J. Chem. Soc. 1935, 443.
- Gamble, D. J. C., G. A. R. Kon, and B. Saunders, J. Chem. Soc. 1935, 644.
- Hawthorne, J. R., and R. Robinson, J. Chem. Soc. 1936, 763.
- 29. Hillemann, H., Ber. 68, 102 1935.
- 30. Hillemann, H., Ber. 69, 2610 1936.
- 31. Kägi, H., and K. Miescher, Helv. Chim. Acta 22, 1683 1939.
- Koebner, A., and R. Robinson, J. Chem. Soc. 1938, 1994.
- 33. Kon, G. A. R., J. Chem. Soc. 1933, 1081.
- Kon, G. A. R., and A. M. Woolman, J. Chem. Soc. 1939, 794.
- 35. Krüber, O., Ber. 67, 1000 1934.
- 36. Mohler, H., and J. Sorge, Helv. Chim. Acta 22, 229 1939.
- Müller, M., Z. physiol. Chem. 233, 223
   1935.
- Riegel, B., M. H. Gold, and M. A. Kubico, J. Am. Chem. Soc. 65, 1772 1943.
- 39. Rochelmeyer, H., Arch. Pharm. 274, 543 1936.

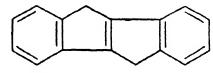
- Ruzicka, L., L. Ehmann, M. W. Goldberg, and H. Hösli, Helv. Chim. Acta 16, 833 1933.
- Sen-Gupta, Science and Culture 3,
   1937; C.A. 31, 7868 1937; Chem.
   Zentr. 1938, I, 4621.
- 42. Tschesche, R., and H. Knick, Z. physiol. Chem. 222, 58 1933.
- 43. Tschesche, R., and H. A. Offe, Ber. 68, 1998 1935.
- 44. Weizmann, C., and E. Bergmann, Scripta Academica Hierosolymitana (Jerusalem) Sci. Report #1, 1938.
- Weizmann, C., E. Bergmann, and T. Berlin, J. Am. Chem. Soc. 60, 1331 1938.

# 5. MISCELLANEOUS POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-20}$

#### C16H12

2,3,6,7-Dibenzobicyclo-[3,3,0]octene-1,5

(Diphensuccindene-10)



M. P., °C 210<sup>10</sup>, 11

x,x-Dihydrobenzo-[jk]-fluorene (a)

M. P., °C 76<sup>25</sup>

- (a) The structure of this compound was not clearly defined in the literature.
- 1,2-Dihydropyrene



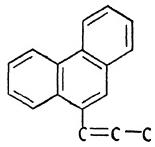
M. P., °C 13214

x,x-Dihydropyrene (a)

M. P., °C 106<sup>29</sup> (a) The structure of this compound was not clearly defined in the literature.

## C17H14

9-(Propen-1'-yl)-phenanthrene



B. P., °C @ 760mm

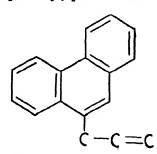
 $n_{\,\mathtt{D}}^{\,20}$ 

1.6928

27.5° 6

 $2.5^{6}$ 

9-(Propen-2'-yl)-phenanthrene



M. P., °C 516 240

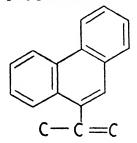
B. P., °C @ 760mm 161-163

1.256

 $n_{\rm p}^{20}$ 

1.62806

## 9-Isopropenylphenanthrene



M. P., °C 386

B. P., °C @ 760mm 163

206

 $n_{\,{\scriptscriptstyle \mathrm{D}}}^{\,20}$ 

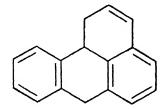
1.67656

## x,x-Dihydrobenzo-[lm]-2-methylfluorene (a)

M. P., °C 127-12839

(a) The structure of this compound was not clearly defined in the literature.

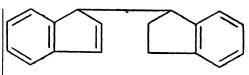
## 8,9-Benzo-7,9a-dihydrophenalene



M. P., °C 81<sup>5</sup> 79–80<sup>34</sup>

C18H16

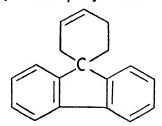
1-(1'-Indanyl)-indene



B. P., °C @ 760mm 190-192

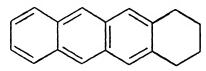
128

Spiro[fluorene-9,1'-cyclohexene-3']



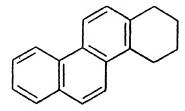
M. P., °C 145.5<sup>41</sup>

## 1,2,3,4-Tetrahydronaphthacene



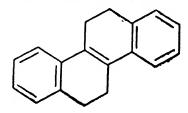
M. P., °C 233<sup>38</sup>

## 1,2,3,4-Tetrahydrochrysene



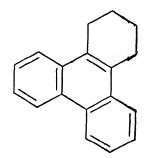
M. P., °C 180.5–181.5<sup>2</sup>

## 5,6,11,12-Tetrahydrochrysene



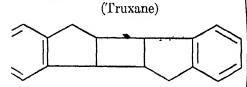
## M. P., °C 105<sup>33</sup>

## 1,2,3,4-Tetrahydrotriphenylene



M. P., °C 122–123° 120–1217

## 1,2,3,4-Di-(3',2'-indano)-cyclobutane



M. P., °C 116<sup>30, 87</sup>

B. P., °C @ 760mm 205 20<sup>37</sup> 206–207 13<sup>30</sup>

## $C_{19}H_{18}$

# Benzo-[lm]-2,4,4-trimethyl-4,4a-dihydrofluorene

M. P., °C 77-79<sup>22</sup>

## 1,2-Benzo-8-methyl-3,4,5,6-tetrahydroanthracene

M. P., °C 70–70.5<sup>19</sup>

## 2-Methyl-1,2,3,4-tetrahydronaphthacene

M. P., °C 203<sup>13</sup>

## 1-Methyl-1,2,3,4-tetrahydrochrysene

M. P., °C 120.5–121<sup>3</sup>

## 2-Methyl-1,2,3,4-tetrahydrochrysene

M. P., °C 141.5–142<sup>2</sup> 3-Methyl-1,2,3,4-tetrahydrochrysene

M. P., °C 130–131³

11-Methyl-1,2,3,4-tetrahydrochrysene

M. P., °C 71-72¹

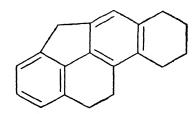
2-Methyl-1,2,3,4-tetrahydrotriphenylene

M. P., °C 116.2-116.8<sup>21</sup>

Cyclohexano-[b]-cyclopentano-[def]-9,10-dihydrophenanthrene

M. P., °C 83–83.5<sup>16</sup>

Cyclopentano-[mno]-1,2,3,4,5,6-hexahydrochrysene



M. P., °C 116.6-117.2<sup>15</sup>

 $C_{20}H_{20}$ 

9-Cyclohexylanthracene

M. P., °C 135–136<sup>42, 43</sup>

9-Phenyl-1,2,3,4,4a,9a-hexahydroanthracene

B. P., °C @ 760mm 235

 $1.5^{23}$ 

1,2-Benzo-8-ethyl-3,4,5,6-tetrahy-droanthracene

M. P., °C 65-67<sup>20</sup>

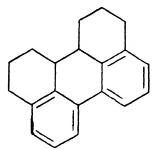
2,8-Dimethyl-1,2,3,4-tetrahydronaphthacene

M. P., °C 214<sup>13</sup>

2,3-Dimethyl-1,2,3,4-tetrahydrotriphenylene

M. P., °C 158–160<sup>16</sup>

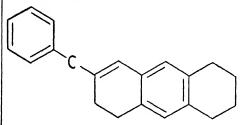
1,2,3,10,11,12,12a,12b-Octahydroperylene



M. P., °C 159–161<sup>45</sup> 119–120<sup>44</sup>

 $C_{21}H_{22}$ 

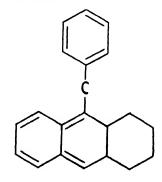
7-Benzyl-1,2,3,4,5,6-hexahydroanthracene



B. P., °C @ 760mm 255–258

1335

9-Benzyl-1,2,3,4,4a,9a-hexahydroanthracene



B. P., °C @ 760mm 255–258

2023. 24

1-Methyl-7-isopropyl-x-isopropenyl-phenanthrene (a)

M. P., °C 64.5-65.5<sup>28</sup>

(a) The structure of this compound was not clearly defined in the literature.

10-Isopropyl-x<sub>4</sub>-tetrahydro-1,2-benzoanthracene (a)

M. P., °C 72.5-73.5<sup>17</sup>

(a) The structure of this compound was not clearly defined in the literature.

3-Methyl-2a,3,4,5,11,12-hexahydrocholanthrene

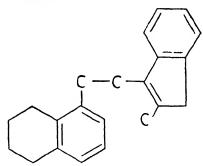
M. P., °C 157<sup>40</sup> 156–157<sup>36</sup>

3-Methyl-6,7,8,9,10,12b-hexahydrocholanthrene

M. P., °C 160-160.5<sup>18</sup>

C22H24

1-(5',6',7',8'-Tetrahydronaphthyl)-2-[3''-(2''-methylindenyl)]ethane



M. P., °C 73<sup>12</sup> B. P. °C @ 760mm

B. P., °C @ 760mm 190

0.312

1-Methyl-1,2-(2',3'-indano)-1,2,3, 4,5,6,7,8-octahydrophenanthrene

M. P., °C 123.5–124<sup>12</sup>

C24H28

1-[1'-(5'-Methyl-3,4-dihydronaphthyl)]-2-[1''-(5''-methyl-1'', 2'',3'',4''-tetrahydronaphthyl)]-ethane

B. P., °C @ 760mm 176–178

 $0.1^{32}$ 

1-[1'-(5'-Methyl-3',4'-dihydronaphthyl)]-2-[1''-(8''-methyl-1'', 2'',3'',4''-tetrahydronaphthyl)]ethane

B. P., °C @ 760mm 185–186

 $0.1^{31}$ 

9,10-Dipentylidene-9,10-dihydroanthracene

M. P., °C 103–108<sup>26</sup>

1,2-Benzo-2a,5-diisobutyl-2a,5-dihydroacenaphthylene (a)

M. P., °C 160<sup>27</sup>

(a) The structure of this compound was not clearly defined in the literature.

5,6,7,8-Dicyclopentano-1,2,3,4tetramethyl-9,10-dihydroanthracene

M. P., °C 255–2574

## C25H30

2-(5,6-Dimethyl-3,4-dihydronaphthyl)-1'-(5',6'-dimethyl-1',2', 3',4'-tetrahydronaphthyl)methane C25H30

246

B. P., °C @ 760mm 205–210

 $0.05^{32}$ 

## C26H32

1-[5'-(1',2',3',4'-Tetrahydronaphthyl)]-2-[3''-(2'',4''-dimethyl-7''-isopropylindenyl)]-ethane

$$\begin{array}{c|c} & & \\ & &$$

B. P., °C @ 760mm 218

 $0.3^{12}$ 

1,2-(2',1'-Naphtho)-5,9a-dimethyl-8-isopropyl-3,4,4a,5,6,7,8,9aoctahydrofluorene

B. P., °C @ 760mm 215–220

 $0.3^{12}$ 

1,2,3,4-Tetramethyl-5,6,7,8-dicyclohexano-9,10-dihydroanthracene

M. P., °C 250–251<sup>4</sup>

References on Miscellaneous Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-20}$ 

- Bachmann, W. E., and R. O. Edgerton,
   J. Am. Chem. Soc. 62, 2550 1940.
- Bachmann, W. E., and W. S. Struve, J. Org. Chem. 4, 456 1939.

247 References

- Bachmann, W. E., and W. S. Struve, J. Org. Chem. 5, 416 1940.
- Backer, H. J, and L. H. Huisman, Rec. trav. chim. 60, 557 1941; C.A. 37, 5982 1943.
- Bally, O., and R. Scholl, Ber. 44, 1656
   1911.
- Bergmann, E., and F. Bergmann, J. Am. Chem. Soc. 59, 1443 1937.
- Bergmann, E , and O. Blum-Bergmann,
   J. Am. Chem. Soc. 59, 1441 1937.
- 8. Bergmann, E., and H. Taubadel, Ber. 65, 463 1931.
- Bradsher, C. K., J. Am. Chem. Soc. 61, 3131 1939.
- Brand, K., and K. O. Müller, Ber 55, 601 1922.
- Brand, K., K.O. Müller, and H. Kessler, Ber. 59, 1962 1926.
- Cook, J. W., C. L. Hewett, W. V. Mayneord, and E. Roc, J. Chem. Soc. 1934, 1727.
- 13. Coulson, E. A., J. Chem. Soc 1935, 77.
- 14. Coulson, E. A., J. Chem Soc 1937, 1298.
- 15. Fieser, L. F., and J. Cason, J. Am Chem. Soc. 62, 1293 1940.
- Fieser, L. F., and W. H. Daudt, J. Am. Chem. Soc. 63, 782 1941.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 59, 1028 1937.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 60, 940 1938.
- Fieser, L. F., and W. S. Johnson, J. Am. Chem. Soc. 61, 168 1939.
- Fieser, L. F., and W. S. Johnson, J. Am. Chem. Soc. 61, 1647 1939.
- Fieser, L. F., and L. M. Joshel, J. Am. Chem. Soc. 61, 2958 1938.
- France, H., S. H. Tucker, and J. Forrest, J. Chem. Soc. 1945, 7.
- 23. Godchot, M., Ann chim. phys. [8] 12, 468 1907.
- 24. Godchot, M., Bull. soc. chim. [4] 1, 121 1907.
- 25. Goldschmiedt, G., Sitzber. Akad. Wiss.

- Wien, Math. naturw. Klasse 81, Abt. I, 415 1880.
- 26. Jüngermann, E., Ber. 38, 2868 1905.
- Lerer, M., Ann. combustibles liquides 10, 455 1935.
- Nyman, G. A., Ann. Acad. Sci. Fennicae A41, No. 5; C.A. 30, 2958 1936; Chem. Zentr. 1936, I, 2348.
- Pestemer, M., and F. Manchen, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 145, Abt. IIb, 312 1936.
- Risi, J., and D. Gauvin, Can. J. Research 13B, 228 1935.
- Ruzicka, L., and K. Hofmann, Helv. Chim. Acta 20, 1155 1937.
- Ruzicka, L., and K. Hofmann, Helv. Chim. Acta 22, 126 1939.
- 33. Salzer, W., Z. physiol. Chem. 274, 39 1942.
- 34. Scholl, R., and C. Scer, Ber. 44, 1671 1911.
- 35. Schroeter, G., Ber. 57, 2003 1924.
- 36. Shear, M. J., Am. J. Cancer 28, 334 1936.
- Stobbe, II., and F. Zschoch, Ber. 60, 457 1927.
- von Braun, J., O. Bayer, and L. F. Fieser, Ann. 459, 287 1927.
- von Braun, J., and G. Manz, Ber. 70, 1603 1937.
- Wieland, H., and E. Dane, Z. physiol. Chem. 219, 240 1933.
- 41. Wieland, II., and O. Probst, Ann. 530, 274 1937.
- 42. Willemart, A., Bull. soc. chim. [5] 6, 201 1939.
- 43. Willemart, A., Compt. rend. 207, 536 1938.
- Zinke, A., and O. Benndorf, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 143, Abt. IIb, 1 1934.
- Zinke, A, and N. Schniderschitsch, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 138, Abt. IIb, 28 1929.

## VIII. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA

## $C_n H_{2n-22}$

- 1. Naphthalene with Phenylalkenyl Substitutions
- 2. Anthracene and Phenanthrene Derivatives of Empirical Formula  $C_nH_{2n-23}$
- 3. Cyclenoanthracenes, Cyclenophenanthrenes and Their Alkyl Derivatives
- 4. Benzofluorenes and Their Alkyl Derivatives
- 5. Pyrene and Its Alkyl Derivatives
- 6. Miscellaneous Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>3n-21</sub>

# 1. NAPHTHALENE WITH PHENYLALKENYL SUBSTITUTIONS, $C_nH_{2n-32}$

## C18H14

## 1-Phenethenylnaphthalene

M. P., °C 72.5-73.5<sup>2</sup>

## 2-Phenethenylnaphthalene (a)

M. P., °C 205–207<sup>12</sup>

(a) The above formula was given for this compound, but the name given in the literature was "Phenyl-α-naphthylethylene."

## 1-Phenyl-1-(1'-naphthyl)-ethene

M. P., °C 60<sup>1. 3</sup> 59.5-60<sup>11</sup>

B. P., °C @ 760mm

350-355<sup>1</sup> 221-224

15<sup>10</sup> 9<sup>11</sup>

195-196

## 1-Phenyl-1-(2'-naphthyl)-ethene

$$\bigcirc c = c$$

M. P., °C 523

B. P., °C @ 760mm

220 24\*

C19H16

1-Phenyl-2-(1'-naphthyl)-propene-1

M. P., °C 1394

# x-Phenyl-x-(1'-naphthyl)-propene-x (a)

M. P., °C 55–616

B. P., °C @ 760mm 232–236

22-236

(a) The structure of this compound was not clearly defined in the literature.

## 1-o-Tolyl-1-(2'-naphthyl)-ethene

M. P., °C 66–66.5<sup>5</sup>

## x-p-Tolyl-x-(1'-naphthyl)-ethene (a)

B. P., °C @ 760mm

224-226

20°

1.0693

21.5°°

(a) The structure of this compound was not clearly defined in the literature.

#### $C_{21}H_{20}$

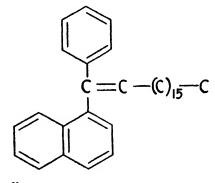
2-Methyl-3-phenyl-3-(x'-naphthyl)butene-1 (a)

M. P., °C 858

(a) The structure of this compound was not clearly defined in the literature.

## C34H46

1-Phenyl-1-(1'-naphthyl)-octadecene-1



 $D_{f 4}^{20}$ 

 $0.948 (a)^7$ 

 $n_{\scriptscriptstyle D}^{20}$ 

1.55587

(a) This constant is an extrapolated value.

References on Naphthalene with Phenylalkenyl Substitutions

- 1. Acree, S. F., Ber. 37, 2753 1904.
- 2. Balla, E., Compt. rend. 198, 947 1934.
- 3. Bergmann, E., and A. Bondi, Ber. 66, 286 1933.
- Cook, J. W., and R. A. E. Galley, J. Chem. Soc. 1931, 2012.

- Fieser, L. F., and M. S. Newman, J. Am. Chem. Soc. 58, 2376 1936.
- 6. Luce, E., Compt. rend. 180, 145 1925.
- Mikeska, L. A., Ind. Eng. Chem. 28, 970 1936.
- 8. Ramart, P., Compt. rend. 179, 634 1924.
- 9. Shurakovskii, E., J. Russ. Phys. Chem.
- Soc. 41, 1687 1909; C.A. 5, 1097 1911; Chem. Zentr. 1910, I, 1144.
- Spilker, A., and W. Schade, Ber. 65, 1686 1932.
- 11. Stoermer, R., and M Simon, Ber. 37, 4163 1904.
- 12. Thiele, J., Ber. 32, 1296 1899.

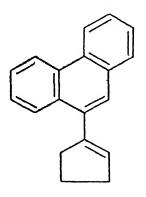
# 2. ANTHRACENE AND PHENANTHRENE DERIVATIVES OF EMPIRICAL FORMULA C<sub>n</sub>H<sub>2n-22</sub>

#### C<sub>18</sub>H<sub>14</sub>

1-(2'-Phenanthryl)-butadiene-1,3

## G 9H16

## 9-(Cyclopenten-1'-yl)-phenanthrene

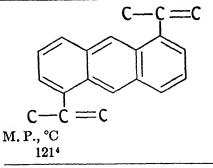


B. P., °C @ 760mm 185

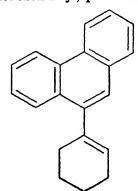
 $0.85^{1}$ 

### C20H18

## 1,5-Diisopropenylanthracene



9-(Cyclohexen-1'-yl)-phenanthrene



M. P., °C 132<sup>1</sup>

## C21H20

1-(Cyclopenten-1'-yl)-2-(9''-phenanthryl)-ethane

M. P., °C 105.5–107<sup>2</sup>

## $C_{22}H_{22}$

1-(2'-Methylcyclopenten-1'-yl)-2-(9''-phenanthryl)-ethane

M. P., °C 73–75<sup>2</sup> 1-(Cyclohexen-1'-yl)-2-(9''-phenanthryl)-ethane

B. P., °C @ 760mm 205–206

 $0.4^{5}$ 

 $C_{26}H_{30}$ 

1,4-Dicyclohexylanthracene

M. P., °C 160–161³ References on Anthracene and Phenanthrene Derivatives of Empirical Formula  $C_nH_{2n-22}$ 

1. Bergmann, E., and F. Bergmann, J.

Am. Chem. Soc. 59, 1443 1937.
 Bergmann, E., and O. Blum-Bergmann, J.
 Am. Chem. Soc. 58, 1678 1936.

Buu-Hol and P. Cagniant, Compt. rend.
 216, 381 1943; C.A. 38, 2334 1944.

4. Coulson, E. A., J. Chem. Soc. 1930, 1931.

Hewett, C. L., Gazz. chim. ital. 67, 728
 1937.

 Kögl, F., H. Erxleben, and L. Jänecke, Ann. 482, 105 1930.

# 3. CYCLENOANTHRACENES, CYCLENOPHENANTHRENES, AND THEIR ALKYL DERIVATIVES, $C_nH_{2n-22}$

## C18H14

2,3-(Cyclohexen-5'-o)-phenanthrene

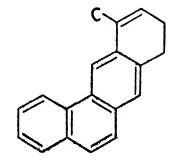
M. P., °C 112-113.5<sup>1</sup>

## C19H16

1,2-(3'-Methylcyclohexen-3'-0)anthracene

M. P., °C 74–754

2,3-(3'-Methylcyclohexen-3'-o)phenanthrene



M. P., °C 80-80.6°

2,3-(6'-Methylcyclohexen-5'-o)phenanthrene

M. P., °C 118-118.5³

## C20H18

2,3-(6'-Ethylcyclohexen-5'-o)phenanthrene

 $C_{22}H_{22}$ 

2,3-(6'-n-Butylcyclohexene-5'-o)phenanthrene

C23H24

2,3-(6'-n-Pentylcyclohexen-5'-0)phenanthrene

М. Р., °С 59-60⁵

## C24H26

## 2,3-(6'-n-Hexylcyclohexen-5'-0)phenanthrene

M. P., °C 47-48<sup>5</sup>

## C25H28

## 2,3-(6'-n-Heptylcyclohexen-5'-o)phenenthrene

## References on Cyclenoanthracenes, Cyclenophenanthrenes, and Their Alkyl Derivatives

- Bachmann, W. E., J. Org. Chem. 3, 434 1938-39.
- Bachmann, W. E., and J. M. Chemerda, J. Am. Chem. Soc. 61, 2358 1939.
- Bachmann, W. E., and A. L. Wilds, J. Am. Chem. Soc. 60, 624 1938.
- Cook, J. W., and A. M. Robinson, J. Chem. Soc. 1938, 505.
- Cook, J. W., and A. M. Robinson, J. Chem. Soc. 1940, 303.
- Fieser, L. F., and W. S. Johnson, J. Am. Chem. Soc. 61, 1647 1939.

# 4. BENZOFLUORENES AND THEIR ALKYL DERIVATIVES, C<sub>n</sub>H<sub>2n-22</sub>

## $C_{16}H_{10}$

## Benzo-[jk]-fluorene

(Fluoranthene) (Idryl)

M. P., °C

110

110.0-110.79

11015, 16, 27, 28

109-11031

10912, 13, 20, 30

108-10925

B. P., °C @ 760mm

39310

382-383 250-251

60<sup>14</sup> 60<sup>30</sup>

74922

250 217

3014

 $D_4^{20}$ 

1.0996 1.1045

18.7° 27 15.2° 27

1.236 (a) (solid)21

 $n_{\rm D}^{20}$ 

1.62768

 $n_{\mathrm{H}\alpha}^{18.727}$ 

1.63314

 $n_{\mathrm{H}\alpha}^{15.2}$  27

1.66107 1.66772 n<sub>Hβ</sub>

1.63693

n<sub>Hβ</sub> n<sup>18.7 27</sup>

1.64269

n<sub>He</sub> 27

(a) The temperature of this determination was not given.

## $C_{17}H_{12}$

## 1,2-Benzofluorene

(Chrysofluorene)

м. Р., °С

186.5

 $189 - 190^{23}$ 

188<sup>2</sup>, 17

187-1881

186-18718

18619

183-1846

182-1837

1828

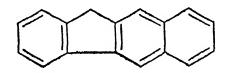
B. P., °C @ 760mm

 $413^{17}$ 

398-400 75828

## 2,3-Benzofluorene

(Isonaphthofluorene)



M. P., ℃

20924

208-20928

2083, 26

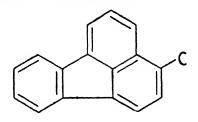
B. P., °C @ 760mm 401-402

75828

### 3,4-Benzofluorene

M. P., °C 124-125<sup>5</sup>

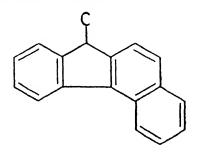
## Benzo-[lm]-2-methylfluorene



M. P., °C

## $C_{18}H_{14}$

## 3,4-Benzo-9-methylfluorene



M. P., °C 80.8–82.0<sup>11</sup>

## C19H16

## 3,4-Benzo-9,9-dimethylfluorene

M. P., °C 148-148.54

## References on Benzofluorenes and Their Alkyl Derivatives

- Bamberger, E., and J. Kranzfeld, Ber. 18, 1931 1885.
- Bamberger and Kranzfeld through C. Graebe, Ann. 335, 135 1904.
- Barnett, E. de B., N. F. Goodway, and J. W. Watson, Ber. 66, 1876 1933.
- Bradsher, C. K, and A. S. Burhans, J. Am. Chem. Soc. 62, 3140 1940.
- Cook, J. W., A. Dansi, C. L. Hewett,
   J. Iball, W. V. Mayneord, and E.
   Roe, J. Chem. Soc. 1935, 1319.
- Cook, J. W., and C. L. Hewett, J. Chem. Soc. 1934, 365.
- Cook, J. W., C. L. Hewett, W. V. Mayneord, and E. Roe, J. Chem. Soc. 1934, 1727.
- Cook, J. W., and R. W. G. Preston, J. Chem. Soc. 1944, 553.
- Davis, W. W., M. E. Krahl, and G. H. A. Clowes, J. Am. Chem. Soc. 62, 3080 1940.
- 10. Decker, H., Ber. 67, 1636 1934.
- Fieser, L. F., and L. M. Joshel, J. Am. Chem. Soc. 62, 957 1940.
- Fittig, R., and F. Gebhard, Ann. 193, 142 1878.
- Fittig, R., and F. Gebhard, Ber. 10, 2141 1877.
- Fittig, R., and H. Liepmann, Ann. 200,
   1 1880,
- Goldschmiedt, G., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 81, Abt. I, 415 1880.
- Goldschmiedt, G., and M. von Schmidt, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 83, Abt. IIb, 7 1881.
- 17. Graebe, C., Ann. 335, 122 1904.

25. von Braun, J., and E. Rath, Ber. 61, 956 1928.

Weger, M., Z. angew. Chem. 22, 338
 1909.

 Weitzenböck, R., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 121, Abt. IIb, 1227 1912.  Winterstein, A., and K. Schön, Z. physiol. Chem. 230, 146 1934.

Winterstein, A., K. Schön, and H. Vetter, Z. physiol. Chem. 230, 158
 1934.

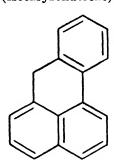
 Zanetti, J. E., and G. Egloff, Ind. Eng. Chem. 9, 474 1917.

# 6. MISCELLANEOUS POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{\bullet n-22}$

## C17H12

## 2,3-Benzophenalan

(Benzanthrene) (Isochrysofluorene)



M. P., °C 84<sup>6, 14</sup> 81–82<sup>18</sup>

# 1,x-Naphthylenephenylenemethane (a)

M. P., °C 7638

(a) The naphthylene bond may be in either the 2- or 8-position.

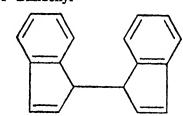
## C18H14

## 1-Phenyl-3-indenylidenepropene-1

$$\bigcirc -c = c - c = \bigcirc$$

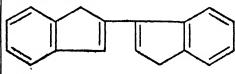
## M. P., °C 190⁵0

## 1,1'-Biindenyl



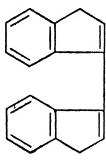
M. P., °C 99–100<sup>23</sup>, \*\* 98<sup>49</sup>

## 2,3'-Biindenyl



M. P., °C 57-58<sup>48</sup> 56<sup>46</sup>

## 3,3'-Biindenyl

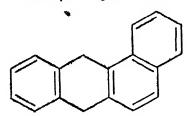


M. P., °C 130.5–131.5<sup>19</sup> 130–131.5<sup>49</sup> 129.5–130.5<sup>49</sup>

## 3,4,7,8-Dibenzo-2,6-dimethylbicyclo-[3,3,0]-octadiene-1,5

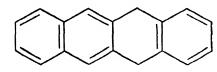
M. P., °C 212<sup>11</sup>

## 1,2-Benzo-9,10-dihydroanthracene



M. P., °C 112-112.5<sup>1</sup>

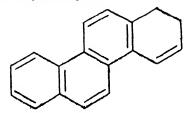
## 5,12-Dihydronaphthacene



M. P., °C 212-213<sup>24</sup> 212<sup>17</sup> 207<sup>28</sup> 206-207<sup>37</sup>

B. P., °C @ 760mm ca. 400<sup>37</sup>

## 1,2-Dihydrochrysene



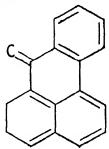
M. P., °C 182.5-184.54

## x,x-Dihydrochrysene (a)

M. P., °C 167-169<sup>44</sup>

(a) The structure of this compound was not clearly defined in the literature.

## 2,3-Benzo-1-methylene-8,9-dihydrophenalan



M. P., °C 155-156<sup>12, 13</sup>

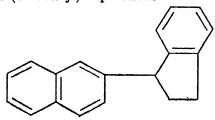
## Naphtho-[1,8-ab]-benzo-[e]-cycloheptane

(Pleiadene)

M. P., °C 116.5<sup>45</sup>

C19H16

2-(1'-Indanyl)-naphthalene

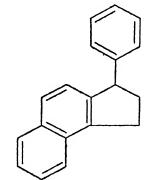


M. P., °C 52-53<sup>21</sup> 47<sup>52</sup>

B. P., °C @ 760mm 229-230

 $13^{52}$ 

1-Phenyl-4,5-benzoindane

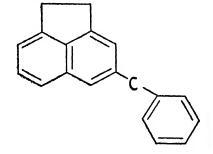


M. P., °C 79<sup>52</sup>

B. P., °C @ 760mm 226-229

 $13^{52}$ 

4-Benzylacenaphthene

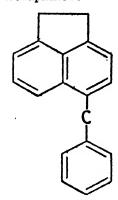


M. P., °C 45–46<sup>26</sup>

B. P., °C @ 760mm 260-265

 $20^{26}$ 

5-Benzylacenaphthene



M. P., °C

 $112-113^{25}$ 

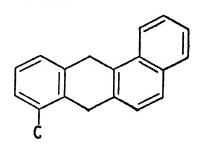
 $111-112^{42} \\ 110-111^{26}, \, {}^{27}$ 

B. P., °C @ 760mm

ca. 340-315<sup>25</sup>
210-215

 $13^{27}$ 

1,2-Benzo-5-methyl-9,10-dihydroanthracene



M. P., °C 128-129.5³

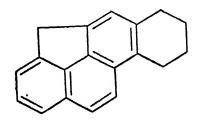
1,2-Benzo-10-methyl-9,10-dihydroanthracene

M. P., °C 94.4-94.9<sup>31</sup>

## Dicyclopentano-[a,fg]-anthracene

M. P., °C 174-17630

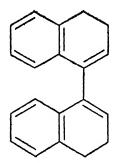
## Cyclopentano-[mno]-1,2,3,4-tetrahydrochrysene



M. P., °C 129-129.4<sup>29</sup> 127.5-128.5<sup>29</sup>

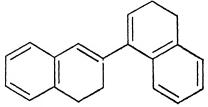
## C20H18

3,4,3',4'-Tetrahydro-1,1'-binaphthyl



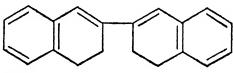
M. P., °C 141<sup>54</sup> 140<sup>8</sup> 139–140<sup>47</sup>

## 3,4,3',4'-Tetrahydro-1,2'-binaphthyl



M. P., °C 87<sup>53</sup>

## 3,4,3'4'-Tetrahydro-2,2'-binaphthyl



M. P., °C 156<sup>51</sup> 153<sup>53</sup>

## x,x-Dihydro-x-benzylfluorene (a)

M. P., °C 7141

(a) The structure of this compound was not clearly defined in the literature.

## 3,4,7,8-Dibenzo-2,6-diethylbicyclo-[3,3,0]-octadiene-1,5

M. P., ℃ 154<sup>11</sup>

3,4,7,8-Dibenzo-2,6-diethylidenebicyclo-[3,3,0]-octane

$$c-c$$

M. P., °C 199.5<sup>11</sup>

1,2-Benzo-5,8-dimethyl-3,4-dihydroanthracene

M. P., °C 82.2-82.8<sup>32</sup>

2,3-Benzo-1,5-dimethyl-1,4-dihydrophenanthrene

M. P., °C 108–109<sup>38</sup>

264

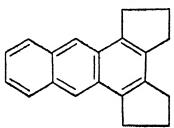
2,8-Dimethyl-5,12-dihydronaphthacene

M. P., °C 217<sup>22</sup>

5,6-Dimethyl-5,6-dihydrochrysene

M. P., °C 104-104.5<sup>40</sup>

1,2,3,4-Dicyclopentanoanthracene



M. P., ℃ 1468

Cyclohexano-[a]-cyclopentano-[de]anthracene

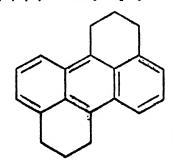
M. P., °C 106-107<sup>84</sup>

2a,3,4,5-Tetrahydrocholanthrene

M. P., °C 107 (a)<sup>2</sup> 101-101.5 (a)<sup>2</sup>

(a) These constants were determined on different crystalline forms.

1,2,3,7,8,9-Hexahydroperylene



M. P., ℃

18956

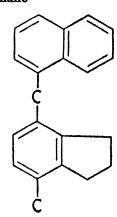
182-18555

 $183 - 184^{57}$ 

18343

 $C_{21}H_{20}$ 

(1'-Naphthyl)-[4'-(7'-methylindanyl)]methane



B. P., °C @ 760mm 221-226

433

3-Methyl-2a,3,4,5-tetrahydrocholanthrene

M. P., °C 97–99°

 $C_{22}H_{22}$ 

7,7'-Dimethyl-3,4,3',4'-tetrahydro-1,1'-binaphthyl 266

M. P., °C

162-16315, 16

x<sub>8</sub>-Octahydro-1,2,3,4-dibenzoanthracene (a)

M. P., °C 1297

- (a) The structure of this compound was not clearly defined in the literature.
- x<sub>8</sub>-Octahydro-1,2,5,6-dibenzoanthracene (a)

M. P., °C 188–190<sup>19, 20</sup>

- (a) The structure of this compound was not clearly defined in the literature.
- **3,4,7,8-Dibenzo-2,6-di-***n*-propyl-bicyclo-[3,3,0]-octadiene-1,5

$$\begin{array}{c} c-c-c \\ \hline \\ c-c-c \end{array}$$

M. P., °C 135-136<sup>10</sup>

3,4,7,8-Dibenzo-2,6-diisopropylbicyclo-[3,3,0]-octadiene-1,5

3,4,7,8-Dibenzo-2,6-di-n-propylidenebicyclo-[3,3,0]-octane

$$\begin{array}{c} c-c-c \\ \hline \\ c-c-c \end{array}$$

M. P., °C 157-158<sup>10</sup>

3,4,7,8-Dibenzo-2,6-diisopropylidenebicyclo-[3,3,0]-octane

$$\begin{array}{c} c-c-c \\ \hline \\ c-c-c \end{array}$$

M. P., °C 18910

C24H26

1,2,3,4-Tetramethyl-5,6,7,8-dicyclopentanoanthracene

M. P., °C 296–297<sup>5</sup>

## 1,2,2a,3,4,4a,5,6,6a,7,8,8a,9,10-Tetradecahydrocoronene

M. P., °C 277-278<sup>36</sup>

## C26H30

## 1,2,3,4-Tetramethyl-5,6,7,8-dicyclohexanoanthracene

M. P., °C 272–273⁵

References on Miscellaneous Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-22}$ 

- 1. Bachmann, W. E., J. Org. Chem. 1, 347 1936-37.
- 2. Bachmann, W. E., J. Org. Chem. 3, 434 1938-39.
- Bachmann, W. E., and L. H. Pence, J. Am. Chem. Soc. 59, 2339 1937.
- Bachmann, W. E., and W. S. Struve,
   J. Org. Chem. 4, 456 1939.
- Backer, H. J., and L. H. H. Huisman, Rec. trav. chim. 60, 557 1941; C.A. 37, 5982 1943.
- Bally, O., and R. Scholl, Ber. 44, 1656
   1911.

 Barnett, E. de B., N. F. Goodway, and C. A. Lawrence, J. Chem. Soc. 1935, 1684

267

- Barnett, E. de B., and C. A. Lawrence,
   J. Chem. Soc. 1935, 1104.
- Bergmann, E., and O Blum-Bergmann,
   J. Am. Chem. Soc. 59, 1573 1937.
- Brand, K., and T. Sasaki, Ber. 58, 2546
   1925.
- Brand, K., and F. Schlager, Ber. 56, 2541 1923.
- Charrier, G., and E. Ghigi, Atti accad. Lincei [6] 16, 257 1932.
- Charrier, G., and E. Ghigi, Gazz. chim. ital. 62, 928 1932.
- Chatterjee, N. N., and H. B. Roy, J. Indian Chem. Soc. 20, 329 1943.
- Chu, E. J. H, and Z.-I. Shen, J. Chinese Chem. Soc. 10, 113 1943; Brit. Chem. Abstracts AII, 1944, 190.
- Chu, E. J. II., and Z.-I. Shen, J. Chinese Chem. Soc 10, 119 1943; C.A. 38, 2951 1943.
- 17. Clar, E., Ber. 75, 1271 1942.
- Clar, E., and F. Furnari, Ber. 65, 1420
   1932.
- 19. Cook, J. W., J. Chem. Soc. 1931, 487.
- 20. Cook, J. W., J. Chem. Soc. 1931, 489.
- Cook, J. W., and G. A. D. Haslewood,
   J. Chem. Soc. 1935, 770.
- Coulson, E. Λ., J. Chem. Soc. 1934, 1406.
- 23. Courtot, C., Ann. chim. [9] 4, 68 1915.
- Dufraisse, C., and R. Horclois, Bull. soc. chim. [5] 3, 1880 1936.
- Dziewoński, C., and E. Dotta, Bull. soc. chim. [3] 31, 373 1904.
- Dziewoński, K., and K. Leonhard, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1928A, 99
- Dziewoński, K., and M. Rychlik, Ber.
   58, 2239 1925.
- Fieser, L. F., J. Am. Chem. Soc. 53, 2329 1931.
- 29. Fieser, L. F., and J. Cason, J. Am. Chem. Soc. **62**, 1293 **1940**.
- Fieser, L. F., and E. B. Hershberg, J. Am Chem. Soc. 59, 394 1937.
- 31. Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 59, 2502 1937.
- 32. Fieser, L. F., and W. S. Johnson, J. Am. Chem. Soc. **61**, 1647 **1939**.
- Ficser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 58, 2482 1936.

- Ficser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 59, 883 1937.
- Fieser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 60, 170 1938.
- Fromherz, H., L. Thaler, and G. Wolf
   Z. Elektrochem. 49, 387 1943.
- 37. Gabriel, S., and E. Leupold. Ber 31 1272 1898.
- 38. Graebe, C., Ber. 27, 952 1894.
- Grignard, V., and C. Courtot, Compt. rend. 154, 362 1912.
- Hewett, C. L., J. Chem. Soc. 1940. 293.
- Kliegl, A., F. Weng, and G. Wiest, Ber.
   1262 1930.
- 42. Lorriman, F. R., J. Am. Chem. Soc. 47, 211 1925.
- 43. Mason, F. A., Ind. Chemist 5, 111 1929.
- 44. Mohler, H., and J. Sorge, Helv. Chim. Acta 22, 229 1939.
- Rieche, A., H. Sauthoff, and O. Müller, Ber. 65, 1371 1932.
- Risi, J., and D. Gauvin, Can. J. Research 13B, 228 1935.

- 47. Schroeter, G., Ber. 58, 713 1925.
- 48. Stobbe, H., and E. Färber, Ber. 57, 1838 1924.
- Straus, F., R. Kühnel, and R. Haensel, Ber. 66, 1847 1933.
- 50. Thiele, J., Ber. 33, 3395 1900.
- von Braun, J., and G. Kirschbaum, Ber.
   54, 597 1921.
- von Braun, J., G. Manz, and E. Reinsch, Ann. 468, 277 1929.
- 53. Weidlich, H. A., Ber. 71, 1201 1938.
- 54. Weidlich, H. A., Ber. 71, 1203 1938.
- Zinke, A., and O. Benndorf, Sitzber.
   Akad. Wiss. Wien, Math. naturw.
   Klasse 143, Abt. IIb, 1 1934.
- Zinke, A., and N. Schniderschitsch, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 138, Abt. IIb, 28 1929.
- Zinke, A., and E. Unterkreuter, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 128, Abt. IIb, 153 1919.

# IX. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-24}$

- 1. Fluorene Derivatives of Empirical Formula C<sub>n</sub>H<sub>2n-21</sub>
- 2. Dihydroanthracenes and Dihydrophenanthrenes with One Phenyl Substitution
- 3. 1,2-Benzoanthracene and Its Alkyl Derivatives
- 4. 3,4-Benzophenanthrene and Its Alkyl Derivatives
- 5. Naphthacene and Its Alkyl Derivatives
- 6. Chrysene and Its Alkyl Derivatives
- 7. Triphenylene and Its Alkyl Derivatives
- 8. Miscellaneous Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-24</sub>

## 1. FLUORENE DERIVATIVES OF EMPIRICAL FORMULA C,H2n-24

## $C_{18}H_{12}$

# 9-(Cyclopentadien-2',4'-ylidene)fluorene

M. P., °C 133–135<sup>7</sup>

## C19H14

## 9-Phenylfluorene

146

 $148.5^{17}$ 

147-14835

146--14837

14727, 37, 38

 $146^{1}$ 

145-14611, 12, 81

145.518, 19, 20

145 (a)

144<sup>4</sup>, 15

 $D_{\blacktriangle}^{20}$ 

1.232

0° 41

(a) The melting point 145 is found in references 23, 24, 25, 28, 30, 33, 40.

# $C_{20}H_{16} \\$ 3-Benzylfluorene

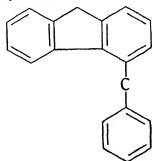
M. P., °C 106.5<sup>10</sup>

1068

104-10613

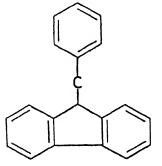
 $102^{14}$ 

## 4-Benzylfluorene



M. P., °C 77<sup>16</sup>

## 9-Benzylfluorene



# M. P., °C 133 134-135<sup>39</sup> 134<sup>34</sup> 133<sup>5</sup> 131-132<sup>2</sup> 130-131<sup>32</sup>

## 9-o-Tolylfluorene

M. P., °C 133\*6

## 9-p-Tolylfluorene

(a) The structure of this compound was not clearly defined in the literature.

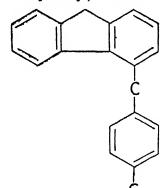
## 9-Methyl-9-phenylfluorene

This series continued on next page

 $C_{21}H_{18}$ 

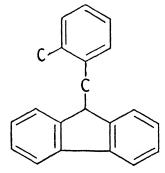
## 2-Phenethylfluorene

## 4-(4'-Methylbenzyl)-fluorene



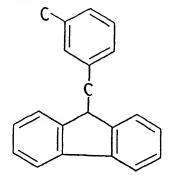
# M. P., °C 72<sup>26</sup>

## 9-(2'-Methylbenzyl)-fluorene



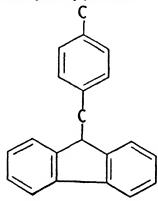
M. P., °C 71–72<sup>29</sup>

## 9-(3'-Methylbenzyl)-fluorene



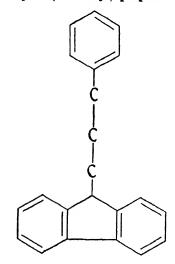
M. P., °C 111-112°

## 9-(4'-Methylbenzyl)-fluorene



#### $C_{22}H_{20}$

## 1-Phenyl-3-(9'-fluoryl)-propane



M. P., °C 75–76<sup>2</sup> 71<sup>22</sup>

## References on Fluorene Derivatives of Empirical Formula C<sub>n</sub>H<sub>2n-24</sub>

- Bachmann, W. E., J. Am. Chem. Soc. 52, 3287 1930.
- 2. Bergmann, E., Ber. 63, 1617 1930.
- Bergmann, E., and A. Bondi, Ber. 64, 1455 1931.
- 4. Boeseken, J., Rec. trav. chim. 22, 301 1903.
- Brown, W. G., and B. A. Bluestein, J. Am. Chem. Soc. 65, 1082 1943.
- Cook, J. W., and R. W. G. Preston, J. Chem. Soc. 1944, 553.
- 7. Courtot, C., Ann. chim. [9] 4, 168 1915.
- 8. Dansi, A., and A. Sempronj, Gazz. chim. ital. 64, 218 1934.
- 9. De Fazi, R., Gazz. chim. ital. 51, I, 328 1921.
- Dziewoński, K., and Z. Reicher, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1931A, 643.
- Fischer, E., and O. Fischer, Ann. 194, 242 1878.
- Fischer, E., and O. Fischer, Ber. 11, 612
   1878.

- Fortner, M., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 111, Abt. IIb, 741 1902.
- Goldschmiedt, G., Sitzber. Akad. Wiss. Wien, Math naturw. Klasse 84, Abt. IIb, 305 1881.
- Gomberg, M., and L. H. Cone, Ber. 37, 3538 1904.
- Gotz, R., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 110, Abt. Hb, 1037 1901.
- 17. Hanriot, M., and O. Saint-Pierre, Bull. soc. chim. [3] 1, 774 1889.
- 18. Hemilian, W., Ber. 11, 202 1878.
- 19. Hemilian, W., Ber. 11 837 1878.
- 20. Kliegl, A., Ber. 38, 284 1905.
- 21. Kovache, A., Ann. chim. [9] 10, 184 1918.
- Kuhn, R., and A Winterstein, Helv. Chim. Acta 11, 123 1928.
- McVicker, W. H., J. K. Marsh, and A. W. Stewart, J. Chem. Soc. 127, 999 1925.
- Meyer, R., and E. Saul, Ber. 25, 3586
   1892.
- 25. Nef, J. U., Ann. 309, 126 1899.
- Pick, H., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 113, Abt. Hb, 649 1904.
- Pinck, L. A., J. Am. Chem. Soc. 55, 1711 1933.
- Ramart-Lucas and J. Hoch, Bull. soc. chim. [5] 2, 1376 1935.
- Sieglitz, A., and H. Jassoy, Ber. 54, 2133
   1921.
- 30. Staudinger, H., Ber. 39, 3060 1906.
- Steiglitz, J., and R. L. Brown, J. Am. Chem. Soc. 44, 1270 1922.
- Thiele, J., and F. Heule, Ann. 347, 290 1906.
- Ullmann, F., and R. von Wurstemberger, Ber. 37, 73 1904.
- Vansheidt, A., and B. Moldavski, Ber. 64, 917 1931.
- Vorlander, D., and A. Pritzsche, Ber. 46, 1793 1913.
- Weiss, R., and E. Knapp, Monatsh. 61, 61 1932.
- 37. Werner, A., and A. Grob, Ber. 37, 2887 1904.
- 38. Wieland, H., K. Heymann, T. Tsatsas, D. Juchum, G. Varvoglis, G. Labriola,

 $C_{20}H_{16}$ 

274

- O. Dobbelstein, and H. S. Boyd-Barrett, Ann. **514**, 145 **1934**.
- 39. Wislicenus, W., and W. Mocker, Ber. 46, 2772 1913.
- Zelinskii, N. D., and M. V. Gaverdovskaya, Ber. 61, 1049 1928.
   Ziegler, K. and F. Ditzel, Ann. 473, 194
- 41. Ziegler, K., and F. Ditzel, Ann. 473, 194 1929.

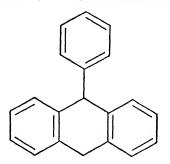
# 2. DIHYDROANTHRACENES AND DIHYDROPHENANTHRENES WITH ONE PHENYL SUBSTITUTION, $C_nH_{2n-24}$

## $C_{20}H_{16}$

## 2-Phenyl-9, 10-dihydroanthracene

M. P., °C 93-96<sup>16</sup>

## 9-Phenyl-9, 10-dihydroanthracene



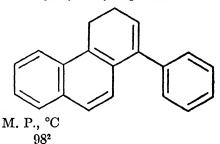
M. P., °C 90-91<sup>11</sup> 90<sup>9</sup> 87<sup>15</sup>

## x-Phenyl-x,x-dihydroanthracene (a)

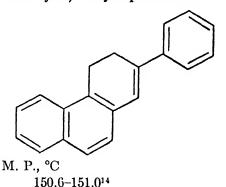
M. P., °C 123<sup>15</sup> 120–120.5<sup>17</sup>

(a) The structure of this compound was not clearly defined in the literature.

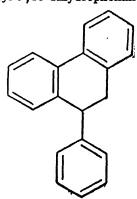
## 1-Phenyl-3,4-dihydrophenanthrene



## 2-Phenyl-3, 4-dihydrophenanthrene



9-Phenyl-9, 10-dihydrophenanthrene



M. P., °C 121.5<sup>10</sup> 84<sup>4</sup>

#### $C_{21}H_{18}$

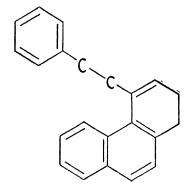
9-Benzyl-9, 10-dihydroanthracene

M. P., °C 119-120<sup>6, 7</sup> 110-111<sup>1</sup>

## $C_{22}H_{20}$

9,9-Dimethyl-10-phenyl-9,10-dihydroanthracene

M. P., °C 145–146<sup>5</sup> 4-Phenethyl-1,2-dihydrophenanthrene



B. P., °C @ 760mm 180

 $n_{\rm p}^{20}$ 

1.6510 26° ³

1-Phenethyl-3, 4-dihydrophenanthrene

M. P., °C 62-63\*

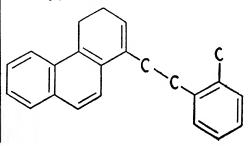
B. P., °C @ 760mm 185-187

 $0.5 - 1^8$ 

 $0.01^{2}$ 

 $C_{23}H_{22}$ 

1-o-Tolyl-2-[1'-(3',4'-dihydrophenan-thryl)]-ethane



M. P., °C 57–58\*

B. P., °C @ 760mm 190-195

0.5 - 18

1-p-Toly1-2-[1'-(3',4'-dihydrophenanthryl)]-ethane

M. P., °C 79.5-818

B. P., °C @ 760mm 200-205

0.5 - 1.08

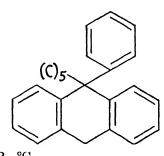
#### $C_{24}H_{24}$

9,9-Diethyl-10-phenyl-9,10-dihydroanthracene

M. P., °C 135-136<sup>13</sup>

#### $C_{25}H_{26}$

9-Pentyl-9-phenyl-9, 10-dihydroanthracene



M. P., °C 85<sup>12</sup>

> References on Dihydroanthracenes and Dihydrophenanthrenes with One Phenyl Substitution

- 1. Bach, C., Ber. 23, 2527 1890.
- Bachmann, W. E., and A. L. Wilds, J. Am. Chem. Soc. 60, 624 1938.
- 3. Bergmann, E., J. Chem. Soc. 1938, 1291.
- Bergmann, E., and F. Bergmann, J. Am. Chem. Soc. 59, 1443 1937.
- Bradsher, C. K., and E. S. Smith, J. Am. Chem. Soc. 65, 1643 1943.
- 6. Cook, J. W., J. Chem. Soc. 1926, 1677.

 $C_{18}H_{12}$ 

- 7. Cook, J. W., J. Chem. Soc. 1926, 2160.
- Drake, N. L., and W. C. McVey, J. Org. Chem. 4, 464 1939.
- Dufraisse, C., L. Velluz, and Mme. L. Velluz, Bull. soc. chim. [5] 5, 1073 1938.
- Earl, J. C., and C. H. Wilson, J. Proc. Roy. Soc. N. S. Wales 65, 178 1931.
- 11. Haack, E., Ber. 62, 1771 1929.
- 12. Jungermann, E., Ber. 38, 2868 1905.

- Kehrmann, F., R. Monier, and M. Ramm, Ber. 56, 169 1923.
- 14. Newman, M. S., J. Org. Chem. 9, 518 1944.
- Schlenk, W., and E. Bergmann, Ann. 463, 1 1928.
- Scholl, R., and W. Neovius, Ber. 44, 1075 1911.
- von Baeyer, Λ., and A. Schillinger, Ann. 202, 51 1880.

#### 3. 1,2-BENZOANTHRACENE AND ITS ALKYL DERIVATIVES, C, H2n-14

#### $C_{18}H_{12}$

#### 1,2-Benzoanthracene

(Naphthanthracene)

M. P., °C

159.7

165-16717

 $161.4 - 161.8^{21}$ 

160.5-16124

 $160 - 160.5^{30}$ 

 $159.5 - 160.5^{28}$ 

16041

159-16012, 27, 30

 $158 - 160^{38}, 42$ 

 $158.5 - 159.5^{16}$ 

1599, 36, 39

158-1596, 7, 13, 22

158-158.5

156--15710

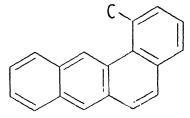
155-15723, 35

 $D_{\star}^{20}$ 

1.245 (solid)37

#### C19H14

#### 1,2-(3'-Methylbenzo)-anthracene



M. P., ℃

138.5-139.234

138.5 -139.021

135.5-136.54

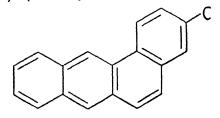
#### 1,2-(4'-Methylbenzo)-anthracene

M. P., °C

 $149 \cdot 150^{13}$ 

 $148 - 119^3$ 

#### 1.2-(5'-Methylbenzo)-anthracene



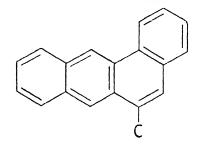
#### 1,2-(6'-Methylbenzo)-anthracene

M. P., °C 194-195<sup>19</sup>

#### 1,2-Benzo-3-methylanthracene

M. P., °C 155<sup>18</sup> 153-154<sup>11</sup>

#### 1,2-Benzo-4-methylanthracene



M. P., °C 124-126<sup>33</sup> 124.5-125.5<sup>14</sup> 124.1-124.6<sup>32</sup>

#### 1,2-Benzo-5-methylanthracene

M. P., °C 158 158–159.4<sup>31</sup>

 $158.5 - 159.1^{33}$ 

157.5-158.5<sup>14</sup> 156.5-157<sup>5</sup>

 $154 - 156^{23}$ 

 $D_4^{20}$ 

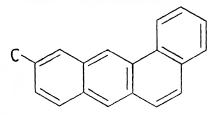
1.231 (solid)<sup>37</sup>

#### 1,2-Benzo-6-methylanthracene

M. P., °C 150.5-151.5<sup>13</sup> 149-151.5<sup>2</sup> 127 (a)<sup>22</sup>

(a) The structural formula indicates that this compound has a para bond in the 9,10-position.

#### 1,2-Benzo-7-methylanthracene

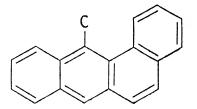


#### 1,2-Benzo-8-methylanthracene

# M. P., °C $145 \text{ (a)}^{22}$ $118-118.5 \text{ (b)}^{30}$ $117-118 \text{ (b)}^{31}$ $117-118^{15}$ $114-117 \text{ (c)}^{2}$

- (a) The structural formula indicates that this compound has a *para* bond in the 9,10-position.
- (b) This compound remelts at 113.5–114. The substance melting at 113.5-114 is considered a different modification in the literature.
- (c) This compound remelts at 112-113.

#### 1,2-Benzo-9-methylanthracene



M. P., °C 138.5 138-139<sup>19</sup> 138.4-138.8<sup>25, 40</sup> 138.0-138.8<sup>21</sup> 137.5-138.5<sup>8</sup>

#### 1,2-Benzo-10-methylanthracene

M. P., °C

140.4

140.5-141.5<sup>19</sup>

140-141<sup>29</sup>

140.2-140.8<sup>33</sup>

140.0-140.5<sup>21</sup>

139.5-140.5<sup>26</sup>

139-140<sup>3, 27</sup>

#### References on $C_{18}II_{12}$ through $C_{19}II_{14}$ Compounds

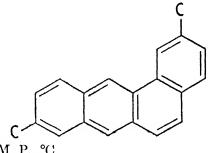
- Bachmann, W. E., J. Org. Chem. 1, 347 1936-37.
- Bachmann, W. E., and J. M. Chemerda,
   J. Org. Chem. 6, 36 1941.
- Bachmann, W. E., and G. D. Cortes, J. Am. Chem. Soc. 65, 1329 1943.
- Bachmann, W. E., and R. O. Edgerton,
   J. Am. Chem. Soc. 62, 2550 1940.
- Bachmann, W. E., and A. L. Wilds, J. Am. Chem. Soc. 60, 624 1938.
- Badger, G. M., and J. W. Cook, J. Chem. Soc. 1939, 802.
- Barnett, E. de B., and M. A. Matthews, Chem. News 130, 339 1925.
- Bradsher, C. K., J. Am. Chem. Soc. 62, 1077 1940.
- 9. Brass, K., and K. Fanta, Ber. 69, 1 1936.
- Burger, A., and E. Mosettig, J. Am. Chem. Soc. 59, 1302 1937.
- 11. Cook, J. W., J. Chem. Soc. 1930, 1087.
- 12. Cook, J. W., J. Chem. Soc. 1931, 2524.
- 13. Cook, J. W., J. Chem. Soc. 1932, 456.
- 14. Cook, J. W., J. Chem. Soc. 1933, 1592.
- Cook, J. W., and C. G. M. De Worms,
   J. Chem. Soc. 1939, 268.
- Cook, J. W., C. L. Hewett, and I. Hieger, J. Chem. Soc. 1933, 395.

- Cook, J. W., and C. A. Lawrence, J. Chem. Soc. 1937, 817.
- Cook, J. W., and A. M. Robinson, J. Chem. Soc. 1938, 505.
- Cook, J. W., A. M. Robinson, and F. Goulden, J. Chem. Soc. 1937, 393.
- 20. Coulson, E. A., J. Chem. Soc. 1935, 77.
- Davis, W. W., M. E. Krahl, and G. H. A. Clowes, J. Am. Chem. Soc. 62, 3080 1940.
- Dziewoński, K., and E. Ritt, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1927A, 181.
- 23. Fieser, L. F., and J. Cason, J. Am. Chem. Soc. 61, 1740 1939.
- Fieser, L. F., and H. J. Creech, J. Am. Chem. Soc. 61, 3502 1939.
- Fieser, L. F., M. Fieser, E. B. Hershberg, M.S. Newman, A.M. Seligman, and M. J. Shear, Am. J. Cancer 29, 260 1937.
- Fieser, L. F., and J. L. Hartwell, J. Am. Chem. Soc. 60, 2555 1938.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 59, 1028 1937.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 59, 2502 1937.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 61, 1272 1939.
- Fieser, L. F., and W. S. Johnson, J. Λm. Chem. Soc. 61, 168 1939.
- 31. Fieser, L. F., and W. S. Johnson, J. Am. Chem. Soc. **61**, 1617 **1939**.
- 32. Fieser, L. F., and R. N. Jones, J. Am. Chem. Soc. 60, 1940 1938.
- Fieser, L. F., and M. S. Newman, J. Am. Chem. Soc. 58, 2376 1936.
- Fieser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 60, 170 1938.
- Haworth, R. D., and C. R. Mavin, J. Chem. Soc. 1933, 1012.
- Hertel, E., and H. W. Bergk, Z. physik. Chem. 33B, 319 1936.
- Iball, J., Z. Krist. 99A, 230 1938; C. A.
   32, 8236 1938.
- Mayer, F., and T. Oppenheimer, Ber.
   51, 510 1918.
- McVicker, W. H., J. K. Marsh, and A. W. Stewart, J. Chem. Soc. 127, 999
- Newman, M. S., J. Am. Chem. Soc. 59, 1003 1937.

- 41. Sturrock, M. G., and T. Lawe, Can. J. Research 17B, 71 1939.
- 42. Weitzenbock, R., and H. Lieb, Sitzber. Akad. Wiss. Wien, Math. naturw.
  - Klasse **121**, Abt. IIb, 225 **1912**.

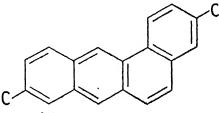
#### $C_{20}H_{16}$

1,2-(4'-Methylbenzo)-6-methylanthracene



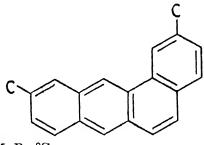
M. P., °C 1646

1,2-(5'-Methylbenzo)-6-methylanthracene



M. P., °C 186–187<sup>6</sup>

1,2-(4'-Methylbenzo)-7-methylanthracene



M. P., °C 236•

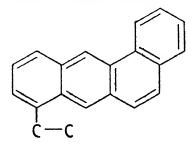
#### 1,2-(5'-Methylbenzo)-7-methylanthracene

M. P., °C 189–190<sup>6</sup>

#### 1,2-(3'-Methylbenzo)-10-methylanthracene

M. P., °C 124-125<sup>21</sup>

# 1,2-Benzo-5-ethylanthracene



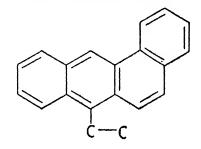
M. P., °C 120<sup>12</sup> 118–119<sup>2</sup>

#### 1,2-Benzo-8-ethylanthracene

M. P., °C 78-79 (a)<sup>17</sup>

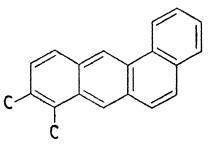
(a) This compound remelts at 82.5-83.

#### 1,2-Benzo-10-ethylanthracene



M P., °C 113.5-114<sup>16</sup> 112-113<sup>16</sup> 112.4-112.8<sup>12</sup>

# 1,2-Benzo-5,6-dimethylanthracene



M. P., °C 187–188<sup>3</sup>, •

282

1,2-Benzo-3,9-dimethylanthracene

M. P., °C 93-93.5<sup>22</sup>

#### 1,2-Benzo-5,7-dimethylanthracene

M. P., °C 124.5–125<sup>3</sup>

#### 1,2-Benzo-4,9-dimethylanthracene

M. P., °C 75.1-75.51°

# 1,2-Benzo-5,8-dimethylanthracene

M. P., °C 133.5–134.5³ 131.2–131.4 (a)<sup>17</sup>

(a) This compound remelts at 134.4-134.7.

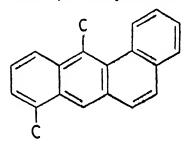
#### 1,2-Benzo-6,7-dimethylanthracene

M. P., °C 174<sup>6</sup> 173–174<sup>5</sup>

#### 1,2-Benzo-4,10-dimethylanthracene

M. P., °C 114-114.4<sup>19</sup>

#### 1,2-Benzo-5,9-dimethylanthracene



M. P., °C 135.0-135.5<sup>15</sup> 135-135.5<sup>26</sup>

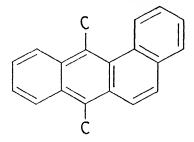
#### 1,2-Benzo-5,10-dimethylanthracene

M. P., °C 147-147.5<sup>20</sup>

#### 1,2-Benzo-8,10-dimethylanthracene

M. P., °C 145.5-146.5<sup>17</sup>

#### 1,2-Dibenzo-9,10-dimethylanthracene



#### $C_{21}H_{18}$

#### 1,2-Benzo-5-n-propylanthracene

$$C-C-C$$

M. P., °C 90-92<sup>2</sup> 91-91.5<sup>10</sup>

#### 1,2-Benzo-10-n-propylanthracene

M. P., °C 107-10816

## 1,2-Benzo-3-isopropylanthracene

M. P., °C 926

#### 1,2-Benzo-5-isopropylanthracene

#### 1,2-Benzo-6-isopropylanthracene

#### 1,2-Benzo-7-isopropylanthracene

131-1325, 6

1256 1,2-Benzo-8-isopropylanthracene

$$C-C-C$$

#### 1,2-Benzo-10-isopropylanthracene

$$c-c-c$$

M. P., °C 94-95<sup>6</sup> 93-93.5<sup>16</sup>

#### 1,2-Benzo-9-methyl-10-ethylanthracene

$$\frac{c-c}{c}$$

M. P., °C 70–71.5<sup>24</sup>

#### 1,2-Benzo-5,9,10-trimethylanthracene

M. P., °C

#### 1,2-Benzo-6,9,10-trimethylanthracene

M. P., °C 157–158<sup>4</sup>

# 1,2-Benzo-7,9,10-trimethylanthracene

M. P., °C 99.5-100<sup>3</sup>

# 1,2-Benzo-8,9,10-trimethylanthracene

M. P., °C 102-103.5<sup>2</sup>

#### $C_{22}H_{20}$

#### 1,2-(6'-Isopropylbenzo)-10-methylanthracene

M. P., °C 98–99<sup>14</sup>

#### 1,2-Benzo-5-n-butylanthracene

M. P., °C 81<sup>11</sup>

## 1,2-Benzo-10-n-butylanthracene

M. P., °C 96.8-97.5<sup>16</sup> 96.4-96.7<sup>12</sup>

# 1,2-Benzo-9-methyl-10-n-propylan-thracene

M. P., °C 99–101<sup>24</sup>

# 1,2-Benzo-9,10-diethylanthracene

M. P., °C 98.5-99.5<sup>1</sup>

#### 1,2-Benzo-5-ethyl-9,10-dimethylanthracene

M. P., °C 107-108<sup>2</sup>

#### 1,2-Benzo-5,6,9,10-tetramethylanthracene

M. P., °C 132–1334

#### $C_{23}H_{22}$

## 1,2-Benzo-5-n-pentylanthracene

М. Р., °С 9**3**<sup>11</sup>

#### 1,2-Benzo-10-n-pentylanthracene

M. P., °C 82.5-83.5<sup>16</sup> 82.6-83.3<sup>13</sup> 1,2-Benzo-9-methyl-10-n-butylanthracene

M. P., °C 71-72<sup>24</sup>

1,2-Benzo-5-n-propyl-9,10-dimethyl-anthracene

 $\mathbf{C}_{24}\mathbf{H}_{24}$ 

1,2-Benzo-5-n-hexylanthracene

M. P., °C 72–73<sup>11</sup>

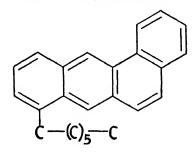
1,2-Benzo-9,10-di-n-propylanthracene

$$c-c-c$$

M. P., °C 100.5–101<sup>2</sup>

C25H26

1,2-Benzo-5-n-heptylanthracene



M. P., ℃ 68<sup>11</sup>

> References on C<sub>20</sub>H<sub>16</sub> through C<sub>25</sub>H<sub>26</sub> Compounds

- Bachmann, W. E., and J. M. Chemerda, J. Am. Chem. Soc. 60, 1023 1938.
- Bachmann, W. E., and J. M. Chemerda, J. Am. Chem. Soc. 61, 2358 1939.
- Bachmann, W. E., and J. M. Chemerda,
   J. Org. Chem. 6, 36 1941.
- Badger, G. M., J. W. Cook, and F. Goulden, J. Chem. Soc. 1940, 16.
- Barry, G., J. W. Cook, G. A. D. Haslewood, C. L. Hewett, I. Hieger, and E. L. Kennaway, Proc. Roy. Soc. (London) 117B, 318 1935.
- 6. Cook, J. W., J. Chem. Soc. 1932, 456.
- 7. Cook, J. W., Proc. Roy. Soc. (London) 111B, 485 1932.

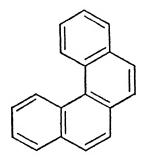
- 8. Cook, J. W., and C. G. M. DeWorms, J. Chem. Soc. 1939, 268.
- 9. Cook, J. W., and G. A. D. Haslewood, J. Chem. Soc. 1934, 428.
- 10. Cook, J. W., and G. A. D. Haslewood, J. Chem. Soc. 1935, 767.
- 11. Cook, J. W., and A. M. Robinson, J. Chem. Soc. 1940, 303.
- 12. Cook, J. W., A. M. Robinson, and F. Goulden, J. Chem. Soc. 1937, 393.
- 13. Davis, W. W., M. E. Krahl, and G. H. A. Clowes, J. Am. Chem. Soc. 62, 3080 1940.
- 14. Fieser, L. F., and R. C. Clapp, J. Am. Chem. Soc. 63, 319 1941.
- 15. Fieser, L. F., M. Fieser, E. B. Hershberg, M. S. Newman, A. M. Seligman, and M. J. Shear, Am. J. Cancer 29, 260 1937.
- 16. Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 59, 1028 1937.

- 17. Fieser, L. F., and W. S. Johnson, J. Am. Chem. Soc. 61, 1647 1939.
- 18. Fieser, L. F., and W. S. Johnson, J. Am. Chem. Soc. 62, 575 1940.
- 19. Fieser, L. F., and R. Jones, J. Am. Chem. Soc. 60, 1940 1938.
- 20. Fieser, L. F., and M. S. Newman, J. Am. Chem. Soc. 58, 2376 1936.
- 21. Fieser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 60, 170 1938.
- 22. Fieser, L. F., and A. M. Seligman, J. Am. Chem. Soc. **61**, 136 **1939**.
- 23. Hartwell, J. L., and H. L. Stewart, J. Natl. Cancer Inst. 3, 277 1942.
- 24. Mikhailov, B. M., and N. G. Černova, Compt. rend. acad. sci. (U.R.S.S.) 20, 579 1938.
- 25. Newman, M. S., J. Am. Chem. Soc. 59, 1003 1937.
- 26. Newman, M. S., J. Am. Chem. Soc. 60 1141 1938.

#### 3,4-BENZOPHENANTHRENE AND ITS ALKYL DERIVATIVES, $C_n II_{2n-24}$

C18H12

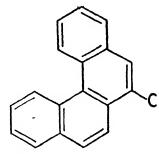
#### 3,4-Benzophenanthrene



M. P., °C  $68^{2}$  $65.6 - 66.2^8$ 

C19H14

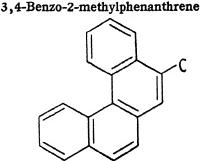
#### 3,4-Benzo-1-methylphenanthrene



M. P., °C 77–787

B. P., °C @ 760mm 210

0.47



M. P., °C 69.5–70<sup>5</sup>

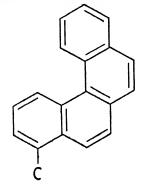
# 3,4-Benzo-6-methylphenanthrene

M. P., °C 80–81<sup>6</sup>

# 3,4-Benzo-7-methylphenanthrene

M. P., °C 54–54.56

# 3,4-Benzo-8-methylphenanthrene



M. P., °C 65–66

 $C_{20}H_{16}$ 

3,4-Benzo-2-ethylphenanthrene

M. P., °C 50.4-51.2°

# 3,4-Benzo-2,9-dimethylphenanthrene

M. P., °C 130.6-131.0<sup>8</sup>

# 3,4-Benzo-6,7-dimethylphenanthrene

M. P., °C 94.5–95.0<sup>4</sup> 94.5–95<sup>3</sup>

#### C21H18

.3,4-Benzo-2-isopropylphenanthrene

M. P., °C 91.5-92.57

 $C_{22}H_{20}$ 

3,4-(6'-Methylbenzo)-9-isopropylphenanthrene

M. P., °C 98-99<sup>1</sup> 3,4-Benzo-2,9-diethylphenanthrene

$$c-c$$

M. P., °C 106.4–107.08

References on 3,4-Benzophenanthrene and Its Alkyl Derivatives

- 1. Adelson, D. E., and M. T. Bogert, J. Am. Chem. Soc. 59, 1776 1937.
- 2. Cook, J. W., J. Chem. Soc. 1931, 2524.
- Fieser, L. F., M. Fieser, and E. B. Hershberg, J. Am. Chem. Soc. 58, 1463 1936.
- Fieser, L. F., M. Fieser, E. B. Hershberg, H. L. Holmes, and M. S. Newman, Science [N.S.] 83, 558 1936.
- 5. Hewett, C. L., J. Chem. Soc. 1936, 596.
- 6. Hewett, C. L., J. Chem. Soc. 1938, 1286.
- 7. Hewett, C. L., J. Chem. Soc. 1940, 293.
- Newman, M. S., and L. M. Joshel, J. Am. Chem. Soc. 60, 485 1938.
- Newman, M. S., and L. M. Joshel, J. Am. Chem. Soc. 62, 972 1940.

5. NAPHTHACENE AND ITS ALKYL DERIVATIVES, C<sub>n</sub>H<sub>2n-24</sub>

## Naphthacene (Tetracene) (Chrysogene) (Rubene)

C18H12

343

3573, 8

34912

341.5-343.07

3411, 2

33714, 15

335-3369

33511, 13

C19H14

#### 2-Methylnaphthacene

M. P., °C 350<sup>6</sup>

 $C_{20}H_{16}$ 

#### 1,6-Dimethylnaphthacene

M. P., °C 138 139<sup>10</sup>

# 2,8-Dimethylnaphthacene

#### 2,9-Dimethylnaphthacene

3626

C21H18

#### 2-Isopropylnaphthacene

M. P., °C 273 2744

# References on Naphthacene and Its Alkyl Derivatives

- 1. Brass, K., and K. Fanta, Ber. 69, 1 1936.
- 2. Clar, E., Ber. 65, 503 1932.
- 3. Clar, E., Ber. 75, 1271 1942.
- 4. Cook, J. W., J. Chem. Soc. 1934, 1412.
- 5. Coulson, E. A., J. Chem. Soc. 1934, 1406.
- 6. Coulson, E. A., J. Chem. Soc. 1935, 77.
- Davis, W. W., M. E. Krahl, and G. H. A. Clowes, J. Am. Chem. Soc. 62, 3080 1940.
- 8. Dufraisse, C., and R. Horelois, Bull. soc. chim. [5] 3, 1880 1936.
- Dziewoński, K., and E. Ritt, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1927A, 181.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 62, 49 1940.
- Gabriel, S., and J. Colman, Ber. 33, 446
   1900.
- 12. Hertel, E., and H. W. Bergk, Z. physik. Chem. 33B, 319 1936.
- Weitzenbock, R., and H. Lieb, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 121, Abt. IIb, 225 1912.
- 14. Winterstein, A., and K. Schon, Naturwissenschaften 22, 237 1934.
- Winterstein, A., K. Schön, and H. Vetter, Z. physiol. Chem. 230, 158 1934.

#### 6. CHRYSENE AND ITS ALKYL DERIVATIVES, C<sub>n</sub>H<sub>2n-24</sub>

#### $C_{18}H_{12}$

#### Chrysene

# M. P., °C

252

255-25673

25561, 66

254.5-2558, 12, 25

254.1-254.49

253.2-254.250

25437, 46

253-25159

253.2-253.820

 $252.5^{53}$ 

25257

251-25262

25171

250 (a)

249-25023

248-25013, 34

 $247.5 - 249.5^7$ 

24924, 74

248-24917, 18, 62

247-2494

246-248.563

2.1836, 39, 54, 56, 68, 69

247-2481, 62

246-24821

24716

24540, 67, 72

B. P., °C @ 760mm 448<sup>41</sup>, <sup>44</sup>, <sup>65</sup> 436.0<sup>35</sup>

 $D_4^{20}$ 

1.273 (solid)8

1.274 (solid)8. 42

Additional Data

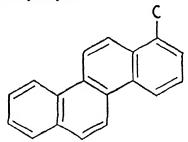
Sublimation Temp. (°C)

190 169 20mm<sup>30</sup> in vac. 41

(a) The melting point 250 is found in references 6, 11, 15, 22, 30, 33, 41, 43, 47, 51, 52, 58, 64, 70, 75.

#### C19H14

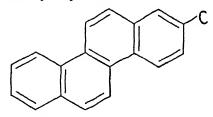
#### 1-Methylchrysene



M. P., °C

 $249.5 - 250^{5}$ 

#### 2-Methylchrysene



M. P., °C

229-2304

#### 3-Methylchrysene

M. P., °C

170-170.5

# 4-Methylchrysene

M. P., °C 151-151.5<sup>4</sup>. <sup>27</sup> 151<sup>39</sup> 149-149.5<sup>5</sup>

#### 5-Methylchrysene

M. P., °C 117.8 118-118.8<sup>2</sup> 118-117.7<sup>20</sup> 117.3-117.7<sup>20</sup> 116.8-117.6<sup>28</sup> 115-117<sup>49</sup>

## 6-Methylchrysene

M. P., °C 160.4 161.0-161.4<sup>48</sup> 160.4-160.9<sup>20</sup> 159-159.8<sup>29</sup> 159.5<sup>14</sup>

#### C20H16

#### 1-Ethylchrysene

M. P., °C 236<sup>31</sup> 183.5-184<sup>5</sup>

#### 2-Ethylchrysene

M. P., °C 126<sup>31, 32</sup>

#### 5-Ethylchrysene

M. P., °C 88–9049

#### 6-Ethylchrysene

$$C-C$$

M. P., °C 126.4-126.8<sup>48</sup>

#### 1,4-Dimethylchrysene

M. P., °C 141.5-142.5<sup>5</sup>

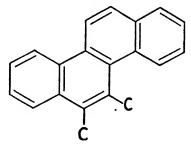
# 2,3-Dimethylchrysene

M. P., °C 215-215.3<sup>25</sup>

# 2,8-Dimethylchrysene

M. P., °C 218<sup>55</sup>

#### 5,6-Dimethylchrysene



M. P., °C 128.6–129.8<sup>49</sup> 128.5–129.2<sup>20</sup> 127–128<sup>58</sup>

# 2,11-Dimethylchrysone

## 4,11-Dimethylchrysene

M. P., °C 147–147.5<sup>26</sup>

#### $C_{21}H_{18}$

#### 4-Isopropylchrysene

M. P., °C 227<sup>19</sup>

#### 6-Isopropylchrysene

$$C-C-C$$

M. P., °C 137<sup>10</sup>

#### 5-Methyl-6-ethylchrysene

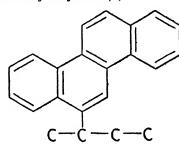
M. P., °C 125.5–126.5<sup>14</sup>

#### 1,2,7-Trimethylchrysene

M. P., °C 281–282<sup>60</sup>

#### $C_{22}H_{20}$

#### 6-sec-Butylchrysene (a)



M. P., °C 10010

(a) The structure of this compound was not clearly defined in the literature.

#### x,x-Diethylchrysene (a)

M. P., °C 145<sup>32</sup>

(a) The structure of this compound was not clearly defined in the literature.

#### 1,2,7,8-Tetramethylchrysene

M. P., °C 298–299<sup>60</sup>

#### References on Chrysene and Its Alkyl Derivatives

- 1. Adler, A., Ber. 12, 1889 1879.
- Bachmann, W. E., and R. O. Edgerton, J. Am. Chem. Soc. 62, 2550 1940.
- Bachmann, W. E., and S. R. Safir, J. Am. Chem. Soc. 63, 2601 1941.
- Bachmann, W. E., and W. S. Struve, J. Org. Chem. 4, 456 1939.
- Bachmann, W. E., and W. S. Struve, J. Org. Chem. 5, 416 1940.
- Bamberger, E., and F. Chattaway, Ber. 26, 1745 1893.
- Barry, G., and J. W. Cook, Am. J. Cancer 20, 58 1934.
- Baxter, G. P., and A. H. Hale, J. Am. Chem. Soc. 58, 510 1936.
- Baxter, G. P., and A. H. Hale, J. Am. Chem. Soc. 59, 506 1937.
- Bergmann, F., and H. E. Eschinazi, J. Am. Chem. Soc. 65, 1413 1943.
- Am. Chem. Soc. **65**, 1413 **1943**. 11. Beschke, E., M. Winograd-Finkel, and
- G. Köhres, Ann. 384, 143 1911.
  12. Bottomley, A. C., and C. C. Twort, Am. J. Cancer 21, 781 1934.
- 13. Bradley, M. J., and S. W. Parr, Chem. &
- Met. Eng. 27, 737 1922. 14. Bradsher, C. K., and A. S. Burhans, J.
- Am. Chem. Soc. **62**, 3140 **1940**. 15. Butenandt, A., and H. Thompson, Ber. **67**, 140 **1934**.
- Chuang, C.-K., Y.-T. Huang, and C.-M. Ma, Ber. 72, 713 1939.
- 17. Cook, J. W., and C. L. Hewett, J. Chem. Soc. 1934, 365.

- Cook, J. W., C. L. Hewett, and I. Hieger, J. Chem. Soc. 1933, 395.
- Cook, J. W., and A. M. Robinson, J. Chem. Soc. 1938, 505.
- Davis, W. W., M. E. Krahl, and G. H. A. Clowes, J. Am. Chem. Soc. 62, 3080 1940.
- Diels, O., and W. Gadke, Ber. 60, 140 1927.
- Diels, O., and Λ. Karstens, Ann. 478, 129
   1930.
- Dieterle, H., and H. Rochelmeyer, Arch. Pharm. 273, 532 1935.
- Dimroth, K., and E. Stockstrom, Ber. 75, 180 1942.
- Fieser, L. F., M. Fieser, and E. B. Hershberg, J. Am. Chem. Soc. 58, 1463 1936.
- Fieser, L. F., M. Fieser, E. B. Hershberg, H. L. Holmes, and M. S. Newman, Science [N.S.] 83, 553 1936.
- Fieser, L. F., and W. S. Johnson, J. Am. Chem. Soc. 61, 1617 1939.
- Fieser, L. F., and L. M. Joshel, J. Am. Chem. Soc. 62, 1211 1940.
- Fieser, L. F., L. M. Joshel, and A. M. Seligman, J. Am. Chem. Soc. 61, 2131 1939.
- Fischer, F., H. Schrader, and W. Meyer, Ges. Abhandl. Kenntnis Kohle 5, 413 1920.
- Funke, K., and E. Muller, J. prakt. Chem. [2] 144, 212 1936.
- Funke, K., and J. Ristie, J. prakt. Chem. [2] 146, 151 1936.
- 33. Graebe, C., Ann. 163, 361 1872.
- 34. Graebe, C., Ber. 7, 782 1874.
- 35. Graebe, C., and J. Walter, Ber. 14, 175 1881.
- Goldschmiedt, G., and M. von Schmidt, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 83, Abt. 11b, 7 1881.
- 37. Hansen, C. J., Ber. 42, 210 1909.
- 38. Hewett, C. L., J. Chem. Soc. 1940, 293.
- Jones, W. E., and G. R. Ramage, J. Chem. Soc. 1938, 1853.
- 40. Kopp, A., Mon. sci. [3] 8, 1147 1878.
- 41. Krafft, F., and H. Weiland, Ber. 29, 2240 1896.
- 42. Krishnan, K. S., and S. Banerjee, Trans. Roy. Soc. (London) 234, 265 1935.
- 43. Kruber, O., Ber. 57, 1008 1924.

- Lagerlöf, D., J. prakt. Chem. [2] 98, 136
   1918.
- Lewis, H. J., G. R. Ramage, and R. Robinson, J. Chem. Soc. 1935, 1412.
- 46. Meyer, H., Z. angew. Chem. 37, 796 1924.
- 47. Meyer, R., Ber. 45, 1609 1912.
- 48. Newman, M. S., J. Am. Chem. Soc. 60, 2947 1938.
- 49. Newman, M. S., J. Am. Chem. Soc. 62, 870 1940.
- Newman, M. S., and J. A. Cathcart, J. Org. Chem. 5, 618 1940.
- Orlov, N. A., and N. D. Likhachev, Ber.
   719 1929.
- Orlov, N. A., and N. D. Likhachev, Ber.
   63, 2179 1930.
- Pascal, P., Bull. soc. chim. [4] 29, 641
   1921.
- Peak, D. A., and R. Robinson, J. Chem. Soc. 1936, 759.
- 55. Ramage, G. R., J. Chem. Soc. 1938, 397.
- Ramage, G. R., and R. Robinson, J. Chem. Soc. 1933, 607.
- Raudnitz, H., F. Petrů, and F. Haurowitz, Z. physiol. Chem. 209, 103 1932.
- Raudnitz, H., F. Petrů, and A. Stadler, Ber. 66, 879 1933.
- Ruzicka, L., L. Ehmann, M. W. Goldberg, and H. Hosli, Helv. Chim. Acta 16, 833 1933.
- Ruzicka, L., A. Grob, and G. Anner, Helv. Chim. Acta 26, 254 1943.

- Ruzicka, L., and H. Hósli, Helv. Chim. Acta 17, 470 1934.
- Ruzicka, L., G. Thomann, E. Brandenberger, M. Furter, and M. W. Goldberg, Helv. Chim. Acta 17, 200 1934.
- Schmid, L., and M. Zentner, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 137, Abt. IIb, 92 1928.
- 64. Schmidt, E., J. prakt. Chem. [2] 9, 241 1874.
- 65. Schweitzer, R., Ann. 264, 193 1891.
- Sturrock, M. G., and T. Lawe, Can. J. Research 17 B, 71 1939.
- Tsukamoto, T., J. Pharm. Soc. Japan 48, 18 1928; Chem. Zentr. 1928, I, 2408.
- 68. Versmann, F., Chem. News 30, 203 1874.
- von Braun, J., and G. Irmisch, Ber. 64, 2461 1931.
- 70 Weger, M., Z. angew. Chem. 22, 338 1909.
- Weitzenbock, R., and H. Lieb, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 121, Abt. 11b, 225 1912.
- Winterstein, A., and K. Schon, Z. physiol. Chem. 230, 146 1934.
- Winterstein, A., K. Schön, and H. Vetter, Z. physiol. Chem. 230, 158 1934.
- Zanetti, J. E., and G. Egloff, Ind. Eng. Chem. 9, 474 1917.
- 75. Zeidler, O., Ann. 191, 285 1878.

#### 7. TRIPHENYLENE AND ITS ALKYL DERIVATIVES, C<sub>n</sub>H<sub>2n-24</sub>

#### C18H12

#### Triphenylene

200<sup>5</sup> 199<sup>1</sup> 198–198.5<sup>11</sup>, <sup>18</sup>

198<sup>4</sup>, <sup>12</sup> 197.7-198<sup>1</sup>

197.5-197.9<sup>7</sup>

 $196.5 - 197.5^3$ 

19716

196-19713, 14

 $196.5^{10}$ 

19611, 15

 $195^{6}$ 

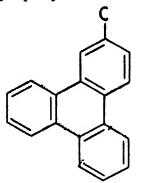
B. P., °C @ 760mm 4251

 $C_{19}H_{14}$ 

#### 1-Methyltriphenylene

M. P., °C 93.4-94.2° 93-94°

#### 2-Methyltriphenylene

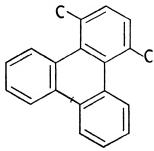


M. P., °C 102.6–103.6° 101–102°

# $\boldsymbol{C_{20}H_{16}}$

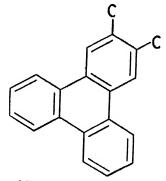
# 1,2-Dimethyltriphenylene

# 1,4-Dimethyltriphenylene



M. P., °C 108.4-109.29

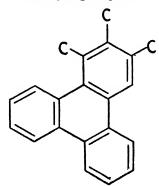
# 2,3-Dimethyltriphenylene



M. P., °C 156.7–157.2<sup>8</sup>

# $C_{21}H_{18} \\$

#### 1,2,3-Trimethyltriphenylene



#### M. P., °C 109.8-110.68

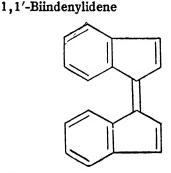
# References on Triphenylene and Its Alkyl Derivatives

- Bachmann, W. E., and H. T. Clarke, J. Am. Chem. Soc. 49, 2089 1927.
- Bachmann, W. E., and G. D. Cortes, J. Am. Chem. Soc. 65, 1329 1943.
- Bachmann, W. E., and W. S. Struve, J. Org. Chem. 4, 472 1939.
- Bergmann, E., and O. Blum-Bergmann,
   J. Am. Chem. Soc. 59, 1441 1937.
- Berthelot, Bull. soc. chim. [2] 22, 437 1874.
- Blum-Bergmann, O., J. Am. Chem. Soc. 60, 1999 1938.
- Davis, W. W., M. E. Krahl, and G. H. A. Clowes, J. Am. Chem. Soc. 62, 3080 1940.

- Fieser, L. F., and W. H. Daudt, J. Am. Chem. Soc. 63, 782 1941.
- Fieser, L. F., and L. M. Joshel, J. Am. Chem. Soc. 61, 2958 1938.
- 10. Kaffer, H., Ber. 66, 1812 1935.
- 11. Mannich, C., Ber. 40, 159 1907.
- Mohler, H., and J. Sorge, Helv. Chim. Acta 22, 229 1939.
- Nenitzescu, C. D., and D. Curcăneanu, Ber. 70, 346 1937.
- Pirrone, F., Gazz. chim. ital. 66, 244
   1936.
- 15. Schmidt, H., and G. Schultz, Ann. 203, 118 1880.
- Sturrock, M. G., and T. Lawe, Can. J. Research 17B, 71 1939.
- Ward, J. J., W. R. Kirner, and H. C. Howard, J. Am. Chem. Soc. 67, 246 1945.
- Weitzenböck, R., and H. Lieb, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 121, Abt. IIb, 225 1912.

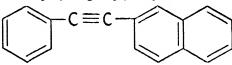
# 8. MISCELLANEOUS POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-24}$

# $C_{18}H_{12}$



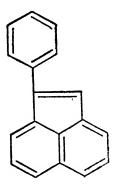
- M. P., °C 232–235 (a)<sup>47</sup>
- (a) This compound melts with decomposition.

#### Phenyl-(2-naphthyl)-ethyne



M. P., °C 117<sup>37</sup> 115–116<sup>38</sup>

#### 1-Phenylacenaphthylene



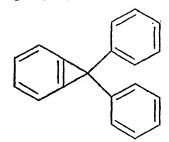
M. P., °C 54-55\*\*

## Benzo-[de]-9-methyleneanthracene

M. P., °C 225<sup>13, 14</sup>

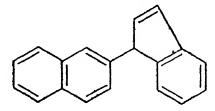
#### $C_{19}H_{14}$

#### 1,1-Diphenyl-2,3-benzocyclopropane



M. P., °C 144<sup>44, 45</sup> 138<sup>34</sup>

## 2-(1'-Indenyl)-naphthalene

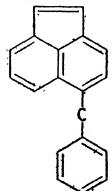


M. P., °C 88<sup>50</sup> 87<sup>16</sup>

B. P., °C @ 760mm 246-250

1450

# 5-Benzylacenaphthylene

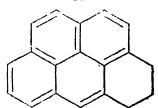


M. P., °C 104-105<sup>20</sup>

#### 3-Isopropenylpyrene

M. P., °C 61.5-62.5<sup>23</sup>. <sup>24</sup>

#### Cyclohexano-[cd]-pyrene



M. P., °C 107-10848

 $C_{20}H_{16}$ 

1-Phenyl-5-(1'-indenylidene)pentadiene-1,3

M. P., °C 182–183¹

1-Phenyl-4-(1'-naphthyl)-butadiene-1,3

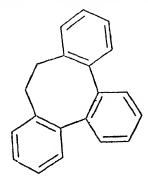
M. P., °C 109<sup>30</sup>

# 3-Phenethenylacenaphthene

M. P., °C 93.2–94.0<sup>25</sup> 3-(2'-Buten-1'-yl)-pyrene

M. P., °C 75 · 76<sup>25</sup>

1,2,3,4,5,6-Tribenzocyclooctane



M. P., °C 111--113<sup>46</sup>

2,3-Cyclopentano-5,6-benzophenalene

M. P., °C 144.5-145<sup>29</sup>

#### 6,12b-Dihydrocholanthrene

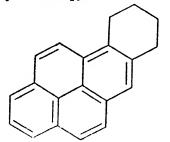
M. P., °C 161.5-162.5<sup>3</sup>

#### Cyclopentano-[x,x]-x,x-dihydrochrysene (a)

M. P., °C 153.2–153.5<sup>27</sup>

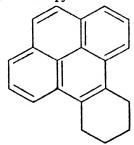
(a) The structure of this compound was not clearly defined in the literature.

#### 1,2-Cyclohexanopyrene



M. P., °C 135<sup>52</sup> 128<sup>36</sup>

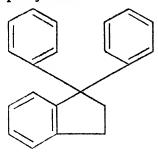
# 4,5-Cyclohexanopyrene



M. P., °C 113<sup>26</sup>

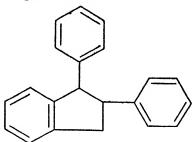
 $C_{21}H_{18}$ 

#### 1,1-Diphenylindane



M. P., °C 67-68<sup>31</sup> 67<sup>32</sup>

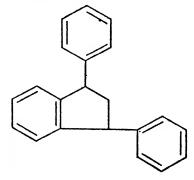
#### 1,2-Diphenylindane



M. P., °C 123-124.5 (a)<sup>8</sup> 89 (a)<sup>8</sup>

(a) These constants were determined on isomeric forms.

#### 1,3-Diphenylindane



M. P., °C 156–157<sup>41, 54</sup> 155–156<sup>42</sup>

#### meso-3-Methyl-6, 12b-dihydrocholanthrene

M. P., °C 138-139<sup>28</sup>

#### 3-Methyl-11,12-dihydrocholanthrene

M. P., °C 154.5–155<sup>28</sup>

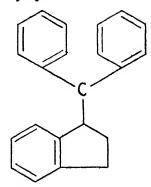
# 3-Methyl-x,x-dihydrocholanthrene (a)

M. P., °C 136–137<sup>2</sup>

(a) The structure of this compound was not clearly defined in the literature.

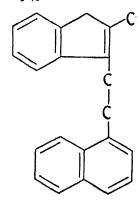
#### C22H20

#### 1-Benzhydrylindane



M. P., °C 85–86<sup>18</sup>

1-(1'-Naphthyl)-2-[3''-(2''-methyl-indenyl)]-ethane



M. P., °C 99–99.54

1-(1'-Naphthyl)-2-[1''-(3'',4''-dihydronaphthyl)]-ethane

B. P., °C @ 760mm 225–227

139

1,2-Diphenyl-1,2,3,4-tetrahydronaphthalene

B. P., °C @ 760mm 183-184

 $1.5^{10}$ 

2,3-Diphenyl-1,2,3,4-tetrahydronaphthalene

M. P., °C 129-129.5<sup>19</sup>

2-Methyl-3-phenethyl-4,5-benzoin-dene

B. P., °C @ 760mm 190–195

 $0.4^{15}$ 

1-Methyl-1,2-(2',3'-indano)-1,2,3,4tetrahydrophenanthrene

M. P., °C 111-1124

B. P., °C @ 760mm 190 -195

0.24

Cyclopentano-[hi]-2,3-dimethyl-x,x-dihydrochrysene (a)

M. P., °C 193.5--194.5<sup>27</sup>

(a) The structure of this compound was not clearly defined in the literature.

Spiro[3,4-benzofluorene-9,1'-cyclohexane]

B. P., °C @ 760mm 225-230

0.059

 $C_{23}H_{22}$ 

#### 1-Methyl-1-phenyl-2-benzylindane

# 117-118.5<sup>11</sup> 1-Phonyl-2-methyl-3-benzylindane

#### B. P., °C @ 760mm 242-245

16°

#### 1,2-Dimethyl-1,3-diphenylindane

#### M. P., °C 97.5–98.5<sup>11</sup>

# 1-(1'-Naphthyl)-2-[3''-(2'',5''-dimethylindenyl)]-ethane

M. P., °C 81-81.54

# 1-(1'-Naphthyl)-2-[3"-(2",6"-dimethylindenyl)]-ethane

B. P., °C @ 760mm 184-186

 $0.05^{4}$ 

1-(1'-Naphthyl)-2-[3''-(2'',7''-dimethylindenyl)]-ethane

M. P., °C 964

1,2-(2',1'-Naphtho)-6,9a-dimethyl-3,4,4a,9a-tetrahydrofluorene

1,2-(2',1'-Naphtho)-7,9a-dimethyl-3,4,4a,9a-tetrahydrofluorene

M. P., °C 134.5–1354 1,2-(2',1'-Naphtho)-8,9a-dimethyl-3,4,4a,9a-tetrahydrofluorene

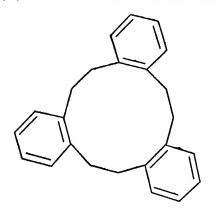
M. P., °C 121-1224

 $C_{24}H_{24}$ 

1,4-Dibenzyl-1,2,3,4-tetrahydronaphthalene

M. P., °C 92-93<sup>21</sup> ?

#### 1,2,5,6,9,10-Tribenzocyclododecane



M. P., °C 184.5<sup>7</sup>

# $\mathbf{C}_{25}\mathbf{H}_{26}$

#### 1-Isopropyl-3-benzhydrylindane

M. P., °C 126.5-127.5<sup>53</sup>

# 1-(1'-Naphthyl)-2-[3''-(4''-methyl-7''-isopropylindenyl)]-ethane

M. P., °C 92–93<sup>17</sup>

This series continued on next page

1-[1'-(7'-Methylnaphthyl)]-2-[1''-(5'',6''-dimethyl-3'',4''-dihy-dronaphthyl)]-ethane

B. P., °C @ 760mm 215-217

0.340

# Spiro[4-isopropyl-7-methylindane-1, 1'-phenalan] (a)

M. P., °C 82-83<sup>17</sup>

(a) The above formula was given for this compound, but the name given in the literature was "7-methyl-4-isopropylhydrindene-1,7'-spiro-7', 8'-dihydrophenalene."

# $C_{28}H_{28}$

1-sec-Butyl-3-benzhydrylindane

$$\begin{array}{c|c} c - c - c \\ \hline \end{array}$$

M. P., °C 104-106<sup>53</sup>

1,2,3,4,5,6,7,8-Tetracyclopentano-9,10-dihydroanthracene

M. P., °C 377-378 Diindano-[3',2',1'-de,3",2",1"-kl]tetradecahydroanthracene

M. P., °C 246–248<sup>49</sup>

Indo-[1',2',3'-de]-(3a",4",5",6",
7",7a"-hexahydroindo)-[1",2",
3"-kl]-1,2,3,4,5,6,7,8-octahydroanthracene

M. P., °C 271–273<sup>49</sup>

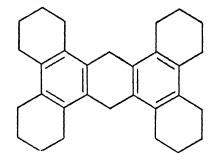
 $C_{28}H_{32}$ 

1,2,3,4-Dicyclopentano-5,6,7,8-dicyclohexano-9,10-dihydroanthracene

M. P., °C 333-334<sup>5</sup>

C<sub>30</sub>H<sub>36</sub>

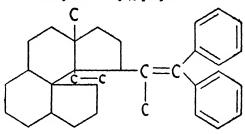
1,2,3,4,5,6,7,8-Tetracyclohexano-9,10-dihydroanthracene



M. P., °C 382–383<sup>6</sup>

C34H44

1,1-Diphenyl-2-[3'-(hexahydroindano-[e-7'',7a'']-3a'-ethyl-7a'-methylhexahydroindanyl)]-propene-1



M. P., °C 178<sup>12</sup>

#### C36H48

1,2,3,4,5,6-Tri-[2',1'-(2a',3',4',5', 5a',6',7',8',8a',8b'-decahydro-acenaphtho)]-benzene

M. P., °C 217-219<sup>51</sup> 215<sup>49</sup>

B. P., °C @ 760mm 360

 $0.2^{19}$ 

(a) The structures given cholestane and derivatives vary. The above name and structure appear in reference 48.

References on Miscellaneous Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-24}$ 

- 1. Alexa, V., Ber. 64, 2250 1931.
- 2. Bachmann, W. E., J. Org. Chem. 1, 347 1936-37.
- 3. Bachmann, W. E , J. Org. Chem. 3, 434 1938-39.
- Bachmann, W. E., J. W. Cook, C. L. Hewett, and J. Iball, J. Chem. Soc. 1936, 54.
- Backer, H. J., and L. H. H. Huisman, Rec. trav. chim. 60, 557 1941; C.A. 37, 5982 1943.
- Backer, H. J., J. Strating, and L. H. H. Huisman, Rec. trav. chim. 58, 761 1939.
- Baker, W., R. Banks, D. R. Lyon, and F. G. Mann, J. Chem. Soc. 1945, 27.
- Bergmann, E., and H. Weisz, Ann. 480, 64 1930.
- Bergmann, F., and H. E. Eschinazi, J. Am. Chem. Soc. 66, 183 1944.

#### C<sub>37</sub>H<sub>50</sub>

#### 3- $\alpha$ -Naphthylcholestadiene (a)

M. P., °C 131–133<sup>48</sup>

Additional Data

 $[\alpha]_{\rm p}^{29} = -49.7^{\circ}$  48

311 References

- Bergmann, F., H. E. Eschinazi, and D. Schapiro, J. Am. Chem. Soc. 64, 557 1942.
- Blum-Bergmann, O., J. Chem. Soc. 1938, 723.
- Butenandt, A., F. Hildebrandt, and H. Brücker, Ber. 64, 2529 1931.
- Charrier, G., and E. Ghigi, Atti accad. Lincei [6] 16, 257 1932.
- Charrier, G., and E. Ghigi, Gazz. chim. ital. 62, 928 1932.
- Cook, J. W., A. Dansi, C. L. Hewett,
   J. Iball, W. V. Mayneord, and E. Roe,
   J. Chem. Soc. 1935, 1319.
- Cook, J. W., and G. A. D. Haslewood, J. Chem. Soc. 1935, 770.
- Cook, J. W., C. L. Hewett, W. V. Mayneord, and E. Roe, J. Chem. Soc. 1934, 1727
- 18. Courtot, C., Ann. chim. [9] 5, 52 1916.
- Crawford, H. M., J. Am. Chem. Soc. 61, 608 1939.
- Dziewoński, K, and K. Leonhard, Bull. intern. acad polon. sci., Classe sci. math. nat. 1928A, 99.
- Dziewoński, K., and J. Moszew, Roczniki Chem. 11, 169 1931; C.A. 26, 131 1932; Chem. Zentr. 1931, I, 2875.
- 22 Dziewoński, K., J. Moszew, S. Lepiankiewicz, and L. Sucheni, Bull. intern. acad polon.sci., Classe sci. math. nat. 1929A, 950.
- Dziewoński, K., and L. Sternbach, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1937A, 81.
- Dziewoński, K., and L. Sternbach, Roczniki Chem. 17, 101 1937; Chem. Zentr. 1937, II, 65.
- Dziewoński, K., and P. Tresiński, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1937A, 579.
- Fieser, L. F., and M. Fieser, J. Am. Chem. Soc. 57, 782 1935.
- Fieser, L. F., M. Fieser, and E. B. Hershberg, J. Am. Chem. Soc. 58, 1463 1936.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 60, 940 1938.

- Fieser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 59, 883 1937.
- Friedmann, E., and W. E. van Heyningen, J. prakt Chem. [2] 146, 163 1936.
- 31. Gagnon, P., Ann. chim. [10] 12, 296 1929.
- Gagnon, P. E., L. Gravel, and L. P. Amiot, Can. J. Research 22B, 32 1944.
- 33. Ghigi, E., Ber. 73, 677 1940.
- 31. Hemilian, W., Ber. 7, 1203 1874.
- Hershberg, E. B., and L. M. Joshel, J. Am. Chem. Soc. 62, 1305 1940.
- Kon, G. A. R., and E. M. F. Roe, J. Chem. Soc. 1945, 143.
- Ruggli, P., and A. Jenny, Helv. Chim. Acta 10, 228 1927.
- Ruggli, P., and M. Reinert, Helv. Chim. Acta 9, 67 1926.
- Ruzicka, L., and H. Hosli, Helv. Chim. Acta 17, 470 1934.
- 41. Schlenk, W., and E. Bergmann, Ann. 479, 58 1930.
- Schlenk, W., and E. Bergmann, Ann. 479, 78 1930.
- 43. Scholl, R., and K. Meyer, Ber. **69**, 152 1936.
- 44. Schwartz, II., Ber 14, 1516 1881.
- 45. Schwarz, H., Bull. soc. chim. [2] 37, 24 1882.
- 16. Shuttleworth, R. G., W. S. Rapson, and E. T. Stewart, J. Chem. Soc. 1944, 71.
- Straus, F., R. Kuhnel, and R. Haensel, Ber. 66, 1847 1933.
- Urushibara, Y., T. Ando, H. Araki, and A. Ozawa, Bull. Chem. Soc. Japan 12, 353 1937.
- 49. von Braun, J., Ber. 67, 214 1934.
- von Braun, J., G. Manz, and E. Reinsch, Ann. 468, 277 1929.
- Ward, J. J., W. R. Kirner, and H. C. Howard, J. Am. Chem. Soc. 67, 246 1945.
- Winterstein, A., H. Vetter, and K. Schön, Ber. 68, 1079 1935.
- 53. Wuest, H.-M., Ann. 415, 291 1918.
- Ziegler, K., H. Grabbe, and F. Ulrich, Ber. 57, 1983 1924.

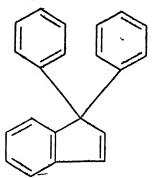
#### X. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA C<sub>n</sub>H<sub>2n-26</sub>

- 1. Indene with Two Phenyl Substitutions
- 2. Binaphthyls, Their Alkyl Derivatives, and Two Naphthyl Substitutions on Alkanes
- 3. Anthracene and Phenanthrene with One Phenyl Substitution
- 4. Cholanthrene and Its Alkyl Derivatives
- 5. Other Benzocyclanoanthracenes, Benzocyclanophenanthrenes, and Their Alkyl Derivatives
- 6. Miscellaneous Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-26</sub>

# 1. INDENE WITH TWO PHENYL SUBSTITUTIONS, C<sub>n</sub>H<sub>2n-26</sub>

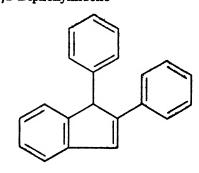
#### $\mathbf{C}_{21}\mathbf{H}_{16}$

#### 1,1-Diphenylindene



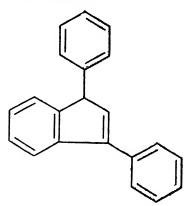
M. P., °C 91-92<sup>13</sup> 90-91<sup>14</sup>

#### 1,2-Diphenylindene



M. P., °C 177 177–178<sup>19, 20, 21</sup> 177<sup>25</sup> 176–177<sup>23, 24</sup> 175<sup>3</sup> 174–175<sup>1</sup>

## 1,3-Diphenylindene



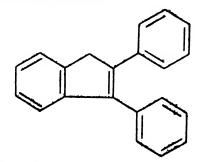
 $\begin{array}{c} \text{M. P., °C} \\ 84\text{--}85 \text{ (a)}^{12} \\ 71\text{--}72^{31} \\ 68\text{--}69 \text{ (a)}^{12} \end{array}$ 

B. P., °C @ 760mm 230

(a) These constants were determined on different crystalline forms.

1531

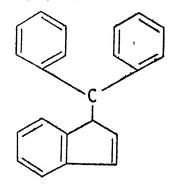
# 2,3-Diphenylindene



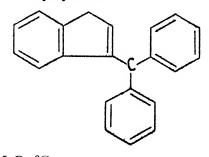
M. P., °C 109.5 111<sup>7</sup> 110<sup>2</sup>

 $\boldsymbol{C_{22}H_{18}}$ 

#### 1-Benzhydrylindene



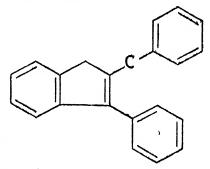
#### 3-Benzhydrylindene



# 1-Phenyl-2-benzylindene

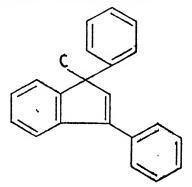
# 2-Benzyl-3-phenylindene

97-996



M. P., °C 101<sup>5</sup> 96-98<sup>4, 8</sup> 96-97.5<sup>8</sup> 92-93<sup>22</sup>

# 1-Methyl-1,3-diphenylindene



M. P., °C 59-60<sup>16, 17</sup>

B. P., °C @ 760mm 205–210

717

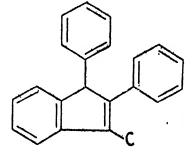
1-Methyl-2,3-diphenylindene

M. P., °C 106.5<sup>17</sup>

1,3-Diphenyl-2-methylindene

M. P., °C 108<sup>15</sup>

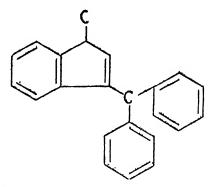
1,2-Diphenyl-3-methylindene



M. P., °C 91<sup>17</sup>

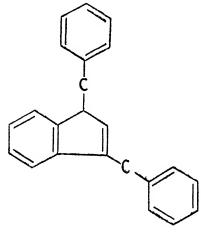
 $C_{23}H_{20}$ 

1-Methyl-3-benzhydrylindene



M. P., °C 162.5-163.5<sup>30</sup>

1,3-Dibenzylindene

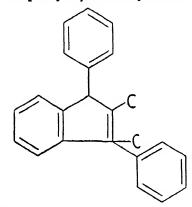


M. P., °C 62–63<sup>27</sup>

#### 1-p-Tolyl-2-phenyl-5-methylindene

M. P., °C 145–146<sup>18</sup>

#### 1,3-Diphenyl-2,3-dimethylindene



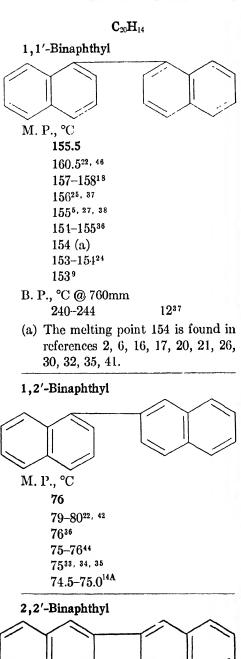
M. P., °C 68.5-69.5°

#### References on Indene with Two Phenyl Substitutions

- Banús, A. G., Anales soc. españ. fís. quím. 26, 372 1928; Chem. Zentr. 1929, I1, 1412.
- Banús, A. G., and F. Calvet, Anales soc. españ. fís. quím. 27, 49 1929; C.A. 23, 3205 1929.
- 3. Bergmann, E., and A. Bondi, Ber. 63, 1158 1930.
- 4. Bergmann, E., and H. Weisz, Ann. 480, 64 1930.

- Bettzieche, F., and A. Ehrlich, Z. physiol. Chem. 160, 1 1926.
- 6. Blum-Bergmann, O., Ann. 484, 26 1930.
- 7. Blum-Bergmann, O., Ber. 65, 109 1932.
- Blum-Bergmann, O., J. Chem. Soc. 1935, 1020.
- Blum-Bergmann, O., J. Chem. Soc. 1938, 723.
- 10. Courtot, C., Ann. chim. [9] 5, 52 1916.
- 11. Courtot, C., Compt. rend. 160, 523 1915.
- Dufraisse, C., and L. Enderlin, Bull. soc. chim. [5] 1, 267 1934.
- 13. Gagnon, P., Ann. chim. [10] 12, 296 1929.
- Gagnon, P. E., L. Gravel, and L. P. Amiot, Can. J. Research 22B, 32 1944.
- Ingold, C. K., and C. L. Wilson, J. Chem. Soc. 1933, 1493.
- Koelsch, C. F., J. Am. Chem. Soc. 56, 1605 1934.
- Koelsch, C. F., and P. R. Johnson, J. Am. Chem. Soc. 65, 567 1943.
- McKenzie, A., and J. R. Myles, Ber. 65, 209 1932.
- 19. Orekhov, A., Ber. 47, 89 1914.
- Orekhov, A., Bull. soc. chim. [4] 25, 598 1919.
- Orekhov, A. P., and R. Grinberg, J. Russ. Phys. Chem. Soc. 48, 1702 1916; C.A. 11, 3275 1917; Chem. Zentr. 1923, I, 1619.
- Orekhov, A. P., and R. Grinberg, J. Russ. Phys. Chem. Soc. 48, 1713 1916;
   C.A. 11, 3275 1917; Chem. Zentr. 1923, I, 1619.
- 23. Orekhov, A., and M. Tiffeneau, Bull. soc. chim. [4] 31, 253 1922.
- 24. Roger, R., and A. McKenzie, Ber. 62, 272 1929.
- 25, Ruggli, P., Ann. 414, 125 1917.
- Schlenk, W., and E. Bergmann, Ann. 463, 1 1928.
- 27. Thiele, J., and S. Bühner, Ann. 347, 249 1906.
- Thiele, J., and K. Merck, Ann. 415, 257 1918.
- Thiele, J., and P. Ruggli, Ann. 393, 61 1912.
- 30. Wuest, H.-M., Ann. 415, 291 1918.
- Ziegler, K., H. Grabbe, and F. Ulrich, Ber. 57. 1983 1924.

# 2. BINAPHTHYLS, THEIR ALKYL DERIVATIVES, AND TWO NAPHTHYL SUBSTITUTIONS ON ALKANES, $C_nH_{2n-26}$



M. P., °C

187

190<sup>3, 28</sup>

187–189<sup>12</sup>

188<sup>43</sup>

187–188<sup>4</sup>

187.5<sup>22</sup>

187 (ii)

186<sup>23, 47</sup>

184<sup>25</sup>

182–184<sup>31</sup>

181–182.5<sup>29</sup>

182<sup>14, 15</sup>

- B. P., °C @ 760mm 452<sup>7</sup>
- (a) The melting point 187 is found in references 1, 8, 10, 11, 13, 17, 19, 33, 34, 35, 36, 39, 40, 45.

#### References on Binaphthyls

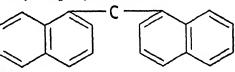
- Bamberger, E., and F. D. Chattaway, Ann. 284, 52 1895.
- Bennett, G. M., and E. E. Turner, J. Chem. Soc. 105, 1057 1914.
- Bodroux, F., Bull. soc. chim. [3] 25, 491 1901.
- Bradley, M. J., and S. W. Parr, Chem. & Met. Eng. 27, 737 1922.
- Brass, K., and R. Patzelt, Ber. 70, 1349 1937.
- Busch, M., and W. Weber, J. prakt. Chem. [2] 146, 1 1936.
- Chattaway, F. D., J. Chem. Soc. 67, 653 1895.
- Chattaway, F. D., and W. H. Lewis, J. Chem. Soc. 65, 869 1894.
- Corbellini, A., and F. Steffenoni, Rend. ist. lombardo sci. [2] 69, 429 1936; Chem. Zentr. 1937, I, 1420.
- Cumming, W. M., and G. Howie, J. Chem. Soc. 1932, 528.
- Cumming, W. M., and G. D. Muir, J. Roy. Tech. Coll. Glasgow 4, 61 1937.

- 12. Dziewoński, K., and J. Moszew, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1928A, 283.
- Ferko, P., Ber. 20, 660 1887.
- 14. Fröschl, N., and J. Harlass, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 140, Abt. IIb, 703 1931.
- 14A. Hooker, S. C., and L. F. Fieser, J. Am. Chem. Soc. 58, 1216 1936.
- Inove, H., and K. Ishimura, Bull. Chem. Soc. Japan 9, 431 1934.
- 16. Julius, P., Ber. 19, 2549 1886.
- 17. Korn, O., Ber. 17, 3019 1884.
- 18. Kursanov, D. N., A. S. Kursanova, and A. N. Blokhina, J. Gen. Chem. (U.S.S.R.) 8, 1786 1938; C. A. 33, 4979 1939.
- 19. Loevenich, J., and A. Loeser, Ber. 60, 320 **1927**.
- 20. Loevenich, J., and K. Sipmann, J. prakt. Chem. [2] 124, 127 1929.
- 21. Lossen, F., Ann. 144, 71 1867.
- 22. Mascarelli, L., Gazz. chim. ital. 58, 791 **1928**.
- 23. Meyer, H., and A. Hoffmann, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 125-126, Abt. IIb, 449 1916-17.
- 24. Müller, E., and T. Töpel, Ber. 72, **273 1939.**
- 25. Ogawa, T., Bull. Chem. Soc. Japan 6, 174 1931.
- 26. Onufrowicz, S., Ber. 23, 3355 1890.
- 27. Pestemer, M., and J. Cecelsky, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 140, Abt. IIb, 541 1931.
- 28. Roux, L., Ann. chim. phys. [6] 12, 289 1887.
- 29. Sachanen (Sakhanin), A. N., and M. D. Tilicheev, Ber. 62, 658 1929.
- 30. Sakellarios, E., and T. Kyrimis, Ber. 57, 322 1924.
- 31. Schroeter, G., Ber. 57, 1990 1924.
- 32. Smith, W., Ber. 4, 888 1871.
- 33. Smith, W., Ber. 10, 1272 1877.
- 34. Smith, W., Ber. 10, 1603 1877.
- 35. Smith, W., J. Chem. Soc. 32, 551 1877.
- 36. Smith, W., J. Chem. Soc. 35, 224 1879.
- 37. Späth, E., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 122, Abt. IIb. 1207 1913.
- 38. Ullmann, F., and J. Bielecki, Ber. 34, 2174 1901.
- 39. Vesely, V., and F. Stursa, Chem. Listy

- 26, 495 1932; C. A. 27, 717 1933; Chem. Zentr. 1933, I, 3079.
- 40. Veselý, V., and F. Štursa, Collection Czechoslov. Chem. Commun. 4, 139 1932.
- 41. Walder, H., Ber. 15, 2166 1882.
- 42. Wegscheider, R., Ber. 23, 3199 1890.
- 43. Wegscheider, R., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 89, Abt. II, 885 1884.
- 44. Weidlich, H. A., Ber. 71, 1201 1938.
- 45. Weitzenböck, R., and C. Seer, Ber. 46, 1994 **1913.**
- 46. Willgerodt, C., and P. Schlösser, Ber. **33**, 692 **1900**.
- 47. Witt, O. N., and J. Dedichen, Ber. 30, 2655 1897.

#### C21H16

#### Di-1, 1'-naphthylmethane



M. P., °C

108

1091, 5, 6, 7, 15

108-10919

106-10720

B. P., °C @ 760mm

245-250

 $D_4^{20}$ 1.0224

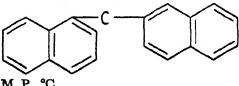
52° 19

41

 $n_{\rm p}^{20}$ 

52° 19 1.686  $n_{\mathrm{H}lpha}^{52}$  19 1.677  $n_{\mathrm{H}\beta}^{52}$  19 1.712

#### Di-1,2'-naphthylmethane



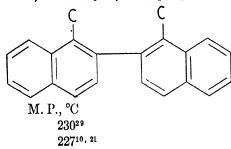
M. P., °C

9628

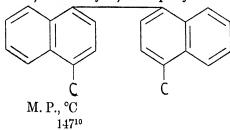
95-96

#### $C_{22}H_{18}$

# 1,1'-Dimethyl-2,2'-binaphthyl



## 4,4'-Dimethyl-1,1'-binaphthyl

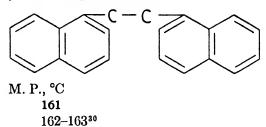


# 1,1-Di-(1'-naphthyl)-ethane

# 1,1-Di-(2'-naphthyl)-ethane

This series continued on next poge

# 1,2-Di-(1'-naphthyl)-ethane



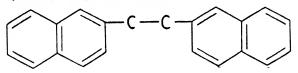
161-16222

161-161.531

1603, 27, 33

159-16011

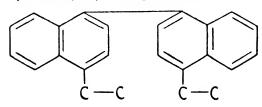
# 1,2-Di-(2'-naphthyl)-ethane



M. P., °C 182–183<sup>30</sup> 182<sup>22</sup> 181–182<sup>32</sup>

#### $\mathbf{C}_{24}\mathbf{H}_{22}$

#### 4,4'-Diethyl-1,1'-binaphthyl (a)



M. P., °C 76–77<sup>18</sup>

(a) The structure of this compound was not clearly defined in the literature.

1,4-Di-(1'-naphthyl)-butane

M. P., °C. 102<sup>16</sup> 101<sup>2</sup>

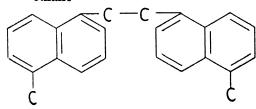
B. P., °C @ 760mm 240

316

x,x-Di-[1'-(2'-methylnaphthyl)]ethane (a)

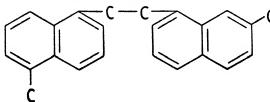
M. P., °C 177–178<sup>17</sup>

- (a) The structure of this compound was not clearly defined in the literature.
  - 1,2-Di-[1'-(5'-methylnaphthyl)]-ethane



M. P., °C 115–117<sup>26</sup>

1-[1'-(5'-Methylnaphthyl)]-2-[1''-7''methylnaphthyl)]-ethane



M. P., °C 74-75<sup>26</sup>

B. P., °C @ 760mm 192-193

 $0.1^{25}$ 

1,2-Di-[1'-(7'-methylnaphthyl)]ethane

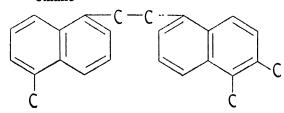
M. P., °C 122.5–123.5<sup>28</sup>

B. P., °C @ 760mm 220–225

 $0.3^{28}$ 

#### $C_{25}H_{24}$

1-[1'-(5'-Methylnaphthyl)]-2-[1''-(5'',6''-dimethylnaphthyl)]-ethane



M. P., °C 128–129<sup>26</sup>

1-[1'-(7'-Methylnaphthyl)]-2-[1''-(5'',6''-dimethylnaphthyl)]-ethane

M. P., °C 108.5–109.5<sup>28</sup>

B. P., °C @ 760mm 225–230

 $0.2^{28}$ 

# Di-1-(3,4-dimethylnaphthyl)-methane

M. P , °C 174–175<sup>17</sup>

#### $C_{26}H_{26}$

# x6-Hexamethyl-x, x'-binaphthyl (a)

M. P., °C 183 <sup>18</sup>

(a) The structure of this compound was not clearly defined in the literature.

# 3,4-Di-(1'-naphthyl)-hexane

1554

# 3,4-Di-(2'-naphthyl)-hexane

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ &$$

М. Р., °C 1984

#### 1,4-Di-(1'-naphthyl)-2,3-dimethylbutane

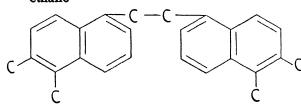
# 1,4-Di-[1'-(4'-methylnaphthyl)]-butane

 $131 - 132^2$ 

# $\begin{array}{c|c} C & C \\ \hline C & C \\ \hline \end{array}$

M. P., °C 126-127<sup>14</sup>

# 1,2-Di-[1'-(5',6'-dimethylnaphthyl)]-ethane



M. P., °C 163-165<sup>26</sup>

#### References on Alkyl Binaphthyls and Two Naphthyl Substitutions on Alkanes

- Anderson, A. R., and W. F. Short, J. Chem. Soc. 1933, 485.
- Bachmann, W. E., J. W. Cook, C. L. Hewett, and J. Iball, J. Chem. Soc. 1936, 54.
- 3. Bamberger, E., and W. Lodter, Ber. 21, 51 1888.
- Campbell, N. R., and F. W. Chattaway, Proc. Roy. Soc. (London) 130B, 435 1942.
- 5. Chichibabin, A. E., Ber. 44, 443 1911.
- Chichibabin, A. E., and O. I. Magidson,
   J. prakt. Chem. [2] 90, 168 1914.

 $C_{20}H_{14}$ 

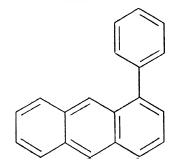
- Chichibabin, A., and O. Magidson, J. Russ. Phys. Chem. Soc. 46, 1389 1914;
   Chem. Zentr. 1915, I, 1123.
- 8. Claus, A., and W. Ruppel, J. prakt. Chem. [2] 41, 48 1890.
- Elbs, K., J. prakt. Chem. [2] 47, 41 1893.
- Fichter, F., and S. Herszbein, Helv. Chim. Acta 11, 1264 1928.
- 11. Friedmann, W., Ber. 49, 277 1916.
- 12. Friedmann, W., Ber. 49, 1352 1916.
- Froschl, N., and J. Harlass, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 140, Abt. IIb, 703 1931.
- Gamble, D. J. C., and G. A. R. Kon,
   J. Chem. Soc. 1935, 443.
- 15. Grabowski, J., Ber. 7, 1605 1874.
- Harper, S. H., G. A. R. Kon, and F. C. J. Ruzicka, J. Chem. Soc. 1934, 121.
- 17. Hewett, C. L., J. Chem. Soc. 1940, 293.
- 18. Kruber, O., Ber. 72, 1972 1939.
- Larsen, R. G., R. E. Thorpe, and F. A. Armfield, Ind. Eng. Chem. 34, 183 1942.
- 20. Magidson, O. I., Ber. 58, 433 1925.

- Mayer, F., and O. Schnecko, Ber. 56, 1408 1923.
- Mayer, F., and A. Sieglitz, Ber. 55, 1835 1922.
- Migita, M., Bull. Chem. Soc. Japan 8, 22 1933.
- 24. Richter, M., Ear. 13, 1728 1880.
- Ruzicka, L, and K. Hofmann, Helv. Chim. Acta 20, 1155 1937.
- Ruzicka, L., and K. Hofmann, Helv. Chim. Acta 22, 126 1939.
- Ruzicka, L., and H. Hosli, Helv. Chim. Acta 17, 470 1934.
- Ruzicka, L., and E. Morgeli, Helv. Chim. Acta 19, 377 1936.
- Scholl, R., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 127, Abt. Hb, 33 1918.
- 30. Shorygin, Ber. 59, 2510 1926.
- 31. Simpson, J. C. E., J. Chem. Soc. 1943, 447.
- Szperl, L, Roczniki Chem. 1923, II, 291; C. A. 18, 1290 1924.
- 33. Weitzenbock, R., and H. Lieb, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 121, Abt. IIb, 225 1912.

# 3. ANTHRACENE AND PHENANTHRENE WITH ONE PHENYL SUBSTITUTION, C<sub>n</sub>II<sub>2n-26</sub>

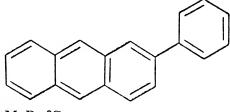
#### C20H14

#### 1-Phenylanthracene



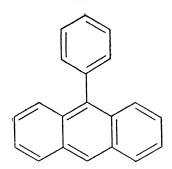
M. P., °C 123<sup>44</sup> 110–112<sup>20, 22</sup>

#### 2-Phenylanthracene



M. P., °C 207-207.5<sup>22</sup>

#### 9-Phenylanthracene



M. P., °C

154.5

155-157<sup>4</sup>

156<sup>27, 36, 41</sup>

155<sup>26</sup>

152-153<sup>40, 42</sup>

152.8<sup>35</sup>

#### x-Phenylanthracene (a)

- B. P., °C @ 760mm 417<sup>29</sup>
- (a) The structure of this compound was not clearly defined in the literature.

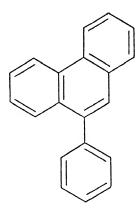
# 1-Phenylphenanthrene

M. P., °C 79–79.5²

# 2-Phenylphenanthrene

M. P., °C 196.6–197.237

## 9-Phenylphenanthrene



M. P., °C 113<sup>43</sup> 110<sup>13</sup> 105-106<sup>28, 32</sup> 104-105<sup>18</sup>

 $\mathbf{C}_{21}\mathbf{H}_{16}$ 

#### 9-Benzylanthracene

# 2-Methyl-9-phenylanthracene

M. P., °C 119<sup>31, 39</sup>

# 9-Methyl-10-phenylanthracene

M. P., °C 116<sup>38</sup> 113.5–114.5<sup>15</sup> 112.5–113.5<sup>19</sup> 113<sup>12</sup> 112<sup>11</sup>

# 9-Benzylphenanthrene

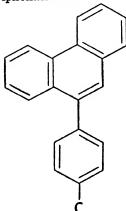
M. P., °C 155-156<sup>30</sup> 154<sup>24</sup> 153-154<sup>1, 2</sup>

# x-Benzylphenanthrene (a)

M. P., °C 107-108² 105-106² 95-96² 91-92⁴⁵ 79-80²

(a) The structures of these compounds were not clearly defined in the literature.

# 9-p-Tolylphenanthrene



M. P., °C 90–91<sup>18</sup>

#### 9-Methyl-10-phenylphenanthrene

M. P., °C 99–100<sup>17</sup>

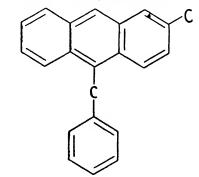
 $C_{22}H_{18}$ 

#### 9-Ethyl-10-phenylanthracene

M. P., °C 110<sup>5</sup> 107–108.5<sup>19</sup>

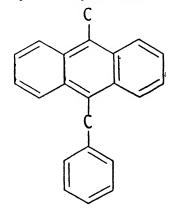
#### 2-Methyl-9-benzylanthracene

#### 2-Methyl-10-benzylanthracene



M. P., °C 1016

#### 9-Methyl-10-benzylanthracene



M. P., °C 167.8–168.6<sup>15</sup>

# x-Methyl-9-tolylanthracene (a)

M. P., °C 191<sup>84</sup>

(a) The structure of this compound was not clearly defined in the literature.

# 1-Phenethylphenanthrene

M. P., °C 86.5–89.5<sup>25</sup>

#### 9-Phenethylphenanthrene

M. P., °C 81.5<sup>13</sup>

# ${\bf 9-Ethyl-10-phenylphen} anthrene$

M. P., °C 161<sup>17</sup>  $C_{23}H_{20}$ 

9-n-Propyl-10-phenylanthracene

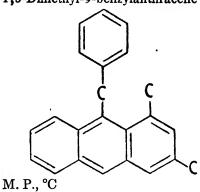
M. P., °C 115–116<sup>5</sup>

# 9-Isopropyl-10-phenylanthracene

M. P., °C 166–167<sup>5</sup>

1268

#### 1,3-Dimethyl-9-benzylanthracene



# 1,3-Dimethyl-10-benzylanthracene

M. P., °C 1498

#### 1,4-Dimethyl-9-benzylanthracene

M. P., °C 1359

#### 2,3-Dimethyl-9-benzylanthracene

# 1,3,9-Trimethyl-10-phenylanthracene

M. P., °C 1658

# 1,3,10-Trimethyl-9-phenylanthracene

M. P., °C 1218

# 2,3,9-Trimethyl-10-phenylanthracene

M. P., °C 119<sup>10</sup> 331

# 9-n-Propyl-10-phenylphenanthrene

M. P., °C 148.5–149.5<sup>16</sup>

#### $C_{24}H_{22}$

## 9-Butyl-10-phenylanthracene

M. P., °C 156<sup>5</sup>

# $\mathbf{C}_{25}\mathbf{H}_{24}$

# 2-Phenyl-9-pentylanthracene

M. P., °C 203-204<sup>5</sup>

#### 2,3,6,7-Tetramethyl-9-benzylanthracene

# 1-(2'-Isopropyl-5'-methylbenzyl)phenanthrene

M. P., °C 115-116<sup>23</sup>

# $\boldsymbol{C_{30}H_{34}}$

#### 9-Phenyl-10-n-decylphenanthrene

M. P., °C 99–100<sup>16</sup>

- References on Anthracene and Phenanthrene with One Phenyl Substitution
- Bachmann, W. E., J. Am. Chem. Soc. 56, 1363 1934.
- Bachmann, W. E., J. Am. Chem. Soc. 57, 555 1935.
- Bachmann, W. E., and A. L. Wilds, J. Am. Chem. Soc. 60, 624 1938.
- Barnett, E. de B., J. W. Cook, and I. G. Nixon, J. Chem. Soc. 1927, 504.
- Barnett, E. de B., J. W. Cook, and J. L. Wiltshire, J. Chem. Soc. 1927, 1724.
- Barnett, E. de B., and N. F. Goodway, J. Chem. Soc. 1929, 1754.
- Barnett, E. de B., N. F. Goodway, and J. W. Watson, Ber. 66, 1876 1933.
- 8. Barnett, E. de B., and C. L. Hewett, Ber. **64**, 1572 **1931**.
- 9. Barnett, E. de B., and J. A. Low, Ber. 64, 49 1931.
- Barnett, E. de B., and F. C. Marrison, Ber. 64, 535 1931.
- 11. Barnett, E. de B., and M. A. Mathews, Ber. **59**, 1429 **1926**.
- 12. Bergmann, E., Ber. 63, 1037 1930.
- Bergmann, E., and F. Bergmann, J. Am. Chem. Soc. 59, 1443 1937.
- Bergmann, E., and S. Fujise, Ann. 480, 188 1930.
- Berliner, E., J. Am. Chem. Soc. 66, 533 1944.
- Bradsher, C. K., and S. T. Amore, J. Am. Chem. Soc. 66, 1280 1944.
- Bradsher, C. K., and R. Rosher, J. Am. Chem. Soc. 61, 1524 1939.
- Bradsher, C. K., and A. K. Schneider,
   J. Am. Chem. Soc. 60, 2960 1938.
- Bradsher, C. K., and E. S. Smith, J. Am. Chem. Soc. 65, 451 1943.
- 20. Charrier, G., and E. Ghigi, Ber. 69, 2211 1936.
- 21. Cook, J. W., J. Chem. Soc. 1926, 2160.
- 22. Cook, J. W., J. Chem. Soc. 1930, 1087.
- Cook, J. W., C. L. Hewett, W. V. Mayneord, and E. Roe, J. Chem. Soc. 1934, 1727.

- 24. Dilthey, W., S. Henkels, and M. Leonhard, J. prakt. Chem. [2] 151, 97 1938.
- 25. Drake, N. L., and W. C. McVey, J. Org. Chem. 4, 464 1939.
- Dufraisse, C., and D. Daniel, Bull. soc. chim. [5] 4, 2063 1937.
- Dufraisse, C., L. Velluz, and Mme. L. Velluz, Bull. soc. chim. [5] 4, 1260 1937.
- Eschinazi, H. E., and F. Bergmann,
   J. Am. Chem. Soc. 65, 1411 1943.
- Friedel, C., and J. M. Crafts, Ann. chim. phys. [6] 1, 449 1884.
- Goldschmiedt, G., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 84, Abt. IIb, 305 1881.
- 31. Hemilian, W., Ber. 16, 2360 1883.
- 32. Koelsch, C. F., J. Am. Chem. Soc. 56, 480 1934.
- Krollpfeiffer, F., and F. Branscheid, Ber. 56, 1617 1923.
- 34. Limpricht, II., Ann. 299, 286 1898.
- 35. Linebarger, C. E., Am. Chem. J. 13, 554 1891.
- McVicker, W. H., J. K. Marsh, and A. W. Stewart, J. Chem. Soc. 127, 999 1925.
- 37. Newman, M. S., J. Org. Chem. 9, 518 1944.
- 38. Schlenk, W., and E. Bergmann, Ann. 463, 1 1928.
- Scholl, R., H. Delmert, and L. Wanka,
   Ann. 493, 56 1932.
- 40. Scholl, R., and J. Donat, Ann. 512, 1 1934.
- 41. Velluz, L., Compt. rend. 206, 1514 1938.
- von Baeyer, A., and A. Schillinger, Ann.
   202, 54 1880.
- Weizmann, C., E. Bergmann, and T. Berlin, J. Am. Chem. Soc. 60, 1331 1938.
- 44. Weizmann, C., E. Bergmann, and L. Haskelberg, J. Chem. Soc. 1939, 391.
- 45. Willgerodt, C., and B. Albert, J. prakt. Chem, [2] 84, 383 1911.

#### 4. CHOLANTHRENE AND ITS ALKYL DERIVATIVES, C<sub>n</sub>H<sub>2n-26</sub>

#### $C_{20}H_{14}$

#### Cholanthrene

M. P., °C 172

173.0-173.525

 $173 - 173.5^{20}$ 

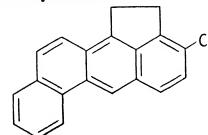
170-17111

170.1-170.612

168.5-1702. 10

#### $C_{21}H_{16}$

#### 3-Methylcholanthrene



M. P., °C

178.9

180.3-180.61

179.5-18015

 $178.5 - 179.5^{21}$ 

 $178.5 - 179^{16, 18, 25}$ 

 $178 - 179^{24}$ 

177-17816

176.5-177.5°

175.3-177.112

176-17717

175-1774

 $175.5^{14}$ 

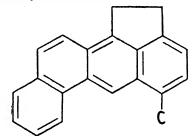
#### 4-Methylcholanthrene

M. P., ℃

 $154.5 - 155^3$ 

153-15413

#### 5-Methylcholanthrene



M. P., °C

160-161.5 (a)<sup>3</sup>

(a) This compound remelts at 164-165.

#### 11-Methylcholanthrene

M. P., ℃

204.2-205.213

# x-Methylcholanthrene (a)

 $D_4^{20}$ 

 $1.277^{23}$ 

(a) The structure of this compound was not clearly defined in the literature.

 $C_{22}H_{18}$ 

#### 3-Ethylcholanthrene

M. P., °C 179.5–180<sup>5, 7</sup>

#### 1,3-Dimethylcholanthrene

M. P., °C 179–180<sup>6</sup>

## 2,3-Dimethylcholanthrene

M. P., °C 169–170<sup>19</sup>, <sup>26</sup>

#### 3,11-Dimethylcholanthrene

M. P., °C 175.8-176.5<sup>18</sup>

# 4,11-Dimethylcholanthrene

M. P., °C 161.7–162.4<sup>13</sup>

#### $C_{23}H_{20}$

#### 3-Isopropylcholanthrene

M. P., °C 188–189<sup>8</sup>

#### $C_{24}H_{22}$

#### 3-tert-Butylcholanthrene

M. P., °C 204–205<sup>22</sup>

#### References on Cholanthrene and Its Alkyl Derivatives

- 1. Bachmann, W. E., J. Org. Chem. 1, 347 1936-37.
- Bachmann, W. E, J. Org. Chem. 3, 434 1938-39.
- Bachmann, W. E., and J. M. Chemerda,
   J. Org. Chem. 6, 50 1941.
- Bergmann, E , and O. Blum-Bergmann,
   J. Am. Chem. Soc. 59, 1573 1937.
- Bruce, W. F., through L. F. Fieser and D. K. Snow, J. Am. Chem. Soc. 60, 176 1938.
- Bruce, W. F., and L. F. Fieser, J. Am. Chem. Soc. 59, 479 1937.
- Bruce, W. F., and S. J. Kahn, J. Am. Chem. Soc. 60, 1017 1938.
- 8. Bruce, W. F., and F. Todd, J. Am. Chem. Soc. **61**, 157 **1939**.

- Cook, J. W., and G. A. D. Haslewood, J. Chem. Soc. 1934, 428.
- Cook, J. W., and G. A. D. Haslewood, J. Chem. Soc. 1935, 770.
- Cook, J. W., G. A. D. Haslewood, and A. M. Robinson, J. Chem. Soc. 1935, 665.
- Davis, W. W., M. E. Krahl, and G. H. A. Clowes, J. Am. Chem. Soc. 62, 3080 1940.
- Fieser, L. F., and D. M. Bowen, J. Am. Chem. Soc. 62, 2103 1940.
- Fieser, L. F., and V. Desreuv, J. Am. Chem. Soc. 60, 2255 1938.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 60, 940 1938.
- Fieser, L. F., and M. Newman, J. Am. Chem. Soc. 57, 961 1935.
- Fieser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 57, 228 1935.
- Fieser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 57, 942 1935.
- Fieser, L. F., and A. Seligman, J. Am. Chem. Soc. 57, 1377 1935.
- Fieser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 57, 2174 1935.
- Fieser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 58, 2182 1936.
- Fieser, L. F., and D. K. Snow, J. Am. Chem. Soc. 60, 176 1938.
- Iball, J., Z. Krist. 94, 7 1936; Chem. Zentr. 1936, II, 3293.
- 24. Sannié, C., Bull. soc chim. [5] **5, 260 1938.**
- Shear, M. J., Am. J. Cancer 26, 322 1936.
- Shear, M. J., Am. J. Cancer 28, 334
   1936.

5. OTHER BENZOCYCLANOANTHRACENES, BENZOCYCLANOPHENANTHRENES, AND THEIR ALKYL DERIVATIVES,  $C_nH_{2n-26}$ 

#### $C_{19}H_{12}$

Benzo-[b]-cyclopentano-[def]-phenanthrene

M. P., °C 122.7–123.1<sup>6</sup> 122.5–123<sup>7</sup> 120–121<sup>8</sup>

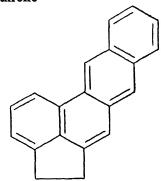
#### $C_{20}H_{14}$

Benzo-[a]-cyclopentano-[de]-anthracene

M. P., °C 138.5–140<sup>12</sup> 137–138<sup>18</sup> 133–134<sup>17</sup> 130<sup>4</sup> Benzo-[a]-cyclopentano-[kl]-anthracene

M. P., °C 176.5–177 <sup>10, 18</sup>

Benzo-[b]-cyclopentano-[jk]-phenanthrene



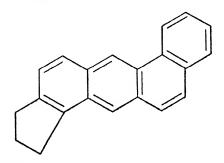
M. P., °C 233.1–234.3<sup>5</sup> 192.5–193.5<sup>13</sup> 192<sup>1</sup>

Benzo-[b]-cyclopentano-[def]-1methylphenanthrene

M. P., °C 181–181.47

 $C_{21}H_{16}$ 

# 1,2-Benzo-5,6-cyclopentanoanthracene



M. P., °C 199-200<sup>2</sup>. <sup>3</sup>

 $D_4^{20}$ 

1.255

18° 14

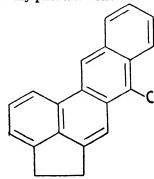
#### 1,2-Benzo-6,7-cyclopentanoanthracene

M. P., °C 164–165³

# Benzo-[a]-cyclopentano-[kl]-7-methyl-anthracene

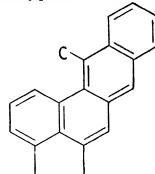
M. P., °C 187.5% 18

# Benzo-[b]-cyclopentano-[jk]-1-methylphenanthrene



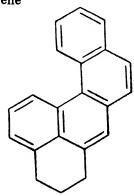
M. P., °C 181.7–182.5<sup>16</sup>

#### Benzo-[b]-cyclopentano-[jk]-4methylphenanthrene



M. P., °C 133–134<sup>15</sup>

# $\label{eq:benzo-continuous} Benzo-[c]-cyclohexano-[jk]-phenanthrene$



M. P., °C 138-138.5<sup>11</sup>

#### $C_{22}H_{18}$

Benzo-[b]-cyclopentano-[jk]-1-ethyl-phenanthrene

M. P., °C 174.5–175<sup>16</sup>

# Benzo-[b]-cyclopentano-[jk]-1,4-dimethylphenanthrene

M. P., °C 164-164.5<sup>15</sup>

 $C_{23}H_{20}$ 

Benzo-[b]-cyclopentano-[jk]-1-n-propylphenanthrene

M. P., °C 165.5-166.5<sup>16</sup>

Benzo-[b]-cyclopentano-[jk]-1-ethyl-4-methylphenanthrene

M. P., °C 157.5–158.5<sup>15</sup>  $C_{24}H_{22}$ 

#### $C_{24}H_{22}$

#### Benzo-[b]-cyclopentano-[jk]-1-nbutylphenanthrene

M. P., °C 128-129<sup>16</sup>

#### Benzo-[b]-cyclopentano-[jk]-1-propyl-4-methylphenanthrene

M. P., °C 119.5–120.5<sup>15</sup>

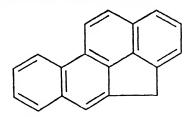
#### References on Other Benzocyclanoanthracencs, Benzocyclanophenanthrenes, and Their Alkyl Derivatives

- 1. Cook, J. W., J. Chem. Soc. 1930, 1087.
- 2. Cook, J. W., J. Chem. Soc. 1931, 499.
- 3. Cook, J. W., J. Chem. Soc. 1931, 2529.
- 4. Dansi, A., Gazz. chim. ital. 67, 85 1937.
- Davis, W. W., M. E. Krahl, and G. H. A. Clowes, J. Am. Chem. Soc. 62, 3080 1940.
- Fieser, L. F., and J. Cason, J. Am. Chem. Soc. 61, 1740 1939.
- Fieser, L. F., and J. Cason, J. Am. Chem. Soc. 62, 432 1940.
- Fieser, L. F., and J. Cason, J. Am. Chem. Soc. 62, 1293 1940.
- Ficser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 57, 942 1935.
- Fieser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 57, 2174 1935.
- Ficser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 58, 478 1936.
- Fieser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 59, 883 1937.
- Geyer, B. P., and S. Zuffanti, J. Am. Chem. Soc. 57, 1787 1935.
- Iball, J, Z. Krist. 94, 7 1936; C. A. 30, 7412 1936; Chem. Zentr. 1936, II, 3293.
- Mikhallov, B. M., and A. N. Blokhina,
   J. Gen. Chem. (U.S.S.R.) 12, 283
   1942; C. A. 37, 3085 1943.
- Miklallov, B. M., and A. N. Blokhina,
   J. Gen. Chem. (U.S.S.R.) 13, 609
   1943; Survey For. Petrol. Liter. 473B,
   Oct. 20, 1944.
- Morelli, E., and A. Dansi, Biochim. e terap. sper. 24, 432 1937.
- 18. Shear, M. J., Am. J. Cancer 28, 334 1936.

# 6. MISCELLANEOUS POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-26}$

 $C_{19}H_{12}$ 

# Cyclopentano-[def]-chrysene



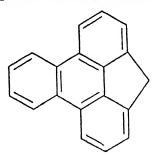
M. P., °C

172.4-172.922

 $172 - 172.5^{22}$ 

 $171.5 - 172.3^{22}$ 

#### Cyclopentano-[def]-triphenylene

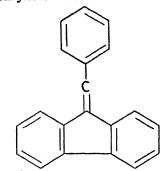


M. P., °C 115-116<sup>37</sup>

# Benzo-[cd]-5-hydropyrene

M. P., °C 13549

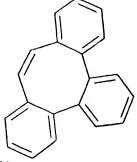
# $\begin{array}{c} C_{20}H_{14} \\ \text{9-Benzylidenefluorene} \end{array}$



M. P., °C 153-154 (a)<sup>42</sup> 76 (b)

- (a) This constant was determined on an isomeric form of the compound.
- (b) The melting point 76 is found in references 32, 34, 38, 41, 46, 47, 48.

#### 1,2,3,4,5,6-Tribenzocyclooctene-7



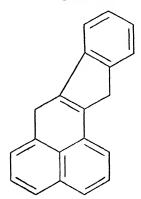
M. P., °C 138.5–139<sup>44</sup>

# 2,3-(2',3'-Indo)-fluorene

(Ellagene)

M. P., °C 19738

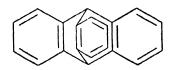
#### 2,3-(3',2'-Indo)-phenalene



M. P., °C 185<sup>11</sup>

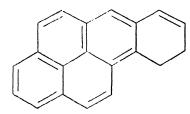
#### 9,10-Endo-o-phenylene-9,10-dihydroanthracene

(Tryptycene)



M. P., °C 254.8-255.2<sup>5</sup>

#### 2,3-(Cyclohexen-3'-o)-pyrene (a)

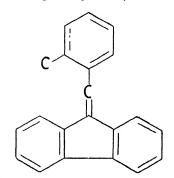


M. P., °C 149.5-150.0<sup>1</sup> 148-149<sup>21</sup>

(a) Reference 31 gives evidence that this compound may possibly be cyclohexano-[def]-chrysene.

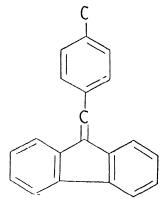
 $C_{21}H_{16}$ 

9-(2'-Methylbenzylidene)-fluorene



M. P., °C 109.5<sup>15</sup>

#### 9-(4'-Methylbenzylidene)-fluorene



M. P., °C 97.5<sup>15</sup>

#### 1,2-Benzo-9-(propen-2'-yl)-anthracene

$$c=c-c$$

M. P., °C 115-116<sup>26</sup> 1,2-Benzo-10-(propen-2'-yl)-anthracene

M. P., °C 125.5–126.5<sup>25</sup>

#### 1,2-Benzo-5-isopropenylanthracene

$$C-C=C$$

M. P., °C 137–139<sup>14</sup>

# 6-Isopropenylchrysene

$$C = C - C$$

M. P., °C 1617

B. P., °C @ 760mm 220

27

1-Methyl-2,3-(2',3'-indo)-fluorene (Isophthalacene)

M. P., °C 22220

1-Methyl-3,4-(3',2'-indo)-fluorene (Phthalacene)

M. P., °C 17330

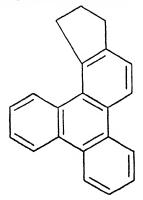
Spiro[3,4-benzocyclopentane-1,9'-fluorene]

M. P., °C 220<sup>21</sup>

## Cyclopentano-[hi]-5-methylchrysene

M. P., °C 169–169.5<sup>2</sup>

# 1,2-Cyclopentanotriphenylene

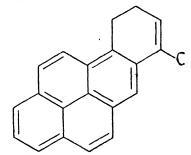


M. P., °C 171-172.5<sup>6</sup> 115-116<sup>52</sup>

# Cyclohexano-[hi]-chrysene

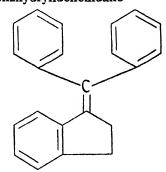
M. P., °C 116.5-117<sup>29</sup>, 43

#### 1,2-(6'-Methylcyclohexen-5'-o)pyrene



M. P., °C 162–163<sup>27</sup> 155–156<sup>23</sup>

# 1-Benzhydrylideneindane



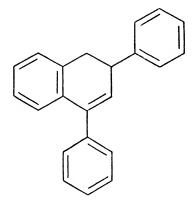
 $C_{22}H_{18}$ 

M. P., °C 9216

#### 4-o-Biphenylyl-1,2-dihydronaphthalene

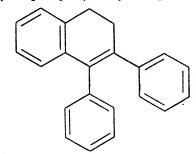
M. P., °C 75.5–76.5°

#### 2,4-Diphenyl-1,2-dihydronaphthalene



M. P., °C 136<sup>12</sup>

# 3,4-Diphenyl-1,2-dihydronaphthalene



M. P., °C  $94-95^8$   $76.5-77^{12}$   $76^{13}$ 

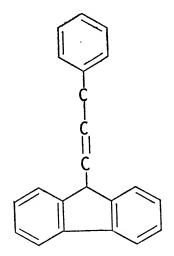
B. P., °C @ 760mm 210–215

 $0.5^{8}$ 

# Di-1-(3,4-dihydronaphthyl)-ethyne

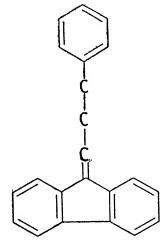
M. P., °C 124<sup>17</sup> 120–121<sup>40</sup>

# 1-(9'-Fluoryl)-3-phenylpropene-1



M. P., °C 88<sup>33, 47</sup>

# 1-Phenyl-3-(9'-fluorylidene)-propane



M. P., °C 81-82<sup>47</sup> 81<sup>33</sup>

# Cyclopentano-[jk]-1-phenyl-3,4-dihy-drophenanthrene

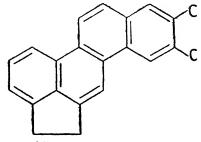
M. P., °C 144<sup>28</sup>

#### 6-(2'-Buten-2'-yl)-chrysene

$$c-c=c-c$$

M. P., °C 159–160°

# Cyclopentano-[hi]-2,3-dimethyl-chrysene

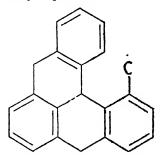


M. P., °C 222.6-223.1<sup>24</sup>

# 5,7,12,14-Tetrahydropentacene

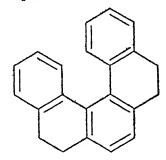
M. P., °C 244-245<sup>39</sup> 240-241<sup>36</sup>

# 2,3-Benzo-4,5-(3'-methylbenzo)-3a, 6-dihydrophenalene



M. P., °C 182<sup>51</sup>

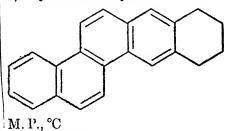
# 3,4,5,6-Dibenzo-1,2,7,8-tetrahy-drophenanthrene



M. P., °C 142<sup>50</sup>

## 2,3-Cyclohexanochrysene

217-21815



Benzo-[hij]-2a,3,4,5-tetrahydrocholanthrene

M. P., °C 227.5-228.5<sup>35</sup>

 $C_{23}H_{20}$ 

1-Methyl-4-o-biphenylyl-1, 2-dihydronaphthalene

B. P., °C @ 760mm 215-218

 $6-7^{10}$ 

2-Methyl-4-o-biphenylyl-1,2-dihydronaphthalene

M. P., °C 77–78<sup>10</sup>

B. P., °C @ 760mm 218-220

8-910

1,4-Dimethyl-9-methylene-10-phenyl-9,10-dihydroanthracene

M. P., °C 1294

6-(2'-Penten-2'-yl)-chrysene

$$C-C=C-C-C$$

M. P., °C 102<sup>7</sup>

B. P., °C @ 760mm 210-215

 $0.1^{7}$ 

5,6-(6'-Methylcyclohexano)-chrysene

M. P., °C 1397

#### $C_{24}H_{22}$

1,2-Dimethyl-4-o-biphenylyl-1,2-dihydrenaphthalene

M. P., °C 78-79.5<sup>10</sup>

B. P., °C @ 760mm 217-218

810

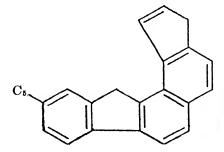
### 6-(2'-Hexen-2'-yl)-chrysene

B. P., °C @ 760mm 230-235

 $0.1^{7}$ 

#### $C_{25}H_{24}$

1,2-(4',5'-Indo)-7-pentylfluorene (Diels' Hydrocarbon II) (a)

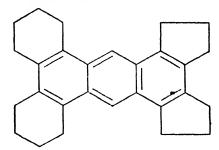


M. P., °C 220<sup>18</sup> 219<sup>19</sup>

(a) Although the above structure was assigned by Diels, it is not generally accepted as definite.

#### C28H30

1,2,3,4-Dicyclopentano-5,6,7,8-dicyclohexanoanthracene



M. P., °C 342-3443

References on Miscellancous Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-26}$ 

- Bachmann, W. E., and M. Carmack, J. Am. Chem. Soc. 63, 1685 1941.
- Bachmann, W. E., and S. R. Safir, J. Am. Chem. Soc. 63, 2601 1941.
- Backer, H. J., and L. H. H. Huisman, Rec. trav. chim. 60, 557 1941; C. A. 37, 5982 1943.
- Barnett, E. de B., and J. A. Low, Ber. 64, 49 1931.
- Bartlett, P. D., M. J. Ryan, and S. G. Cohen, J. Am. Chem. Soc. 64, 2649 1942.
- Bergmann, E., and O. Blum-Bergmann,
   J. Am. Chem. Soc. 58, 1678 1936.
- Bergmann, F., and H. E. Eschinazi, J. Am. Chem. Soc. 65, 1413 1943.
- Bergmann, F., H. E. Eschinazi, and D. Schapiro, J. Am. Chem. Soc. 64, 557 1942.
- Bradsher, C. K., and L. Rapoport, J. Am. Chem. Soc. 65, 1646 1943.
- Bradsher, C. K., and L. Rapoport, J.
   Am. Chem. Soc. 66, 1281 1944.

- Buu-Hoï and P. Cagniant, Compt. rend. 216, 346 1943; C. A. 38, 3977 1944.
- Crawford, H. M., J Am Chem. Soc. 61, 608 1939.
- Crawford, H. M., J. Am. Chem. Soc. 64, 727 1942.
- Cook, J. W., and C. G. M. DeWorms, J. Chem. Soc. 1939, 268.
- Cook, J. W., and W. Graham, J. Chem. Soc. 1944, 329.
- 16. Courtot, C., Ann. chim. [9] 5, 52 1916.
- Dane, E., O. Hoss, A. W. Bindseil, and J. Schmitt, Ann. 532, 39 1937.
- Diels, O., W. Gadke, and P. Kording, Ann. 459, 1 1927.
- Diels, O., and Λ. Karstens, Λnn. 478, 129 1930.
- Errera, G., Gazz. chim. ital. 38, II, 588 1908.
- 21. Fecht, H., Ber. 40, 3883 1907.
- Fieser, L. F., and J. Cason, J. Am. Chem. Soc. 62, 1293 1940.
- Fieser, L. F., and M. Fieser, J. Am. Chem. Soc. 57, 782 1935.
- Fieser, L. F., M. Fieser, and E. B. Hershberg, J. Am. Chem. Soc. 58, 1463 1936.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 59, 1028 1937.
- Fieser, L. F., and E. B. Hershberg,
   J. Am. Chem. Soc. 62, 49 1940.
- Fieser, L. F., and M. S. Newman, J. Am. Chem. Soc. 57, 1602 1935.
- Fieser, L. F., and M. A. Peters, J. Am. Chem. Soc. 54, 4373 1932.
- Fieser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 58, 478 1936.
- 30. Gabriel, S., Ber. 17, 1389 1884.
- Kon, G. A. R., and E. M. F. Roe, J. Chem. Soc. 1945, 143.
- Kovache, A., Ann. chim. [9] 10, 184
   1918.

- Kuhn, R., and A. Winterstein, Helv. Chim. Acta 11, 123 1928.
- Lettre, H., H. Barnbeck, and W. Lege, Ber. 69, 1151 1936.
- Mikhailov, B. M., and N. G. Chernova,
   J. Gen. Chem. (U.S.S.R.) 12, 276
   1942; C. A. 37, 3083 1943.
- Mills, W. H., and M. Mills, J. Chem. Soc. 101, 2194 1912.
- Mohler, H., and J. Sorge, Helv. Chim. Acta 22, 229 1939.
- Nierenstein, M., and C. W. Webster, J. Am. Chem. Soc. 67, 691 1945.
- Philippi, E., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 123, Abt. IIb, 21 1914.
- Pinkney, P. S., and C. S. Marvel, J. Am. Chem. Soc. 59, 2669 1937.
- 41. Reddelien, G., J. prakt. Chem. [2] 91, 213 1915.
- 42. Schlenk, W., and E. Bergmann, Ber. 62, 745 1929.
- 43. Shear, M. J., Am. J. Cancer 28, 334 1936.
- Shuttleworth, R. G., W. S. Rapson, and E. T. Stewart, J. Chem. Soc. 1944, 71.
- 45. Sieglitz, A, Ber. 52, 1513 1919.
- 46. Thiele, J, Ber. 33, 851 1900.
- 47. Thicle, J., and F. Heule, Ann. 347, 290 1906.
- 48. Ullmann, F., and R. von Wurstemberger, Ber. 38, 4105 1905.
- Vollmann, H., H. Becker, M. Corell, and H. Streeck, Ann. 531, 1 1937.
- 50. Weidlich, H. A., Ber. 71, 1203 1938.
- Weiss, R., and F. Müller, Monatsh.
   65, 129 1934.
- Weizmann, C., and E. Bergmann, Scripta Academica Hierosolymitana (Jerusalem) Sci. Report No 1 1938.

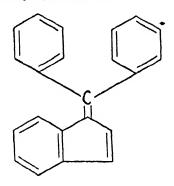
### XI. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA C<sub>n</sub>H<sub>2n-28</sub>

- 1. Indene Derivatives of Empirical Formula  $C_nH_{2n-28}$
- 2. Naphthalene with Two Phenyl or One Biphenylyl Substitutions
- 3. Benzopyrenes and Their Alkyl Derivatives
- 4. Miscellaneous Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-28</sub>

# 1. INDENE DERIVATIVES OF EMPIRICAL FORMULA C,H2n-28

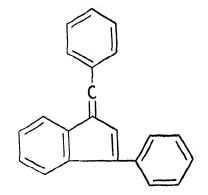
### $C_{22}H_{16}$

# 1-Benzhydrylideneindene



M. P., °C 114.5<sup>7</sup> 111-112<sup>2</sup>. <sup>4</sup>

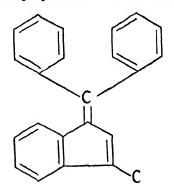
# 1-Benzylidene-3-phenylindene



M. P., °C 77.5⁵

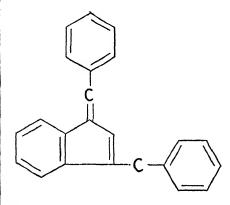
# $\mathbf{C_{23}H_{18}}$

# 1-Benzhydrylidene-3-methylindene



M. P., °C 120-121°

# 1-Benzylidene-3-benzylindene



M. P., °C 137–138³ 137.5° 137–137.5°

 $C_{24}H_{20}$ 

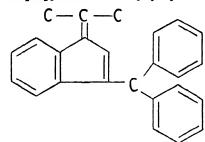
1-Benzylidene-3-(4'-methylbenzyl)-indene

M. P., °C 106–107<sup>1</sup>

1-(4'-Methylbenzylidene)-3-benzylindene

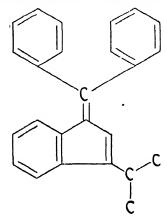
M. P., °C 98–100¹ 94\*  $C_{25}H_{22}$ 

1-Isopropylidene-3-benzyhydrylindene



M. P., °C 174–175°

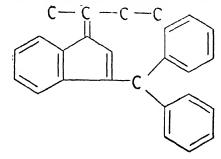
1-Benzhydrylidene-3-isopropylindene



M. P., °C 127-128°

 $C_{26}H_{24}$ 

 $1\hbox{-}sec\hbox{-}Butylidene\hbox{-}3\hbox{-}benzhydrylindene$ 



M. P., °C 149.5–151°

# 1-Benzhydrylidene-3-sec-butylindene

$$\begin{array}{c|c} & & \\ & &$$

M. P., °C 91–929

# 1-Benzylidene-3-(4'-isopropylbenzyl)-indene

 $\substack{\text{M. P., °C} \\ 105-107^1}$ 

### 1-(4'-Isopropylbenzylidene)-3benzylindene

$$c - c c$$

M. P., °C 93–94¹

References on Indene Derivatives of Empirical Formula  $C_nH_{2n-23}$ 

- 1. Bernthsen, W., Ann. 415, 274 1918.
- 2. Courtot, C., Ann. chim. [9] 4, 168 1915.
- 3. Courtot, C., Ann. chim. [9] 5, 52 1916.
- Grignard, V., and C. Courtot, Compt. rend. 152, 272 1911.
- Mayer, F., A. Sieglitz, and W. Ludwig, Ber. 54, 1397 1921.
- Thiele, J., and A. Bühner, Ann. 347, 249 1906.
- Thiele, J., and K. Merck, Ann. 415, 257 1918.
- Wislicenus, W., and W. Hentrich, Ann. 436, 9 1924.
- 9. Wuest, H.-M., Ann. 415, 291 1918.

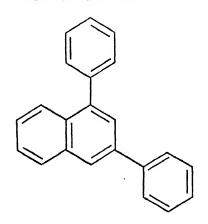
# 2. NAPHTHALENE WITH TWO PHENYL OR ONE BIPHENYLYL SUBSTITUTIONS, $C_nII_{2n-28}$

### $C_{22}H_{16}$

# 1,2-Diphenylnaphthalene

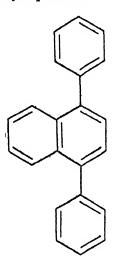
M. P., °C 114<sup>5</sup> 109.5–110<sup>11</sup>

### 1,3-Diphenylnaphthalene



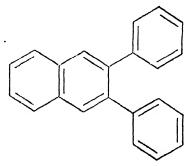
M. P., °C 70-71<sup>11</sup>

### 1,4-Diphenylnaphthalene



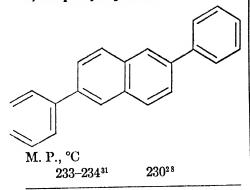
M. P., °C 135–137<sup>36</sup> 135–136<sup>13, 14</sup> 134–136<sup>1</sup> 133<sup>31</sup> 132–133<sup>30</sup>

# 2,3-Diphenylnaphthalene



M. P., °C 86–87<sup>11</sup>

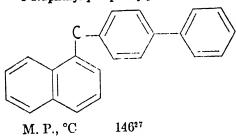
# 2, 6-Diphenylnaphthalene



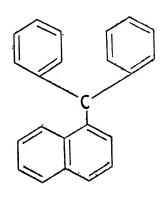
### 2.7-Diphenylnaphthalene

 $C_{23}H_{18}$ 

# 1-Naphthyl-p-biphenylylmethane



### 1-Benzhydrylnaphthalene



150 152<sup>25</sup> 150.5<sup>33</sup>, <sup>34</sup> 150 (a) 149.5<sup>22</sup> 149 (b)<sup>6</sup>, <sup>7</sup>, <sup>8</sup>, <sup>23</sup> 148–149<sup>10</sup> 134 (b)<sup>6</sup>, <sup>7</sup>, <sup>8</sup>, <sup>23</sup>

M. P., °C

 $D_4^{20}$ 

1.190

()° 37

- (a) The melting point 150 is found in references 1, 2, 3, 20, 24, 32, 38.
- (b) These constants were determined on different crystalline forms.

# 2-Naphthyl-p-biphenylylmethane

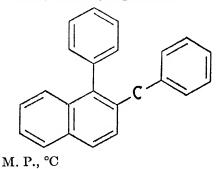
M. P., °C

 $111^{26}$ 

### 2-Benzhydrylnaphthalene

M. P., °C 77-78<sup>19</sup> 73-74<sup>25</sup>

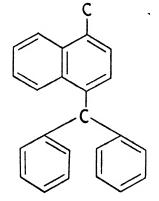
### 1-Phenyl-2-benzylnaphthalene



87–88°

 $\boldsymbol{C_{24}H_{20}}$ 

### 1-Methyl-4-benzhydrylnaphthalene

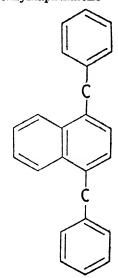


M. P., °C 149<sup>6</sup>

# Phenyl-p-tolyl-1-naphthylmethane

# 1, 4-Dibenzylnaphthalene

13312



M. P., °C 88<sup>15, 16, 17</sup>

This series continued on next page

### 2, 6-Dibenzylnaphthalene

### 1,8-Dibenzylnaphthalene

M. P.,  $^{\circ}$ C 146.5 $^{15, 16, 35}$ 

# 2,7-Dibenzylnaphthalene

# $\boldsymbol{x}$ , $\boldsymbol{x}$ -Dibenzylnaph thalene (a)

M. P., °C 132<sup>15</sup>

B. P., °C @ 760mm

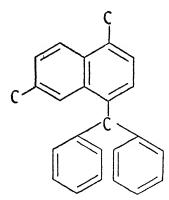
265-275

 $9^{29}$ 

(a) The structure of this compound was not clearly defined in the literature.

#### $C_{25}H_{22}$

### 1,6-Dimethyl-4-benzhydrylnaphthalene



M. P., °C 134–135<sup>6</sup>

References on Naphthalene with Two Phenyl or One Biphenylyl Substitutions

- 1. Acree, S. F., Ber. 37, 616 1904.
- 2. Acree, S. F., Ber. 37, 625 1904.
- 3. Acree, S. F., Ber. 37, 2753 1904.
- Allen, C. F. II., and L. Gilman, J. Am. Chem. Soc. 58, 937 1936.
- Bergmann, F., H. E. Eschinazi, and D. Schapiro, J. Am. Chem. Soc. 64, 557 1942.
- Bistrzycki, A., and G. Krause, Helv. Chim. Acta 16, 100 1933.
- 7. Bodroux, F., Bull. soc. chim. [1] 17, 318 1915.
- 8. Bodroux, F., Compt. rend. 161, 131 1915.
- 9. Borsche, W., P. Hofmann, and II. Kühn, Ann. 554, 23 1943.
- Bowden, S. T., and W. E. Harris, J. Chem. Soc. 1939, 307.
- Crawford, H. M., J. Am. Chem. Soc. 61, 608 1939.
- Dilthey, W., J. prakt. Chem. [2] 109, 273 1925.
- Dufraisse, C., and R. Priou, Bull. soc. chim. [5] 5, 502 1938.
- Dufraisse, C., and R. Priou, Bull. soc. chim. [5] 5, 611 1938.

 Dziewoński, K., and J. Moszew, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1928 A, 283.

- Dziewoński, K., and J. Moszew, Roczniki Chem. 11, 169 1931; C.A. 26, 131 1932; Chem. Zentr. 1931, I, 2875.
- Dziewoński, K., J. Moszew, S. Lepiankiewicz, and L. Sucheni, Bull. intern. acad. polon. sci., Classe sci. mathnat. 1929 A, 950.
- Dziewoński, K., and S. Wodelski, Roczniki Chem. 12, 366 1932; C.A.: 27, 2145 1933.
- Gomberg, M., and F. W. Sullivan, Jr.,
   J. Am. Chem. Soc. 44, 1810 1922.
- 20. Hemilian, V., Ber. 13, 678 1880.
- 21. Hey, D. H., and S. E. Lawton, J. Chem. Soc. 1940, 374.
- Ipatieff, V. N., and B. N. Dolgov, Bull. soc. chim. [4] 45, 950 1929.
- 23. Lehne, A., Ber. 13, 358 1880.
- McKenzie, A., and W. S. Dennler, J. Chem. Soc. 125, 2105 1924.
- McMullen, T. C., J. Am. Chem. Soc. 44, 2055 1922.
- Migita, M., Bull. Chem. Soc. Japan 7, 382 1932.
- Migita, M., Bull. Chem. Soc. Japan 8, 22 1933.
- Pokrovskaya, E. S., and T. G. Stepanseva, J. Gen. Chem. (U. S. S. R.)
   1953 1939; C.A. 34, 4731 1940.
- Price, C. C., and J. M. Ciskowski, J. Am. Chem. Soc. 60, 2199 1938.
- Price, C. C., H. M. Shafer, M. F. Huber, and C. Bernstein, J. Org. Chem. 7, 517 1942.
- 31. Price, C. C., and A. T. Tomisck, J. Am. Chem. Soc. **65**, 439 **1943**.
- Ullmann, F., and A. Murav'ev-Vinogradov, Ber. 38, 2213 1905.
- Vansheidt, A., and B. Moldavski, Ber.
   917 1931.
- Vansheidt, A., and B. Moldavski, J. Gen. Chem. (U. S. S. R.) 1, 304 1931;
   Chem. Zentr. 1931, II, 3208.
- 35. von Bogusky, J. J., Ber. 39, 2866 1906.
- Weiss, R., and A. Abeles, Monatsh. 61, 162 1932.
- 37. Ziegler, K., and F. Ditzel, Ann. 473, 194 1929.
- 38. Zsuffa, M., Ber. 43, 2915 1910.

# 3. BENZOPYRENES AND THEIR ALKYL DERIVATIVES, C<sub>n</sub>H<sub>2n-28</sub>

### $C_{20}H_{12}$

### 1,2-Benzopyrene

 $D_4^{20}$ 1.351 (solid)<sup>12</sup>
1.282 (b) (solid)<sup>13</sup>

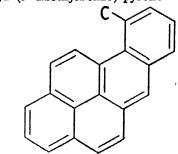
- (a) This compound was given as 1,2-Benzopyrene in the literature, but no structural formula or indication of the numbering system used was given.
- (b) The temperature of this determination was not given.

# 4,5-Benzopyrene

M. P., °C 178–179<sup>5</sup> 178<sup>4</sup> 174<sup>11</sup>

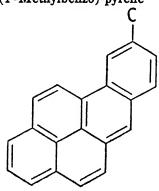
### C21H14

# 1,2-(3'-Methylbenzo)-pyrene



M. P., °C 190-191<sup>13</sup> 190-190.8<sup>1</sup>

### 1, 2-(4'-Methylbenzo)-pyrene



### M. P., °C

139-140 (a)<sup>1</sup> 138-139 (b)<sup>8</sup>

B. P., °C @ 760mm

175

 $0.01^{1}$ 

- (a) This compound remelts at 140-140.4.
- (b) This compound remelts at 140-140.2.

### 1,2-(5'-Methylbenzo)-pyrene

M. P., °C

147.5-148 (a)<sup>1</sup>

146.5-147.58

146.5-147 (a)<sup>8</sup>

B. P., °C @ 760mm 220

 $0.01^{1}$ 

(a) This compound remelts at 147.6-148.1.

# 4,5-(3'-Methylbenzo)-pyrene

M. P., °C 219.5–220<sup>10</sup> 217.5–218<sup>10</sup>

### 3-Methyl-1, 2-benzopyrene

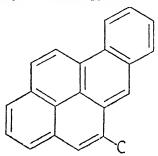
M. P., °C

216.6-217.36

216.2-216.79

215.7-216.29

### 4-Methyl-1, 2-benzopyrene



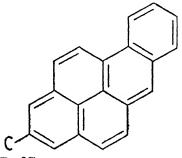
M. P., °C

171-172.516

171-17220

171-171.59

# 7-Methyl-1, 2-benzopyrene



M. P., °C 157.5–159.5³

#### $C_{22}H_{16}$

1, 2-(5', 6'-Dimethylbenzo)-pyrene

M. P., °C 215–216<sup>1</sup>

#### 8-Ethyl-1, 2-benzopyrene

M. P., °C 112<sup>21</sup>

### References on Benzopyrenes and Their Alkyl Derivatives

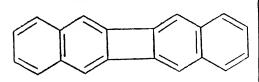
- Bachmann, W. E., and M. Carmack, J. Am. Chem. Soc. 63, 2494 1941.
- Bachmann, W. E., M. Carmack, and S. R. Safir, J. Am. Chem. Soc. 63, 1682 1941.
- Bergmann, E., and O. Blum-Bergmann,
   J. Am. Chem. Soc. 58, 1678 1936.

- 4. Clar, E., Ber. 76, 609 1943.
- Cook, J. W., C. L. Hewett, and I. Hieger, J. Chem. Soc. 1933, 395.
- Davis, W. W., M. E. Krahl, and G. H. A. Clowes, J. Am. Chem. Soc. 62, 3080 1940.
- Fieser, L. F., and M. Fieser, J. Am. Chem. Soc. 57, 782 1935.
- Fieser, L. F., and E. B. Hershberg,
   J. Am. Chem. Soc. 60, 1658 1938.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 60, 2512 1938.
- Fieser, L. F., and M. S. Newman, J. Am. Chem. Soc. 57, 1602 1935.
- Ghigi, E., Atti congr. intern. chim.
   10th Congr. Rome 3, 178 1939; C.A.
   33, 9318 1939; Chem. Zentr. 1939,
   II, 3984.
- Iball, J., Z. Krist. 94, 7 1936; C.A. 30, 7412 1936; Chem. Zentr. 1936, 11, 3293.
- Kon, G, A. R., and E. M. F. Roe, J. Chem. Soc. 1945, 143.
- Mayneord, W. V., and E. Roe, Biochem. J. 30, 707 1936.
- Miescher, G., F. Almasy, and K. Klaui, Biochem. Z. 287, 189 1936.
- Mohler, II., and J. Sorge, Helv. Chim. Acta 22, 229 1939.
- Sannié, C., Biochem. J. 30, 704 1936.
- 18. Shear, M. J., Am J. Cancer 26, 322 1936.
- Vollmann, H., H. Becker, M. Corell, and H. Streeck, Ann. 531, 1 1937.
- Weizmann, C., and E. Bergmann, Scripta Academica Hierosolymitana (Jerusalem) Sci. Report No. 1 1938.
- Windaus, A., and K. Raichle, Ann. 537, 157 1939.
- Winterstein, A., and K. Schön, Z. physiol. Chem. 230, 146 1934.
- 23. Winterstein, A., and H. Vetter, Z. physiol. Chem. 230, 169 1934.
- Winterstein, A., II. Vetter, and K. Schon, Ber. 68, 1079 1935.

# 4. MISCELLANEOUS POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-28}$

### $C_{20}H_{12}$

# 1,2,3,4-Di-(3',2'-naphtho)-cyclobutane



M. P., °C  $365 \pm 2^{50}$ 

### Perylene

26538

### $C_{21}H_{14} \\$

# Phenyl-(9'-fluoryl)-ethyne

M. P., °C 98–100³⁴

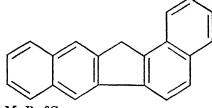
# 1,2,3,4-Dibenzofluorene

M. P., °C 158-160<sup>57</sup> 158-159<sup>7</sup> 157.5-158.5<sup>43</sup>

### 1,2,5,6-Dibenzofluorene

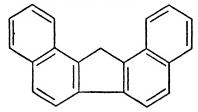
M. P., °C 171-172<sup>21</sup>

### 1,2,6,7-Dibenzofluorene



M. P., °C 294<sup>24</sup>

# 1,2,7,8-Dibenzofluorene

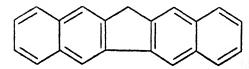


 $\begin{array}{c} \text{M. P., °C} \\ 242.5^{53} \\ 234^{13} \end{array}$ 

 $232^{14}$   $230^{10}$ 

228-230<sup>36</sup>

### 2,3,6,7-Dibenzofluorene



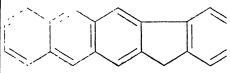
M. P., °C 282.5–283.5<sup>42</sup> 281–282<sup>24</sup>

### x,x,x,x-Dibenzofluorene (a)

M. P., °C 306<sup>4</sup> 190.5<sup>52</sup>

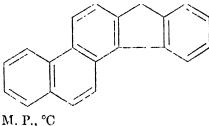
(a) The structure of these compounds was not clearly defined in the literature.

### 2,3-(3',2'-Indo)-anthracene



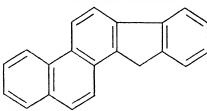
M. P., °C 317<sup>5</sup>

### 1,2-(3',2'-Indo)-phenanthrene



M. P., ℃ 18911

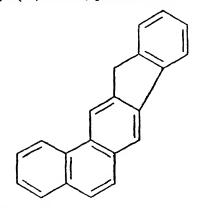
# $1\;\text{\tt,2-(2'\,\tt,3'-Indo)-phenanthrene}$



M. P., °C 338–339<sup>20</sup> 328<sup>2</sup>

327-32822, 28

### 2,3-(3',2'-Indo)-phenanthrene

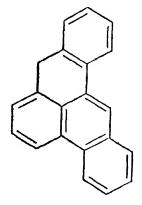


M. P., °C 226-226.5<sup>21</sup>

# 8,9-(2', 1'-Naphtho)-phenalene

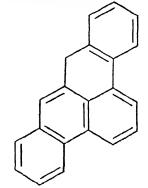
M. P., °C 149--149.5<sup>28</sup>

# 2,3,5,6-Dibenzophenalene



M. P., °C 128-129<sup>15</sup>

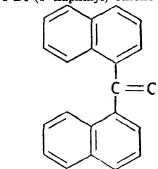
# 2,3,7,8-Dibenzophenalene



M. P., °C 171-172.5<sup>15</sup>

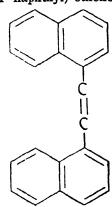
# $\mathbf{C}_{22}\mathbf{H}_{16}$

# 1,1-Di-(1'-naphthyl)-ethene



M. P., °C 107<sup>47</sup>

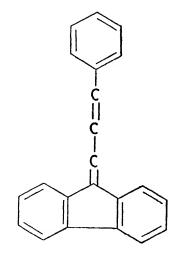
# 1,2-Di-(1'-naphthyl)-ethene



### 1,2-Di-(2'-naphthyl)-ethene

$$\begin{array}{c} \text{M. P., °C} \\ 255^{61} \\ 254 \cdot 255^{29} \cdot ^{60} \\ 253 - 254^{54} \end{array}$$

# 1-Phenyl-3-(9'-fluorylidene)-propene-1



$$\begin{array}{c} \text{M. P., °C} \\ 155^{35} \\ 154\text{--}155^{62} \\ 154.5^{55} \end{array}$$

# 1-Phenyl-1-(9'-phenanthryl)-ethene

M. P., °C 142<sup>6</sup>  $C_{22}H_{16}$ 

### 1-Phenyl-2-(9'-phenanthryl)-ethene

M. P., °C 118<sup>6</sup>

# x-Phenyl-x, x-dihydrobenzo-[jk]-fluorene (a)

M. P., °C 148<sup>56</sup>

(a) The structure of this compound was not clearly defined in the literature.

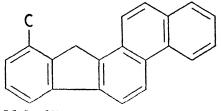
### 1,2-(2',1'-Naphtho)-5-methylfluorene

M. P., °C 275-276<sup>2</sup> 275<sup>51</sup> 1,2-(2',1'-Naphtho)-6-methylfluorene

1,2-(2',1'-Naphtho)-7-methylfluorene

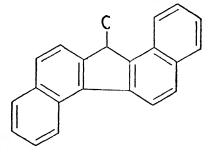
M. P., °C 33·!--336² 301<sup>43</sup>

1,2-(2',1'-Naphtho)-8-methylfluorene



M. P., °C 281-281.5<sup>2</sup>

# 1,2,5,6-Dibenzo-9-methylfluorene



1,2-(2',3'-Indo)-3-methylphenanthrene

1,2,3,4-Dibenzo-9,10-dihydroan-thracene

M . P., °C 202–203³

1,2,5,6-Dibenzo-9,10-dihydroan-thracene

M. P., °C 218.5-219.5<sup>1</sup> 196-198<sup>19</sup>

1 , 2-Benzo-x , x-dihydronaphthacene (a)

M. P., °C 157–160.5<sup>3</sup>

(a) The hydrogens may be in either the 5,8- or 9,10-positions.

5,14-Dihydropentacene

M. P., °C 273<sup>16</sup>

6,13-Dihydropentacene

M. P., °C 271-272<sup>41</sup> 270<sup>17, 39, 40</sup>

 $C_{23}H_{18}$ 

1,2,3,4-Dibenzo-6-methyl-9,10-dihydroanthracene

M. P., °C 207–209.5<sup>3</sup> 11,12-Benzo-5-methyl-5,6-dihydrochrysene

M. P., °C 151–152°

1,4-o-Endophenylene-2,3-benzo-1,4,4a,9a-tetrahydrofluorene

M. P., °C 118<sup>37</sup>

C21H20

1,2,5,6-Dibenzo-9,10-dimethyl-9,10-dihydroanthracene

M. P., °C 277-278 (a)<sup>18</sup> 207-209 (b)<sup>18</sup> (a) This constant was determined on the *trans* isomer of the compound.

(b) This constant was determined on the cis isomer of the compound.

1,2,2a,3,4,4a,5,6-Octahydrocoronene

M. P., °C 269-270<sup>30</sup>

### $C_{25}H_{22}$

1,2-(2',1'-Naphtho)-5-methyl-8isopropylfluorene

M. P., °C 198<sup>22, 28</sup>

 $C_{26}H_{24}$ 

9-Cyclohexyl-10-phenylanthracene

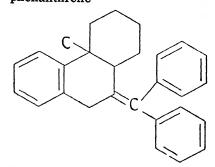
M. P., °C 231–232<sup>59, 59</sup>

 $C_{27}H_{26}$ 

- 2,7-Dibenzyl-x<sub>4</sub>-tetrahydrofluorene
  (a)
- M. P., °C 90 91<sup>27</sup>
- (a) The structure of this compound was not clearly defined in the literature.

### $C_{28}H_{28}$

4a-Methyl-10-benzhydrylidene-1,2,3,4,4a,9,10,10a-octahydrophenanthrene



M. P., °C 126<sup>31</sup>

#### C30H32

- 1-Methyl-7-isopropyl-x, x-diphenyl-x<sub>6</sub>-hexahydrophenanthrene (a)
- M. P., °C 82<sup>32</sup>
- (a) The structure of this compound was not clearly defined in the literature.

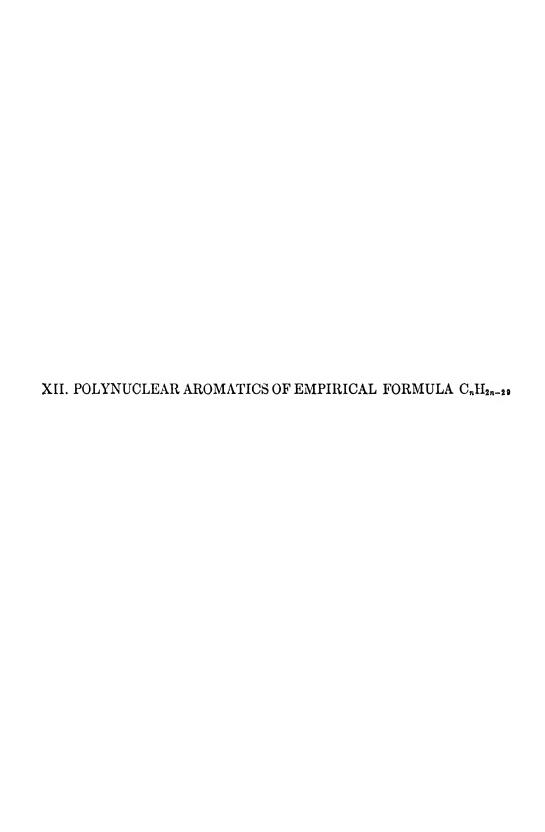
References on Miscellaneous Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-28}$ 

- 1. Bachmann, W. E., J. Org. Chem. 1, 347 1936-37.
- Bachmann, W. E., J. W. Cook, C. L. Hewett, and J. Iball, J. Chem. Soc. 1936, 54.
- Bachmann, W. E., and L. H. Pence, J. Am. Chem. Soc. 59, 2339 1937.
- Bamberger, E., and F. D. Chattaway, Ann. 284, 52 1895.
- Barnett, E. de B., N. F. Goodway, and J. W. Watson, Ber. 66, 1876 1933.
- Bergmann, E., and F. Bergmann, J. Am. Chem. Soc. 59, 1443 1937.
- Bergmann, E, and F. Bergmann, J. Am. Chem. Soc. 60, 1805 1938.
- 8. Billows, E., Rivista min. crist. ital. 30, 31 1903.
- Bradsher, C. K., and L. Rapoport, J. Am. Chem. Soc. 66, 1281 1944.
- Buu-Hoi and P. Cagniant, Compt. rend. 216, 299 1943; C.A. 38, 2330 1944.
- Buu-Hoi and P. Cagniant, Rev. sci. 80, 319 1942; C A. 39, 3276 1945.
- Calcott, W. S., J. M. Tinker, and V. Weinmayr, J. Am. Chem. Soc. 61, 949 1939.
- Chichibabin, A. E., and O. I. Magidson,
   J. prakt. Chem. [2] 90, 168 1914.
- Chichibabin, A., and O. Magidson, J. Russ. Phys. Chem. Soc. 46, 1389 1914;
   Chem. Zentr. 1915, I, 1123.
- 15. Clar, E., Ber. 76, 609 1943.
- 16. Clar, E., and F. John, Ber. 62, 3021 1929.
- 17. Clar, E., and F. John, Ber. 63, 2967 1930.
- 18. Cook, J. W., J. Chem. Soc. 1931, 489.
- 19. Cook, J. W., J. Chem. Soc. 1933, 1592.
- 20. Cook, J. W., J. Chem. Soc. 1941, 685.
- Cook, J. W., A. Dansi, C. L. Hewett, J. Iball, W. V. Mayneord, and E. Roe, J. Chem. Soc. 1935, 1319.

369 References

- Cook, J. W., C. L. Hewett, W. V. Mayneord, and E. Roe, J. Chem. Soc. 1934, 1727.
- Cook, J. W., C. L. Hewett, W. V. Mayneord, and E. Roe, J. Soc. Chem. Ind. 53, 569 1934.
- Cook, J. W., and R. W. G. Preston, J. Chem. Soc. 1944, 553.
- Corbellini, A., and F. Steffenoni, Rend. ist. lombardo sci. [2] 69, 429 1936;
   Chem. Zentr. 1937, I, 1420.
- Davis, W. W., M. E. Krahl, and G. H. A. Clowes, J. Am. Chem. Soc. 62, 3080, 1940.
- Dziewoński, K., M. Dominikowna, L. Galuszkowna, and W. Muz, Roczniki Chem. 13, 283 1933; C.A. 27, 4533 1933; Chem. Zentr. 1933, II, 1524.
- Elbs, K., J. prakt. Chem. [2] 47, 44
   1893.
- 28A. Fieser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 58, 478 1936.
- 29. Friedmann, W., Ber. 49. 1352 1916.
- Fromherz, H., L. Thaler, and G. Wolf,
   Z. Elektrochem. 49, 387 1943.
- 31. Grewe, R., Ber. 76, 1076 1943.
- 32. Heiduschka, A., and H. Grimm, Arch. Pharm. 250, 33 1912.
- 33. Hertel, E., and H. W. Bergk, Z. physik. Chem. 33B, 319 1936.
- Hurd, C. D., and F. L. Cohen, J. Am. Chem. Soc. 53, 1068 1931.
- 35. Kuhn, R., and A. Winterstein, Helv. Chim. Acta 11, 116 1928.
- 36. Magidson, O. I., Ber. 58, 433 1925.
- 37. Mameli, E., Gazz. chim. ital. 67, 669 1937.
- 38. Marschalk, C., Bull. soc. chim. [1]
  43, 1388 1928.
- Marsehalk, C., Bull soc. chim. [5] 4, 1381 1937.
- 40. Marschalk, C., Bull. soc. chim. [5] 4, 1535 1937.
- 41. Marschalk, C., Bull. soc. chim [5] 5, 156 1938.
- 42. Martin, R. H., J. Chem. Soc. 1941, 679.
- 43. Mohler, II., and J. Sorge, Helv. Chim. Acta 22, 229 1939.

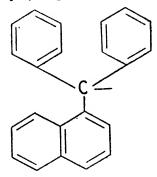
- Morgan, G. T., and J. G. Mitchell,
   J. Chem. Soc. 1934, 536.
- Nenitzescu, C. D., and D. Λ. Isăcescu, Ber. 63, 2484 1930.
- Pestemer, M., and J. Cecelsky, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 140, Abt. Hb, 541 1931.
- Pfeiffer, P., and P. Schneider, J. prakt Chem. [2] 129, 129 1931.
- Pongratz, A., and A. Halabarda, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 139, Abt. Hb, 433 1930.
- 49. Rådulescu, D., G. Ostrogovich, and F. Barbulescu, Ber. 34, 2240 1931.
- Rosenhauer, E., F. Braun, R. Punmerer, and G. Riegelbauer, Ber. 7J, 2281 1937.
- Ruzieka, L., and M. W. Goldberg, Helv. Chim. Acta 20, 1245 1937.
- Schmidlin, J., and M. Huber, Ber. 43, 2824 1910.
- Schmidlin, J., and P. Mossini, Ber. 42, 2377 1909.
- 54. Szperl, L., Roczniki Chem. 1923, II, 291; C.A. 18, 1290 1924.
- Thiele, J., and F. Heule, Ann. 347, 290 1906.
- von Braun, J., and G. Manz, Ber. 70, 1603 1937.
- Weizmann, C., and E. Bergmann, Scripta Academica Hierosolymitana (Jerusalem) Sci. Report No. 1 1938.
- Willemart, A., Bull. soc. chim. [5] 6, 204 1939.
- Willemart, A., Compt. rend. 207, 536
   1938.
- Wislicenus, W., and H. Wren, Ber. 33, 502 1905.
- Wood, J. H., J. A. Bacon, A. W. Meibohm, W. H. Throckmorton, and G. P. Turner, J. Am. Chem. Soc. 63, 1334 1941.
- Ziegler, K., and F Crossmann, Ann.
   89 1934.
- Zinke, A., F. Stimler, and E. Reuss, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 143, Abt. Hb, 329 1936.
- 64. Zinke, A., and H. Troger, Ber. 74, 107 1941.



# XII. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-29}$

 $C_{23}H_{17}$ 

Diphenyl-(1-naphthyl)-methyl



M. P., °C 135–137<sup>1, 2</sup>

- References on Polymuclear Aromatics of Empirical Formula C<sub>n</sub>II<sub>2n-29</sub>
- Gomberg, M., and C. S. Schoepfle, J. Am. Chem. Soc. 39, 1652 1917.
- Gomberg, M., and C. Schoepfle, J. Am. Chem. Soc. 41, 1655 1919.

# XIII. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nII_{2n-30}$

- 1. Picene and Its Alkyl Derivatives
- 2. Dibenzoanthracenes, Dibenzophenanthrenes, and Their Alkyl Derivatives
- 3. Miscellaneous Polynuclear Aromatics of Empirical Formula  $C_n H_{2n-30}$

# 1. PICENE AND ITS ALKYL DERIVATIVES, $C_nH_{2n-30}$

### $C_{22}H_{14}$

#### Picene

M. P., °C

365

366.0-366.56

365-36611,14,18

 $363.5 - 361.5^{3}$ 

3641,2,16

B. P., °C @ 760mm 518-5204

### $C_{23}H_{16}$

### 5-Methylpicene

M. P., °C 251.6-252.2<sup>6</sup>

# 13-Methylpicene

M. P., °C 203.6-204.4<sup>6</sup>

### $C_{24}H_{18}$

### 2,11-Dimethylpicene

M. P., °C

293-29412

# 4,9-Dimethylpicene

M. P., °C 380-381<sup>10</sup> 370-371<sup>7</sup>

### 4,11-Dimethylpicene

305-306<sup>9, 9</sup> C<sub>.5</sub>H<sub>.0</sub>

### 3,4,9-Trimethylpicene

30610, 13

M. P., °C 380 381<sup>10</sup> 372-373<sup>7</sup>

### 3,4,11-Trimethylpicene

M. P., °C 309-310<sup>10</sup> 308-310<sup>12</sup>

B. P., °C @ 760mm 270

 $0.1^{12}$ 

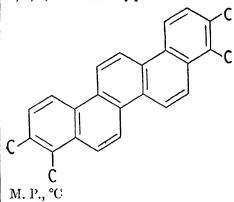
# x,x,x-Trimethylpicene (a)

M. P., °C 306<sup>15</sup> 305-306<sup>17</sup> 298<sup>5</sup>

(a) The structure of this compound was not clearly defined in the literature.

### $C_{26}H_{22}$

### 3,4,9,10-Tetramethylpicene



# References on Picene and Its Alkyl Derivatives

400-40110

- Bamberger, E., and F. D. Chattaway, Ann. 284, 52 1895.
- Bamberger, E., and F. Chattaway, Ber. 26, 1751 1893.
- Davis, W. W., M. E. Krahl, and G. H. A. Clowes, J. Am. Chem. Soc. 62, 3080 1940.
- 1. Graebe, C., and J. Walter, Ber. 14, 175 1881.
- Jacobs, W., and O. Isler, J. Biol. Chem. 119, 155 1937.
- Newman, M. S., J. Org. Chem. 9, 518 1944.

- Ruzicka, L., M. Goldberg, and K. Hofmann, Helv. Chim. Acta 20, 325 1937.
- Ruzicka, L., A. Grob, and G. Anner, Helv. Chim. Acta 26, 254 1943.
- Ruzicka, L., and K. Hofmann, Helv. Chim. Acta 20, 1155 1937.
- Ruzicka, L., and K. Hofmann, Helv. Chim. Acta 22, 126 1939.
- Ruzicka, L., and H. Hosli, Helv. Chim. Acta 17, 470 1934.
- Ruzicka, L., and E. Morgeli, Helv. Chim. Acta 19, 377 1936.

- Ruzicka, L., H. Schellenberg, and M. W. Goldberg, Helv. Chim. Acta 20, 791 1937.
- Ruzicka, L., G. Thomann, E. Brandenberger, M. Furter, and M. W. Goldberg, Helv. Chim. Acta 17, 200 1934.
- Spring, F. S., and T. Vickerstaff, J. Chem. Soc. 1937, 249.
- 16. Weidlich, H. A., Ber. 71, 1203 1938.
- Wieland, H., A. Hartman, and H. Dietrich, Ann. 522, 191 1936.
- Winterstein, A., K. Schön, and H. Vetter, Z. physiol. Chem. 230, 158 1934.

# 2. DIBENZOANTHRACENES, DIBENZOPHENANTHRENES, AND THEIR ALKYL DERIVATIVES, $C_nH_{2n-30}$

375

### $C_{22}H_{14}$

### 1,2,3,4-Dibenzoanthracene

M. P., °C

2057

203-20116

 $200-202^2$ 

200-201.54

### 1,2,5,6-Dibenzoanthracene

M. P., °C

264.5

 $266.6 \cdot 266.9^{17}$ 

 $266 - 266.5^{20}$ 

 $265.5 - 266.0^{26}$ 

 $263.5 - 261.5^{25}$ 

262.7-264.017

2626, 29

261--26222

261-261.516, 18

26024

 $D_{4}^{20}$ 

 $1.282 (a)^{24}$ 

(a) The temperature of this determination was not given.

### 1,2,7,8-Dibenzoanthracene

M. P., ℃

198.0-198.417

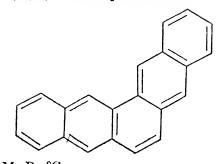
19614

 $195 - 196^{27}$ 

### 2,3,5,6-Dibenzophenanthrene

M. P., °C 261<sup>11</sup> 137-138<sup>23</sup>

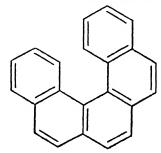
### 2,3,6,7-Dibenzophenanthrene



 $\begin{array}{c} \text{M.P., °C} \\ 257^{\,8} \\ 255-257^{\,30} \\ 250-251^{\,10} \end{array}$ 

Additional Data Sublimation Temp. (°C) 230 12mm<sup>10</sup>

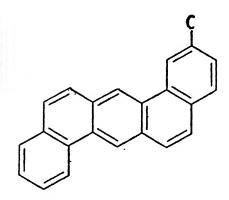
# 3,4,5,6-Dibenzophenanthrene



M. P., °C 177-178<sup>15</sup> 177<sup>28</sup> 145-146<sup>29</sup>

### C23H16

1,2-(4'-Methylbenzo)-5,6-benzoanthracene



M. P., °C 256–257.5<sup>12, 13</sup>

### 1,2-(5'-Methylbenzo)-5,6-benzoanthracene

M. P., °C 245<sup>12, 13, 16</sup> 244–245<sup>19</sup> 377 C<sub>28</sub>H<sub>16</sub>

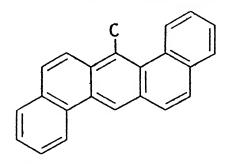
# 1,2,3,4-Dibenzo-6-methylanthracene

M. P., °C 157.5–1584

### 1,2,5,6-Dibenzo-4-methylanthracene

M. P., °C 184-185<sup>16</sup>

### 1,2,5,6-Dibenzo-9-methylanthracene



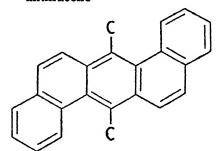
M. P., °C 192–194.5<sup>21</sup>

### 2,3-(4'-Methylbenzo)-6,7-benzophenanthrene

M. P., °C 315-316°

### $C_{24}H_{18}$

1,2,5,6-Dibenzo-9,10-dimethylanthracene



M. P., °C 205.5-206.5<sup>3, 13</sup> 203-204<sup>1</sup>

### 1,2,7,8-Dibenzo-9,10-dimethylanthracene

M. P., °C 205.5-206.5<sup>12</sup> 2,3-(5'-Methylbenzo)-5,6-(4'-methylbenzo)-phenanthrene

M. P., °C 228<sup>11</sup>

2,3-(4'-Methylbenzo)-6,7-(5'methylbenzo)-phenanthrene (a)

M. P., °C 325<sup>5</sup>

(a) More information about this compound is found in references 8 and 10. The physical constants in these references were found to be in error.

2,3,6,7-Di-(5'-methylbenzo)phenanthrene (a)

M. P., °C 325<sup>5</sup>

(a) More information about this compound is found in references 8 and 10. The physical constants in these references were found to be in error.

#### C<sub>30</sub>H<sub>30</sub>

1,2,5,6-Dibenzo-9,10-di-n-butylanthracene

M. P., °C 143.5–144.5<sup>12, 13</sup>

References on Dibenzoanthracenes, Dibenzophenanthrenes, and Their Alkyl Derivatives

- Akin, R. B., and M. T. Bogert, J. Am. Chem. Soc. 59, 1564 1937.
- Bachmann, W. E., J. Am. Chem. Soc. 56, 1363 1934.
- Bachmann, W. E., and J. M. Chemerda,
   J. Am. Chem. Soc. 61, 2358 1939.
- Bachmann, W. E., and L. H. Pence, J. Am. Chem. Soc. 59, 2339 1937.

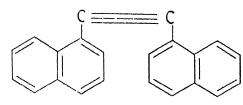
 $C_{22}H_{14}$ 

- Clar, E., "Aromatische Kohlenwasserstoffe," Springer-Verlag, Berlin, 1941.
- 6. Clar, E., Ber. 62, 350 1929.
- Clar, E., through Cook, Hieger, Kennaway, and Mayneord, Proc. Roy. Soc. London 111B, 455 1932.
- 8. Clar, E., and F. John, Ber. 64, 981 1931.
- Clar, E., F. John, and R. Avenarius, Ber. 72, 2139 1939.
- Clar, E., F. John, and B. Hawran, Ber.
   62, 940 1929.
- Clar, E., and H. D. Wallenstein, Ber. 64, 2076 1931.
- 12. Cook, J. W., J. Chem. Soc. 1931, 487.
- 13. Cook, J. W., J. Chem. Soc. 1931, 489.
- 14. Cook, J. W., J. Chem. Soc. 1932, 1472.
- 15. Cook, J. W., J. Chem. Soc. 1933, 1592.
- Cook, J. W., I. Hieger, E. L. Kennaway, and W. V. Mayneord, Proc. Roy. Soc. London 111B, 455 1932.
- Davis, W. W., M. E. Krahl, and G. H. A. Clowes, J. Am. Chem. Soc. 62, 3080 1940.

- Earle, W. R., and C. Voegtlin, Am. J. Cancer 34, 373 1938.
- Fieser, L. F., and E. M. Dietz, Ber. 62, 1827 1929.
- Fieser, L. F., and E. B. Hershberg, J. Am. Chem. Soc. 60, 1893 1938.
- Fieser, L. F., and G. W. Kilmer, J. Am. Chem. Soc. 61, 862 1939.
- Fieser, L. F., and M. S. Newman, J. Am. Chem. Soc. 53, 2376 1936.
- 23. Hewett, C. L., J. Chem. Soc. 1933, 1286.
- Krishnan, K. S., and S. Banerjee, Z. Krist. 91, 170 1935; C A. 29, 7731 1935; Chem. Zentr. 1935, II, 2890.
- Lorenz, E., and M. J. Shear, Am. J. Cancer 26, 333 1936.
- 26. Shear, M. J., Am. J. Cancer 26, 322 1936.
- Waldmann, H., J. prakt. Chem. [2] 135, 1 1932.
- 28. Weidlich, H. A., Ber. 71, 1203 1939.
- Weitzenbock, R., and A. Klinger, Sitzber, Akad. Wiss. Wien, Math. naturw. Klasse 127, Abt. 119, 119 2918.
- Winterstein, A., and K. Schon, Z. physiol. Chem. 230, 146 1934.

# 3. MISCELLANEOUS POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-30}$

# $C_{22}H_{14}$ Di-(1'-naphthyl)-ethyne



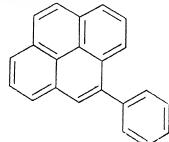
M. P., °C 225<sup>25</sup> 171<sup>33</sup>, <sup>34</sup>

### x-Phenylbenzo-[jk]-fluorene (a)

M. P., °C 144<sup>48</sup>

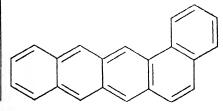
(a) The structure of this compound was not clearly defined in the literature.

### 4-Phenylpyrene



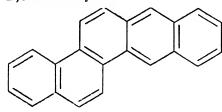
M. P., °C 169<sup>47</sup>

### 1,2-Benzonaphthacene



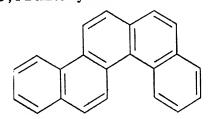
M. P., °C 263-264<sup>3, 12, 15, 16</sup>

### 2,3-Benzochrysene



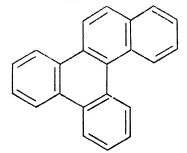
M. P., °C 293-294<sup>12, 16</sup> 293<sup>14</sup>

### 3,4-Benzochrysene



M. P., °C 128<sup>4, 50</sup> 126–127<sup>28</sup> 122<sup>49</sup>

# 5,6-Benzochrysene

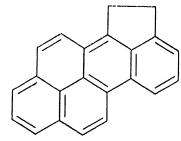


M. P., °C 115–116° 114.5–115<sup>26, 27</sup>

### Pentacene

M. P., °C 270-271<sup>38, 39</sup>

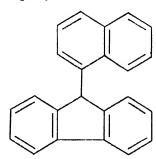
# Benzo-[hij]-cholanthrene



M. P., °C 255-256<sup>2</sup> 252-253<sup>36, 37</sup>

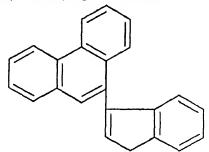
### $C_{23}H_{16}$

# 9-(1'-Naphthyl)-fluorene



M. P., °C 113<sup>31</sup> 109–110<sup>46</sup> 103.5<sup>45</sup>

### 9-(3'-Indenyl)-phenanthrene

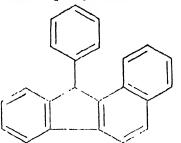


M. P., °C 121.5<sup>5</sup>

B. P., °C @ 760mm 230

 $0.7^{5}$ 

### 1,2-Benzo-9-phenylfluorene



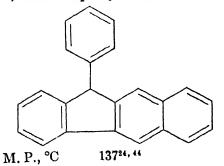
M. P., °C 196<sup>13, 31</sup> 195.5<sup>44</sup> 195<sup>6</sup> 194<sup>17</sup>

 $D_4^{20}$ 

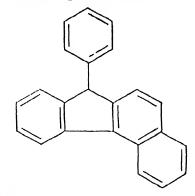
1.242

0° 53

# 2,3-Benzo-9-phenylfluorene

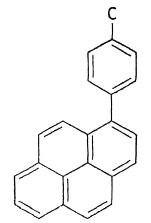


# 3,4-Benzo-9-phenylfluorene



M. P., °C 135–138<sup>13</sup>

### 1-p-Tolylpyrene



 $\begin{array}{c} \text{M. P., °C} \\ \text{155-156}^{\text{41}} \end{array}$ 

# 1,2-Benzo-8-isopropenylpyrene

$$c=c$$

 $C_{23}H_{16}$ 

M.P., °C 114-115<sup>51</sup>

# 5,6-Benzo-11-methylchrysene

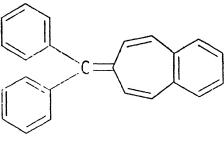
M.P., °C 163.5-161<sup>10</sup>

### 5,6-Benzo-12-methylchrysene

M . P., °C 150.5–151.5<sup>10</sup>

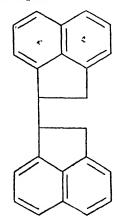
### $C_{24}H_{18}$

### 1,2-Benzo-5-benzhydrylidenecycloheptadiene-3,6



M. P., °C 9243

# 1,1'-Biacenaphthenyl



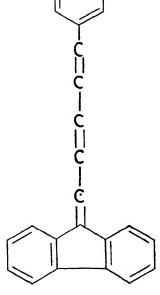
M. P., °C 120<sup>18,19,21</sup>

### x,x'-Biacenaphthenyl (a)

M. P., °C 174<sup>23</sup>

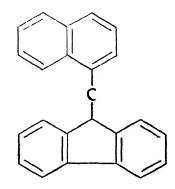
(a) The structure of this compound was not clearly defined in the literature.

#### 1-Phenyl-5-(9'-fluorylidene)-pentadiene-1,3

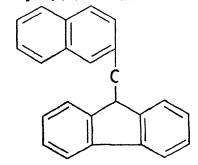


M. P., °C 155.5<sup>32</sup>

## (1-Naphthyl)-(9'-fluoryl)-methane

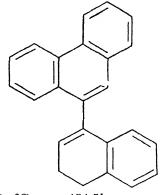


M. P., °C 135<sup>40</sup> 133–134<sup>42</sup> (2-Naphthyl)-(9'-fluoryl)-methane



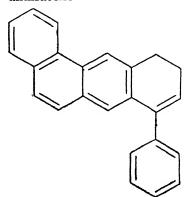
M. P., °C 16442

# 9-[4'-(1',2'-Dihydronaphthyl)]phenanthrene



M. P., °C 181.5<sup>5</sup>

#### 7,8-Benzo-5-phenyl-1,2-dihydroanthracene



M. P., °C 125-126<sup>1</sup>

#### 1,2-Benzo-5-phenyl-9,10-dihydroanthracene

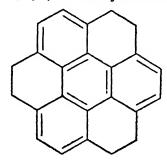
M. P., °C 96–96.5³

#### Crackene (a)

M. P., °C 309-310<sup>7</sup> 308<sup>29, 30</sup>

(a) The structure of this compound was not clearly defined in the literature. It may be a dimethyl picene or a dinaphthylbutene.

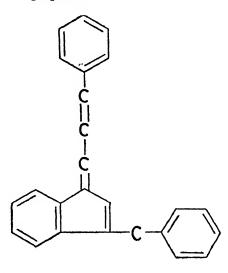
#### 1,2,5,6,9,10-Hexahydrocoronene



M. P., °C 271-272<sup>22</sup>

C25H20

1-Phenyl-3-(3'-benzylindenylidene)propene-1



M. P., °C 161-162<sup>52</sup>

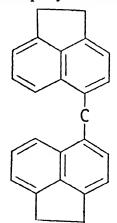
#### 4-Benzhydrylacenaphthene

M. P., °C 17635

#### 5-Benzhydrylacenaphthene

M. P., °C 1678

#### Di-5,5'-acenaphthylmethane



M. P., °C 140–141<sup>20</sup>

#### 1,2-(4'-Methylbenzo)-3-methyl-9phenylfluorene

M. P., °C 215<sup>17</sup>

1,2-Benzo-5,6-[3',2'-(5',6',7',8'-tetrahydronaphtho)]-fluorene

M. P., °C 160<sup>11</sup>

B. P., °C @ 760mm 330

211

# $C_{26}H_{22}$

x-Benzhydrylidene-x,x-dimethylx,x-benzocycloheptadiene-x,x
(a)

M. P., °C 128.5-129<sup>43</sup>

(a) The structure of this compound was not clearly defined in the literature.

# Phenyl-p-tolyl-(4-acenaphthyl)-methane

M. P., °C 209<sup>35</sup>

#### C<sub>36</sub>H<sub>42</sub>

Didecahydronaphtho-[3', 4', 5'-abc, 3'', 4'', 5''-jkl]-1, 2, 2a, 3, 4, 4a, 5, 6, 6a, 7, 8, 12c, 12d, 12e-tetradecahydrocoronene

M. P., °C 262–263<sup>22</sup>

References on Miscellaneous Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-0}$ 

- Bachmann, W. E., and J. T. Bradbury, J. Org. Chem. 2, 175 1937-38.
- Bachmann, W. E., and M. Carmack,
   J. Am. Chem. Soc. 63, 1685 1941.
- Bachmann, W. E., and L. H. Pence,
   J. Am. Chem. Soc. 59, 2339 1937.
- 4. Bergmann, E., J. Chem. Soc. 1938, 1291.
- Bergmann, E., and F. Bergmann, J. Am. Chem. Soc. 59, 1443 1937.
- Bergmann, E., and H. A. Wolff, Ber. 63, 1176 1930.
- 7. Börnstein, E., Ber. 39, 1238 1906.
- Bowden, S. T., and W. E. Harris, J. Chem. Soc. 1939, 307.
- Bradsher, C. K., and L. Rapoport, J. Am. Chem. Soc. 65, 1646 1943.
- Bradsher, C. K., and L. Rapoport, J. Am. Chem. Soc. 66, 1281 1944.

- 11. Buu-Hoï and P. Cagniant, Rev. sci. 80, 384 1942; C.A. 39, 3277 1945.
- 12. Clar, E., Ber. 62, 1574 1929.
- 13. Clar, E., Ber. 63, 512 1930.
- Clar, E., and L. Lombardi, Ber. 65, 1411 1932.
- 15. Cook, J. W., J. Chem. Soc. 1931, 499.
- Cook, J. W., I. Hieger, E. L. Kennaway, and W. V. Mayneord, Proc. Roy. Soc. (London) 111B, 455 1932.
- Dilthey, W., J. prakt. Chem. [2] 109, 273 1925.
- Doliński, J., and K. Dziewoński, Ber. 48, 1917 1915.
- Doliński, J., and K. Dziewoński, Bull. intern. acad. sci. Cracovie 1914 A, 389.
- Dziewoński, K., W. Kahl, W. Koczorswska, and A. Wulffsohn, Roczniki Chem. 13, 154 1933; C.A. 27, 3709 1933; Chem. Zentr. 1933, I, 3566.
- Dziewoński, K., and C. Paschalski, Ber. 47, 2680 1914.
- Fromherz, H., L. Thaler, and G. Wolf,
   Z. Elektrochem. 49, 387 1943.
- Garvey, B. S., Jr., L. F. Halley, and C. F. H. Allen, J. Am. Chem. Soc. 59, 1827 1937.
- Gomberg, M., and F. W. Sullivan, Jr.,
   J. Am. Chem. Soc. 44, 1810 1922.
- 25. Grabowski, J., Ber. 11, 298 1878.
- 26. Hewett, C. L., Gazz. chim. ital. 67, 728 1937.
- 27. Hewett, C., J. Chem. Soc. 1938, 193.
- 28. Hewett, C. L., J. Chem. Soc. 1938, 1286.
- Klaudy, J., and I. Fink, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 108, Abt. 11b, 788 1899.
- 30. Klaudy, J., and J. Fink, Chem. Ztg. 24, 60 1900.
- 31. Kovache, A., Ann. chim. [9] 10, 184 1918.
- 32. Kuhn, R., and A. Winterstein, Helv. Chim. Acta 11, 116 1928.
- 33. Leroy, J. Λ., Bull. soc. chim. [3] 7, 644 1892.
- Leroy, J. A., Compt. rend. 113, 1056 1891.
- Lorriman, F. R., J. Am. Chem. Soc. 47, 211 1925.
- Mikhallov, B. M., and N. G. Chernova,
   J. Gen. Chem. (U. S. S. R.) 12, 276
   1942; C.A. 37, 3083 1943.

387 References

- Mikhallov, B. M., and N. G. Chernova,
   J. Gen. Chem. (U. S. S. R.) 12, 525
   1942; Survey For. Petrol. Liter. Aug.
   20, 1943.
- Philippi, E., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 123, Abt. IIb, 21 1914.
- Philippi, E., Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 138 suppl., Abt. IIb, 638 1929.
- Schlenk, W., and E. Bergmann, Ann.
   479, 42 1930.
- Scholl, R., K. Meyer, and J. Donat, Ber. 70, 2180 1937.
- 42. Sieglitz, A., and H. Jassoy, Ber. 54, 2133 1921.
- 43. Staudinger, H., and N. Kon, Ann. 384, 38 1911.
- 44. Ullmann, F., and A. Murav'ev-Vinogradov, Ber. 38, 2213 1905.

- Ullman, F., and R. von Wurstemberger, Ber. 38, 4105 1905.
- Vansheidt, A., and B. Moldavskii, Ber.
   1362 1930.
- Vollmann, H., H. Becker, M. Corell, and H. Streeck, Λnn. 531, 1 1937.
- 48. von Braun, J., and G. Manz, Ber. 70, 1603 1937.
- 49. Weidlich, H. A., Ber. 71, 1203 1938.
- Weizmann, C., and E. Bergmann, Scripta Academica Hierosolymitana (Jerusalem), Sci. Report No. 1 1938.
- Windaus, A., and K. Raichle, Ann. 537, 157 1939.
- Wislicenus, W., and W. Hentrich, Ann. 436, 9 1924.
- 53. Ziegler, K., and F. Ditzel, Ann. 473, 194 1929.

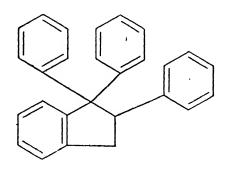
#### XIV. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA C<sub>n</sub>H<sub>2n-32</sub>

- 1. Indane with Three Phenyl Substitutions
- 2. Fluorene with Two Phenyl Substitutions
- 3. Dihydroanthracenes and Dihydrophenanthrenes with Two Phenyl Substitutions
- 4. Miscellaneous Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-32}$

# 1. INDANE WITH THREE PHENYL SUBSTITUTIONS, $C_nH_{2n-32}$

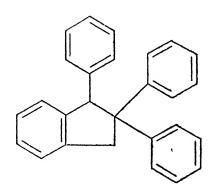
#### $C_{27}H_{22}$

#### 1,1,2-Triphenylindane



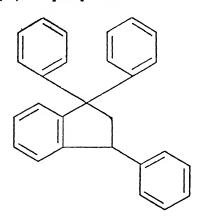
M. P., °C 837

#### 1,2,2-Triphenylindane



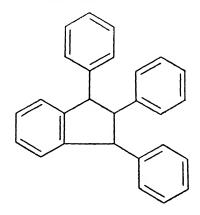
M. P., °C 140<sup>7</sup>

#### 1,1,3-Triphenylindane



M. P., °C 112<sup>2</sup>· <sup>7</sup> 111–112<sup>7</sup>

# 1,2,3-Triphenylindane



M. P., °C 154<sup>6</sup> 153<sup>7</sup>

#### C28H24

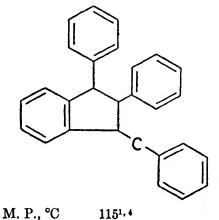
#### 1-Phenyl-3-benzhydrylindane

M. P., °C 137<sup>7</sup> 135 (a)<sup>10</sup> 133-135<sup>5</sup> 133<sup>9</sup> 131<sup>7</sup>

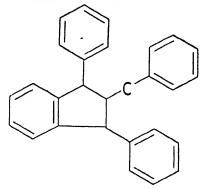
107 (b)10

- (a) This constant was determined on the stabile form of the compound.
- (b) This constant was determined on the labile form of the compound.

## 1,2-Diphenyl-3-benzylindane

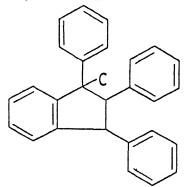


#### 1,3-Diphenyl-2-benzylindane



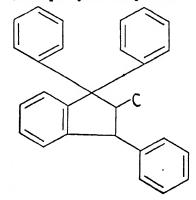
M. P., °C 121–122.5<sup>4</sup>

## 1-Methyl-1,2,3-triphenylindane

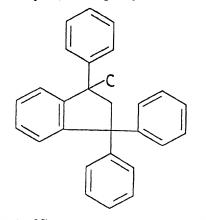


M. P., °C 163-165³ 163-1644

#### 1,1,3-Triphenyl-2-methylindane



#### 1-Methyl-1,3,3-triphenylindane



M. P., °C 143<sup>2</sup>· <sup>8</sup>

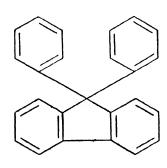
#### References on Indane with Three Phenyl Substitutions

- Bergmann, E., and W. Schreiber, Ann. 500, 118 1933.
- 2. Bergmann, E., and H. Weiss, Ann. 480, 49 1930.
- 3. Bergmann, E., and H. Weiss, Ann. 480, 59 1930.
- Bergmann, E., and H. Weiss, Ann. 480, 64 1930.
- 5. Blum-Bergmann, O., Ann. 484, 26 1930.
- Kohler, E. P., and W. E. Mydans, J. Am. Chem. Soc. 54, 4667 1932.
- Schlenk, W., and E. Bergmann, Ann. 463, 1 1928.
- Schoepfle, C. S., and J. D. Ryan, J. Am. Chem. Soc. 52, 4021 1930.
- 9. Wieland, II., and H. Kloss, Ann. 470, 201 1929.
- Zal'kind, Y., and A. Kruglov, Ber. 61, 2306 1928.

# 2. FLUORENE WITH TWO PHENYL SUBSTITUTIONS, C<sub>n</sub>H<sub>2n-32</sub>

#### $C_{25}H_{18}$

#### 9,9-Diphenylfluorene



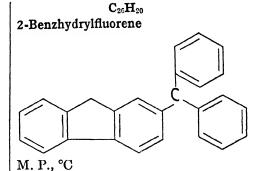
M. P., °C 222-223<sup>16</sup> 222<sup>18</sup>

219-2201

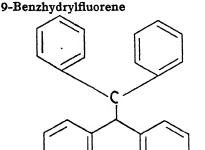
 $D_4^{20}$ 

1.220

0019



147–148<sup>2</sup>



M. P., °C

217-2187. 20 2179, 10, 12

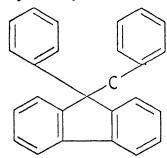
216-2173

187-189 (a)<sup>3</sup>

187 (a)12

(a) This constant was determined on an isomeric form of the compound.

# 9-Phenyl-9-benzylfluorene



M. P., °C 13911, 14 136-1378 136-136.515 1366 125-126 (a)14

(a) This constant was determined on an isomeric form of the compound.

#### 9,9-Dibenzylfluorene

M. P., °C 150-15113 147-14817

#### 1,1-Diphenyl-2-(9'-fluoryl)-ethane

M. P., °C  $107^{12}$ 

#### $C_{27}H_{22}$

#### 3,6-Dibenzylfluorene

M. P., °C 1254,5

#### References on Fluorene with Two Phenyl Substitutions

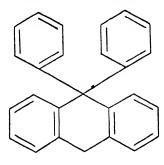
- 1. Bachmann, W. E., J. Am. Chem. Soc. 52, 3287 1930.
- Bachmann, W. E., R. Hoffman, and F. Whitehead, J. Org. Chem. 8, 320 1943.
- 3. Bergmann, E., Ber. 63, 1617 1930.
- Dziewoński, K., M. Dominikowna, L. Galuszkowna, and W. Muz, Roczniki Chem. 13, 283 1933; C.A. 27, 4533 1933; Chem. Zentr. 1933, II, 1524.
- Dziewoński, K., and A. Obtulowicz, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1930A, 399.
- Gomberg, M., and L. H. Cone, Ber. 39, 2957 1906.
- 7. Kaufmann, V., Ber. 29, 73 1896.
- 8. Kliegl, A., Ber. 38, 284 1905.
- 9. Kliegl, A., Ber 64, 2420 1931.
- Klinger, H., and C. Lonnes, Ber. 29, 734 1896.

- Klinger, H., and C. Lonnes, Ber. 29, 2152 1896.
- Schlenk, W., and E. Bergmann, Ann.
   463, 1 1928.
- Schlenk, W., and E. Bergmann, Ann. 479, 58 1930.
- Schlenk, W., and E. Bergmann, Ber.
   62, 745 1929.
- Schmidt, R. E., B. Stein, and C. Bamberger, Ber. 62, 1890 1929.
- Sergeev, P. G., J. Russ. Phys. Chem. Soc. 61, 1421 1929; Chem. Zentr. 1930, II, 391.
- Thiele, J., and F. Heule, Ann. 347, 290 1906.
- Ullmann, F., and R. von Wurstemberger, Ber. 33, 4105 1905.
- 19. Ziegler, K., and F. Ditzel, Ann. 473, 194 1929.
- Ziegler, K., and W. Schafer, Ann. 511, 101 1934.

# 3. DIHYDROANTHRACENES AND DIHYDROPHENANTHRENES WITH TWO PHENYL SUBSTITUTIONS, $C_nH_{2n-32}$

#### C26H20

## 9,9-Diphenyl-9,10-dihydroanthracene



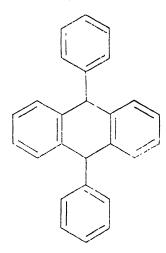
M. P. °C

 $200^{1}$ 

 $195-196^{10}$ 

194-195<sup>2</sup>, 4

9,10-Diphenyl-9,10-dihydroanthracene



M. P., °C 227–228 (a)<sup>7</sup> 208<sup>5</sup>

198.5–199.5 (a)<sup>7</sup>

199 (b)<sup>13</sup>

190 (b)18

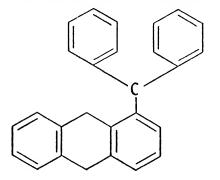
 $164.2^{8, 11}$   $159^{12}$ 

- B. P., °C @ 760mm 437<sup>11</sup>
- (a) These constants were determined on isomeric forms.
- (b) These constants were determined on different crystalline forms.
- 9,10-Diphenyl-9,10-dihydrophenanthrene

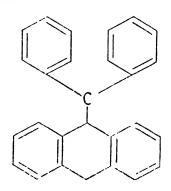
M. P., °C 130–131<sup>13</sup>

 $\boldsymbol{C_{27}H_{22}}$ 

1-Benzhydryl-9, 10-dihydroanthracene

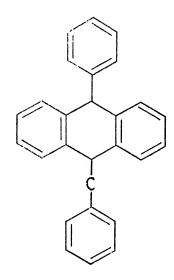


M. P., °C 131–132<sup>14</sup> 9-Benzhydryl-9, 10-dihydroanthracene



M. P., °C 207.5<sup>13</sup>

9-Phenyl-10-benzyl-9, 10-dihydroanthracene



M. P., °C 119<sup>13</sup>

#### 9-Phenyl-10-o-tolyl-9, 10-dihydroanthracene

M. P., °C 186–187<sup>7</sup>

# 2-Methyl-9, 10-diphenyl-9, 10-dihy-droanthracene

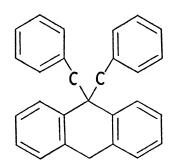
M. P., °C 179<sup>6</sup>

# 9-Methyl-9, 10-diphenyl-9, 10-dihydroanthracene

M. P., °C 171<sup>13</sup>

#### $C_{28}H_{21}$

9,9-Dibenzyl-9,10-dihydroanthracene



M. P., °C 178<sup>3</sup> 115<sup>9</sup>

9,10-Dibenzyl-9,10-dihydroanthracene

M. P., °C 118<sup>3</sup>

- x,x-Dimethyl-9,10-diphenyl-9,10-dihydroanthracene (a)
- M. P., °C 185<sup>12</sup>
- (a) The structure of this compound was not clearly defined in the literature.

#### $C_{30}H_{23}$

x,x-Dimethyl-9,10-dibenzyl-9,10-dihydroanthracene (a)

- M. P., °C 215<sup>12</sup>
- (a) The structure of this compound was not clearly defined in the literature.

References on Dihydroanthracenes and Dihydrophenanthrenes with Two Phenyl Substitutions

- Barnett, E. de B., J. W. Cook, and I. G. Nixon, J. Chem. Soc. 1927, 504.
- 2. Bergmann, E., Ber. 63, 1617 1930.
- 3. Bergmann, E., and S. Fujise, Ann. 480, 189 1930.
- Blicke, F. F., and R. A. Patelski, J. Am. Chem. Soc. 58, 559 1936.
- Dufraisse, C., L. Velluz, and Mme. L. Velluz, Bull. soc. chim. [5] 5, 1073 1938.
- 6. Guyot, A., and C. Staehling, Bull. soc. chim. [3] 33, 1105 1905.
- 7. Haack, E., Ber. 62, 1771 1929.
- Haller, A., and A. Guyot, Bull soc chim. [3] 31, 795 1904.
- 9. Hallgarten, F., Ber. 21, 2508 1888.
- Liebermann, C., and S. Lindenbaum, Ber. 38, 1799 1905.
- Linebarger, C. E., Am. Chem. J. 13. 556 1891.
- 12. Rây, J. N., J. Chem. Soc. 117, 1335 1920.
- Schlenk, W., and E. Bergmann, Ann. 463, 1 1923.
- Scholl, R., and J. Donat, Ann. 512, 1 1934.
- 4. MISCELLANEOUS POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA C<sub>n</sub>II<sub>2n--32</sub>

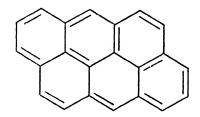
#### $C_{22}H_{12}$

#### Dinaphthylanthrylene (a)

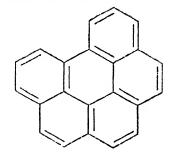
M. P., °C 270<sup>23</sup>

(a) This name was given in the literature with the formula:

#### Dibenzo-[cd,jk]-pyrene (Anthranthrene)



#### Benzo-[ghi]-perylene



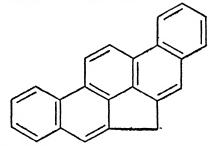
M. P., °C 278-281<sup>12</sup> 277-278.5<sup>12</sup> 273<sup>5, 7</sup> 272<sup>35</sup>

 $C_{23}H_{14}$ 

#### Benzo-[bc]-cholanthrene

M. P., °C 266–267<sup>21</sup>

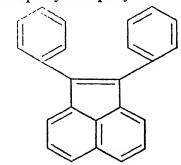
#### Cyclopentano-[ghi]-picene



M. P., °C 277<sup>31</sup>

#### C24H16

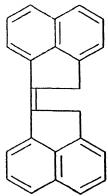
## 1,2-Diphenylacenaphthylene



M. P., °C 164.5-166<sup>24</sup> 161-162<sup>32</sup> 161.3<sup>2</sup> 159.5-161<sup>33</sup>

#### 1,1'-Biacenaphthylidene

(Biacene)



M. P., °C 277<sup>15, 16</sup> 275-276<sup>17</sup> 271-273<sup>18</sup>

#### (1-Naphthyl)-(9'-fluorylidene)methane

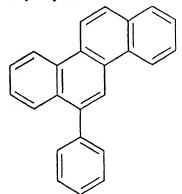
M. P., °C 149-150<sup>28</sup>

## 1,2-Benzo-5-phenylanthracene

M. P., °C 151-152<sup>1</sup> 1,2-Benzo-6-phenylanthracene

M. P., °C 240–241<sup>10</sup>

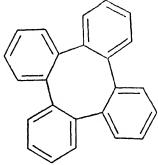
#### 6-Phenylchrysene



M. P., °C 192 -192.5<sup>30</sup>

## 1,2,3,4,5,6,7,8-Tetrabenzocyclooctane

(Tetraphenylene)



M. P., °C 233<sup>26</sup>

#### Tetraphenylene (a)

M. P., °C

 $304 - 305^6$ 

(a) The structure of this compound was not clearly defined in the literature.

#### Indo-[3,4-ab]-benzo-[h]-anthracene

M. P., °C 231 233<sup>11</sup>

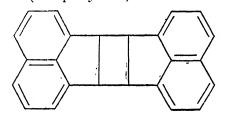
#### x,x-Dihydro-x,x-dibenzopyrene (a)

M. P., °C

 $313.5 \cdot 314.5^{31}$ 

(a) The structure of this compound was not clearly defined in the literature.

#### 1,2,3,4-Di-(2',1'-acenaphtheno)cyclobutane (α-Heptacyclene)



M. P., °C 306-307<sup>19, 20</sup>

#### $\beta$ -Heptacyclene (a)

M. P., °C

 $234^{20}$ 

 $232 - 234^{19}$ 

(a) The structure of this compound was not clearly defined in the literature.

#### $C_{25}H_{13}$

#### 1,2-Benzo-10-benzylanthracene

M. P., °C 195–196<sup>10</sup>

This series continued on next page

#### 2-Benzylchrysene

#### Dibenzo-[a,jk]-4-ethyl-10-hydropyrene (a)

#### M. P., °C 116.5–117.5<sup>13</sup>

(a) The structure of this compound was not clearly defined in the literature.

### $\mathbf{C}_{26}\mathbf{H}_{20}$

#### 1,2-Di-p-tolylacenaphthylene

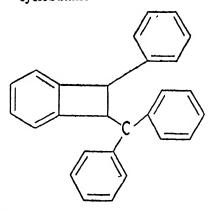
M. P., °C

#### 1-Phenyl-7-(9'-fluorylidene)-heptatriene-1,3,5

M. P., °C 166<sup>25</sup>

#### C27H22

#### 1-Phenyl-2-benzhydryl-3, 4-benzocyclobutane



M. P., °C 184<sup>27</sup>

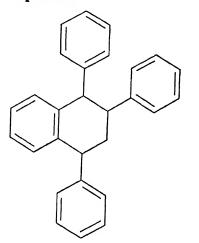
#### $C_{28}H_{24}$

#### 1,2,3-Triphenyl-1,2,3,4-tetrahydronaphthalene

M. P., °C 136–137³

This series continued on next page

#### 1,2,4-Triphenyl-1,2,3,4-tetrahydronaphthalene



M. P., °C 127⁴

References on Miscellaneous Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-32}$ 

- Bachmann, W. E., and J. T. Bradbury, J. Org. Chem. 2, 175 1937.
- Bachmann, W. E., and E. J.-H. Chu, J. Am. Chem. Soc. 58, 1118 1936.
- Bergmann, E., and O. Zwecker, Ann. 487, 155 1931.
- 4. Blum, O., Ber. 62, 881 1929.
- Brass, K., and K. Fanta, Ber. 69, 1 1936.
- Chandra-Sircar, A., and J. N. Majumder, J. Indian Chem. Soc. 5, 417 1928.
- 7. Clar, E., Ber. 65, 846 1932.
- 8. Clar, E., Ber. 72, 1645 1939.
- 9. Clar, E., Ber. 76, 328 1943.
- 10. Cook, J. W., J. Chem. Soc. 1930, 1087.
- 11. Cook, J. W., J. Chem. Soc. 1931, 499.
- Cook, J. W., and N. Percy, J. Soc. Chem. Ind. 64, 27 1945.
- Cook, J. W., and A. M. Robinson, J. Chem. Soc. 1940, 303.

- Corbellini, A., and F. Steffenoni, Rend. ist. lombardo sci. [2] 69, 429 1936; Chem. Zentr. 1937, I, 1420.
- Doliński, J., and K. Dziewoński, Ber. 48, 1917 1915.
- Doliński, J., and K. Dziewoński, Bull. intern. acad. sci. Cracovie 1914 A, 389.
- Dziewoński, K., and L. Gizler, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1937 A, 441.
- Dziewoński, K., and T. Lityński, Ber. 58, 2539 1925.
- Dziewoński, K., and C. Paschalski, Ber. 46, 1986 1913.
- Dziewoński, K., and G. Rapalski, Ber. 45, 2491 1912.
- Fieser, L. F., and E. B. Seligman, J. Am. Chem. Soc. 57, 1681 1935.
- Funke, K., and E. Muller, J. prakt. Chem. [2] 144, 242 1936.
- 23. Grabowski, J., Ber. 11, 298 1878.
- Koelsch, C. F., and H. J. Richter, J. Am. Chem. Soc. 59, 2165 1937.
- Kuhn, R., and A. Winterstein, Helv. Chim. Acta 11, 116 1928.
- Rapson, W. S., R. G. Shuttleworth, and J. N. van Niekerk, J. Chem. Soc. 1943, 326.
- Schlenk, W., and E. Bergmann, Ann. 463, 1 1928.
- Schlenk, W., and E. Bergmann, Ann. 479, 42 1930.
- Scholl, R., and K. Meyer, Ber. 67, 1229 1934.
- Simpson, J. C. E., J. Chem. Soc. 1943, 447.
- Waldmann, H., and G. Pitschak, Ann. 527, 183 1937.
- Wittig, G., and K. Henkel, Ann. 542, 130 1939.
- Wittig, G., M. Leo, and W. Wiemar, Ber. 64, 2405 1931.
- Zil'berman, G. B., J. Gen. Chem. (U. S. S. R.) 7, 234 1937; C.A. 31, 4314 1937;
   Chem. Zentr. 1937, I, 4787.
- Zinke, A., U. Noculak, R. Skrabal, and H. Troger, Ber. 73, 1187 1940.

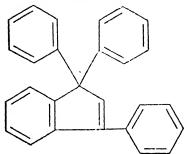


- 1. Indene with Three Phenyl Substitutions
- 2. Anthracene or Phenanthrene with Two Phenyl Substitutions
- 3. Miscellaneous Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-34</sub>

#### 1. INDENE WITH THREE PHENYL SUBSTITUTIONS, C<sub>n</sub>H<sub>2n-84</sub>

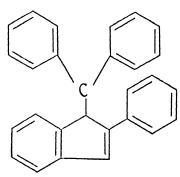
#### $C_{27}H_{20}$

#### 1,1,3-Triphenylindene



M. P., °C 135<sup>12, 16</sup> 134–135<sup>15</sup> 132–134<sup>2, 9</sup> 133<sup>17</sup>

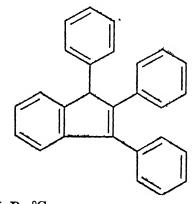
# 1-Benzhydryl-2-phenylindene



 $C_{28}H_{22}$ 

M. P., °C 1754

#### 1,2,3-Triphenylindene



M. P., °C

135<sup>12, 18, 17</sup>

132–134<sup>6</sup>

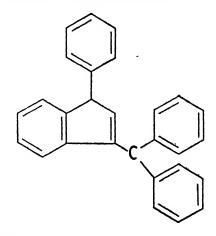
130–132<sup>9</sup>

D<sub>4</sub><sup>20</sup>

1.176

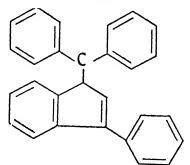
0°18

### 1-Phenyl-3-benzhydrylindene



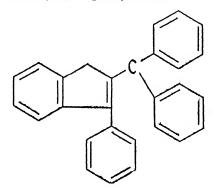
M. P., °C 131<sup>15</sup> 124–125<sup>4</sup>

## 1-Benzhydryl-3-phenylindene



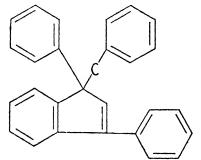
M. P., °C 174.5-175<sup>4</sup> 171<sup>15</sup>

#### 2-Benzhydryl-3-phenylindene

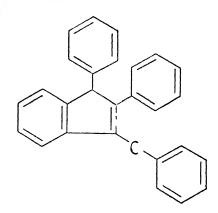


M. P., °C 162–163.5<sup>10</sup>

## 1-Benzyl-1, 3-diphenylindene

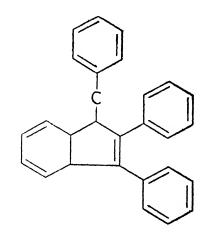


#### 1,2-Diphenyl-3-benzylindene



M. P., °C 143-144<sup>1</sup> 118.5-119.5<sup>6</sup>

## 1-Benzyl-2,3-diphenylindene



M. P., °C 118.5–120<sup>5</sup>

#### 1,2-Diphenyl-3-p-tolylindene

M. P., °C 116-117<sup>7</sup>

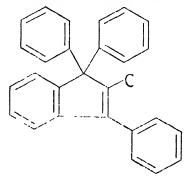
## 1-p-Tolyl-2,3-diphenylindene

M. P., °C 154–156<sup>7</sup>

#### 1-Methyl-1,2,3-triphenylindene

M. P., °C 96-98<sup>11</sup>

#### 1,1,3-Triphenyl-2-methylindene



M. P., °C 157-159.5<sup>11</sup> 150<sup>3</sup>

#### x-Methyl-x,x,x-triphenylindene (a)

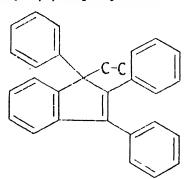
M. P., °C

11814

(a) The structure of this compound was not clearly defined in the literature.

#### $C_{29}H_{24}$

#### 1-Ethyl-1,2,3-triphenylindene



M. P., °C 108<sup>12</sup>

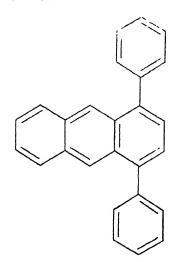
#### References on Indene with Three Phenyl Substitutions

- Bergmann, E., and W. Schreiber, Ann. 500, 118 1933.
- Bergmann, E., and H. Weiss, Ann. 480, 49 1930.
- 3. Bergmann, E., and H. Weiss, Ann. 480, 61 1930.
- 4. Blum-Bergmann, O., Ann. 484, 26 1930.
- 5. Blum-Bergmann, O., Ber. 65, 109 1932.
- Koelsch, C. F., J. Am. Chem. Soc. 54, 2045 1932.
- Koelsch, C. F., J. Am. Chem. Soc. 56, 1337 1934.
- Koelsch, C. F., J. Am. Chem. Soc. 56, 1605 1934.

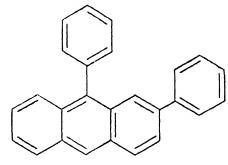
- Koelsch, C. F., and P. R. Johnson, J. Am. Chem. Soc. 65, 567 1943.
- Koelsch, C. F., and R. V. White, J. Am. Chem. Soc. 65, 1639 1943.
- Koelseh, C. F., and R. V. White, J. Org. Chem. 6, 602 1941.
- 12. Kohler, E. P., Am. Chem. J. 40, 217 1908.
- Kohler, E. P., and W. E. Mydans, J. Am. Chem. Soc. 54, 4667 1932.
- Kohler, E. P., and E. M. Nygaard, J. Am. Chem. Soc. 52, 4128 1930.
- Schlenk, W., and E. Bergmann, Ann. 463, 1 1928.
- 16. Ziegler, K., Ann. 434, 34 1923.
- Ziegler, K., and F. Crossmann, Ber. 62, 1768 1929.
- Ziegler, K., and F. Ditzel, Ann. 473, 194 1929.

# 2. ANTHRACENE OR PHENANTHRENE WITH TWO PHENYL SUBSTITUTIONS, $C_nH_{2n-34}$

#### C<sub>.6</sub>H<sub>18</sub> 1,4-Diphenylanthracene



#### 2,9-Diphenylanthracene



M. P., °C 165–166<sup>25</sup>

This series continued on next page.

#### 9,10-Diphenylanthracene

M. P., °C

247

250-25119

249-25023, 26

24817

247-24816, 23

24725, 34

24638

245-24629

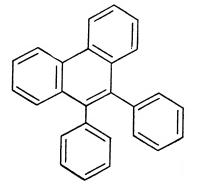
242-24328

241-24311

 $214 (a)^{34}$ 

(a) This constant was determined on an isomeric form of the compound.

#### 9,10-Diphenylphenanthrene



M. P., °C 236

24035

23718, 30

235-235.514

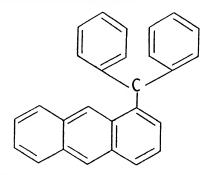
23536

23412, 15

233-23440

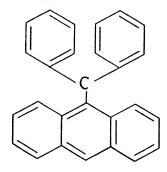
#### $C_{27}H_{20}$

#### 1-Benzhydrylanthracene



M. P., °C 173–175³7

#### 9-Benzhydrylanthracene



M. P., °C 204–2054

#### 9-Phenyl-10-benzylanthracene

M. P., °C 155<sup>7</sup> 154<sup>27</sup> 151<sup>34</sup>

## 9-Phenyl-10-o-tolylanthracene

M. P., °C 261–262<sup>23</sup> 257–258<sup>34</sup> 172–173 (a)<sup>34</sup>

(a) This constant was determined on an isomeric form of the compound.

#### 9-Phenyl-10-m-tolylanthracene

M. P., °C 182–183<sup>34</sup>

#### 9-Phenyl-10-p-tolylanthracene

M. P., °C 192<sup>22</sup>

#### 1-Methyl-9, 10-diphenylanthracene

M. P., °C 194<sup>34</sup>

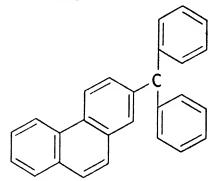
#### 2-Methyl-9, 10-diphenylanthracene

M. P., °C 242-243<sup>20</sup> 214<sup>34</sup> 213<sup>21</sup>

#### 1-Benzhydrylphenanthrene

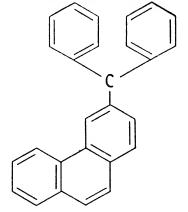
M. P., °C 175–176<sup>3</sup>

#### 2-Benzhydrylphenanthrene



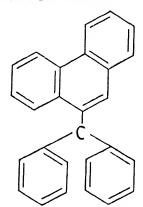
M. P., °C 151–152³

#### 3-Benzhydrylphenanthrene



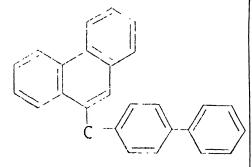
M. P., °C 122–123²

#### 9-Benzhydrylphenanthrene



M. P., °C 175–176<sup>1</sup> 174--175<sup>3</sup>

#### p-Biphenylyl-9-phenanthrylmethane



M. P., °C 192–193³

#### $C_{28}H_{22}$

#### 9,10-Dibenzylanthracene

M. P., °C
242
245<sup>6</sup>
243-245<sup>2</sup>
211-242<sup>13</sup>
211<sup>31</sup>
240<sup>13</sup>
239-240<sup>31, ©2</sup>

#### 9,10-Di-o-tolylanthracene

M. P., °C 347–348<sup>41</sup>

#### 9,10-Di-m-tolylanthracene

M. P., °C 22241

#### 9,10-Di-p-tolylanthracene

M. P., °C 279<sup>41</sup> 278–279<sup>26</sup>

#### 1,4-Dimethyl-9,10-diphenylanthracene

M. P., °C 1899

#### 9,10-Dibenzylphenanthrene

M. P., °C 180–1824²

 $C_{29}H_{24}$ 

1,3-Dimethyl-9-phenyl-10-benzylanthracene

M. P., °C 1378

2,3-Dimethyl-9-phenyl-10-benzylanthracene

M. P., °C 16310 C<sub>30</sub>H<sub>26</sub>

9,10-Di-(2',4'-dimethylphenyl)anthracene

M. P., °C 290<sup>17</sup>

2,3,6,7-Tetramethyl-9,10-diphenylanthracene

M. P., °C 312<sup>5</sup> 284–285<sup>33</sup>

#### 1-Methyl-7-isopropyl-x,x-diphenylphenanthrene (a)

- M. P., °C 200<sup>24</sup>
- (a) The structure of this compound was not clearly defined in the literature.

# References on Anthracene or Phenanthrene with Two Phenyl Substitutions

- Bachmann, W. E., J. Am. Chem. Soc. 56, 1363 1934.
- Bachmann, W. E., and J. M. Chemerda, J. Org. Chem. 4, 583 1939.
- Bachmann, W. E., and M. C. Kloetzel,
   J. Org. Chem. 2, 356 1937-38.
- Bachmann, W. E., and M. C. Kloetzel, J. Org. Chem. 3, 55 1938-39.
- Barnett, E. de B., J. Chem. Soc. 1939, 348.
- Barnett, E. de B., and J. W. Cook, J. Chem. Soc. 1928, 566.
- Barnett, E. de B., J. W. Cook, and J. L. Wiltshire, J. Chem. Soc. 1927, 1724.
- 8. Barnett, E. de B., and C. L. Hewett, Ber. 64, 1572 1931.
- Barnett, E. de B., and J. A. Low, Ber. 64, 49 1931.
- Barnett, E. de B., and F. C. Marrison, Ber. 64, 535 1931.
- Barnett, E. de B., and M. A. Mathews, Ber. 59, 1429 1926.
- 12. Bergmann, E., Ber. 63, 1617 1930.
- Bergmann, E., and S. Fujise, Ann. 480, 188 1930.
- 14. Biltz, H., Ber. 38, 203 1905.
- Bradsher, C. K., and R. Rosher, J. Am. Chem. Soc. 61, 1524 1939.
- Bradsher, C. K., and E. S. Smith, J. Am. Chem. Soc. 65, 451 1943.
- 17. Clar, E., Ber. 64, 2194 1931.
- 18. Dufraisse, C., and R. Priou, Bull. soc. chim. [5] 5, 611 1938.
- Dufraisse, C., L. Velluz, and Mme. L. Velluz, Bull. soc. chim. [5] 5, 1073 1938.

- Duveen, D., and A. Willemart, J. Chem. Soc. 1939, 116.
- Guyot, A., and C. Staehling, Bull. soc. chim. [3] 33, 1105 1905.
- Guyot, A., and F. Vallette, Ann. chim. phys. [8] 23, 363 1911.
- 23. Haack, E., Ber. 62, 1771 1929.
- Heiduschka, A., and H. Grimm, Arch. Pharm. 250, 33 1912.
- 25. Hirshberg, Y., and L. Haskelberg, Trans. Faraday Soc. 39, 45 1943.
- Ingold, C. K., and P. G. Marshall, J. Chem. Soc. 1923, 3080.
- Julian, P., and W. Cole, J. Am. Chem. Soc. 57, 1607 1935.
- Kehrmann, F., R. Monier, and M. Ramm, Ber. 56, 169 1923.
- Koelsch, C. F., J. Org. Chem. 3, 456
   1938.
- Lettré, H., H. Barnbeck, and W. Lege, Ber. 69, 1151 1936.
- Lippman, E., and R. Fritsch, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 113, Abt. IIb, 429 1904.
- Lippman, E., and I. Pollack, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 111, Abt. 11b, 402 1902.
- Muller, E., and E. Topel, Ber. 72, 273
   1939.
- Schlenk, W., and E. Bergmann, Ann. 463, 1 1928.
- Schmidliu, J., and R. von Escher, Ber.
   43A, 1153 1910.
- Schoepfle, C. S., and J. D. Ryan, J. Am. Chem. Soc. 54, 3687 1932.
- 37. Scholl, R., and J. Donat, Ann. 512, 1 1934.
- Simonis, H., and P. Remmert, Ber. 48, 206 1915.
- Weizmann, C., E. Bergmann, and L. Haskelberg, J. Chem. Soc. 1939, 391.
- Werner, A., and A. Grob, Ber. 37, 2887
   1904.
- 41. Willemart, A., Bull. soc. chim. [5] 4, 510 1937.
- 42. Willgerodt, C., and B. Albert, J. prakt. Chem. [2] 84, 383 1911.

# 3. MISCELLANEOUS POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-34}$

#### C24H14

#### Di-(1-naphthyl)-butadiyne

#### 1,2-(2',1'-Acenaphtho)-anthracene

M. P., °C 290–291<sup>63</sup>

17133

#### Benzo-[a]-indo-[3,2,1-kl]-anthracene

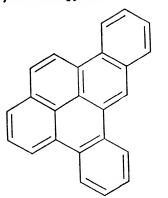
M. P., °C 178–179<sup>29</sup>

#### 1,2-Benzocholanthrylene

M. P., °C 181-181.3<sup>29</sup> 180.2-180.6<sup>76</sup>

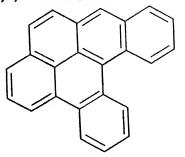
#### 1,2-(3',2'-Naphtho)-pyrene

#### 1,2,4,5-Dibenzopyrene



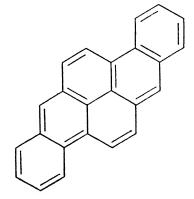
M. P., °C 238–239<sup>31</sup> 225<sup>21</sup>

#### 2,3,4,5-Dibenzopyrene



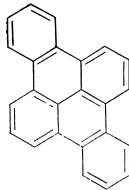
M. P., °C  $226-227^{16}$ 

### 1,2,6,7-Dibenzopyrene



M. P., °C 315<sup>66</sup> 308<sup>18</sup>

# 4,5,9,10-Dibenzopyrene



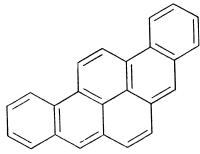
 $\begin{array}{c} \text{M. P., °C} \\ 353-355^{54} \\ 340-342^{21} \end{array}$ 

# x,x,-Dibenzopyrene (a)

M. P., °C 320–320.5<sup>77</sup>

(a) The structure of this compound was not clearly defined in the literature.

# Benzo-[ghi]-picene



M. P., °C 281.5–282<sup>58</sup> 280<sup>18</sup>

 $C_{25}H_{16}$ 

#### Spiro[fluorene-9,9'-fluorene]

M. P., °C 198–199<sup>22</sup>

#### 1,2-Benzo-5,6-(2',1'-naphtho)fluorene

M. P., °C 286–287<sup>14</sup>

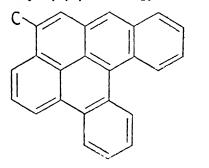
#### 1,2-Benzo-5,6-(3',2'-naphtho)fluorene

M. P., °C 225<sup>14</sup> This series continued on next page

#### 1,2-Benzo-6,7-(3',2'-indo)-anthracene

M. P., °C 302–304<sup>24</sup>

#### 9-Methyl-2,3,4,5-dibenzopyrene



M. P., °C 225–226<sup>15</sup>

#### $C_{26}H_{18}$

# ${\tt 1,2-} Dibenzyl ideneace na ph thene$

M. P., °C 14246

### 9,9'-Bifluoryl

 $\begin{array}{c} \text{M. P., °C} \\ \textbf{246.5} \\ \textbf{249^{44}} \\ \textbf{246-247^{43, 69, 70}} \\ \textbf{246^{6, 28, 30}} \\ \textbf{215^{51}} \\ \textbf{241-245^{49, 74}} \\ \textbf{242-241^{25}} \end{array}$ 

241<sup>47, 61</sup>
240<sup>39, 55, 59</sup>

### 9-Benzhydrylidenefluorene

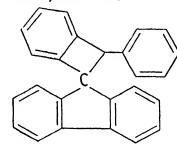
 $\begin{array}{c} \text{M. P., °C} \\ \textbf{229} \\ 229.5^{37, \ 38, \ 40} \\ 229-229.5^{63} \\ 228^{60} \\ 225^{32} \\ 224-225^{55} \\ 223.5-224.5^{76} \\ 213 \ (a)^{55} \end{array}$ 

(a) This constant was determined on a stereoisomer of the compound.

1-(1'-Naphthyl)-3-(9''-fluorylidene)propene-1

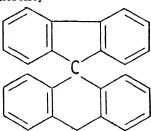
M. P., °C 174–175<sup>56</sup>

Spiro[2-phenyi-3,4-benzocyclobutane-1,9'-fluorene]



M. P., °C 23541

Spiro[9,10-dihydroanthracene-9,9'-fluorene]



M. P., °C 207<sup>22</sup>

### Spiro[9,10-dihydrophenanthrene-9,9'-fluorene]

M. P., °C 195<sup>62</sup>

### x,x-Dihydrohexacene (a)

M. P., °C 369-370<sup>45</sup> 357-358<sup>19</sup>

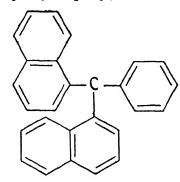
(a) The hydrogens are in either the 5, 16- or the 6, 15-positions.

### 1,2,3,4,5,6-Tribenzo-9,10-dihydroanthracene

M. P., °C 281–283<sup>1</sup>

C27H20

### Phenyldi-(1'-naphthyl)-methane



M. P., °C 288<sup>36</sup> 204<sup>57</sup>

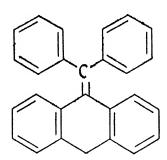
### 9-Fluorylbenzhydrylidenemethane

M. P., °C 111–112<sup>55</sup>

This series continued on next page

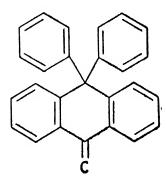
### 2,2'-Difluorylmethane

### 9-Benzhydrylidene-9, 10-dihydroanthracene



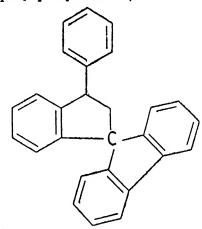
M. P., °C 256<sup>50</sup>

### 9,9-Diphenyl-10-methylene-9,10dihydroanthracene



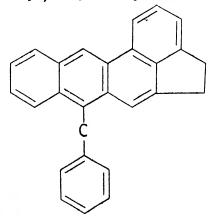
M. P., °C 192<sup>2</sup> 187<sup>5</sup>

### Spiro[3-phenylindane-1,9'-fluorene]



M. P., °C 125–127<sup>42</sup>

### Indo-[4',3'-ab]-10-benzylanthracene



M. P., °C 199–200<sup>28</sup>

### $C_{28}H_{22}$

1,4-Di-(3'-methylindenylidenemethyl)-benzene

M. P., °C 224<sup>48</sup>

### 1-Phenyl-3-benzhydrylideneindane

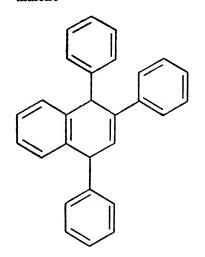
M. P., °C 130-131.5 (a)<sup>11</sup> 130-131 (a)<sup>11</sup> 115-117 (a)<sup>11</sup> 115 (a)<sup>11</sup> 115<sup>71</sup>

(a) These constants were determined on isomeric forms.

### 1,2,3-Triphenyl-1,4-dihydronaphthalene

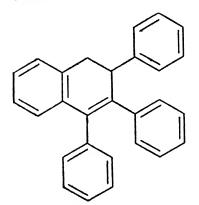
M. P., °C 1658

### 1,2,4-Triphenyl-1,4-dihydronaphthalene



M. P., °C 142.5 - 144<sup>10</sup>

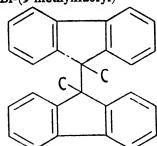
### 2,3,4-Triphenyl-1,2-dihydronaphthalene



B. P., °C @ 760mm 215

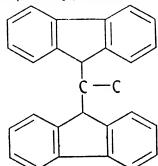
 $0.5^{9}$ 

### 9,9'-Bi-(9-methylfluoryl)



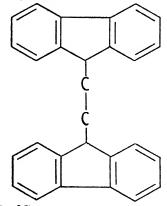
M. P., °C 209<sup>64, 65</sup> 206<sup>72</sup>

### 1,1-Di-(9'-fluoryl)-ethane



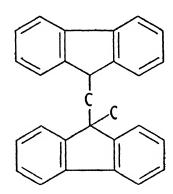
M. P., °C 262-263<sup>52</sup>

### 1,2-Di-(9'-fluoryl)-ethane



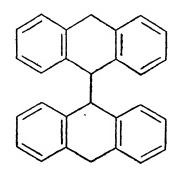
M. P., °C 224–225<sup>72, 73</sup> 218–219<sup>7</sup>

### 9-Fluoryl-[9'-(9'-methylfluoryl)]methane

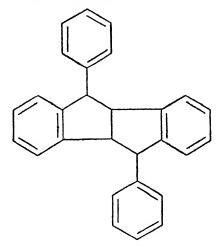


M. P., °C 171–171.5<sup>72</sup>

### Bi-9,9'-(9,10-dihydroanthryl)



M. P., °C 255<sup>55</sup> 248–249<sup>53</sup> 2,3,6,7-Dibenzo-4,8-diphenylbicyclo-[3,3,0]-octane



 $\begin{array}{c} \text{M. P., °C} \\ 207\text{--}208 \text{ (a)}^{12} \\ 166\text{--}167 \text{ (a)}^{12} \end{array}$ 

(a) These constants were determined on isomeric forms.

This series continued on next page

### Dibenzo-[de, op]-1,2,3,8,9,10-hexahydropentacene

M. P., °C 255–256<sup>20</sup>

#### C29H24

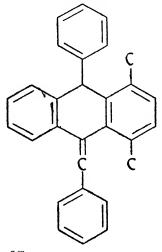
### x-Benzyl-x-benzhydrylindene (a)

M. P., °C 137–138 (b)<sup>75</sup>

- 130 131 (b) 75

  (a) The structure of this compound was not clearly defined in the literature.
- (b) These constants probably represent two different compounds.

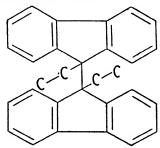
### 1,4-Dimethyl-9-phenyl-10-benzylidene-9,10-dihydroanthracene



M. P., °C 170³

#### $C_{30}H_{26}$

### 9,9'-Bi-(9-ethylfluoryl)



M. P., °C 210<sup>64, 65</sup>

This series continued on next page

2,3,6,7-Dibenzo-4,8-dibenzylbicy-clo-[3,3,0]-octane

M. P., °C 141<sup>13</sup>

2,3,6,7-Dibenzo-4,8-di-m-tolylbicyclo-[3,3,0]-octane

M. P., °C 150<sup>12</sup> 2,3,6,7-Dibenzo-4,8-di-p-tolylbicyclo-[3,3,0]-octane

M. P., °C 188–189 (a)<sup>12</sup> 145–146 (a)<sup>12</sup>

(a) These constants were determined on isomeric forms.

 $C_{32}H_{30}$ 

9,9-Bi-(10,10-dimethyl-9,10-dihydro-anthryl)

M. P., °C 3154

#### C34H34

1,2-Di-[2'-(1'-phenyl-1',2',3',4'-tetrahydronaphthyl)]-ethane

M. P., °C 169.5-170.5<sup>34</sup>

#### C36H38

1,2-(2',1'-Acenaphtho) -3,4,5,6-di-[2'',1''-(2a'',3'',4'',5'',5a'',6'', 7'',8'',8a'',8b''-decahydroacenaphtho)]-benzene

M. P., °C 245<sup>67</sup>

References in Miscellaneous Polynuclear Aromatics of Empirice! Formula  $C_nH_{2n-24}$ 

- Bachmann, W. E., and L. H. Pence, J. Am. Chem. Soc. 59, 2339 1937.
- Barnett, E. de B., and J. W. Cook, J. Chem. Soc. 1928, 566.

- Barnett, E. de B., and J. A. Low, Ber. 64, 49 1931.
- 4. Barnett, E. de B., and M. A. Matthews, Ber. 59, 767 1926.
- Bergmann, E., and A. Bondi, Ber. 66, 286
   1933.
- Bergmann, E., and S. Fujise, Ann. 483, 65 1930.
- Bergmann, E., H. Hoffmann, and D. Winter, Ber. 66, 46 1933.
- 8. Bergmann, E., and O. Zwecker, Ann. 487, 155 1931.
- Bergmann, F., D. Schapiro, and H. E. Eschinazi, J. Am. Chem. Soc. 64, 559 1942.
- 10. Blum, O., Ber. 62, 881 1929.
- 11. Blum-Bergmann, O., Ann. 484, 26 1930.
- Brand, K., and W. Mühl, J. prakt. Chem. [2] 110, 1 1925.
- Brand, K., and K. O. Müller, Ber. 55, 601 1922.
- Buu-Hoï and P. Cagniant, Rev. sci. 80, 381 1945; C. A. 39, 3277 1945.
- Clar, E., "Aromatische Kohlenwasserstoffe," Springer-Verlag, Berlin, 1941, p. 213.
- 16. Clar, E., Ber. 63, 112 1930.
- 17. Clar, E., Ber. 69, 1671 1936.
- 18. Clar, E., Ber. 72, 1645 1939.
- 19. Clar, E., Ber. 72, 1817 1939.
- 20. Clar, E., Ber. 73, 409 1940.
- 21. Clar, E., Ber. 76, 609 1943.
- Clarkson, R. G., and M. Gomberg, J. Am. Chem. Soc. 52, 2881 1930.

- 23. Cook, J. W., J. Chem. Soc. 1930, 1087.
- 24. Cook, J. W., J. Chem. Soc. 1931, 499.
- Cook, J. W., C. L. Hewett, and I. Ilieger, J. Chem. Soc. 1933, 395.
- Davis, W. W., M. E. Krahl, and G. H. A. Clowes, J. Am. Chem. Soc. 62, 3080 1940.
- Dziewoński, K., and M. Panek, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1927A, 745.
- Eckert, A., J. prakt. Chem. [2] 121, 278
   1929.
- Fieser, L. F., and A. M. Seligman, J. Am. Chem. Soc. 57, 2174 1935.
- 30. Graebe, C., and H. Stindt, Ann. 291, 1 1896.
- Grechkin, N. P., and A. E. Arbusov, Compt. rend. acad. sci. U. R. S. S. 32, 50 1941; Survey For. Petrol. Liter. No. 342, Jan. 16, 1942.
- 32. Grignard, V., and C. Courtot, Compt. rend. 152, 1493 1911.
- Grignard, V., and H. Perrichon, Ann. chim. [10] 5, 5 1926.
- 31. Hewett, C. L., J. Chem. Soc. 1936, 596.
- 35. Ingold, C. K., and J. A. Jessop, J. Chem. Soc. 1930, 713.
- Kashtanov, L. I., J. Gen. Chem. (U. S. S. R.) 4, 1204 1934; C. A. 29, 3328 1935.
- 37. Kaufmann, V., Ber. 29, 73 1896.
- 38. Kaufmann, V., Chem. Ztg. 19, 1033 1895.
- 39. Kliegl, A., Ber. 62, 1327 1929.
- 40. Klinger, H., and C. Lonnes, Ber. 29, 734 1896.
- 41. Klinger, II., and C. Lonnes, Ber. 29, 2152 1896.
- 42. Koelsch, C. F., J. Am Chem. Soc. 55, 3394 1933.
- Maitland, P., and S. H. Tucker, J. Chem. Soc. 1929, 2559.
- 44. Manchot, W., and P. Krische, Ann. 337, 170 1904.
- 45. Marschalk, C., Bull. soc. chim. [5] 6, 1112 1939.
- Maxim, N., Bull. soc. chim. [4] 45, 1137
   1929.
- 47. Mayer, F., Ber. 46, 2579 1913.
- Mayer, F., A. Sieglitz, and W. Ludwig, Ber. 54, 1397 1921.
- Miller, H. F., and J. Bachman, J. Am. Chem. Soc. 57, 766 1935.
- Nenitzescu, C. D., I. Gavât, and D. Cocora, Ber. 72, 819 1939.

- Oliveri-Mandalá, E., G. Caronna, and E. Deleo, Gazz. chim. ital. 68, 327 1938.
- Pummerer, R., and G. Dorfmüller, Ber. 46, 2386 1913.
- 53. Sachse, H., Ber. 21, 2512 1888.
- Sako, S., Bull. Chem. Soc. Japan 9, 55
   1934.
- Schlenk, W., and E. Bergmann, Ann. 463, 1 1928.
- Schlenk, W., and E. Bergmann, Ann. 479, 42 1930.
- Schoepfle, C. S., J. Am. Chem. Soc. 44, 188 1922.
- Scholl, R., and H. Neumann, Ber. 55, 118 1922.
- Schönberg, A., A. Stephenson, H. Kaltschmitt, E. Peterson, and H. Schulten, Ber. 66, 237 1933.
- Staudinger, H., and N. Kon, Ann. 384, 38 1911.
- Stollé, R., H. Munzel, and F. Wolf, Ber. 46, 2339 1913.
- Suszko, J., and R. Schillak, Roczniki Chem. 14, 1216 1934; C. A. 29, 6231 1935; Chem. Zentr. 1935, I, 2361.
- Szperl, L., and T. Wierusz-Kowalski, Chem. Polski 15, 19, 23, 28; Chem. Zentr. 1918, I, 908.
- Vansheidt, A., and B. Moldavski, Ber. 64, 917 1931.
- Vansheidt, A., and B. Moldavski, J. Gen. Chem. (U. S. S. R.) 1, 304 1931;
   Chem. Zentr. 1931, II, 3208.
- 66. Vollmann, H., H. Becker, M. Corell, and H. Streeck, Ann. 531, 1 1937.
- 67. von Braun, J., Ber. 67, 214 1934.
- 68. von Braun, J., and G. Manz, Ann. 496, 170 1932.
- 69. Weissgerber, R., Ber. 34, 1659 1901.
- 70. Weissgerber, R., Ber. 41, 2913 1908.
- 71. Wieland, II., and II. Kloss, Ann. 470, 201 1929.
- 72. Wieland, H., and O. Probst, Ann. 530, 274 1937.
- 73. Wislicenus, W., Ber. 48, 617 1915.
- Wislicenus, W., and K. Russ, Ber. 43, 2719 1910.
- 75. Wuest, H.-M., Ann. 415, 291 1918.
- 76. Zedlitz, O., Ber. 64, 2424 1931.
- Zilberman, G. B., J. Gen. Chem. (U. S. S. R.) 7, 234 1937; C. A. 31, 4314 1937;
   Chem. Zentr. 1937, I, 4787.



### XVI. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA C<sub>n</sub>H<sub>2n-26</sub>

### $C_{24}H_{12}$

#### Coronene

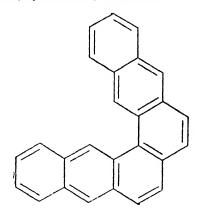
M. P., °C 435<sup>35</sup> 429–430<sup>68</sup>

C26H16

### 9,9'-Bifluorylidene

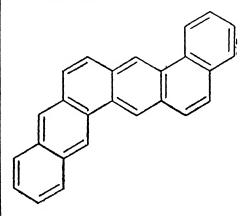
18441, 63

### 1,2-(4',3'-Anthro)-anthracene



M. P., °C 308<sup>24</sup> 304<sup>26</sup>

### 1,2-Benzo-5,6-(3',2'-naphtho)anthracene



M. P., °C 281-282<sup>25</sup> 281<sup>82</sup>

### 1,2,3,4,5,6-Tribenzoanthracene

### 1,2,7,8-Dibenzonaphthacene

M. P., °C 345<sup>28</sup>

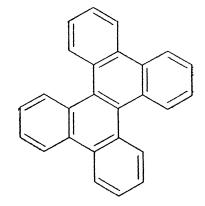
### 1,2,9,10-Dibenzonaphthacene

### 2,3-(4',3'-Naphtho)-chrysene (3,4,8,9-Dibenztetraphene)

### 2,3,8,9-Dibenzochrysene

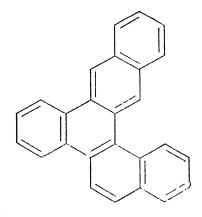
M. P., °C 389<sup>26</sup>

### 5,6,11,12-Dibenzochrysene



M. P., °C 218<sup>76</sup> 215<sup>8</sup>

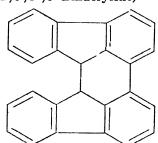
### 2,3,5,6-Dibenzotriphenylene



M. P., °C 185–186<sup>13</sup>

### Difluoro-[1,9-ab,8',9'-de]-cyclohexane

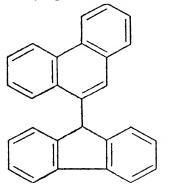
(1,9,8',9'-Bifluorylene)



M. P., °C 218<sup>31</sup>, <sup>33</sup>

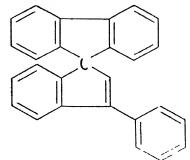
### $C_{27}H_{18}$

9-(9'-Fluoryl)-phenanthrene



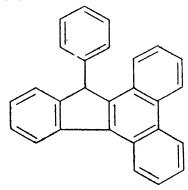
M. P., °C 196–197¹

### Spiro[3-phenylindene-1,9'-fluorene]



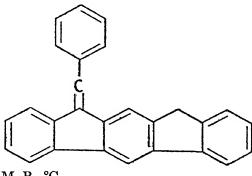
M. P., °C 194–197<sup>47</sup> 140–141<sup>45</sup>

### 1,2,3,4-Dibenzo-9-phenylfluorene



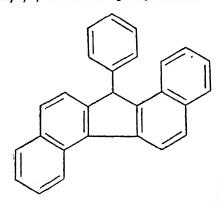
M. P., °C 210–211<sup>1</sup>· <sup>46</sup> 209<sup>48</sup>

2,3-(2',3'-Indo)-9-benzylidenefluorene



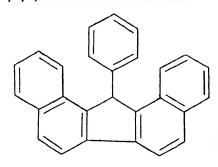
M. P., °C

### 1,2,5,6-Dibenzo-9-phenylfluorene



M. P., °C 219-219.5<sup>12</sup>

### 1,2,7,8-Dibenzo-9-phenylfluorene



M. P., °C 273<sup>67</sup> 148.5–149.5<sup>12</sup>

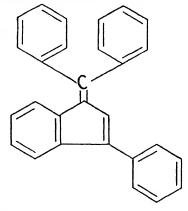
1,2,3,4,5,6-Tri-(3',2'-indo)benzene (Truxene) (a)

M. P., °C 391<sup>71</sup> 369-370<sup>74</sup> 368<sup>2</sup> 365<sup>78</sup> 364-365<sup>30</sup> 330<sup>73</sup>

(a) This compound was thought by some investigators to be 2,3,2', 3'-Biindenylene or 2,3,3',2'-Biindenylene.

### C28H20

### 1-Benzhydrylidene-3-phenylindene



M. P., °C 207–208<sup>17</sup> 207<sup>18, 19</sup> 205-206<sup>80, 87, 88</sup>

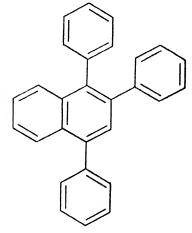
### 1-Benzylidene-2,3-diphenylindene

M. P., °C 184-185<sup>58, 59, 61</sup>

### 1,2,3-Triphenylnaphthalene

 $\begin{array}{c} \text{M. P., °C} \\ 153-154^{11} \\ 152-153.5^{21} \\ 151^{65} \end{array}$ 

### 1,2,4-Triphenylnaphthalene



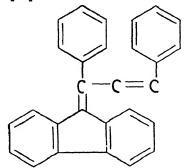
M. P., °C 158–159<sup>15</sup>

### 1-(9'-Fluorylidene)-1,2-diphenylpropene-2

$$\begin{array}{c} c = c \\ \end{array}$$

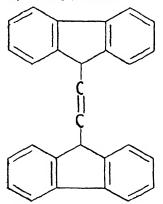
M. P., °C 197–1984•

1-(9'-Fluorylidene)-1,3-diphenylpropene-2



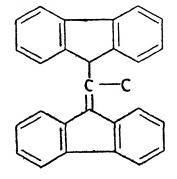
M. P., °C 183–184°

1,2-Di-(9'-fluoryl)-ethene



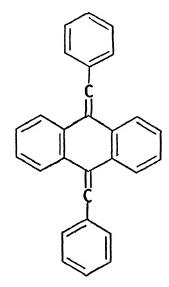
 $\begin{array}{c} \text{M. P., °C} \\ 267-268^{83} \\ 267^{81} \end{array}$ 

9-Fluoryl-9'-fluorylidenemethylmethane



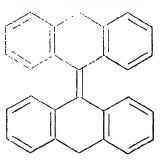
M. P., °C >350<sup>62</sup>

9,10-Dibenzylidene-9,10-dihydroanthracene



M. P., °C 237-210<sup>51</sup> 231-236<sup>51</sup> 199-200<sup>7</sup>

9,9'-Bi-9,10-dihydroanthrylidene

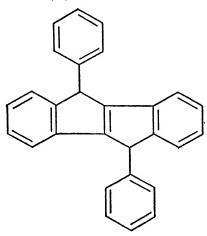


M. P., °C 298–30055

### 9,9'-Bi-9,10-dihydrophenanthrylidene

M. P., °C 3036

### 2,3,6,7-Dibenzo-4,8-diphenylbicy-clo-[3,3,0]-octene-1,5

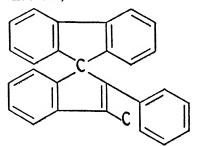


M. P., °C 285–286<sup>20</sup>

### Spiro[2-methyl-3-phenylindene-1,9'-fluorene]

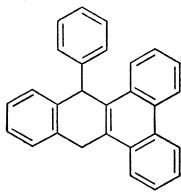
M. P., °C 173–174.5<sup>49</sup>

### Spiro[2-phonyl-3-methylindene-1,9'-fluorene]



M. P., °C 152.5-153.5<sup>49</sup>

### 1,2,3,4-Dibenzo-9-phenyl-9,10dihydroanthracene



M. P., °C 19210

### 9,10,9',10'-Bi-(9,10-dihydroanthryl-

ene)

(Paranthracene)

M. P., °C

243.5

2444, 11, 78

242-24461

242-24386

24052

 $D_4^{20}$ 

 $1.265^{61}$ 

1.261 (a)14

(a) The temperature of this determination was not given.

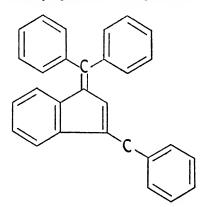
### 1,2,3,4-Di-(3',2'-indo)-5,6-[3'',2''-(5"-methylindo)]-benzene

### $\mathbf{C}_{29}\mathbf{H}_{22}$

### 1-Benzylidene-3-benzhydrylindene

M. P., °C 131.5–132.5<sup>85</sup> 130<sup>28</sup>

### 1-Benzhydrylidene-3-benzylindene

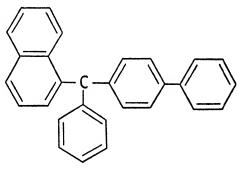


M. P., °C 130–130.5<sup>85</sup>

### 1-Benzylidene-2-benzyl-3-phenylindene

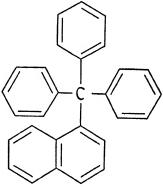
M. P., °C 177.5-179<sup>16</sup> 174-175<sup>60</sup>

### 1-Naphthyl-p-biphenylylphenylmethane



M. P., °C 281–282³

### 1-Naphthyltripheny methane



M. P., °C 193<sup>79</sup>

### $C_{30}H_{24}$

1-Naphthyl-(4'-methyl-p-biphenylyl)phenylmethane

M. P., °C 171<sup>37</sup>

1-(9'-Anthryl)-2-[9''-(9'',10''-dihy-droanthryl)]-ethane

M. P., °C 308<sup>5</sup> 2,3,6,7-Dibenzo-4,8-di-m-tolylbicyclo-[3,3,0]-octene-1,5

M. P., °C 179–180<sup>20</sup>

2,3,6,7-Dibenzo-4,8-di-p-tolylbicyclo-[3,3,0]-octene-1,5

M. P., °C 20020

### 1,2-Cyclohexano-9,10-diphenylanthracene

M. P., °C 235-236<sup>56</sup>

### 6,11-Diphenyl-1,2,3,4-tetrahydronaphthacene

M. P., °C 29829

### x<sub>6</sub>-Hexahydroheptacene (a)

M. P., °C 354<sup>54</sup>

(a) The structure of this compound was not clearly defined in the literature.

### 1,2,3,4,5,6-Tri-[3',2'-(5'-methyl-indo)]-benzene

M. P., °C

### $\alpha$ -Diphenyltruxane (a)

M. P., °C 201–20275

(a) The structure of this compound was not clearly defined in the literature.

### $\gamma$ -Diphenyltruxane (a)

M. P., °C 169--171<sup>75</sup>

(a) The structure of this compound was not clearly defined in the literature.

#### C32H28

1-(Di-p-tolylmethylene)-3-p-tolyl-5methylindene

M. P., °C 192–193<sup>22</sup>

### $C_{36}H_{36}$

1,2,3,4,5,6-Tri-[3',2'-(x<sub>6</sub>-hexahy-dronaphtho)-cyclobutano]-benzene (a)

M. P., °C 124–128<sup>32</sup>

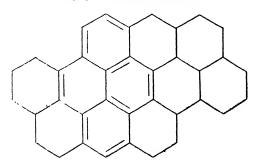
(a) The structure of this compound was not clearly defined in the literature.

### $C_{34}H_{32}$

3,9-Dibenzyl-4,5,6,6a,6b,7,8,9-octahydroperylene

M. P., °C 15489

Didecahydronaphtho-[2',1',8'-bcd, 2'',1'',8''-klm]-1,2,2a,3,4, 4a,5,6-octahydrocoronene



M. P., °C 340<sup>35</sup>

References on Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-v6</sub>

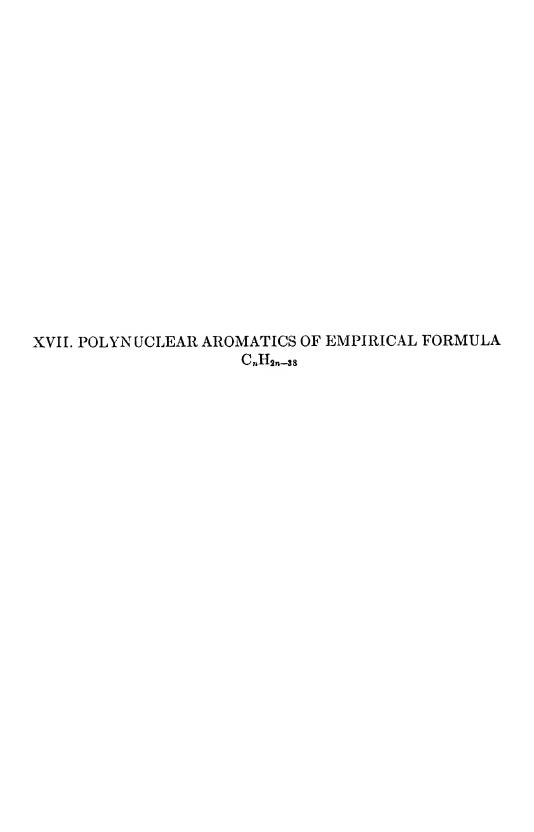
- Bachmann, W. E., J. Am. Chem. Soc. 56, 1363 1934.
- Backer, H. J., and H. B. J. Schrink, Rec. trav. chim. 50, 1066 1931.
- Banus, A. G., and J. Monche, Anales soc. españ. fis. quím. 33, 655 1935; Chem. Zentr. 1936, II, 2130.
- Barberio, M., Gazz. chim. ital. 33, II, 460 1903.
- Barnett, E. de B., and M. A. Matthews, Ber. 59, 767 1926.
- Bergmann, E., and F. Bergmann, J. Am. Chem. Soc. 59, 1443 1937.
- Bergmann, E., and S. Fujise, Ann. 480, 188 1930.
- 8. Bergmann, E., and S. Fujise, Ann. 483, 65 1930.
- 9. Bergmann, E, and T. Ukai, Ber. 66, 54 1933.
- Bergmann, E., and O. Zwecker, Ann. 487, 155 1931.
- Bergmann, F., D. Schapiro, and H. E. Eschinazi, J. Am. Chem. Soc. 64, 559 1942.
- Berliner, E., J. Am. Chem. Soc. 64, 2894 1942.
- Beyer, II., and J. Richter, Ber. 73, 1319
   1940.
- Bhatnagar, S. S., M B. Nevgi, and R. N. Mathur, Z. Physik 100, 141 1936.

15. Blum, O., Ber. 62, 881 1929.

- Blum-Bergmann, O., J. Chem. Soc. 1935, 1020.
- 17. Brand, K., Ber. 54, 1987 1921.
- Brand, K., and L. W. Berlin, Ber. 57, 816 1924.
- Brand, K., and O. Horn, J. prakt. Chem. 115, 351 1927.
- 20. Brand, K., and W. Mühl, J. prakt. Chem. [2] 110, 1 1925.
- 21. Brand, K., and H. W. Stephan, Ber. 72, 2168 1939.
- Brand, K., and G. Wendel, J. prakt. Chem. [2] 115, 335 1927.
- 23. Clar, E., Ber. 76, 119 1943.
- Clar, E., H. Wallenstein, and R. Avenarius, Ber. 62, 950 1929.
- 25. Cook, J. W., J. Chem. Soc. 1931, 499.
- Cook, J. W., I. Hieger, E. L. Kennaway, and W. V. Mayneord, Proc. Roy. Soc. (London) 111B, 455 1932.
- 27. Courtot, C., Ann. chim. [9] 4, 168 1915.
- 28. Courtot, C., Ann. chim. [9] 5, 52 1916.
- 29. Dufraisse, C., and R. Horelois, Bull'soc. chim. [5] 3, 1894 1936.
- 30. Dziewoński, K., and J. Podgórska, Bull. intern. acad. sci. Cracovic 1915A, 8.
- Dziewoński, K., and J. Suszko, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1921A, 67.
- Dziewoński, K., and J. Suszko, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1923A, 17.
- Dziewoński, K., and J. Suszko, Roczniki Chem. 1, 387 1921; C. A. 17, 1159 1921.
- Fieser, L. F., and E. M. Dietz, Ber. 62, 1827 1929.
- Fromherz, H., L. Thaler, and G. Wolf,
   Z. Elektrochem. 49, 387 1943.
- Gomberg, M., and J. C. Pernert, J. Am. Chem. Soc. 48, 1372 1926.
- 38. Graebe, C., Ber. 25, 3146 1892.
- 39. Graebe, C., and H. Stindt, Ann. 291, 1 1896.
- Grignard, V., and C. Courtot, Compt. rend. 152, 1493 1911.
- Hantseh, A., and W. H. Glover, Ber. 39, 4153 1906.
- 42. Kliegl, A., Ber. 43, 2188 1910.
- 43. Kliegl, A., Ber. 64, 2420 1931.
- 44. Klinger, H., and C. Lonnes, Ber. 29, 2157 1896.

- 45. Koelsch, C. F., J. Am. Chem. Soc. 55, 3394 1933.
- 46. Koelsch, C. F., J. Am. Chem. Soc. 56, 480 1934.
- Koelsch, C. F., and P. R. Johnson, J. Am. Chem. Soc. 65, 567 1943.
- Koelsch, C. F., and R. H. Rosenwald,
   J. Am. Chem. Soc. 59, 2170 1937.
- Koelsch, C. F., and R. V. White, J. Org. Chem. 6, 602 1941.
- Kuhn, R., and A. Winterstein, Helv. Chim. Acta 11, 116 1928.
- 51. Lippman, E., and R. Fritsch, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 113, Abt. IIb, 429 1904.
- Luther, R., and F. Weigert, Sitzber. kgl. preuss. Akad. Wiss. 1904, 828.
- Manchot, W., and P. Krische, Ann. 337, 170 1904.
- Marschalk, C., Congr. chim. ind., 18th Congr., Nancy 1938, 976; Chem. Zentr. 1939, II, 3078.
- Matthews, M. A., J. Chem. Soc. 1926, 236.
- Mikhallov, B. M., and N. G. Chernova,
   J. Gen. Chem. (U. S. S. R.) 12, 288
   1942; C. A. 37, 3086 1943.
- Nierenstein, M., and C. W. Webster, J. Am. Chem. Soc. 67, 691 1945.
- 58. Orekhov, A., Ber. 47, 89 1914.
- Orekhov, A., J. Russ. Phys. Chem. Soc.
   48, 1702 1916; C. A. 11, 3275 1917;
   Chem. Zentr. 1923, I, 1619.
- Orekhov, A. P., and R. Grinberg, J. Russ. Phys. Chem. Soc. 48, 1713 1916;
   C. A. 11, 3275 1917; Chem. Zentr. 1923,
   I, 1619.
- Orndorff, W. R., and F. K. Cameron, Am. Chem. J. 17, 658 1895.
- Pummerer, R., and G. Dorfmüller, Ber. 46, 2386 1913.
- 63. Reddelien, G., Ber. 53, 355 1920.
- 64. Ruggli, P., Ann. 414, 125 1917.
- Schlenk, W., and E. Bergmann, Ann. 463, 1 1928.
- 66. Schmidt, J., and H. Wagner, Ber. 43, 1796 1910.
- Schoepfle, C. S., J. Am. Chem. Soc. 44, 188 1922.

- Scholl, R., and K. Meyer, Ber. 65, 902
   1932.
- Schönberg, A., A. Stephenson, H. Kaltschmitt, E. Peterson, and H. Schulten, Ber. 66, 237 1933.
- Schönberg, A., and T. Stolpp, Ann. 483, 90 1930.
- 71. Seka, R., and W. Kelleman, Ber. 75, 1730 1942.
- 72. Smedley, I., J. Chem. Soc. 87, 1249 1905.
- 73. Stobbe, H., and E. Farber, Ber. 57, 1838 1924.
- Stobbe, II., and F. Zschoch, Ber. 60, 457
   1927.
- 75. Stoermer, R., and G. Foerster, Ber. 52, 1255 1919.
- Suszko, J., and R. Schillak, Roczniki Chem. 14, 1216 1934; C. A. 29, 6231 1935; Chem. Zentr. 1935, I, 2361.
- Vanscheidt, A. A., J. Russ. Phys. Chem. Soc. 58, 249 1926; C. A. 21, 3616 1927; Chem. Zentr. 1927, I, 92.
- 78. Weger, M., Z. angew. Chem. 22, 338 1909.
- Wieland, H., K. Heymann, T. Tsatsas,
   D. Juchum, G. Varvoglis, G. Labriola,
   O. Dobbelstein, and H. S. Boyd-Barrett, Ann. 514, 145 1934.
- 80. Wieland, II., and H. Kloss, Ann. 470, 201 1929.
- 81. Wieland, H., and E. Krause, Ann. 443, 129 1925.
- Winterstein, A., and K. Schön, Z. physiol. Chem. 230, 146 1934.
- 83. Wislicenus, W., Ber. 48, 617 1915.
- 84. Wittig, G., and W. Lange, Ann. 536, 266 1938.
- 85. Wuest, H.-M., Ann. 415, 291 1918.
- Yamaguchi, S., J. Chem. Soc. Japan 9.
   303 1934; Chem. Zentr. 1934, II, 4045.
- 87. Zal'kind, Y., and A. Kruglov, Ber. 61, 2306 1928.
- Zal'kind, Y., and A. Kruglov, J. Russ. Phys. Chem. Soc. 61, 803 1929; C. A. 23, 4678 1929.
- 89. Zinke, A., and O. Benndorf, Sitzber.
  Akad. Wiss. Wien, Math. naturw.
  Klasse 143, Abt. IIb, 1 1934.



### XVII. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_n H_{2n-38}$

### C26H14

Diindo-[3,2,1-de,3',2',1'-kl]-anthracene (Rubicene)

M. P., °C  $306^{31}$ 306 (a)36, 37 306 (b)26 305 (b)<sup>27</sup>

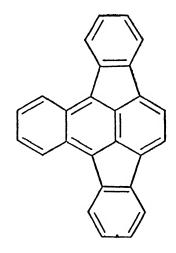
(a) The structural formula of this compound was given in the literature as

work proved the above formula to be correct for rubicene.

(b) This compound is given the formula C<sub>26</sub>H<sub>12</sub> in the literature. (a) This compound sinters at 330.

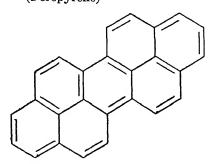
The structural formula is given as follows:

Diindo-[3,2,1-de,3',2',1'-mn]anthracene (Isorubicene)



M. P., ℃ 335 (a)18

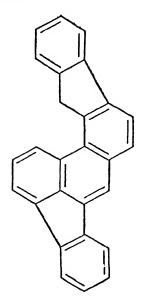
### Dibenzo-[cd,lm]-perylene (Peropyrene)



M. P., °C 374–375<sup>20</sup>

 $C_{27}H_{16}$ 

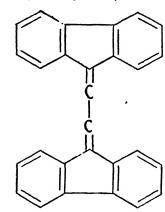
### Diindo-[3,2-c,1',2',3'-jk]-phenan-threne



M. P., °C 198–1994

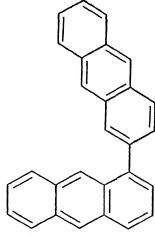
 $C_{28}H_{18}$ 

### 1,2-Di-(9'-fluorylidene)-ethane



M. P., °C 372-374<sup>45</sup> 360<sup>43</sup>

### 1,2'-Bianthryl



Additional Data Sublimation Temp. (°C) 33010

### 2,2'-Bianthryl

448

M. P., °C 355-355.5<sup>39</sup>

### 9,9'-Bianthryl

M. P., °C

319-32025

308-31016, 19

30438

300-3028

30028, 32, 35, 40

### Isodianthryl (a)

M. P., °C

3125

(a) The structure of this compound was not clearly defined in the literature.

### 1,1'-Biphenanthryl

M. P., °C 224–22630

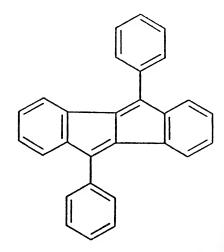
### 9,9'-Biphenanthryl

184-1853

### x, x'-Biphenanthryl (a)

- M. P., °C 308<sup>84</sup>
- (a) The structure of this compound was not clearly defined in the literature.

## 3,4,7,8-Dibenzo-2,6-diphenylbicy-clo-[3,3,0]-octadiene-1,5

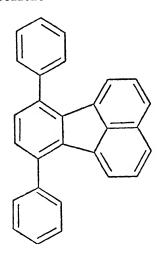


M. P., °C 259–260<sup>11</sup> 259<sup>13</sup> 1,4-Diphenyl-2,3-(2',1'-acenaphtho)-5-methylenecyclopentadiene-1,3

M. P., °C 225-226<sup>23</sup>

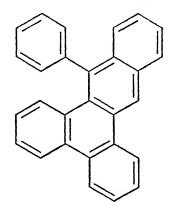
This series continued on next page

### 1,4-Diphenyl-2,3-(2',1'-acenaphtho)benzene



M. P., °C 162–163<sup>22</sup> 160–161<sup>1</sup>

### 2,3-Benzo-1-phenyltriphenylene



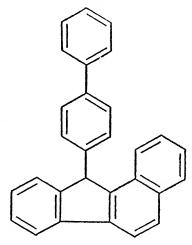
M. P., °C 2276, 9

### 1,2,11,12-Dibenzo-3,10-dihydroperylene

M. P., °C 269-270<sup>17</sup>

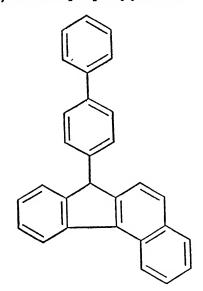
### C29H20

### 1,2-Benzo-9-p-biphenylylfluorene



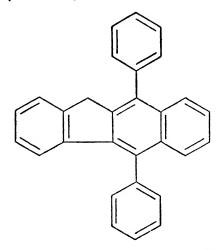
M. P., °C 276<sup>28</sup>A

### 3,4-Benzo-9-p-biphenylylfluorene



M. P., °C 275–276<sup>21</sup>

### 2,3-Benzo-1,4-dipheny:fluorene



M. P., °C 19942

### $C_{30}H_{22}$

9-Phenyl-10-(1'-naphthyl)-9,10dihydroanthracene

M. P., °C 225<sup>29</sup>

x, x-(x', x''-Diphenanthryl)-ethane
(a)

M. P., °C 252.5–254.5<sup>41</sup>

(a) The structure of this compound was not clearly defined in the literature.

3,4,7,8-Dibenzo-2,6-di-o-tolylbicyclo-[3,3,0]-octadiene-1,5

M. P., °C 240<sup>12</sup>

3,4,7,8-Dibenzo-2,6-di-m-tolylbicyclo-[3,3,0]-octadiene-1,5

M. P., °C 184–185<sup>12</sup> 184<sup>13</sup> 3,4,7,8-Dibenzo-2,6-di-p-tolylbicyclo-[3,3,0]-octadiene-1,5

M. P., °C 271–272<sup>11</sup> 271<sup>13</sup>

3,4,7,8-Dibenzo-2,6-dibenzylidenebicyclo-[3,3,0]-octane

M. P., °C 25514, 15

### 5,12-Diphenyl-5,12-dihydronaphthacene

M. P., °C 206-207<sup>24</sup>

### 6,12-Diphenyl-4b,10b-dihydrochrysene

M. P., °C 265–266\*\*

### $C_{32}H_{26}$

### 9,9'-Bi-2,3-dimethylanthryl

### 10,10'-Bi-1,3-dimethylanthryl

M. P., °C 284<sup>4</sup>

This series continued on next page

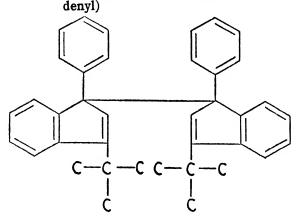
### 1,4-Di-(9'-phenanthryl)-butane

# 3,4,7,8-Dibenzo-2,6-di-(3',4'-dimethylphenylbicyclo-[3,3,0]-octadiene-1,5

M. P., °C 212<sup>12</sup> This series continued on next page

## $C_{38}H_{38}$

## 1,1'-Bi-(1-phenyl-3-tert-butylin-



M. P., °C 150–151<sup>2</sup>

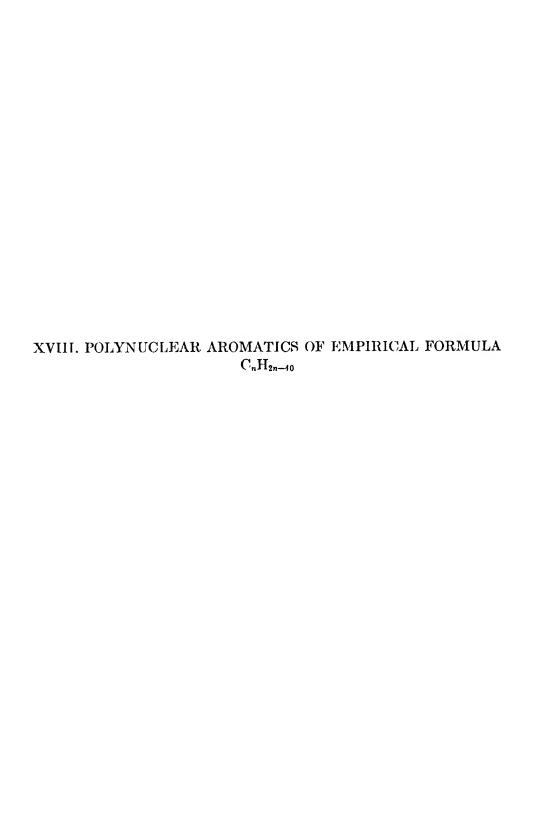
References on Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-38}$ 

- Abramov, V. S., and N. P. Tsyplenkova, Bull. acad. sci. U. R. S. S. 1944, 60; Survey For. Petrol. Liter. Jan. 26, 1945.
- Althausen, D., and C. S. Marvel, J. Am. Chem. Soc. 54, 1174 1932.
- 3. Bachmann, W. E., J. Am. Chem. Soc. 56, 1363 1934.
- 4. Barnett, E. de B., Ber. 65, 1563 1932.
- Barnett, E. de B., and N. F. Goodway, J. Chem. Soc. 1929, 813.
- Bergmann, E., and T. Berlin, J. Chem. Soc. 1939, 493.
- Bergmann, E., and O. Blum-Bergmann,
   J. Am. Chem. Soc. 58, 1678 1936.
- 8. Bergmann, E., and W. Schuchardt, Ann. 487, 225 1931.
- Bergmann, E., and O. Zwecker, Ann. 487, 155 1931.
- 10. Berton, A., Compt. rend. 208, 1898 1939.
- 11. Brand, K., Ber. 45, 3071 1912.
- Brand, K., H. Ludwig, and L. W. Berlin,
   J. prakt. Chem. [2] 110, 26 1925.
- 13. Brand, K., and W. Muhl, J. prakt. Chem. [2] 110, 1 1925.
- 14. Brand, K., and K. O. Müller, Ber. 55, 601 1922.
- Brand, K., and F. Schläger, Ber. 56, 2541
   1923.

- 16. Brass, K., and K. Fanta, Ber. 69, 1 1936.
- Clar, E., "Aromatische Kohlenwasserstoffe," Springer-Verlag, Berlin, 1941, p. 235.
- 18. Clar, E., Ber. 64, 2194 1931.
- 19. Clar, E., Ber. 65, 503 1932.
- 20. Clar, E., Ber. 76, 458 1943.
- Dilthey, W., J. prakt. Chem. [2] 109, 273 1925.
- Dilthey, W., S. Henkels, and A. Schaefer, Ber. 71, 974 1938.
- Dilthey, W., and P. Huchtemann, J. prakt. Chem. [2] 154, 238 1940.
- Dufraisse, C., and R. Horclois, Bull. soc. chim. [5] 3, 1894 1936.
- Dufraisse, C., L. Velluz, and Mme. L. Velluz, Bull. soc. chim. [5] 5, 600 1938.
- Dziewoński, K., and J. Suszko, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1921A, 67.
- Dziewoński, K., and J. Suszko, Roczniki Chem. 1, 387 1921; C.A. 17, 1459 1921.
- 28. Eckert, A., and A. Hofmann, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 124, Abt. IIb, 125 1915.
- 28A. Gomberg, M., and W. E. Gordon, J. Am. Chem. Soc. 57, 119 1935.
- Guyot, A., and C. Stachling, Bull. soc. chim. [3] 33, 1105 1905.
- 30. Ioffe, I. S., J. Gen. Chem. (U.S.S.R.) 3, 524 1933; Chem. Zentr. 1935, I, 391.

- Khmelevski, V. I., and I. Y. Postovskii,
   J. Gen. Chem. (U.S.S.R.) 9, 620 1939;
   C.A. 33, 7777 1939.
- Liebermann, C., and A. Gimbel, Ber.
   1854 1887.
- 33. Marvel, C. S., and W. J. Peppel, J. Am. Chem. Soc. 61, 895 1939.
- 34. Meyer, H., and A. Hoffmann, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 125-126, Abt. IIb, 449 1916-1917.
- 35. Minaev, W., and B. Fedorov, Ber. 62, 2489 1929.
- 36. Pummerer, R., Ber. 45, 294 1912.
- Pummerer, R., and H. M. Ulrich, Ber.
   58, 1806 1925.

- Schlenk, W., and E. Bergmann, Ann. 463, 1 1928.
- 39. Scholl, R., Ber. 52, 1829 1919.
- 40. Schulze, K. E., Ber. 18, 3034 1885.
- Tarbell, D. S., and V. P. Wystrach, J. Am. Chem. Soc. 65, 2149 1943.
- Weiss, R., and A. Beller, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 141, Abt. IIb, 511 1932.
- 43. Wieland, H., and E. Krause, Ann. 443, 129 1925.
- 44. Wieland, H., and O. Probst, Ann. 530, 274 1937.
- 45. Wislicenus, W., Ber. 48, 617 1915.
- 46. Zinke, A., and E. Ziegler, Ber. 74, 115 1941.



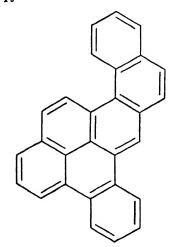
# XVIII. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-40}$

 $\mathbf{C_{28}H_{16}}$ 

$${\bf 1,2\text{-}Di\text{-}(9'\text{-}fluorylidene)\text{--}ethene}$$

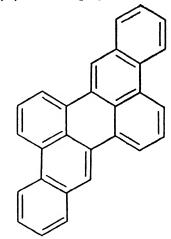
M. P., °C 330<sup>32</sup>

## 1,2-(4',3'-Naphtho)-4,5-benzopyrene



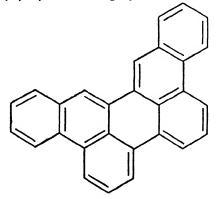
M. P., °C 242–244<sup>24</sup>

## 2,3,8,9-Dibenzoperylene



M. P., °C 310–343<sup>26, 27</sup> 299–300<sup>39</sup>

## 2,3,10,11-Dibenzoperylene



M. P., °C 343-345 (a)<sup>39</sup> 343 (a) <sup>9, 14</sup> 315-318 (a)<sup>26, 27</sup>

(a) Clar ("Aromatische Kohlenwasserstoffe," Springer-Verlag, Berlin, 1941, p. 236) indicates that

these are three different compounds, but does not assign a definite structure to the compound in references 26, 27, and 39.

## $C_{29}H_{18}$

## 2,3-[3',2'-(1'-Phenylindo)]-pyrene

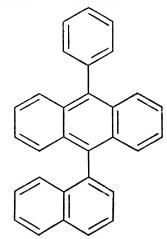
M. P., °C 279–280<sup>15</sup>

#### $C_{30}H_{20}$

#### 1,4-Di-(9'-fluorylidene)-butene-2

M. P., °C 340<sup>34</sup> 331–332<sup>33</sup>

## 9-Phenyl-10-(1'-naphthyl)-anthracene



M. P., °C 244–245<sup>4</sup> 229<sup>25</sup>

#### 1,2-Di-(9'-anthryl)-ethene

M. P., °C 337<sup>38</sup>

## 1,2-Di-(3'-phenanthryl)-ethene

## 2,3,6,7-Dibenzo-4,8-dibenzylidenebicyclo-[3,3,0]-octene-1,5

## 2,3-(9',10'-Phenanthro)-1,4-diphenyl-5-methylenecyclopentadiene-1,3

M. P., °C 239–240<sup>1</sup>

#### 1,2-Benzo-9,10-diphenylanthracene

M. P., °C 192<sup>13</sup>

## 5,11-Diphenylnaphthacene (Diphenylrubene)

M. P., °C 301-302<sup>21</sup>

### 5,12-Diphenylnaphthacene

M. P., °C 207–208<sup>20</sup>

## x,x-Diphenylnaphthacene (a)

M. P., °C 258<sup>22</sup>

(a) The structure of this compound was not clearly defined in the literature.

## 5,6-Diphenylchrysene

M. P., °C 208-209<sup>17</sup>

#### 6,12-Diphenylchrysene

M. P., °C 285<sup>35</sup>

# 2,3,4,5-Dibenzo-1-phenyl-1,3a-dihydropyrene

M. P., °C 257–258<sup>13</sup>

#### $C_{31}H_{22}$

1,1,3-Triphenyl-3-(1'-naphthyl)propadiene

M. P., °C 142 (a)<sup>16</sup> 100–102 (a)<sup>16</sup>

(a) These constants were determined on different crystalline forms.

1,3-Diphenyl-1-pentadeuterophenyl-3-(1'-naphthyl)-propadiene

M. P., °C  $142 (a)^{16} 100-101 (a)^{16}$ 

These constants were determined (a) on different crystalline forms.

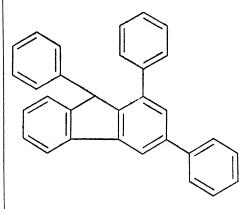
Tri-(2-naphthyl)-methane

19111

M. P., °C 178–179<sup>12</sup>

## 4,5-Benzo-1,1,3-triphenylindene

## 1,3,9-Triphenylfluorene

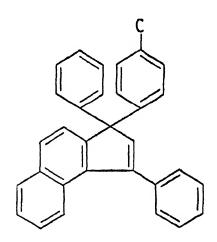


M. P., °C 149<sup>81</sup>

## $\mathbf{C}_{32}\mathbf{H}_{24}$

# 1,3-Diphenyl-1-p-tolyl-3-(1'-naph-thyl)-propadiene

M. P., °C 119–120<sup>16</sup> 4,5-Benzo-1,3-diphenyl-1-p-tolylindene



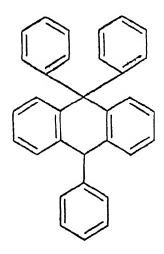
M. P., °C 172–175<sup>16</sup>

(9-Fluoryl)-triphenylmethane

M. P., °C 258-260<sup>28</sup> 248-256 (in nitrogen)<sup>2</sup> 234-235<sup>2</sup> Phenyl-p-biphenylyl-(9'-fluoryl)-methane

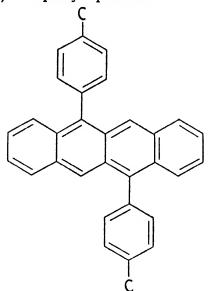
M. P., °C 253–254°

9,9,10-Triphenyl-9,10-dihydroanthracene



M. P., °C 230³ 226–227²<sup>7</sup> 223–225²<sup>9</sup>

## 5,11-Di-p-toly naphthacene



M. P., °C 335–336<sup>36</sup>

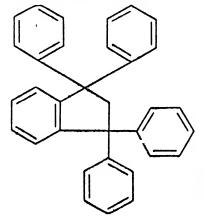
## x,x-Dibenzylchrysene (a)

M. P., °C 254<sup>23</sup>

(a) The structure of this compound was not clearly defined in the literature.

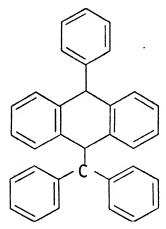
## $C_{33}H_{26}$

## 1,1,3,3-Tetraphenylindane



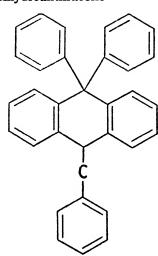
M. P., °C 191–19230

## 9-Phenyl-10-benzhydryl-9,10-dihydroanthracene



M. P., °C 217<sup>5</sup>

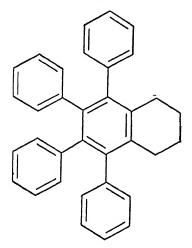
## 9,9-Diphenyl-10-benzyl-9,10dihydroanthracene



M. P., °C 192<sup>7</sup>

#### $C_{34}H_{28}$

## 5,6,7,8-Tetraphenyl-1,2,3,4-tetrahydronaphthalene



M. P., °C 271–272<sup>19</sup>

## x,x-Diphenyl-x<sub>6</sub>-hexahydropen tacene (a)

M. P., °C

(a) The structure of this compound was not clearly defined in the literature, but the phenyl groups are probably in the 6- and 13-positions.

#### C36H32

## Tetraindene (a)

M. P., °C 109–110<sup>10</sup>

(a) The structure of this compound was not clearly defined in the literature.

#### References on Polynuclear Aromatics of Empirical Formula $C_nH_{2n-40}$

- Allen, C. F. H., and A. Bell, J. Am. Chem. Soc. 64, 1253 1942.
- Bachmann, W. E., J. Am. Chem. Soc. 55, 2135 1939.
- Barnett, E. de B., J. W. Cook, and I. G. Nixon, J. Chem. Soc. 1927, 501.
- Barnett, E. de B., J. W. Cook, and J. L. Wiltshire, J. Chem. Soc. 1927, 1724.
- 5. Bergmann, E., Ber. 63, 1037 1930.
- 6. Bergmann, E., Ber. 63, 1617 1930.
- Bergmann, E., and S. Fujise, Ann. 480, 188 1930.
- 8. Brand, K., and K. O. Müller, Ber. 55, 601 1922.
- 9. Brass, K., and K. Fanta, Ber. 69, 1 1936.
- 10. Bruson, H. A., Ber. 60, 1094 1927.
- 11. Chichibabin, A. E., Ber. 44, 1105 1911.
- Chichibabin, A. E., and S. I. Koryagin,
   J. prakt. Chem. [2] 88, 505 1913.
- 13. Clar, E., Ber. 63, 112 1930.
- 14. Clar, E., Ber. 65, 846 1932.
- 15. Clar, E., Ber. 69, 1671 1936.
- Clemo, G. R., R. Raper, and A. C. Robson, J. Chem. Soc. 1939, 431.
- Cook, J. W., and R. A. E. Galley, J. Chem. Soc. 1931, 2012.
- 18. Dilthey, W., and P. Huchtemann, J. prakt. Chem. [2] 154, 238 1940.
- 19. Dilthey, W., W. Schommer, W. Hoschen, and H. Dietrichs, Ber. 68, 1159 1935.
- Dufraisse, C., and R. Horclois, Bull. soc. chim. [5] 3, 1894 1936.
- Dufraisse, C., and M. Loury, Compt. rend. 200, 1673 1935.
- Enderlin, L., Compt. rend. 203, 192 1936.
- Funke, K., E. Müller, and L. Vadasz,
   J. prakt. Chem. [2] 144, 265 1936.
- Grechkin, N. P., and A. E. Arbusov, Compt. rend. acad. sci. U.R.S.S. 32, 50 1941; Survey For. Petrol. Liter. No. 342, Jan. 16, 1942.
- Guyot, A., and C. Staehling, Bull. soc. chim. [3] 33, 1105 1905.
- Ioffe, I. S., J. Gen. Chem. (U.S.S.R.)
   524 1933; Chem. Zentr. 1935, I,
   391.
- Ioffe, I., Trudy Leningrad. Khim. Tekh. Inst. No. 1, 61 1934; C.A. 29, 7969 1935.

- 28. Koelsch, C. F., J. Am. Chem. Soc. 55, 3394 1933.
- Koelsch, C. F., J. Org. Chem. 3, 456
   1938.
- Koelsch, C. F., and R. H. Rosenwald,
   J. Am. Chem. Soc. 59, 2170 1937.
- Kohler, E. P., and L. W. Blanchard, J. Am. Chem. Soc. 57, 367 1935.
- 32. Kuhn, R., and G. Platzer, Ber. 73, 1410 1940.
- Kuhn, R., and K. Wallenfels, Ber. 71, 1889 1938.
- 34. Kuhn, R., and A. Winterstein, Helv. Chim. Acta 11, 116 1928.

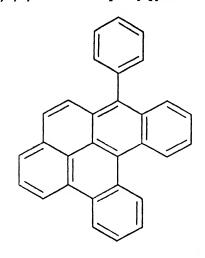
- 35. Marvel, C. S., and W. J. Peppel, J. Am. Chem. Soc. **61**, 895 **1939**.
- Weizmann, A., J. Org. Chem. 8, 285
   1943.
- 37. Wittig, G., and M. Leo, Ber. 64, 2395 1931.
- Wood, J. H., J. A. Bacon, A. W. Meibohm, W. H. Throckmorton, and G. P. Turner, J. Am. Chem. Soc. 63, 1334 1941.
- 39. Zinke, A., and E. Ziegler, Ber. 74, 115 1941.

# XIX. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-42}$

## XIX. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA C<sub>n</sub>H<sub>2</sub>, -42

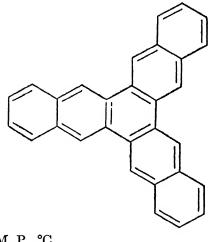
## C30H18

## 2,3,4,5-Dibenzo-1-phenylpyrene



M. P., °C 242–243<sup>11</sup>

## 1,2,3,4-Di-(3',2'-naphtho)-anthracene



M. P., °C 39219 39019 38819 38719

## 1,2-Benzo-5,6-(2',3'-phenanthro)anthracene

M. P., °C 398-39914

## 2,3-Benzo-8,9-(4',3'-naphtho)chrysene

M. P., °C

435-440 (a)16

(a) This compound melts with decomposition.

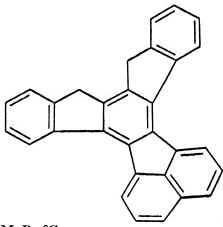
#### 1,2,8,9-Dibenzopentacene

M. P., °C

 $410 (a)^{18}$ 

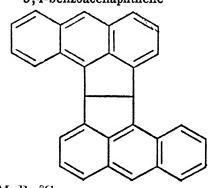
(a) This compound melts with decomposition.

## 1,2-(2',3'-Indo)-3,4-(2',1'-acenaphtho)-fluorene



M. P., °C 299<sup>17</sup>

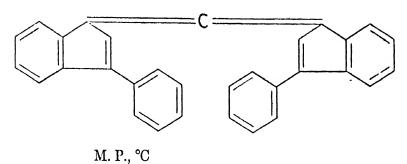
## 1,2-[2',1'-(3,4'-Benzoacenaphtheno)]-3,4-benzoacenaphthene



M. P., °C 349<sup>12</sup>

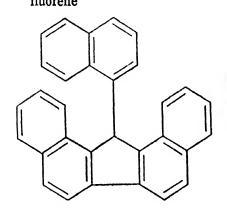
Ca1H20

## Di-(3-phenylindenylidene)-methane



205-2067

# 1,2,7,8-Dibenzo-9-(1'-naphthyl)fluorene



M. P., °C

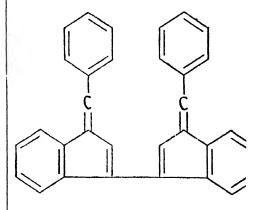
# 2,3,6,7-Dibenzo-x-(2'-naphthyl)fluorene (a)

M. P., °C 235-237 (in carbon dioxide)<sup>10</sup>

(a) The structure of this compound was not clearly defined in the literature.

## $C_{32}H_{22}$

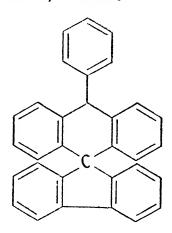
## Bi-3,3'-(1-Benzylideneindenyl)



M. P., °C 217.5<sup>21</sup>

This series continued on next page

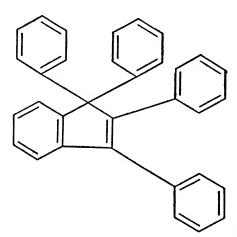
## Spiro[9-phenyl-9, 10-dihydroanthracene-10,9'-fluorene]



M. P., °C 267–268<sup>15</sup>

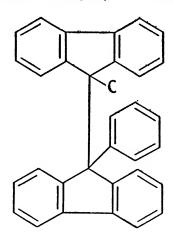
## $C_{33}H_{24} \\$

## 1,1,2,3-Tetraphenylindene



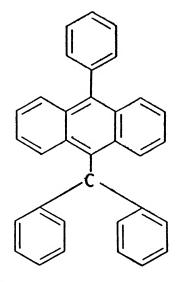
M. P., °C 149–150<sup>18</sup>

## 9-Phenyl-9'-methyl-9,9'-b'fluorenyl



M. P., °C 161<sup>22</sup>

## 9-Phenyl-10-benzhydrylanthracene



M. P., °C 154–1554

# 9-Phenyl-10-benzhydrylidene-9, 10-dihydroanthracene

M. P., °C 2184

## 9,9-Diphenyl-10-benzylidene-9,10dihydroanthracene

#### $C_{34}H_{26}$

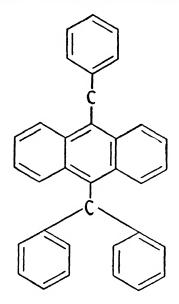
## 1,3-Di-(9'-fluorylmethyl)-benzene

M. P., °C 119–120<sup>20</sup>

## 1,4-Di-(9'-fluoryimethyl)-benzene

## 9-Benzyl-10-benzhydrylanthracene

This series continued on next page

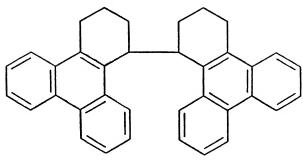


M. P., °C 236³

## 6,13-Diphenyl-5,7,12,14-tetrahydropentacene

#### C<sub>36</sub>H<sub>30</sub>

## 4,4'-Bi-(1,2,3,4-tetrahydrotriphenylenyl)



M. P., °C 300⁵

#### References on Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-12</sub>

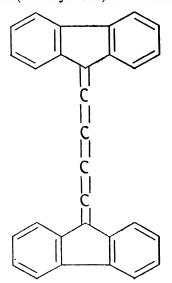
- Allen, C. F. H., and A. Bell, J. Am. Chem. Soc. 64, 1253 1942.
- Barnett, E. de B., and J. W. Cook, J. Chem. Soc. 1928, 566.
- Barnett, E. de B, and N. F. Goodway, J. Chem. Soc. 1929, 1754.
- 4. Bergmann, E, Ber. 63, 1037 1930.
- Bergmann, E., and O. Blum-Bergmann, J. Am. Chem. Soc. 59, 1441 1937.
- Bergmann, E., and S. Fujise, Λnn. 480, 188 1930.
- 7. Blum-Bergmann, O., Ann. 492, 277 1932.
- 8. Chichibabin, A. E., J. prakt. Chem. [2] 84, 760 1911.
- Chichibabin, A. E., J. Russ. Phys. Chem. Soc. 43, 1022 1911; Chem. Zentr. 1911, II, 1929.
- Chichibabin, A. E., and S. I. Koryagin,
   J. prakt. Chem. [2] 88, 505 1913.

- 11. Clar, E., Ber. 63, 112 1930.
- 12. Clar, E., Ber. 72, 2134 1939.
- 13. Clar, E., Ber. 76, 257 1943.
- Clar, E., F. John, and R. Avenarius, Ber. 72, 2139 1939.
- Clarkson, R. G., and M. Gomberg, J. Am. Chem. Soc. 52, 2881 1930.
- 16. Cook, J. W., J. Chem. Soc. 1931, 499.
- Dilthey, W., and S. Henkels, J. prakt. Chem. [2] 149, 85 1937.
- Koelsch, C. F., J. Am. Chem. Soc. 54, 3384 1932.
- Pummerer, R., A. Lüttringhaus, R. Fick, S. Pfoff, G. Riegelbauer, and E. Rosenhauer, Ber. 71, 2569 1938.
- Sieglitz, A., and H. Jassoy, Ber. 51, 2133 1921.
- Straus, F., R. Kühnel, and R. Haensel, Ber. 66, 1847 1933.
- Ziegler, K., and W. Schäfer, Ann. 511, 101 1934.

# XX. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-41}$

#### XX. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA C, H2, -44

# $C_{30}H_{16} \\$ 1,4-Di-(9'fluorylidene)-butatriene



M. P., °C 441–442°

Diindo-[3,2,1-fg,3',2',1'-op]naphthacene (Diphenylenerubene)

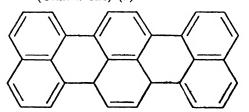
M. P., °C
465<sup>3</sup>, <sup>5</sup>
Additional Data

Sublimation Temp. (°C)

270–275 0.1 mm<sup>5</sup>

This series continued on next page

#### Dinaphtho-[1,8-ab,1',8'-hi]-pyrene (Chalcacene) (a)



M. P., °C 358-3606, 7

(Rhodacene) (b)

## M. P., °C 338-3406

- (a) This name is given to the form of the compound having a benzenoid structure.
- (b) This name is given to the form of the compound having a quinoid structure.

## Dinaphtho-[3,2,1-cd,3',2',1'-jk]pyrene

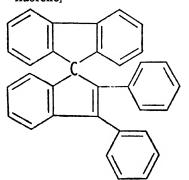
M. P., °C 372-3734 This series continued on next page

M. P., °C

382-3904

 $C_{33}H_{22}$ 

Spiro [2,3-diphenylindene-1,9'fluorene]

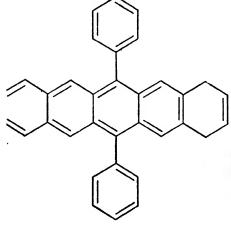


M. P., °C

174-175°

 $C_{31}H_{24} \\$ 

## 6,13-Diphenyl-1,4-dihydropentacene



. P., °C 295¹

## 6,13-Diphenyl-5,14-dihydropentacene

M. P., °C 217-2481

## 3,4-Dibenzylperylene

l M. P., ℃

329-33018

#### C<sub>85</sub>H<sub>26</sub>

1,3-Diphenyl-1,3-di-(1'-naphthyl)propene-1

M. P., °C 167–169<sup>11</sup> 1,1,2-Tetraphenyl-2-(1'-naphthyl)ethane

M. P., °C 194-196 (in nitrogen)² 184-194 (in air)²

References on Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-44}$ 

- Allen, C. F. H., and A. Bell, J. Am. Chem. Soc. 64, 1253 1942.
- Bachmann, W. E., J. Am. Chem. Soc. 55, 2135 1939.

## $C_{36}H_{28}$

## 1,8-Dibenzhydrylnaphthalene

M. P., °C 243<sup>12</sup>

- 3. Badoche, M., Ann. chim. [10] 20, 200 1933.
- 4. Clar, E., Ber. 76, 328 1943.
- Dufraisse, C., and R. Girard, Bull. soc. chim. [5] 1, 1359 1934.
- 6. Dziewoński, K., Ber. 53, 2173 1920.
- Dziewoński, K., Bull. intern. acad. polon. sci., Classe sci. math. nat. 1919 A, 99.
- 8. Koelsch, C. F., J. Am. Chem. Soc. 54, 3384 1932.

- 9. Kuhn, R., and K. Wallenfels, Ber. 71, 783 1938.
- Kuhn, R., and K. Wallenfels, Ber. 71, 1510 1938.
- Maitland, P., and W. H. Mills, J. Chem. Soc. 1936, 987.
- Wittig, G., and H. Petri, Ber. 68, 924
   1935.
- Zinke, A., and O. Benndorf, Monatsh.
   56, 153 1930; C.A. 25, 292 1931; Chem.
   Zentr. 1931, I, 276.

# XXI. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-46}$

## XXI. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-46}$

C<sub>30</sub>H<sub>14</sub>

#### Dibenzo-[bc, ef]-coronene

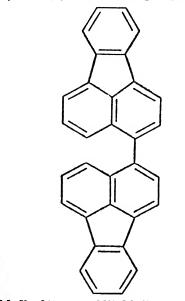
Additional Data Sublimation Temp. (°C) ca 500

1 mm<sup>13</sup>

## Dibenzo-[bc,kl]-coronene

Additional Data Sublimation Temp. (°C) 450-500<sup>12</sup>

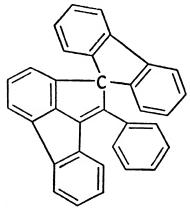
# $C_{32}H_{18} \\$ 4,4'-Bi-(1,2-benzoacenaphthyl)



M. P., °C 327-329<sup>22</sup>

## C23H20

Spiro[indo-[3',2',1'-cd]-2-phenylin-dene-1,9''-fluorene]



M. P., °C 196-197°

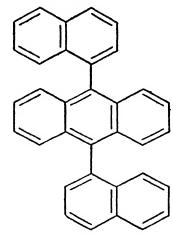
#### $C_{34}H_{22}$

1,3-Di-(9'-fluorylidenemethyl)benzene

M. P., °C 178-179<sup>16</sup>

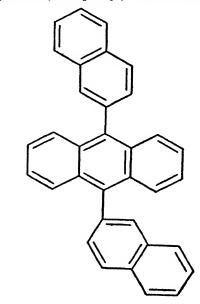
1,4-Di-(9'-fluorylidenemethyl)benzene

M. P., °C 209–210<sup>15</sup> 9,10-Di-(1'-naphthyl)-anthracene



M. P., °C 430–431<sup>24</sup> 430<sup>23</sup>

9,10-Di-(2'-naphthyl)-anthracene



M. P., °C 379<sup>23</sup> 378–379<sup>24</sup>

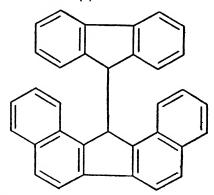
Bi-9,9'-(1,2-benzofluoryl)

M. P., °C 257<sup>20</sup> 221<sup>21</sup>

### 2,3-(2',1'-Acenaphtho)-1,4,5triphenylbenzene

M. P., °C 195–196<sup>7</sup>

## 1,2,7,8-Dibenzo-9-(9'-fluoryl)-fluorene (a)

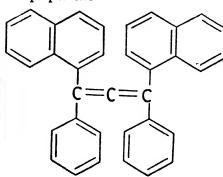


M. P., °C 270<sup>20, 21</sup>

(a) The structure of this compound was not clearly defined in the literature.

#### $C_{35}H_{24}$

## 1,3-Diphenyl-1,3-di-(1'-naphthyl)propadiene



M. P., °C 242-244 (a)<sup>10, 11</sup> 240<sup>5</sup> 158-159 (b) (c)<sup>10, 11</sup>

Additional Data

$$[\alpha]_{5461}^{17} = +437^{\circ} \text{ (b)}_{10, 11}^{10}$$
  
 $[\alpha]_{5461}^{17} = -438^{\circ} \text{ (c)}_{10, 11}^{10}$ 

(a) This constant was determined on the dl-form of the compound.

- (b) This constant was determined on the d-form of the compound.
- (c) This constant was determined on the l-form of the compound.

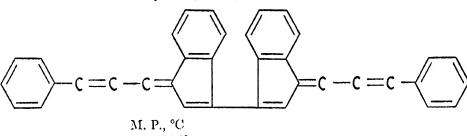
## Di-(x,x'-phenanthryl)-phenylmethane (a)

M. P., °C 1658

(a) The structure of this compound was not clearly defined in the literature.

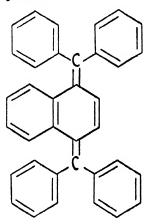
#### $C_{36}H_{26}$

### Bi-3,3'-[1-(3"-phenylpropen-2"ylidene)-indenyl]



25119

## 1,4-Dibenzhydrylidene-1,4-dihydronaphthalene



M. P., °C 262-26317 This series continued on next page

# 1-(1'-Naphthyl)1,2-di-p-biphenyl-ylethene

M. P., °C 209-214<sup>14</sup>

# 1-Benzhydrylidene-4-(1'-naphthyl-phenylmethylene)-cyclohexadiene -2,5

M. P., °C

240-241 (in carbon dioxide)<sup>4</sup> 238-239 (in carbon dioxide)<sup>1</sup>

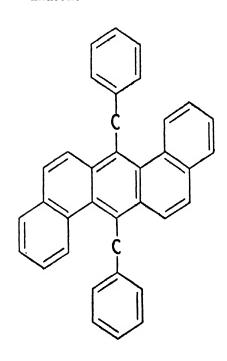
x-Benzhydrylidene-x,x-diphenylx,x-benzocycloheptadiene-x,x (a)

M. P., °C 181-182<sup>18</sup>

- (a) The structure of this compound was not clearly defined in the literature.
- 2,3,6,7-Dibenzo-4,8-di-(2'-naphthyl)bicyclo-[3,3,0]-octane

M. P., °C 225³

## 1,2,5,6-Dibenzo-9,10-dibenzylanthracene



## M. P., °C 195–201 (a)<sup>6</sup>

(a) This compound melts with decomposition.

## x,x,x-Triphenyl-x,x-dihydronaphthacene (a)

M. P., °C 208–209²

(a) The structure of this compound was not clearly defined in the literature.

#### References on Polynuclear Aromatics of Empirical Formula $C_nII_{2n-46}$

- 1. Allard, S., Compt. rend. 199, 423 1934.
- Badoche, M., Bull. soc. chim. [5] 5, 164 1938.
- 3. Brand, K., and K. Trebing, Ber. 56, 2545 1923.
- 4. Chichibabin, A. E., Ber. 41, 2770 1908.
- Clemo, G. R., R. Raper, and A. C. Robson, J. Chem. Soc. 1939, 431.
- 6. Cook, J. W., J. Chem. Soc. 1931, 489.
- 7. Dilthey, W., S. Henkels, and A. Schaefer, Ber. 71, 974 1938.
- 8. Frankforter, G., and W. Kritchevsky, J. Am. Chem. Soc. 37, 385 1915.
- Koelsch, C. F., J. Am. Chem. Soc. 54, 4744 1932.
- Maitland, P., and W. H. Mills, J. Chem. Soc. 1936, 987.
- 11. Maitland, P., and W. II. Mills, Nature 135, 994 1935.
- 12. Scholl, R., and K. Meyer, Ber. 65, 902 1932.
- 13. Scholl, R., and K. Meyer, Ber. 67, 1229 1934.
- Shilov, E. A., and F. K. Yudin, J. Gen. Chem. (U.S.S.R.) 9, 167 1939; C. A. 33, 6283 1939.
- 15. Sieglitz, A., Ber. 52, 1513 1919.
- 16. Sieglitz, A., Ber 53, 1232 1920.
- 17. Staudinger, H., Ber. 41, 1355 1908.
- 18. Staudinger, H., and H. Kon, Ann. 384, 38 1911.
- Straus, F., R. Kuhnel, and R. Haensel, Ber. 66, 1847 1933.
- 20. Vansheidt, A., Ber. 59, 2092 1926.
- Vansheidt, A. A., J. Russ. Phys. Chem. Soc. 53, 69 1926; C. A. 21, 581 1927; Chem. Zentr. 1926, II, 24, 28.
- von Braun, J., and G. Manz, Ber. 70, 1603 1937.
- Willemart, A., Bull. soc. chim. [5] 4, 357 1937.
- 24. Willemart, A., Compt. rend. 201, 1201 1935.

# XXII. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-48}$

## XXII. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA C<sub>n</sub>H<sub>2n-48</sub>

C32H16

Dinaphtho-[2,1,8-cde,2',1',8'-lmn]perylene

M. P., °C 213-214<sup>17</sup>

#### $C_{34}H_{20}$

9,9'-Bi-(2,3-benzofluorylidene)

M. P., °C 232<sup>15</sup>

sym-Di- $\alpha$ -naphthylenediphenyleneethene (a)

M. P., °C 317-318<sup>16</sup>

(a) The structure of this compound was not clearly defined in the literature.

## $|unsym-Di-\alpha-naphthylenediphenylene-$ ethene (a)

M. P., °C 315<sup>16</sup>

(a) The structure of this compound was not clearly defined in the literature.

## (4,5-Diphenylcyclopenten-4-o)-[cd]perylene

M. P., °C 315-316<sup>21</sup>

This scries continued on next page

# 2,3,8,9-Di-(4',3'-naphtho)-chrysene

# $C_{36}H_{24} \\$ 1,3,5-Tri-(1'-naphthyl)-benzene

190.5-1915

C.6H24

492

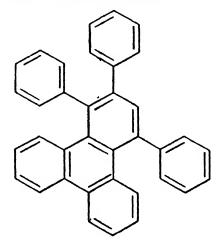
# 1,3,5-Tri-(2'-naphthyl)-benzene

M. P., °C 234–235<sup>5</sup>

- x,x-Di-p-biphenylylacenaphthylene
  (a)
- M. P., °C 189.5–190.5<sup>1</sup>
- (a) The structure of this compound was not clearly defined in the literature.
- 5,6,11-Triphenylnaphthacene (a) (Triphenylrubene)

- M. P., °C 236-237<sup>3</sup>, 4, 7, 9
- (a) This compound was named Diphenylrubrene in reference 7.

# 1,2,4-Triphenyltriphenylene



M. P., °C

# 3,9-Di-(1'-phenethenyl)-perylene

# 257-25821

# (4,5-Di-p-tolylcyclopenten-4-o)-[cd]perylene

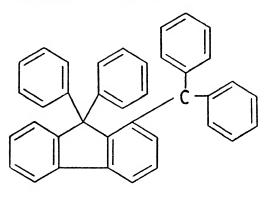
м. Р., °С 326.5-327.512

C38H28

494

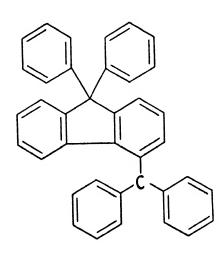
 $C_{38}H_{28}$ 

# 1-Benzhydryl-9,9-diphenylfluorene



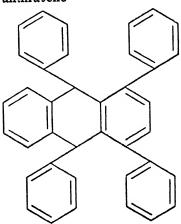
M. P., °C 219-220<sup>14</sup>

# 4-Benzhydryl-9,9-diphenylfluorene



M. P., °C 219.5–221<sup>19</sup> 219–220<sup>14</sup>

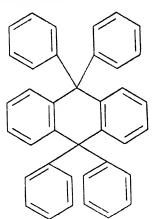
# 1,4,9,10-Tetraphenyl-9,10-dihydroanthracene



M. P., °C 217 (a)<sup>18</sup> 205 (a)<sup>18</sup>

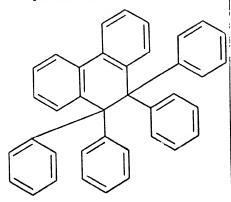
(a) These constants were determined on stereoisomers.

9,9,10,10-Tetraphenyl-9,10-dihy-droanthracene



M. P., °C 159<sup>13, 20</sup>

9,9,10,10-Tetraphenyl-9,10-dihydrophenanthrene



M. P., °C 339.5–34119

C<sub>39</sub>H<sub>30</sub>

1,3,5-Tri-[1'-(4'-methylnaphthyl)]benzene

M. P., °C 1858

1,1,2,2-Tetraphenyl-1-(2'-fluoryl)-ethane

M. P., °C 187-190 (in nitrogen)<sup>2</sup> 168-176<sup>2</sup>

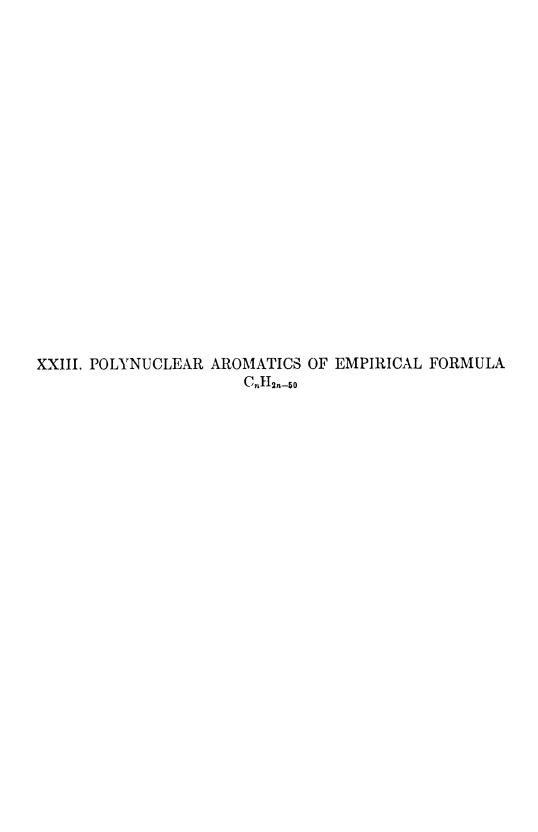
### 1,1,2,2-Tetraphenyl-1-(9'-fluoryl)ethane

M. P., °C 205<sup>11</sup>

References on Polynuclear Aromatics of Empirical Formula  $C_n II_{2n-48}$ 

- Bachmann, W. E., and E. J.-H. Chu, J. Am. Chem. Soc. 58, 1118 1936.
- Bachmann, W. E., R. Hoffman, and F. Whitehead, J. Org. Chem. 8, 320 1943.
- 3. Badoche, M., Bull. soc. chim. [5] 3, 2040 1936.
- Badoche, M., Bull. soc. chim. [5] 5, 164 1938.
- Clapp, D. B., and A. A. Morton, J. Am. Chem. Soc. 58, 2172 1936.
- Dilthey, W., S. Henkels, and A. Schaefer, Ber. 71, 974 1938.
- Dufraisse, C., and M. Badoche, Compt. rend. 193, 242 1931.
- Dziewoński, K., and M. Marusińska, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1938A, 316.
- 9. Enderlin, L., Ann. chim. [11] 10, 5 1938.
- Fieser, L. F., and E. M. Dietz, Ber. 62, 1827 1929.
- Klinger, H., and C. Lonnes, Ber. 29, 734 1896.

- Pongretz, A., and G. Markgraf, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 144, Abt. IIb, 308 1935.
- 13. Rây, J. N., J. Chem. Soc. 117, 1335 1920.
- Sergeev, P. G., J. Russ. Phys. Chem. Soc. 61, 1421 1929; C.A. 24, 1365 1930; Chem. Zentr. 1930, II, 391.
- Thiele, J., and A. Vansheidt, Ann. 376, 269 1910.
- Vansheidt, A. A., J. Russ. Phys. Chem.
   Soc. 58, 249 1926; C.A. 21, 3616 1927;
   Chem. Zentr. 1927, I, 92.
- 17. Vollmann, H., H. Becker, M. Corell, and H. Streeck, Ann. 531, 1 1937.
- Weizmann, C., and E. Bergmann, J. Chem. Soc. 1939, 494.
- 19. Wittig, G., and H. Petri, Ann. 505, 17 1933.
- Wohl, A., and E. Wertyporoch, Ann. 481, 30 1930.
- Zinke, A., and O. Benndorf, Monatsh.
   56, 153 1930; C.A. 25, 292 1931; Chem.
   Zentr. 1931, I, 276.



# POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA C, H2n-50

# $C_{34}H_{18} \\$

Tetrabenzo-[de,hi,op,st]-pentacene (a)

M. P., °C 580 (in sealed tube)<sup>8</sup>

(a) This compound exists in solution with free valences in the 5a- and 12a-positions.

# Dinaphtho-[3,2,1-cd,3',2',1'-lm]perylene

(Isoviolanthrene)

M. P., °C 510<sup>7</sup>

Additional Data Sublimation Temp. (°C) 460 (in carbon dioxide)<sup>28</sup>

# Dinaphtho-[4,3,2-cd,3',2',1'-lm]perylene (Violanthrene)

# Tetrabenzo-[a,cd,j,lm]-perylene

M. P., °C 331-332<sup>7</sup>

# 3,4,7,8-Dibenzo-2,6-di-(2'-naph-thyl)-bicyclo-[3,3,0]-octadiene-1,5 M. P., °C 266<sup>5</sup>

 $C_{36}H_{22}$ 

6,6'-Bi-(1,2-benzoanthryl) (a)

M. P., °C 310<sup>10</sup>

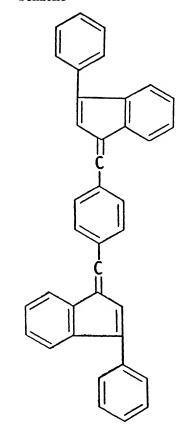
(a) The structure of this compound may be 7,7'-Bi-(1,2-benzoanthryl).

# Indo-[3',2',1'-fg]-11,12-diphenylnaphthacene

(Diphenylphenylenerubene)

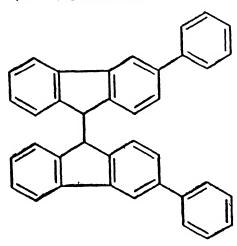
M. P., °C 270–271<sup>11</sup>  $C_{38}H_{23}$ 

1,4-Di-(3'-phenylindenylidenemethyl)
-benzene



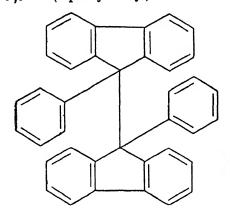
M. P., °C 231-232<sup>15</sup>

# 9,9'-Bi-(3-phenylfluoryl)



M. P., °C 190–193°

# 9,9'-Bi-(9-phenylfluoryl)



M. P., °C 256<sup>21, 22</sup> 254<sup>19</sup> 248–250<sup>20</sup>

 $D_4^{20}$ 

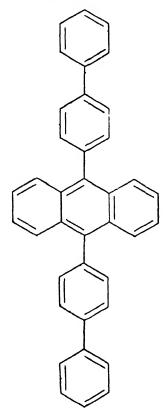
1.266

0° 27

# 9,9'-Bi-(x-phenylfluoryl) (a)

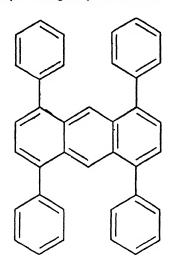
(a) The structure of this compound was not clearly defined in the literature.

# 9,10-Di-p-biphenylylanthracene



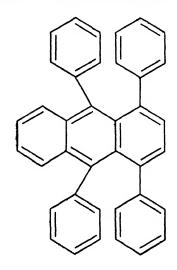
M. P., °C 415<sup>12</sup>

# 1,4,5,8-Tetraphenylanthracene



M. P., °C 370<sup>13</sup>

# 1,4,9,10-Tetraphenylanthracene

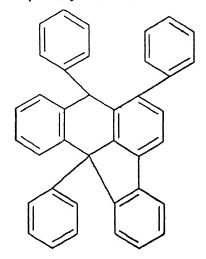


M. P., °C 205<sup>25</sup>• <sup>26</sup> 204<sup>24</sup>

# 2,3,9,10-Tetraphenylanthracene

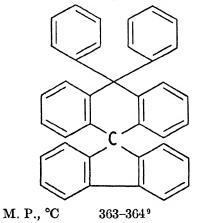
M. P., °C 324-325<sup>1</sup>

# Indo-[3',2',1'-de]-1,9,10-triphenyl-9,10-dihydroanthracene



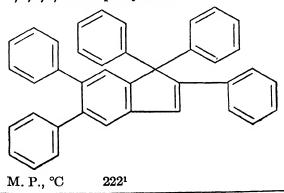
M. P., °C 322<sup>25</sup>, 26

# Spiro[9,9-diphenyl-9,10-dihydroanthracene-10,9'-fluorene]



C<sub>39</sub>H<sub>28</sub>

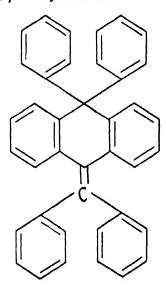
1,1,2,5,6-Pentaphenylindene



1,1,3,5,6-Pentaphenylindene

# 1,2,3,5,6-Pentaphenylindene

# 9,9-Diphenyl-10-benzhydrylidene-9,10-dihydroanthracene



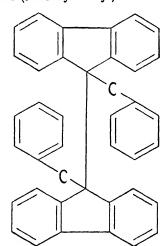
M. P., °C 286³ 276⁴

# C40H30

# 1,3-Diphenyl-2-(4'-benzhydryl-phenyl)-indene

M. P., °C 173-175<sup>14</sup>

# 9,9'-Bi-(9-benzylfluoryl)

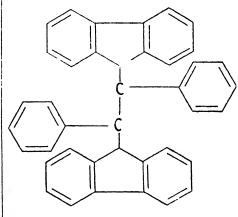


M. P., °C 203-203.5<sup>23</sup> 203<sup>22</sup>

 $\mathbf{x}$ , $\mathbf{x}'$ -Bi- $(\mathbf{x}$ -p-tolylfluoryl) (a) M. P., °C

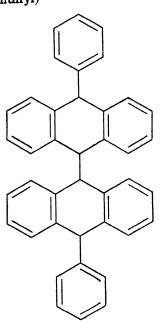
M. P., °C 216<sup>16</sup> (a) The structure of this compound was not clearly defined in the literature.

# 1,2-Diphenyl-1,2-di-(9'-fluoryl)ethane



M. P., °C 321<sup>17</sup>

# 9,9'-Bi-(10-phenyl-9,10-dihydroanthryl)

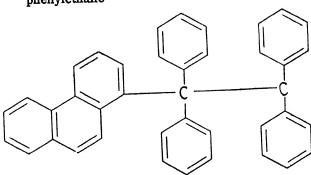


# 1-(9'-Phenanthryl)-1,1,2,2-tetraphenylethane

M. P., °C 152-155 (in nitrogen)<sup>2</sup> 149-152 (in air)<sup>2</sup>

# M. P., °C 260<sup>18</sup>

# 1-(1'-Phenanthryl)-1,1,2,2-tetraphenylethane



M. P., °C 125-135 (a) (in vacuum)<sup>2</sup> 123-134 (a)<sup>2</sup>

(a) This compound melts with decomposition.

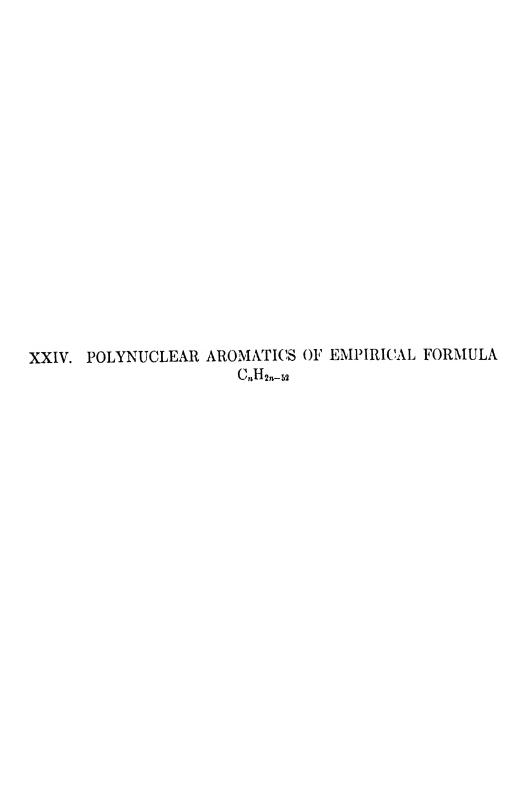
References

- References on Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-50}$
- Allen, C. F. H., and J. W. Gates, Jr., J. Am. Chem. Soc. 65, 2129 1943.
- Bachmann, W. E., R. Hoffman, and F. Whitehead, J. Org. Chem. 8, 320 1943.
- Barnett, E. de B., J. W. Cook, and I. G. Nixon, J. Chem. Soc. 1927, 504.
- 4. Bergmann, E, Ber. 63, 1037 1930.
- 5. Bowden, S. T., J. Chem. Soc. 1939, 26.
- 6. Brand, K, and K. Trebing, Ber. 56, 2545 1923.
- 7. Clar, E., Ber. 76, 458 1943.
- 8. Clar, E., and A. Guzzi, Ber. 65, 1521 1932.
- Clarkson, R. G., and M. Gomberg, J. Am. Chem. Soc. 52, 2881 1930.
- 10. Cook, J. W., J. Chem. Soc. 1931, 499.
- 11. Dufraisse, C., and M. Badoche, Compt. rend. 193, 529 1931.
- Duveen, D., and A. Willemart, J. Chem. Soc. 1939, 116.
- 13. Hirshberg, Y. and L. Haskelberg, Trans. Faraday Soc. 39, 45 1943.
- Koelsch, C. F., J. Am. Chem. Soc. 58, 1331 1936.
- Mayer, F., A Sieglitz, and W. Ludwig, Ber. 54, 1397 1921.

 Scherf, H. W., J. Am. Chem. Soc. 58, 576 1936.

507

- Schlenk, W., and E. Bergmann, Ann. 463, 1 1928.
- 18. Schlenk, W., and E. Bergmann, Ann. 464, 1 1928.
- Schlenk, W., A. Herzenstein, and T. Weickel, Ber. 43, 1753 1910.
- Thomas, J. C., S. T. Bowden, and W. C. Jones, J. Chem. Soc. 1930, 473.
- Vansheidt, A, and B. Moldavskii, Ber.
   63, 1362 1930.
- Vansheidt, A., and B. Moldavskiï, Ber. 64, 917 1931.
- Vansheidt, A., and B. Moldavskii, J. Gen. Chem. (U. S. S. R.) 1, 304 1931;
   Chem. Zentr. 1931, II, 3208.
- Weizmann, C., and E. Bergmann, J. Chem. Soc. 1939, 494.
- Weizmann, C., and E. Bergmann, Scripta Academica Hierosolymitana (Jerusalem) Sci. Rept. No. 1 1938.
- Weizmann, C., E. Bergmann, and L. Haskelberg, J. Chem. Soc. 1939, 301
- Ziegler, K., and F. Ditzel, Ann. 473, 194 1929.
- 28. Zinke, A., W. Penecke, and F. Hanus, Ber. 69, 624 1936.



# XXIV. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-52}$

 $\mathbf{C}_{36}\mathbf{H}_{20}$ 

2,3-(4',3'-Naphtho)-10,11-(2',1'-naphtho)-perylene

M. P., °C ,240<sup>3</sup>

 $C_{38}H_{24}$ 

1,2,7,8-Dibenzo-9-[9'-(1',2'-benzo-fluoryl)]-fluorene (a)

M. P., °C 267<sup>5, 6</sup>

(a) The structure of this compound was not clearly defined in the literature.

Dispiro[fluorene-9,9'-(9',10'-dihydro-anthracene)-10',9''-fluorene]

M. P., °C 471–474¹

 $C_{40}H_{28}$ 

9,10-Dibenzylidene-9,10-dihydroanthracene

M. P., °C 305<sup>2</sup> 302–303<sup>4</sup> References on Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-52</sub>

- Clarkson, R. G., and M. Gomberg, J. Am. Chem. Soc. 52, 2881 1930.
- 2. Padova, R., Compt. rend. 148, 290 1909.
- 3. Schiedt, B., Ber. 71, 1248 1938.
- 4. Staudinger, H., Ber. 41, 1355 1908.
- 5. Vansheidt, A., Ber. 59, 2092 1926.
- Vansheidt, A. A., J. Russ. Phys. Chem. Soc. 58, 69 1926; C. A. 21, 581 1927; Chem. Zentr. 1926, II, 2428.

# XXV. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-54}$

# XXV. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-54}$

### C56H 18

1,2,3,4,5,6-Tri-(2',1'-acenaphtho)benzene (Decacyclene)

M. P., °C 388 390<sup>24</sup> 389.5<sup>16, 17</sup> 387<sup>9, 10, 11, 12, 13</sup>

### C38H22

# Tri- $\alpha$ -naphthylenephenylene-ethene (a)

M. P., °C 300<sup>22</sup>

(a) The structure of this compound was not clearly defined in the literature.

# 2,3,5,6-Di-(2',1'-acenaphtho)-1,4-diphenylbenzene

M. P., °C 403<sup>5</sup>

: mairie distribution de la company de la co

### C40H26

### Tetranaphthyl (a)

M. P., °C 235<sup>1</sup>

(a) The structure of this compound was not clearly defined in the literature.

# 9,9'-Bi-(10-phenylanthryl)

M. P., °C 390–3918

# 1,2,3,4-Tetraphenyl-5,6-(2',1'-acenaphtho)-benzene

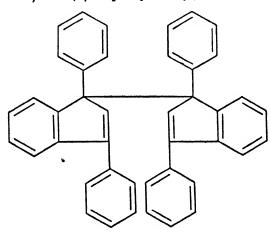
M. P., °C 314<sup>5</sup>

# 1,3,6,8-Tetraphenylpyrene

M. P., °C 299–300<sup>23</sup>

 $\mathbf{C}_{42}\mathbf{H}_{30}$ 

# 1,1'-Bi-(1,3-diphenylindenyl)



M. P., °C 195–196<sup>14</sup> 190–191<sup>15</sup>

# 2,2'-Bi-(1,3-diphenylindenyl)

M. P., °C 249-250 (a)<sup>7</sup> 224-225 (a)<sup>7</sup>

(a) These constants were determined on stereoisomers.

# 1,1,2,2-Tetra-(1'-naphthyl)-ethane

M. P., °C 285–286<sup>20</sup> 277<sup>3</sup>· <sup>4</sup> 276–277<sup>25</sup>

# 1,1,2,2-Tetra-(2'-naphthyl)-ethane

x<sub>4</sub>-Tetraphenyl-x,x-dihydronaphthacene (a)

(Tetraphenyldihydrorubene)

M. P., °C 230–231<sup>6</sup>

(a) The structure of this compound was not clearly defined in the literature.

# $C_{14}H_{34}$

# 1,6-Diphenyl-3,4-di-(9'-fluoryl)hexadiene-1,5

- M. P., °C 204 (a)<sup>18</sup> 160-161 (a)<sup>21</sup>
- (a) These constants were determined on isomeric forms.

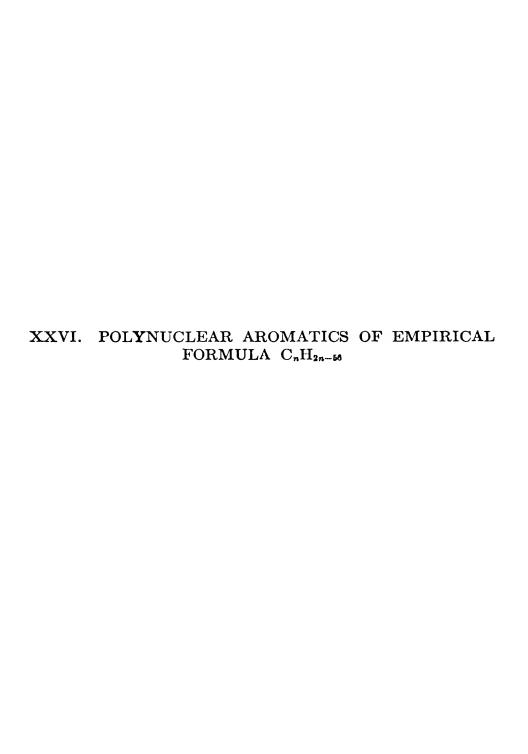
### 1,4-Diphenyl-1,4-di-(9'-phenanthryl)butane

517

References on Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-44}$ 

- Bamberger, E, and F. D. Chattaway, Ann. 284, 52 1895.
- Bergmann, E., and F. Bergmann, J. Am. Chem. Soc. 59, 1443 1937.
- Chichibabin, A. E., and O. I. Magidson,
   J. prakt. Chem. [2] 90, 168 1914.
- Chichibabin, A., and O. Magidson, J. Russ. Phys. Chem. Soc. 46, 1389
   1914: Chem. Zentr. 1915. I. 1123
- 1914; Chem. Zentr. 1915, I, 1123.
  5. Dilthey, W., S. Henkels, and A. Schaefer, Ber. 71, 974 1938.
- 6. Dufraisse, C., and R. Horclois, Bull. soc. chim. [5] 3, 1880 1236.
- Dufraisse, C., B. Masumoto, and R. Buret, Bull. soc. chim. [4] 51, 74 1932.
- Dufraisse, C., L. Velluz, and Mme.
   I. Velluz, Bull. soc. chim. [5] 5, 600
   1938.
- 9. Dziewoński, K., Ber. 36, 962 1903.
- Dziewoński, K., Bull. s.c. chim. [3]
   29, 374 1903.
- Dziewoński, K., Bull soc. chim. [3]
   31, 925 1904.
- Dziewoński, K., and Z. Leyko, Ber. 47, 1679 1914.

- Dziewoński, K., and S. Suknarowski, Ber. 51, 457 1918.
- 14. Goebel, M. T., and C. S. Marvel, J. Am. Chem. Soc. 55, 3712 1933.
- Halley, L. F., and C. S. Marvel, J. Am. Chem. Soc. 54, 4450 1932.
- Padoa, M., Atti accad. Lincei [5] 20, I,3451911.
- Padoa, M., Gazz. chim. ital. 44, II, 517 1914.
- Schlenk, W., and E. Bergmann, Ann. 463, 1 1928.
- Schmidlin, J., and M. Huber, Ber. 43, 2824 1910.
- Schmidlir, J., and P. Massini, Ber. 42, 2377 1909.
- Thiele, J, and F. Heule, Ann. 347, 290 1905.
- Vansheidt, A. A., J. Russ. Phys. Chem. Soc. 58, 249 1926; C. A. 21, 3616 1927; Chem. Zentr. 1927, I, 92.
- Vollmann, H., H. Becker, M. Corell, and H. Streeck, Ann. 531, 1 1937.
- Ward, J. J., W. R. Kirner, and H. C. Howard, J. Am. Chem. Soc. 67, 246 1945.
- 25. Wuis, P. J., and D. Mulder, Rec. trav. chim. 57, 1385 1938.



# XXVI. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-56}$

 $C_{36}H_{16}$  Dinaphtho-[2,1,8-bcd,2',1',8'-klm]-coronene

Additional Data
Sublimation Temp. (~C)
ca 500 1 mm<sup>18</sup>

 $C_{42}H_{28}$ 

1,1,2,2-Tetra-(1'-naphthyl)-ethene

 $C_{40}H_{24}$ 

Difluoro-[2',1',9'-abc,2",1",9"-fgh]-5,10-diphenylcyclodec-adiene-4,9

M. P., °C 225<sup>11</sup>

# 5,6,11,12-Tetraphenylnaphthacene (Rubrene) (a)

(Tetraphenylrubene)

M. P., °C

333

3351, 9

3347, 9

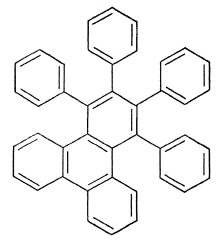
33220

33112. 14, 16

 $330^{2}$ 

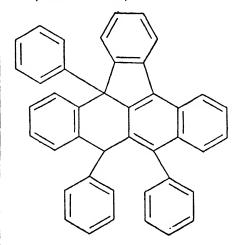
(a) This compound was thought by some investigators to be 2,2'-Bi-(1,3-diphenyl-4,5-benzocyclopentadien-3,5-ylidene). However, later work indicates that the above structure is correct.

# 1,2,3,4-Tetraphenyltriphenylene



M. P., °C 292–293<sup>4</sup>

# Indo-[1',2',3'-op]-5,6,11-triphenyl-6,11-dihydronaphthacene (Pseudorubrene)



M. P., °C 278<sup>1</sup>. 9 210 (a)<sup>15</sup>

(a) This compound remelts at 278.

Diindano-[3',2',1'-fg,3"',2"',1"'-op]-5,11-diphenyl-5,6,10,11-tetrahydronaphthacene

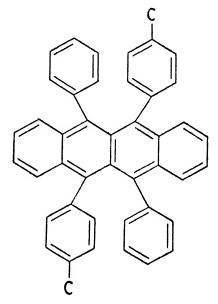
M. P., °C 338 (a)<sup>1</sup> 302 (a)<sup>1</sup>

(a) These constants were determined on stereoisomers.

 $C_{42}H_{28}$ 

5,11-Diphenyl-6,12-di-p-tolylnaphthacene

(Bis-p-tolyl-diphenylrubene) (Dimethylrubrene)



M. P., °C 205<sup>5</sup>

2-Methyl-5,6,12-triphenyl-11-p-tolylnaphthacene

(Dimethylrubrene)

M. P., °C 273<sup>5</sup>

2,8-Dimethyl-5,6,11,12-tetraphenylnaphthacene

(Dimethylrubrene)

M. P., °C 321<sup>5</sup> 315 (a)<sup>17, 19</sup>

(a) The structure of this compound was not clearly defined in the literature.

Pseudobis-(p-tolyl)-diphenylnaphthacene (a)

M. P., °C 294-295 (b)<sup>9</sup> 293-294 (b)<sup>10</sup> 271-272 (b)<sup>9, 10</sup>

- (a) The structure of this compound was not clearly defined in the literature.
- (b) There constants were determined on isomeric forms.

### C16H36

2,8-Dimethyl-5,11-di-p-tolyl-6,12-diphenylnaphthacene
(Dimethyldiphenylditolylrubene)

M. P., °C 242<sup>6</sup> 523 C<sub>48</sub>H<sub>40</sub>

### $C_{48}H_{40}$

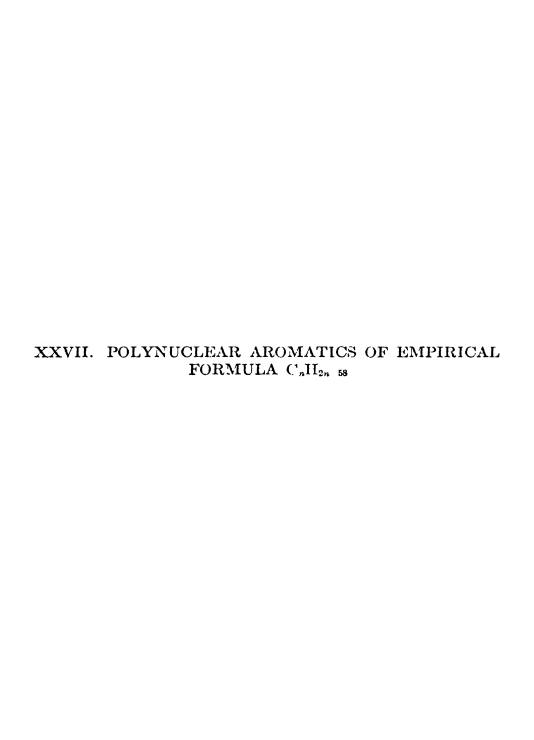
### x<sub>12</sub>-Dodecahydrofluorocyclene (a)

- M. P., °C 3268
- (a) The structure of this compound was not clearly defined in the literature.

### References on Polynuclear Aromatics of Empirical Formula $C_nH_{2n-56}$

- 1. Badoche, M., Ann. chim. [10] 20, 200 1933.
- Bowen, E. J., and F. Steadman, J. Chem. Soc. 1934, 1098
- Chichibabin, A. E., and O. I. Magidson,
   J. prakt. Chem. [2] 90, 168 1914.
- Dilthey, W., S. Henkels, and A. Schaefer, Ber. 71, 974 1938.
- 5. Dufraisse, C., and M. Lowry, Compt. rend. 194, 1664 1932.
- 6. Dufraisse, C., and J. A. Monier, Compt. rend. 196, 1327 1933.
- Dufraisse, C., and L. Velluz, Bull. soc. chim. [5] 3, 1905 1936.

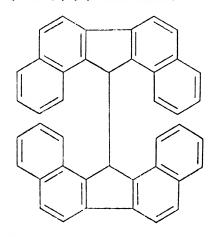
- Dziewoński, K., and L. Gızler, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1937A, 441.
- 9. Enderlin, I., Ann clam. [11] 10, 5 1938.
- Enderlin, L., Compt. rend. 193, 1432
   1931.
- 11. Ferrario, E., Chem. Ztg. 35, 325 1911.
- Hurd, C. D., and F. L. Cohen, J. Am. Chem. Soc. 53, 1068 1931.
- 13. Magidson, O. I., Ber. 58, 433 1925.
- 14. Moureu, C, and C. Dufraisse, Z. physik. Chem. 130, 472 1927.
- Moureu, C., C. Dutreisse, and G. Berchet, Compt. rend. 185, 1085 1927.
- Moureu, C., C. Dufraisse, and P. M. Dean, Compt. rend. 182, 1440 1926.
- Moureu, C., C. Dufrasse, and A. Willemart, Compt. rend. 187, 266 1928.
- 18. Scholl, R., and K. Meyer, Ber. 67, 1229 1934.
- 19. Willemart, A., Ann. chim. [10] 12, 345 1929.
- Wittig, G., and D. Waldi, J prakt. Chem. [2] 160, 242 1942; C. A. 37, 5399 1943.



# XXVII. POLYNUCLEAR AROMATICS OF EMPIRICAL FORMULA $C_nH_{2n-58}$

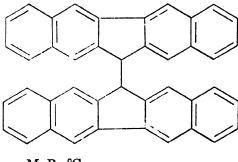
### C42H26

9,9'-Bi-(1,2,7,8-dibenzofluoryl)



M. P., °C 353-355<sup>7</sup>. 8

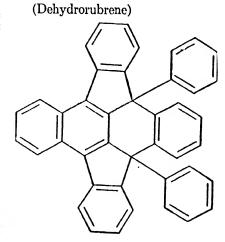
9,9'-Bi-(2,3,6,7-dibenzofluoryl)



M. P., °C 310<sup>3</sup> Diindo-[1',2',3'-fg,1",2",3"-op]-5,11-diphenyl-5,11-dihydronaphthacene (Dehydrorubrene)

M. P., °C 455<sup>2</sup>

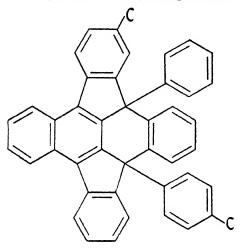
Diindo-[1',2',3'-fg,3",2",1"-op]-5,12-diphenyl-5,12-dihydronaphthacene



M. P., °C 430<sup>5</sup> 427–428<sup>4</sup>

### C44H30

Indo-[1',2',3'-fg]-(6"-methylindo)[3",2",1"-op]-5-p-tolyl-12phenyl-5,12-dihydronaphthacene

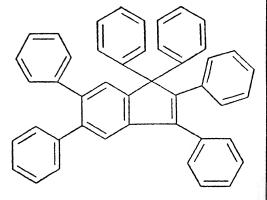


M. P., °C 340 (a)<sup>5</sup>

(a) This compound remelts at 370.

# $C_{45}H_{32}$

# 1,1,2,3,5,6-Hexaphenylindene



M. P.. ℃

#### C46H34

1-(1'-Naphthylphenylmethyl)-4-(1'naphthyldiphenylmethyl)-benzene (a)

M. P., °C 234–235<sup>6</sup>

(a) The structure of this compound was not clearly defined in the literature.

References on Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-58</sub>

- Allen, C. F. H., and J. W. Gates, Jr., J. Am. Chem. Soc. 65, 2129 1943.
- Badoche, M., Ann. chim. [10] 20, 200 1933.
- Cook, J. W., and R. W. G. Preston, J. Chem. Soc. 1944, 553.
- 4. Dufraisse, C., and L. Velluz, Bull. soc. chim. [5] 3, 1905 1936.

- Enderlin, I.., Ann. chim. [11] 10, 5
   1938.
- Gomberg, M., and C. S. Schoepfle, J. Am. Chem. Soc. 41, 1655 1919.
- 7. Vansheidt, A., Ber. 59, 2092 1926.
- Vansheidt, A. A., J. Russ. Phys. Chem. Soc. 58, 69 1926; C. A. 21, 581 1927; Chem. Zentr. 1926, II, 2428.

M. P., °C 430<sup>5</sup> 427–428<sup>4</sup>

#### C44H30

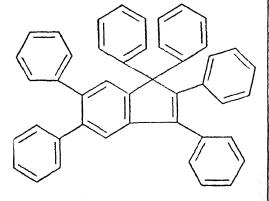
Indo-[1',2',3'-fg]-(6"-methylindo)[3",2",1"-op]-5-p-tolyl-12phenyl-5,12-dihydronaphthacene

M. P., °C 340 (a)<sup>5</sup>

(a) This compound remelts at 370.

#### $C_{45}H_{32}$

1,1,2,3,5,6-Hexaphenylindene



MP.ºC

#### C46H34

1-(1'-Naphthylphenylmethyl)-4-(1'naphthyldiphenylmethyl)-benzene (a)

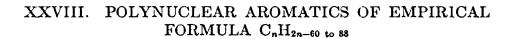
M. P., °C 234–235<sup>6</sup>

(a) The structure of this compound was not clearly defined in the literature.

References on Polynuclear Aromatics of Empirical Formula C<sub>n</sub>H<sub>2n-58</sub>

- Allen, C. F. H., and J. W. Gates, Jr., J. Am. Chem. Soc. 65, 2129 1943.
- Badoche, M., Ann. chim. [10] 20, 200
   1933.
- Cook, J. W., and R. W. G. Preston, J. Chem. Soc. 1944, 553.
- Dufraisse, C., and L. Velluz, Bull. soc. chim. [5] 3, 1905 1936.

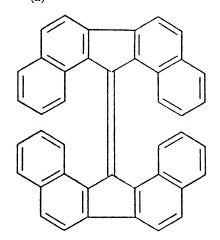
- Enderlin, L., Ann. chim. [11] 10, 5
   1938.
- Gomberg, M., and C. S. Schoepfle, J. Am. Chem. Soc. 41, 1655 1919.
- 7. Vansheidt, A., Ber. 59, 2092 1926.
- Vansheidt, A. A., J. Russ. Phys. Chem. Soc. 58, 69 1926; C. A. 21, 581 1927; Chem. Zentr. 1926, II, 2428.



# XXVIII. POLYNUCLEAR AROMATICS OF ETIPIRICAL FORMULA C<sub>n</sub>H<sub>2n-60 to 88</sub>

 $C_nH_{2n-60}$   $C_{42}H_{24}$ 

9,9'-Bi-(1,2,7,8-dibenzofluorylidene)
(a)



M. P., °C 357<sup>24</sup> 308-310<sup>18, 19</sup>

(a) The structure of this compound was not clearly defined in the literature.

#### C48H36

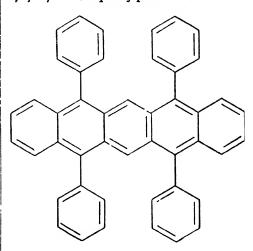
#### x<sub>8</sub>-Octahydrofluorocyclene (a)

M. P., °C 336-337°

(a) The structure of this compound was not clearly defined in the literature.

 $C_nH_{2n-62}$   $C_4cH_{30}$ 

5,7,12,14-Tetraphenylpentacene



M. P., °C 306-308<sup>1</sup>

# $C_n H_{2n-64}$

#### $C_{48}H_{32}$

x<sub>4</sub>-Tetrahydrofluorocyclene (a)

M. P., °C

348-3499

(a) The structure of this compound was not clearly defined in the literature.

#### C49H34

1,3-Diphenyl-9-(x',x',x'-triphenyl-phenyl)-fluorene (a)

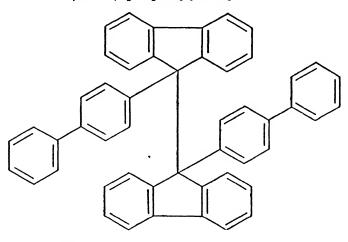
M. P., °C 221<sup>15</sup>  $C_{49}H_{84}$  530

(a) The structure of this compound was not clearly defined in the literature.

## $C_nH_{2n-66}$

#### C50H34

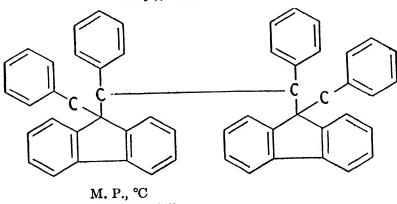
# 9,9'-Bi-(9-p-biphenylylfluoryl)



M. P., °C 175-176<sup>23</sup>

## $\mathbf{C}_{54}\mathbf{H}_{42}$

# 1,2-Diphenyl-1,2-di-[9'-(9'-benzyl-fluoryl)]-ethane



304-30622

 $C_nH_{2n-68}$ 

#### $C_{48}H_{28}$

1,2,3,4,5,6,7,8-Tetra-(2',1'acenaphtheno)-cyclooctadiene-1,5 (a) (Fluorocyclene)

M. P., °C 396–397°, 12 395–39610

(a) The double bonds may be in the 2and 6-positions.

#### $C_{50}H_{32}$

5,11-Diphenyl-6,12-di-(2'-naphthyl)-naphthacene (a)

M. P., °C 280<sup>20</sup>, <sup>25</sup>

(a) This compound was named Dibenzorubrene in the literature.

532

 $C_n H_{2n-70}$ 

C<sub>54</sub>H<sub>18</sub>

1,2-Diphenyl-1,2-di-[9'-(10'-phenyl-anthryl)]-ethane (a)

M. P., °C 252-253<sup>14</sup>

(a) The structure of this compound was not clearly defined in the literature.

# 1,1,2,2-Tetraphenyl-1,2-di-(3'-phenanthryl)-ethane

M. P., °C 150-152 (a) (in nitrogen)<sup>2</sup>

(a) This compound melts with decomposition.

# 1,1,2,2-Tetraphenyl-1,2-di-(9'-phenanthryl)-ethane

M. P., °C 223-225 (in nitrogen)<sup>21</sup> 210-212<sup>21</sup>  $C_{55}H_{40}$ 

Di-[x-(x'-fluorylmethylfluoryl)]methane (a)

M. P., °C 246-247<sup>11</sup>

(a) The structure of this compound was not clearly defined in the literature.

 $C_n H_{2n-72}$ 

 $C_{54}H_{36}$ 

5,11-Diphenyl-6,12-di-p-biphenylyl-naphthacene

M. P., °C

#### C56H40

## 1,2,3,4-Tetraphenyl-1,2,3,4-diendo-(9',10'-anthracene)-cyclobutane

#### $C_nH_{2n-74}$

18416, 17

## $C_{50}H_{38}$

x,x-Di-[x'-(phenanthryl)-phenylmethyl]-phenanthrene (a)

M. P., °C 195–197<sup>13</sup>

(a) The structure of this compound was not clearly defined in the literature.

536

# $C_nH_{2n-76}$ $C_{54}H_{32}$

## Leucacene

M. P., °C 250-252<sup>5</sup>, 6

# $C_{56}H_{36}$

1,2,3,4-Tetraphenyl-1,4,2,3-diendo-(9',10'-anthracene)-cyclobutadiene-1,3

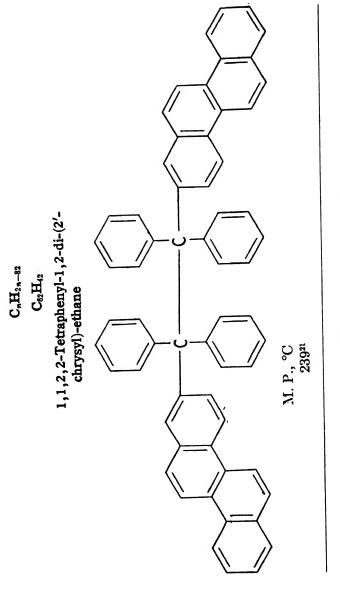
M. P., °C 19716

 $\mathbf{C}_n\mathbf{H}_{2n-78}$ 

C57H36

1,2,3,4,5,6-Tri-[2',1'-(4'-benzyl-acenaphtho)]-benzene

M. P., °C 270<sup>8</sup> 267–270<sup>7</sup>



This series continued on next page

#### $C_nH_{2n-88}$

#### C66H44

#### 2,6,8,12-Tetraphenyl-5,11-di-pbiphenylylnaphthacene

M. P., °C 320 (a)<sup>3</sup>

(a) This compound remelts at 380.

References on Polynuclear Aromatics of Empirical Formula  $C_nH_{2n-80 \text{ to } 88}$ 

- Allen, C. F. H., and A. Bell, J. Am. Chem. Soc. 64, 1253 1942.
- Bachmann, W. E., and M. C. Kloetzel,
   J. Org. Chem. 2, 356 1937-38.
- Duveen, D., and A. Willemart, Compt. rend. 207, 1226 1938.
- Duveen, D., and A. Willemart, Compt. rend. 208, 1587 1939.
- 5. Dziewoński, K., Ber. 53, 2173 1920.
- Dziewoński, K., Bull. intern. acad. polon. sci., Classe sci. math. nat. 1919A, 99.

- 7. Dziewoński, K., Bull. intern. acad. sci. Cracovie 1904A, 201.
- 8. Dziewoński, K., Bull. soc. chim. [3] 31, 925 1904.
- Dziewoński, K., and L. Gizler, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1937A, 441.
- Dziewoński, K., and Z. Leyko, Ber. 47, 1679 1914.
- Dziewoński, K., and M. Panek, Bull. intern. acad. polon. sci., Classe sci. math. nat. 1927A, 745.
- Dziewoński, K., and S. Suknarowski, Ber. 51, 457 1918.

- Frankforter, G., and W. Kritchevsky, J. Am. Chem. Soc. 37, 385 1915.
- Julian, P., and W. Cole, J. Am. Chem. Soc. 57, 1607 1935.
- Kohler, E. P., and L. W. Blanchard, J. Am. Chem. Soc. 57, 367 1935.
- Lippmann, E., and R. Fritsch, Ann. 351, 52 1907.
- Lippmann, E., and R. Fritsch, Sitzber. Akad. Wiss. Wien, Math. naturw. Klasse 113, Abt. IIb, 429 1904.
- 18. Magidson, O. I., Ber. 58, 433 1925.
- Magidson, O., Trans. Sci. Chem. Pharm. Inst. (U. S. S. R.) 16, 33 1926; C. A. 23, 1633 1929; Chem. Zentr. 1928, I, 58.

- Moureu, C., C. Dufraisse, and A. Willemart, Compt. rend. 187, 266 1928.
- 21. Müller, E., and W. Kruck, Ber. 71, 1778 1938.
- Schlenk, W., and E. Bergmann, Ann. 479, 58 1930.
- 23. Schlenk, W., and A. Herzenstein, Ann. 372, 21 1910.
- Vansheidt, A. A., J. Russ. Phys. Chem. Soc. 58, 249 1926; C. A. 21, 3616 1927; Chem. Zentr. 1927, I, 92.
- Willemart, A., Ann. chim. [10] 12, 345 1929.